Commission Directive 2004/73/EC of 29 April 2004 adapting to technical progress for the twenty-ninth time Council Directive 67/548/EEC on the approximation of the laws, regulations and administrative provisions relating to the classification, packaging and labelling of dangerous substances (Text with EEA relevance)

COMMISSION DIRECTIVE 2004/73/EC

of 29 April 2004

adapting to technical progress for the twenty-ninth time Council Directive 67/548/ EEC on the approximation of the laws, regulations and administrative provisions relating to the classification, packaging and labelling of dangerous substances

(Text with EEA relevance)

THE COMMISSION OF THE EUROPEAN COMMUNITIES,

Having regard to the Treaty establishing the European Community,

Having regard to Council Directive 67/548/EEC of 27 June 1967 on the approximation of the laws, regulations and administrative provisions relating to the classification, packaging and labelling of dangerous substances⁽¹⁾, and in particular Article 28 thereof,

Whereas:

- (1) Annex I to Directive 67/548/EEC contains a list of dangerous substances, together with particulars of the classification and labelling of each substance. That list needs to be updated to include further notified new substances and further existing substances as well adapting the existing entries to technical progress such as setting environmental concentration limits for certain substances. Accordingly it is also necessary to delete entries for certain substances and to split some entries because the classification no longer applies to all the substances under those entries. The labelling of substances containing 1,3-butadiene should be changed in order to reflect that that substance will be classified as a mutagen by the present Directive.
- Annex V to Directive 67/548/EEC lays down the methods for the determination of the physicochemical properties, toxicity and ecotoxicity of substances and preparations. It is appropriate to amend that Annex in order to obtain a reduction to a minimum of the number of animals used for experimental purposes, in accordance with Council Directive 86/609/EEC of 24 November 1986 on the approximation of laws, regulations and administrative provisions of the Member States regarding the protection of animals used for experimental and other scientific purposes⁽²⁾. The methods for sub-chronic oral toxicity in Chapters B.1, B.4, B.5, B.31 and B.35 should be revised accordingly. Furthermore, Chapter B.42 should be added to Annex V in order to make available a refined method on sub-chronic oral toxicity. Finally, Chapter A.21 on physico-chemical properties, Chapter B.43 on sub-chronic oral toxicity and Chapters C.21 to C.24 on environmental toxicity should be added in order to allow for the determination of properties which are not yet sufficiently covered by the methods in Annex V.

(3) The measures provided for in this Directive are in accordance with the opinion of the Committee on the Adaptation to Technical Progress of the Directives for the Elimination of Technical Barriers to Trade with Dangerous Substances and Preparations,

HAS ADOPTED THIS DIRECTIVE:

Article 1 U.K.

Directive 67/548/EEC is amended as follows:

- (1) Annex I is amended as follows:
 - (a) note K in the foreword is replaced by the text set out in Annex 1 A;
 - (b) the entries corresponding to the entries set out in Annex 1B to this Directive are replaced by the text set out in that Annex;
 - (c) the entries set out in Annex 1C to this Directive are inserted in accordance with the order of the entries set out in Annex I to Directive 67/548/EEC;
 - (d) the entries with index numbers 604-050-00-X, 607-050-00-8, 607-171-00-6 and 613-130-00-3 are deleted;
 - the entry with index number 048-002-00-0 is replaced by the entries with index numbers 048-002-00-0 and 048-011-00-X set out in Annex 1D to this Directive;
 - (f) the entry with index number 609-006-00-3 is replaced by the entries with index numbers 609-006-00-3 and 609-065-00-5 set out in Annex 1D to this Directive:
 - (g) the entry with index number 612-039-00-6 is replaced by the entries with index numbers 612-039-00-6 and 612-207-00-9 as set out in Annex 1D.
- (2) Annex V is amended as follows:
 - (a) the text set out in Annex 2A to this Directive is added as Chapter A.21;
 - (b) chapter B.1bis is replaced by the text set out in Annex 2B to this Directive;
 - chapter B.1tris is replaced by the the text set out in Annex 2C to this Directive;
 - (d) chapter B.4 is replaced by the text set out in Annex 2D to this Directive;
 - (e) chapter B.5 is replaced by the text set out in Annex 2E to this Directive;
 - (f) chapter B.31 is replaced by the text set out in Annex 2F to this Directive;
 - (g) chapter B.35 is replaced by the text set out in Annex 2G to this Directive;
 - (h) the text set out in Annex 2H to this Directive is added as Chapter B.42 and B.43;
 - (i) the text set out in Annex 2I to this Directive is added as Chapter C.21 to C.24.

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Article 2 U.K.

- 1 Member States shall bring into force the laws, regulations and administrative provisions necessary to comply with this Directive by 31 October 2005 at the latest. They shall forthwith communicate to the Commission the text of those provisions and a correlation table between those provisions and this Directive. When Member States adopt those provisions, they shall contain a reference to this Directive or be accompanied by such a reference on the occasion of their official publication. Member States shall determine how such reference is to be made.
- 2 Member States shall communicate to the Commission the main provisions of national law which they adopt in the field covered by this Directive.

Article 3 U.K.

This Directive shall enter into force on the twentieth day following its publication in the *Official Journal of the European Union*.

Article 4 U.K.

This Directive is addressed to the Member States.

Done at Brussels, 29 April 2004.

For the Commission

Margot WALLSTRÖM

Member of the Commission

ANNEX 1A U.K.

Note K:

The classification as a carcinogen or mutagen need not apply if it can be shown that the substance contains less than 0.1 % w/w 1,3-butadiene (Einecs No 203-450-8). If the substance is not classified as a carcinogen or mutagen, at least the S-phrases (2-)9-16 should apply. This note applies to certain complex oil-derived substances in Annex I.

Index	chemical		EC No	CAS	Classific	atlæbelling		
No	name	related to		No			Limits	related to
		substanc	es					preparations
006-005-	otheram tetrameth disulphide		205-286-2	2137-26-8	Xn; R20/22-4 Xi; R36/38 R43 N; R50-53	20/22-36/ S:	/ 20-60- 61 ≤ C < 25 %: Xn, N;	6/38-43-48/22-50/53 3-48/22-50/53 2-50/53
006-006-	hydrogen cyanide % hydrocyan acid%		200-821-6	574-90-8	T+; R26/27/20 N; R50-53	26/27/28- S:	R26/27/2	38-45-60-61

				2,5 % ≤ C < 7 %: T, N; R23/24/2: 1 % ≤ C < 2,5 %: T, N; R23/24/2: 0,25 % ≤ C < 1 %: Xn; R20/21/2: 0,1 % ≤ C < 0,25 %:Xn; R20/21/2:	5-52-53 2-52-53
006-012-0@i2am (ISO) zinc t dimet	205-288-3137-30-4 nate	Xn;	T+; N R: 222-26-37- S: (1/2-)22-2	R22-26-3 202%36/3 < C < 25 %: T+, N; R26-37-4 10 % < C < 20 %: T+, N;	7-41-43-48/22-50-53 7/39-45-60-61 1-43-48/22-50-53 3-48/22-50-53 3-50-53 0-53

							0,025 % ≤ C < 0,1 %: N; R51-53 0,0025 % ≤ C < 0,025 %: R52-53
006-021-0	Offiluron (ISO) 3-(3,4- dichlorop methoxy- methylure	1-	206-356-5	5330-55-2	Repr. Cat. 2; R61 Repr. Cat. 3; R62 Carc. Cat. 3; R40 Xn; R22-48/2; N; R50-53	S: 53-45-60-	48/22-62-50/53 -61
006-044-0	067proturo 3-(4- isopropylj dimethylu	phenyl)-1,		134123-59	-6arc. Cat. 3; R40 N; R50-53	Xn; N R: 40-50/53 S: (2-)36/37	C ≥ 2,5 %: Xn, N; R40-50-53 60%
006-072-0 X	<i>N,N</i> -	niocarbama arb		552888-80	- % n; R22 R43 N; R51-53	Xn; N R: 22-43-51/ S: (2-)24-37	
006-089-0	00k2orine dioxide		233-162-8	310049-04	- 0 ; R8 R6 T+; R26 C; R34 N; R50	O; T+; N R: 6-8-26-34 S: (1/2-)23-2	C≥5%: T+; N; R26-34-50 -50% ≤ C <5%: 26-2\$A\$6/37/39-38-45-61 R26-36/37/38-50

			0,5 % < C < %: T; R23-3 0,2 % C < 0 %: T; R23-5 0,02 % < C < 0,2 % Xn; N R20-5	1 N; 36/37/38-50 \$\leq 5,5 N; 50 %
006-089-0¢hlorine X dioxide %	233-162-810049-04	C; R34 R N; R50 2. S	85-34-50 R25-3 10 % 1/2-)23-26285 %: C, R22-3 3 % ≤ < 10 ° Xn; N	N; 34-50 ≤ 36/37/39-45-61 N; 34-50 6 C %: 5/6: 5/7:
007-001-Ф агб monia, anhydrous	231-635-37664-41-	T; R23 R C; R34 10 N; R50 S	0-23-34- ₹023-3 5: 5% ≤ 1/2-)9-16	N; 34-50 6/37/39-45-61 Γ;R23-34
007-008-0thyldrazine E	206-114-9302-01-2	Carc. Cat. R 2; R45 4. T; S	5-10-23/244252 3: 10 % (3-45-60-61 < 2 %: T, R45-2 3 % ≤ C < 1 %: T,	N; 13424325035343-50/53 \$\leq 5 \\ N; 100/2 1/22-34-43-51/53 \$\leq 0 \\ N; 100/2 1/22-36/38-43-51/53

							%: T, N; R45-43-5 1 % ≤ C < 2,5 %: T; R45-43-5 0,25 % ≤ C < 1 %: T; R45-52/5 0,1 % ≤ C < 0,25 %: T; R45	2/53
007-010-0	09e4lium nitrite		231-832-4	17758-09-(OO; R8 T; R25 N; R50	O; T; N R: 8-25-50 S: (1/2-)45-6	C ≥ 25 %: T, N; R25-50 5 % ≤ C 6 ₹ 25 %: T; R25 1 % ≤ C < 5 %: Xn; R22	
007-011-0 X	φotassium nitrite		231-832-4	17758-09-(0O; R8 T; R25 N; R50	O; T; N R: 8-25-50 S: (1/2-)45-6	C ≥ 25 %: T, N; R25-50 5% ≤ 5C< 25 %:T; R25 1 % ≤ C< 5 %:Xn; R22	
007-013-0	0 0-2 - dimethylh	E lydrazine		540-73-8	Carc. Cat. 2; R45 T; R23/24/2; N; R51-53	R: 45-23/24/	3 % ≤ C < 25 %: T;	4/25-51/53 1/22-52/53
007-017-0	00s2butyl nitrite	Е	208-819-7	7542-56-3	F; R11	F; T		

					Xn; R20/22 Carc. Cat. 2; R45 Muta. Cat. 3; R68	R: 11-20/22- S: 53-45	45-68	
007-027-0	bis(3,3- bis((1-	ntylidenim	420-190-2)ureido)he	Xn; R21/22-44 C; R34 xR43 N; R50-53	21/22-34- S:	43-48/21-5 5-36/37/39	
008-003-0	ObyArogen peroxide solution %	B .	231-765-0	7722-84-1	R5 O; R8 C; R35 Xn; R20/22	S:	C ≥ 70 %: C; -R20/22-3: 50 % ≤ 26:28736/3 %: C; R20/22-3: 35 % ≤ C < 50 %: Xn; R22-37/3: 8 % ≤ C < 35 %: Xn; R22-41 5 % ≤ C < 8 %: Xi; R36 Footnote: C ≥ 70 %: R5, O; R8 50 % ≤ C < 70 %: O; R8	7/39-45 4
009-015-0	0 ⊕u7 phuryl difluoride		220-281-5	52699-79-8	3T; R23 Xn; R48/20 N; R50	T; N R: 23-48/20- S: (1/2-)45-6		
015-002-0	0 6eđ phosphor	ıs	231-768-7	77723-14-()F; R11 R16 R52-53	F R: 11-16-52/ S: (2-)7-43-6		

015-014-0	0:i butyl phosphate		204-800-2	2126-73-8	Carc. Cat. R40 Xn; R22 Xi; R38	35,n R: 22-38-40 S: (2-)36/37-	-46	
015-015-0	phosphate tritolyl phosphate o-o-o, o- o-m, o- o-p, o- m-m, o- m-p, o- p-p		201-103-5	578-30-8	T; R39/23/24 N; R51-53	39/23/24/ S:	C ≥ 25 %: T, N; 2839/23324 2,5 % ≤ 3:28245-6 %: T; R39/23/24 1 % ≤ C < 2,5 %: T; R39/23/24 0,2 % ≤ C < 1 %: Xn; R68/20/2	1 4/25-52/53 4/25
015-016-0	Prieresyl phosphate tritolyl phosphate m-m-m, m-m-p, m-p-p,		201-105-0	578-32-0	Xn; R21/22 N; R51-53	Xn; N R: 21/22-51/ S: (2-)28-61	R21/22-5	
015-020-0	Masvinpho (ISO) 2- methoxyc methylvin dimethyl phosphate	arbonyl-1- iyl	232-095-	17786-34-	7T+; R27/28 N; R50-53	S:	C ≥ 7 %: T+, N; 5827/28-50 1 % ≤ 28536737-4 %: T, N; R24/25-50 0,1 % ≤ C<1 %: Xn, N; R21/22-50 0,0025 % ≤ C < 0,1 %: N; R50-53 0,0025 % ≤ C < 0,0025	5-60-61 0-53

015-021-0	OtriOhlorfor (ISO) dimethyl 2,2,2- trichloro- hydroxyet	200-149-3	352-68-6	Xn; R22 R43 N; R50-53	Xn; N R: 22-43-50/ S: (2-)24-37	R22-43-50-53	
015-027-0	(ISO) O,O,O,O- tetraethyl		23689-24-5	5T+; R27/28 N; R50-53	S:	$C \ge 7\%$: T+, N; SE27/28-50-53 $1\% \le 2C-36737-45-60-61$ %: $T, N;R24/25-50-530,1\% \le C < 1\%:Xn, N;R21/22-50-530,025\% \le C< 0,1%$: $N;R50-530,0025\% \le C< 0,025\% \le C< 0,025\% \le C< 0,025\% \le C< 0,025\% \le C< 0,0025\% \le C< 0,0025\% \le C< 0,0025\% \le C$	

015-032-0	OprOthoate (ISO) O,O- diethyl isopropylophosphoro	carbamoyl		22275-18-:	5T+: R27/28 R52-53	T+ R: 27/28-52/ S: (1/2-)28-3	0,0025 %: R52-53	1
015-033-0		nethyl	206-052-2	2298-02-2	T+; R27/28 N; R50-53	S:	$C \ge 7$ %: T+, N; SR27/28-5: $1\% \le 36/37745-6$ %: T, N; R24/25-5: $0,1\% \le C < 1\%$: Xn, N; R21/22-5: 0,025 % $\le C$ < 0,1 %: N; R50-53 0,0025 % $\le C$ < 0,025 %: N; R51-53 0,00025 %: C < 0,0025 %: R51-53	0-61 0-53
015-034-0	Parathion (ISO) O,O- diethyl O-4- nitrophen phosphore		200-271-7	756-38-2	T +;R26/28 T; 24-48/25 N; R50-53	24-26/28- S:	86.08 %45-6 ≤ C < 25 %: T+, N; R21-26/2 7 % ≤ C < 10%: T+, N;	8-48/25-50-53

015-035-0)∳a 7athion	206-050-1	298-00-0	R5	T+; N	%:T, N; R21-23/25-48/22-50-53 1 % ≤ C < 3 %: T, N; R23/25-48/22-50-53 0,25 % ≤ C < 1 %: Xn, N; R20/22-50-53 0,1 % ≤ C < 0,25 %: Xn, N; R20/22-51-53 0,025 % ≤ C < 0,1 %: N; R51-53 0,0025 % ≤ C < 0,025 %: C < 0,25 %: Xn, N; R20/22-51-53 C ≥ 25
	- methyl (ISO) O,O- dimethyl O-4- nitrophen phosphore			R10 T+; R26/28 T; R24 Xn; R48/22 N; R50-53	S:	%: T 64/2848/22-50/53 R24-26/28-48/22-50-53 36/03%45-60-61 ≤ C < 25 %: T+, N; R21-26/28-48/22-50-53 7 % ≤ C < 10 %: T+, N; R21-26/28-50-53 3 % ≤ C < 7 %: T, N; R21-23/25-50-53 1 % ≤ C < 3 %: T, N; R23/25-50-53 0,25 % ≤ C < 1 %: Xn, N; R20/22-50-53

							$\begin{array}{l} 0.1 \% \\ \leq C < \\ 0.25 \%: \\ Xn, N; \\ R20/22-5 \\ 0.025 \\ \% \leq C \\ < 0.1 \\ \%: N; \\ R51-53 \\ 0.0025 \\ \% \leq C \\ < 0.025 \\ \%: \\ R52-53 \\ \end{array}$	1-53
015-041-0 X	Othalathion (ISO) 1,2-bis (ethoxyca ethyl O,O-dimethyl phosphore	rbonyl)	204-497-7	7121-75-5	Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)24-60	R22-50-5	3
015-042-0	Ohforthior (common name not adopted by ISO) O-(3- chloro-4- nitrophen O,O- dimethyl phosphore	yl)	207-902-5	5500-28-7	Xn; R20/21/22 N; R50-53	Xn; N 2R: 20/21/22- S: (2-)13-60	R20/21/22	2-50-53

015-047-0	(ISO) O,O,O',O tetraethyl S,S'- methylene			3563-12-2	T; R25 Xn; R21 N; R50-53	S:	C≥25 %: T, N; S\$21-25-50-53 3 % ≤C 6/275-46:-60-61 Xn, N; R22-50-53 0,0025 % ≤ C < 3 %: N; R50-53 0,00025 % ≤ C < 0,0025 %: N; R51-53 0,000025 %: N; R51-53 0,000025 % ≤ C < 0,00025 %: Si, R51-53 0,000025 %: Si, R52-53
015-052-0 X	Ofenchlorp (ISO) O,O- dimethyl O-2,4,5- trichlorop phosphore	henyl	206-082-6	5299-84-3	Xn; R21/22 N; R50-53	Xn; N R: 21/22-50/ S: (2-)25-36/	
015-055-0	00a6ed (ISO) 1,2- dibromo-2 dichloroed dimethyl phosphate	thyl	206-098-3	3300-76-5	Xn; R21/22 Xi; R36/38 N; R50	Xn; N R: 21/22-36/ S: (2-)36/37-	R21/22-36/38-50
015-063-0 X	Odioxathion (ISO) 1,4- dioxan-2,i diyl- <i>O,O</i> , tetraethyl di(phosph	3-	201-107-7 te)	778-34-2	T+; R26/28 T; R24 N; R50-53	T+; N R: 24-26/28- S: (1/2-)28-3	C ≥ 25 %: T 56/№ R24-26/28-50-53 66/97-45-60-61 C < 25 %:T+,N; R21-26/28-50-53 3 % ≤ C < 7

							%:T, N; R21-23/2: 1 % ≤ C < 3 %: T, N; R23/25-50 0,1 % ≤ C < 1 %: Xn, N; R20/22-50 0,025 % ≤ C < 0,1 %: N; R50-53 0,00025 % ≤ C < 0,0025 %: N; R51-53 0,00025 %: C < 0,0025 %: C < 0,0025 %: S; R51-53 0,00025 %: R51-53)-53
015-065-0		hinyl)ethy	11	2703-37-9	T+; R26/27/28	T+; N		
	O,O- dimethyl phosphore		1]		N; R51-53	26/27/28- S: (1/2-)13-2		
015-076-0	θοθαsan O,O- diethyl O-(4- methylcon yl) phosphore			299-45-6	T+; R26/27/28 N; R50-53	26/27/28- S:	$C \ge 7$ %: T+, N; SN0266727/28 $1\% \le 28245760-6$ %: T, N; R23/24/23 $0,1\% \le C$ C < 1%: Xn, N; R20/21/23 0,025 % $\le C$ < 0,1 %: N; R50-53 0,0025 % $\le C$ < 0,025 % $\le C$ < 0,025 %: N; R51-53	1 5-50-53

							0,00025 % \le C \le 0,0025 %: R52-53
015-078-0	Memeton- methylsul S-2- ethylsulpl dimethyl phosphore	phon nonylethyl	241-109-5	517040-19	- 6 ; R25 Xn; R21 N; R51-53	T; N R: 21-25-51/ S: (1/2-)22-2	53
015-083-0	06 Asulide (ISO) O,O- diisoprop 2- phenylsul phosphore	yl phonylami		1741-58-2	Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)24-36	-60-61
015-084-0	ObHorpyrit (ISO) O,O- diethyl O-3,5,6- trichloro- pyridyl phosphore	os 2-	220-864-4	12921-88-2	2T; R25 N; R50-53	T; N R: 25-50/53 S: (1/2-)45-6	C≥25 %: T, N; R25-50-53 3 % ≤ C 60-65 %: Xn, N; R22-50-53 0,0025 % ≤ C < 3 %: N; R50-53 0,00025 % ≤ C < 0,0025 %: N; R51-53 0,000025 % ≤ C < 0,00025 %: Since the control of the control
015-095-0	Ondthamic (ISO) O,S- dimethyl phosphore	ophos amidothioa		010265-92	- 6 +; R26/28 T; R24 N; R50	T+; N R: 24-26/28- S: (1/2-)28-3	50 86/37-45-61
015-096-0 X		oton; hinyl)ethy odithioate		12497-07-6	T+; R28 T; R24 N; R50-53	T+; N R: 24-28-50/ S: (1/2-)28-3	C ≥ 25 %: T 56, N; R24-28-50-53 6/37-45-60-61 C < 25

						%:T +, N; R21-28-5 3 % ≤ C < 7 %:T, N; R21-25-5 1 % ≤ C < 3 %: T, N; R25-50-5 0,25 % ≤ C < 1 %: Xn, N; R22-50-5 0,1 % ≤ C < 0,25 %:Xn, N; R22-51-5 0,025 % ≤ C <	0-53 3
015-097-0	θtenthoat (ISO) ethyl 2- (dimethox phenylace	xyphosphir	02597-03-7 o)-2-	7Xn; R21/22 N; R50-53	Xn; N R: 21/22-50/ S: (2-)22-36/	R21/22-5	0-53
015-100-0 X	(ISO) α-	phosphino etonitrile	314816-18 o)	-Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)36-60	C ≥ 25 %: Xn, N; R22-50-5 •0,025 % ≤ C < 25 %: N; R50-53	3

						0,0025 % ≤ C < 0,025 %: N; R51-53 0,00025 % ≤ C < 0,0025 %: R52-53	
S-	domethyl	211-987-4	1732-11-6	Xn; R21/22 N; R50-53	Xn; N R: 21/22-50/ S: (2-)22-36	R21/22-50)-53
015-105- 0t -iphenyl phosphite		202-908-4	1101-02-0	Xi; R36/38 N; R50-53	Xi; N R: 36/38-50/ S: (2-)28-60	R36/38-50	
015-107-00tBopropi (ISO) ethyl-S,S dipropyl phosphor		236-152-	113194-48	T+; R26/27 T; R25 R43 N; R50-53	T+; N R: 25-26/27- S: (1/2-)27/2	43-50/53 8-36/37/39	9-45-60-61

		1	1		1			
O-bro dic O, dir ph	SO) 4- omo-2,5 chloroph <i>O</i> - methyl osphoro	- nenyl othioate	218-277-3	32104-96-3	3Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)36-60	R22-50-5	3
1- ph 3- (di	SO)	yl yphosphir		57700-17-6	6T; R24/25 N; R50-53	S:	C≥25 %: T, N; 5824/25-56 3 % ≤ C 36/375-46:6 Xn, N; R21/22-56 2,5 % ≤ C < 3 %: N; R50-53 0,25 % ≤ C < 2,5 %: N; R51-53 0,025 % ≤ C < 0,25 %: R52-53	0-61
O- cya O-	SO) 4- anopher ethyl		oate	13067-93	-T; R25-39/2; Xn; R21 Xi; R36 N; R51-53		39/25-51/5 7-45-61	53
S- chl O, die	SO) loromet		246-538-1	24934-91	- 6 +; R27/28 N; R50-53	T+; N R: 27/28-50/ S: (1/2-)28-3	53 86/37-45-6	0-61

015-115-0)0hl orthiop (ISO)	bhos	244-663-6	521923-23	-¶+; R28 T; R24 N; R50-53	T+; N R: 24-28-50/ S: (1/2-)28-3	53 86/37-45-60-61
015-122-(X	ethoxy-2- ethylpyrir yl <i>O,O</i> -			38260-54	-Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)60-61	R22-50-53
015-123-0	Cefiamipho (ISO) ethyl-4- methylthi- tolyl isopropyl phosphora	o- <i>m</i> -	244-848-7	122224-92	-6+; R28 T; R24 N; R50-53	T+; N R: 24-28-50/ S: (1/2-)23-2	C ≥ 25 %: T 56, N; R24-28-50-53 28-36

015-126-0	Official temporal (ISO) 7- chlorobicy dien-6- yl dimethyl phosphate	yclo(3.2.0)		023560-59	- 0 ; R25 N; R50-53	T; N R: 25-50/53 S: (1/2-)23-2	%: N; R51-53 0,0025 % ≤ C < 0,025 %: R52-53 C ≥ 25 %: T, N; R25-50-53 3 % ≤ C 28-23-%6-60-61 Xn, N; R22-50-53 0,25 % ≤ C < 3 %: N; R50-53 0,025 % ≤ C < 0,25 %: N; R51-53 0,0025 % ≤ C < 0,025 %: N; R51-53 0,0025 %: C < 0,025 %: R52-53
015-127-0	Opprobenson S-benzyl diisopropy phosphoro	yl	247-449-()26087-47	- % n; R22 N; R51-53	Xn; N R: 22-51/53 S: (2-)61	
015-128-0	S- ethylsulph O,O-	ninylmethy		5827-05-4	T+; R27 T; R25 N; R50-53	T+; N R: 25-27-50/ S: (1/2-)28-3	C ≥ 25 %: T 58, N; R25-27-50-53 66/97-45-60-61 < 25 %: T+, N; R22-27-50-53 3 % ≤ C < 7 %: T, N; R22-24-50-53 1 % ≤ C < 3 %: T, N; R24-50-53 0,25 % ≤ C < 1 %:

015-129-(Os& fenpho	S	246-814-	125311-71-	- ₩·	T; N	Xn, N; R21-50-5 0,1 % ≤ C < 0,25 %: Xn, N; R21-51-5 0,025 % ≤ C < 0,1 %: N; R51-53 0,0025 % ≤ C < 0,025 %: S R52-53	
	(ISO) O-ethyl O-2- isopropox isopropyl	ycarbonyl phosphora	phenyl- midothioat	e	R24/25 N; R50-53	R: 24/25-50/ S: (1/2-)36/3	C≥25 %: T, N; 5824/25-50 3 % ≤ C 7≈45-60-6 Xn, N; R21/22-50 0,25 % ≤ C < 3 %: N; R50-53 0,025 % ≤ C < 0,25: N; R51-53 0,0025 % ≤ C < 0,025 %: C < 0,025 %: R51-53	1
015-131-0	Osoxathion (ISO) O,O- diethyl O-5- phenyliso ylphospho	xazol-3-	242-624-8	318854-01	- S ; R24/25 N; R50-53	T; N R: 24/25-50/ S: (1/2-)28-3	53 86/37-45-6	0-61
015-132-0	(chloroph <i>O,O</i> -dimethylp	enylthiomo hosphoroc bophenoth	lithioate	953-17-3	T; R24/25 N; R50-53	S:	C≥25 %: T, N; 5\\\\ 24/25-50 3 % ≤ C 66/\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	0-61

							0,025 % \(\le C < 3 \) %: N; R50-53 0,0025 % \(\le C < 0,025 \) %: N; R51-53 0,00025 % \(\le C < 6,0025 \) %: R52-53	
015-133-(X	θiperopho (ISO) S-2- methylpip dipropyl phosphore	eridinocar	bonylmeth	24151-93 yl- <i>O,O</i> -	-Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)60-61	R22-50-5	3
015-134-(pitimipho methyl (ISO) O-(2- diethylam methylpyl yl) O,O- dimethyl phosphoro	ino-6- rimidin-4-	249-528-5	529232-93	-Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)60-61		
015-135-0	bromo-2- chlorophe O- ethylS- propyl phosphoro profenofo (ISO)	othioate	255-255-2	241198-08-	-Xn; R20/21/22 N; R50-53	Xn; N 2R: 20/21/22- S: (2-)36/37-	R20/21/22	2-50-53

							%: N; R51-53 0,00025 % \le C \le 0,0025 %: R52-53
015-136-0	isopropyl- [[(ethylan isopropyl 3-	nino)metho	oxyfosfino		-4; R25 N; R6t0-63 te; oxy]isocrot	S: (1/2-)37-4	C≥25 %: T, N; R25-50-53 3 % ≤ C 5-66-6d: Xn, N; R22-50-53 0,25 % ≤ C < 3 %: N; R50-53 0,025 % ≤ C < 0,25 %: N; R51-53 0,0025 % ≤ C < 0,025 %: C < 0,025 %: C < 0,025 %: R52-53
015-138-0	ปุ๊นInalpho (ISO) <i>O,O</i> - diethyl- <i>O</i> quinoxalin yl phosphoro	- n-2-	237-031-6	513593-03	- 8 ; R25 Xn; R21 N; R50-53	S:	C≥25 %: T, N; S\$21-25-50-53 3 % ≤ C 6/23-45:-60-61 Xn, N; R22-50-53 0,025 % ≤ C < 3 %: N; R50-53 0,0025 % ≤ C < 0,025 %: N; R51-53 0,00025 % ≤ C < 0,0025 %: S; R51-53 0,00025 %: C < 0,0025 %: S;

015-139-0	N Fart		235 062 9	\$13071-79	0 .+·	T+; N	C ≥ 7 %:	
013-139-0	butylthior	nethyl	433-303-6	φ13U/1-/9·	R27/28	1+, N R:	T+, N;	
	0,0-	11011111			N;		5B 27/28-50)-53
		osphorodit	hioate		R50-53	S:	$1\% \le 1$, ,,
	terbufos	ospiiorouit	moate		N30-33		7G45760-6	1
	(ISO)					(1/2-)30/3	%: T, N;	1
	(130)						R24/25-50	52
							l I)- 33
							$\begin{vmatrix} 0.1 \% \le \\ C < 1 \% \end{aligned}$	
							l I	
							Xn, N;	
							R21/22-50)-53
							0,025	
							% ≤ C	
							< 0,1	
							%: N;	
							R50-53	
							0,0025	
							$\% \leq C$	
							< 0,025	
							%: N;	
							R51-53	
							0,00025	
							$ \% \le C < $	
							0,0025	
							% :	
							R52-53	
015-154-0	00-4		240-718-	316672-87	-0Xn·	С	C ≥ 25	
010 10 .		ylphosphoi		100,20,	R20/21	R:	%: C;	
	acid) -pp			C; R34		53225 0321-34	1-52/53
	ethephon				R52-53	S:	10 % ≤	
	1						28E-362357/3	9-45-61
						(-/-)	%: C;	
							R34	
							5 % ≤	
							C < 10	
							%: Xi;	
							R36/37/38	}
							120 0/0 // 0	-
015-179-0			422-720-8	8166242-5				
	condensat	ion			3; R40	R:		
	product				Xn;		43-48/22-	50/53
	of:				R22-48/22			
		droxymeth	ylphospho	nium	C; R34	(1/2-)26-3	6/37/39-4	5-60-61
	chloride,				R43			
	urea and				N;			
	distilled				R50-53			
				1				
	hydrogen	ated						
	hydrogena C16-18	ated						
	hydrogens C16-18 tallow							
	hydrogena C16-18							
016 001 (hydrogens C16-18 tallow alkylamin	e	221 077	27702 06	IE±. D12	F. T.		
016-001-0	hydrogens C16-18 tallow	e	231-977-3	37783-06-4		F+; T+; N		

				N; R50	R: 12-26-50 S: (1/2-)9-16 45-61	5-36-38-
016-008-0@ramoniu polysulph		32-989-1	9080-17-5	5R31 C; R34 N; R50	C; N R: 31-34-50 S: (1/2-)26-4	C ≥ 25 %: C, N; R31-34-50 5 % ≤ ££6125 %: C; R31-34 1 % ≤ C < 5 %: Xi; R31-36/38
016-012-0disulphur dichloride sulfur monochlo		33-036-2	10025-67	R14 T; R25 Xn; R20 R29 C; R35 N; R50	T; C; N R: 14-20-25- S: (1/2-)26-3	C ≥ 25 %: T, 29. N5. 50 R20-25-35-50 R20-25-35-50 R20-25-35-61 C < 25 %: C; R22-35 5 % ≤ C < 10 %: C; R22-34 3 % ≤ C < 5 %: Xn; R22-36/37/38 1 % ≤ C < 3 %: Xi; R36/37/38
016-013-00ulphur X dichloride		34-129-0	10545-99	C; R34 Xi; R37 N; R50	C; N R: 14-34-37- S: (1/2-)26-4	10 % ≤ ££6125 %: C; R34 5 % ≤ C < 10 %: Xi; R36/37/38
016-014- 00ப5 phur tetrachlor	ide		13451-08	- R 14 C; R34 N; R50	C; N R: 14-34-50	C ≥ 25 %: C, N; R34-50

016-021-0m3thaneth methyl mercaptar		200-822-1	174-93-1	F+; R12 T; R23 N; R50-53	S: (1/2-)26-4 F+; T; N R: 12-23-50/ S: (2-)16-25	R34 5 ≤ C < 10 %: Xi; R36/37/38
016-023-0dianethyl sulphate	E	201-058-1	177-78-1	Carc. Cat. 2; R45 Muta. Cat. 3; R68 T+; R26 T; R25 C; R34 R43	` '	C ≥ 25 %: T+;

							R43- R68 1 % ≤ C < 3 %: T; R45- R23- R43- R68 0,1 % ≤ C < 1 %: T; R45- R20- R68 0,01 % ≤ C < 0,1 %: T; R45- R68
016-059-0	004,007, N', N', N', tetramethy mine dihydroch	yldithiobis		917339-60- lia	-Xn; R22 Xi; R36 R43 N; R50-53	Xn; N R: 22-36-43- S: (2-)26-36	
017-003-0	0 6 ഷങium chlorate		236-760-7	713477-00	Φ; R9 Xn; R20/22 N; R51-53	O; Xn; N R: 9-20/22-5 S: (2-) 13-27-61	1/53
017-004-(ှာ dassium chlorate	ı	223-289-7	73811-04-9	OO; R9 Xn; R20/22 N; R51-53	O; Xn; N R: 9-20/22-5 S: (2-)13-16	
017-005-(ØeØlium chlorate		231-887-4	17775-09-9	OO; R9 Xn; R22 N; R51-53	O; Xn; N R: 9-22-51/5 S: (2-) 13-17-46-	
017-011-0	000dium hypochlor solution % Cl active	B rite,	231-668-3	37681-52-9	9C; R34 R31 N; R50	C; N R: 31-34-50 S: (1/2-)28-4	10 % ≤

017-012-00alc	ium ochlorite	231-908-7	77778-54-3	3O; R8 Xn; R22 R31 C; R34 N; R50	8-22-31-3 S:	%: Xi; R31-36/38 C≥25 %: C, N; 4₹20-34-50 10 % ≤ 3€37259-45-61 %: C; R34 3 % ≤ C < 10 %: Xi; R37/38-41 0,5 % ≤ C < 3 %: Xi; R36
024-001-00100 (VI) triox		215-607-8	31333-82-0	O; R9 Carc. Cat. 1; R45 Muta. Cat. 2; R46 Repr. Cat. 3; R62 T+; R26 T; R24/25-4 C; R35 R42/43 N; R50-53	R: 45-46-9-2 S: 53-45-60-	$\begin{array}{l} C \geq 25 \\ \%: T \\ +, N; \\ \mathcal{R}24/26\mathbf{-35}\mathbf{-32}/43/48/23\mathbf{-48}\mathbf{-38}/23\mathbf{-50}/53\mathbf{-}10 \% \\ 61C < \\ 25 \%: \\ T+, N; \\ R21/22\mathbf{-26}\mathbf{-35}\mathbf{-42}/43\mathbf{-45}\mathbf{-46}\mathbf{-48}/23\mathbf{-51}/53\mathbf{-}7 \% \leq \\ C < 10 \\ \%: T \\ +, N; \\ R21/22\mathbf{-26}\mathbf{-34}\mathbf{-42}/43\mathbf{-45}\mathbf{-46}\mathbf{-48}/20\mathbf{-51}/53\mathbf{-}5 \% \leq \\ C < 7 \\ \%: T, N; \\ R21/22\mathbf{-23}\mathbf{-34}\mathbf{-42}/43\mathbf{-45}\mathbf{-46}\mathbf{-48}/20\mathbf{-51}/53\mathbf{-}3 \% \leq \\ C < 5 \\ \%: T, N; \\ R21/22\mathbf{-23}\mathbf{-36}/37/38\mathbf{-42}/43\mathbf{-45}\mathbf{-46}\mathbf{-48}/20\mathbf{-51}/53\mathbf{-}3 \% \leq \\ C < 3 \\ \%: T, N; \\ R23\mathbf{-36}/37/38\mathbf{-42}/43\mathbf{-45}\mathbf{-46}\mathbf{-48}/20\mathbf{-51}/53\mathbf{-}1 \% \leq \\ C < 2,5 \\ \%: T; \\ R23\mathbf{-36}/37/38\mathbf{-42}/43\mathbf{-45}\mathbf{-46}\mathbf{-48}/20\mathbf{-52}/53\mathbf{-}3 \% \leq C < \\ 1 \%: T; \\ R23\mathbf{-36}/37/38\mathbf{-42}/43\mathbf{-45}\mathbf{-46}\mathbf{-48}/20\mathbf{-52}/53\mathbf{-}3 \% \leq C < \\ 1 \%: T; \\ R20\mathbf{-45}\mathbf{-46}\mathbf{-52}/53\mathbf{-}3 \% \leq C < 0,25 \end{array}$

						%: T;	
						R20-45-4	6
024-002-0perassiu dichrom	ım E nate	231-906-0	57778-50-9	Carc. Cat.		C ≥ 25 %: T	3
				2: R45 Muta. Cat. 2;	R: 45-46-60- S:	+, N; 6148-26-8 6 10 %	5-06-24-22/26-38/22/30/513 /23
				R46 Repr. Cat. 2;	53-45-60-	61C < 25 %: T+, N;	
				R60-61 T+; R26 T;			0-61-22-26-34-42/43-48/23-51,
				R25-48/23 Xn; R21 C; R34	3	T+, N;	0-61-22-26-36/37/38-42/43-48/
				R42/43 N; 50-53		C < 7 %: T, N;	0. (1. 22. 22. 24/27/29. 42/42. 49
						3 % ≤ C < 5	0-61-22-23-36/37/38-42/43-48/
						2,5 %	0-61-22-23-42/43-48/20-51/53
							0-61-23-42/43-48/20-51/53
						1 % ≤ C < 2,5 %: T;	
						R45-46-60 0,5 % < C <	0-61-23-42/43-48/20-52/53
						1 %: T; R45-46-	42/43-52/53
						0,25 % ≤ C < 0,5	T2/T3-32/33
						%: T; R45-46-20	0-42/43-52/53
						0,2 % ≤ C < 0,25 %: T;	
						R45-46-20 0,1 % \le C < 0,2	0-42/43
						%: T; R45-46-20	0
024-003-00mlmon		232-143-	17789-09-:	E; R2 O; R8	E; T+; N R:	%: T	3 25 26 34 42/43 48/23 50/53
					43-40-00-	101-2-8-21	-25-26-34-42/43-48/23-50/53

224 004 (224 100 2		Muta. Cat. 2; R46 Repr. Cat. 2; R60-61 T+; R26 T; R25-48/23 Xn; R21 C; R34 R42/43 N; R50-53	53-45-60-	+, N; 6R45-46-60-61-21-25-26-34-42/43-48/23-10 % ≤ C < 25 %: T+, N; R45-46-60-61-22-26-34-42/43-48/23-50, 7 % ≤ C < 10 %: T+, N; R45-46-60-61-22-26-36/37/38-42/43-48/, 5 % ≤ C < 7 %: T, N; R45-46-60-61-22-23-36/37/38-42/43-48/, 3 % ≤ C < 5 %: T, N; R45-46-60-61-22-23-42/43-48/20-51/53, 2,5 % ≤ C < 3 %: T, N; R45-46-60-61-23-42/43-48/20-51/53, 1 % ≤ C < 2,5 %: T; R45-46-60-61-23-42/43-48/20-52/53, 0,5 % ≤ C < 1 %: T; R45-46-60-61-20-42/43-52/53, 0,25 % ≤ C < 0,5 %: T; R45-46-20-42/43-52/53, 0,1 % ≤ C < 0,2 %: T; R45-46-20-42/43, 0,1 % ≤ C < 0,2 %: T; R45-46-20
024-004-0	dichromat anhydrate	234-190-3	10588-01	Carc. Cat. 2; R45 Muta. Cat. 2;	R:	C≥25 %: T +, N; +, N; +, M348-26-25-26-26-24-23/25-38/213/30/313/23 10 % -61 C < 25 %:

			Repr. Cat. 2; R60-61 T+; R26 T; R25-48/2 Xn; R21 C; R34 R42/43 N; 50-53	3	7 % ≤ C < 10 %: T+, N; R45-46-66 5 % ≤ C < 7 %: T, N; R45-46-66 2,5 %: T, N; R45-46-66 1 % ≤ C < 2,5 %: T; R45-46-66 0,5 %: C < 1 %: T; R45-46-66 0,25 % ≤ C < 0,5 %: T;	0-61-22-23-36 0-61-22-23-42 0-61-23-42/43 0-61-23-42/43 0-61-20-42/43	3-48/20-52/53 3-52/53
024-004-0\$edium dichroma dihydrate	234-190-3	37789-12-0	Carc. Cat R45 Muta. Cat. 2; R46 Repr. Cat. 2; R60-61 T+; R26	R:	10 % 61 C < 25 %: T+, N; R45-46-60 7 % ≤ C		5- 48/23/30/33 8/23 1-42/43-48/23-51/
			T; R25-48/2 Xn; R21	3	< 10 %: T+, N; R45-46-60	0-61-22-26-36	5/37/38-42/43-48/

					C; R34 R42/43 N; R50-53		$\begin{array}{l} 3~\% \leq \\ C < 5 \\ \%: T, N; \\ R45-46-6 \\ 2,5~\% \\ \leq C < 3 \\ \%: T, N; \\ R45-46-6 \\ 1~\% \leq \\ C < 2,5 \\ \%: T; \\ R45-46-6 \\ 0,5~\% \\ \leq C < \\ 1~\%: T; \\ R45-46-6 \\ 0,25 \\ \% \leq C \\ < 0,5 \\ \%: T; \end{array}$	
024-011-0	0a+fimonium bis(1- (3,5- dinitro-2- oxidopher (N- phenylcar naphthola	nylazo)-3- bamoyl)-2		2	F; R11 N; R50-53	F; N R: 11-50/53 S: (2-)33-60	-61	
024-018-0)ବୈଷାଧି um chromate	E	231-889-5	57775-11-3	Carc. Cat. 2; R45 Muta. Cat. 2; R46 Repr. Cat.2; R60-61 T+; R26 T; R25-48/2;	R: 45-46-60- S: 53-45-60-	R45-46-6 610 % ≤ C < 25 %: T+, N;	3 26-34-42/43-48/23-50/53 0-61-21-25-26-34-42/43-48/23- 0-61-22-26-34-42/43-48/23-51,

					Xn; R21 C; R34 R42/43 N; R50-53		5 % ≤ C < 7 %: T, N;	0-61-22-26-36/37/38-42/43-48 0-61-22-23-36/37/38-42/43-48
							C < 5 %: T, N; R45-46-6 2,5 % ≤ C < 3	0-61-22-23-42/43-48/20-51/53
							%: T, N; R45-46-6 1 % ≤ C < 2,5	0-61-23-42/43-48/20-51/53
							%: T; R45-46-6 0,5 % ≤ C < 1 %: T;	0-61-23-42/43-48/20-52/53
							R45-46-6 0,25 % \le C < 0,5	0-61-20-42/43-52/53
							%: T; R45-46-2 0,2 % ≤ C < 0,25 %: T;	0-42/43-52/53
							R45-46-2 0,1 % ≤ C < 0,2 %: T; R45-46-2	
027-004-0	00etbalt dichloride	E ;	231-589-4	17646-79-9	Carc. Cat. 2; R49 Xn; R22 R42/43 N;	R:	C ≥ 25 %: T, N; 484902534 2,5 % ≤	1 2/43-50/53
					R50-53		%: T, N; R49-22-4 1 % ≤ C < 2,5 %: T;	2/43-51/53
							R49-42/4 0,25 % ≤ C < 1 %: T; R49-52/5 0,01 %	
							≤ C <	

	I	I	I	l	ı		0.25.0/+	I
							0,25 %: T; R49	
027-005-0	Oe balt sulphate	Е	233-334-2	210124-43	-Carc. Cat. 2; R49 Xn; R22 R42/43 N; R50-53	R:	2,5 % ≤	3-52/53
029-002-0 X	Odicopper oxide copper (I) oxide		215-270-7	71317-39-1	Xn; R22 N; 50-53	Xn; N R: 22-50/53 S: (2-)22-60	-61	
030-001-0	Deinc powder - zinc dust (pyrophor	ric)	231-175-3	37440-66-6	6F; R15-17 N; R50-53	F; N R: 15-17-50/ S: (2-)43-46		
030-002-0	Daific powder - zinc dust (stabilized	d)	231-175-3	37440-66-6	6N; R50-53	N R: 50/53 S: 60-61		
030-003-0)ผู้เนิด chloride		231-592-0)7646-85-7	7Xn; R22 C; R34 N; R50-53	S:	C ≥ 25 %: C, N; \$322-34-5 10 % ≤ 66/37/239-4 %: C, N; R34-51/5. 5 % ≤ C < 10 %: Xn, N; R36/37/33 2.5 % ≤ C < 5 %: N; R51/53	5-60-61 3

							0.25 % ≤ C < 2.5 %: R52/53	
030-006-0	sulphate (hydrous) (mono-, hexa-and hepta hydrate) [1] zinc sulphate (anhydrou [2]		[1]	37446-19-7 [1] 37733-02-0 [2]	R41	Xn; N R: 22-41-50/ S: (2-)22-26	53 -39-46-60-	61
033-001-0 X)@rsenic		231-148-6	57440-38-2	2T; R23/25 N; R50-53	T; N R: 23/25-50/ S: (1/2-)20/2	53 1-28-45-6	0-61
033-002-0	compound with the exception of those specified elsewhere in this Annex				T; R23/25 N; R50-53	S:	C ≥ 25 %: T, N; 5823/25-50 2,5 % ≤ C28245-60 %: T, N; R23/25-51 0,25 % ≤ C < 2,5 %: T; R23/25-52 0,2 % ≤ C < 0,25 %: T; R23/25 0,1 % ≤ C < 0,2 %: T; R23/25 %: T; R23/25	0-61 1/53
042-002-0	Othtakis(donium) hexa-μ- oxotetra- μ3- oxodi- μ5- oxotetrado	ecaoxoocta			543 R23 Xi; R41 R53	T R: 23-41-53 S: (1/2-)26-3	37/39-45-6	1

048-001-0	0 อสีmium				Xn;	Xn; N	C≥	1
	compound	ds,			R20/21/22		25 %:	
	with the				N;	20/21/22-		
	exception				R50-53	S:	R20/21/22	2-50/53
	of					(2-)60-61	2,5 %	
	cadmium						≤ C <	
	sulphosel	enide					25 %:	
	(xCdS.yC						Xn, N;	
	mixture	use),					R20/21/22	2-51/53
	of						0,25	2-31/33
	cadmium						% ≤ C	
							< 2.5	
	sulphide							
	with						%: Xn;	. 50/50
	zinc						R20/21/22	2-52/53
	sulphide						0,1 % ≤	
	(xCdS.yZ	nS),					C < 0.25	
	mixture						%: Xn;	
	of						R20/21/22	2
	cadmium							
	sulphide							
	with							
	mercury							
	sulphide							
	(xCdS.yH	σS)						
	and	50),						
	those							
	specified							
	elsewhere							
	in this							
	Annex							
048-003-0	0a6 mium		224-729-0)4464-23-	7T;	T; N	$C \ge 25$	
	diformate				R23/25	R:	%: T, N;	
	cadmium				R33			3-50/53-68
	Cuaminam				Xn; R68	S:	10 % ≤	2 2 0 1 2 2 0 0
					N;	(1/2-)22-4		
					R50-53	(1/2-)22	%: T, N;	
					K30-33			3-51/53-68
								5-51/55-06
							2,5 %	
							≤ C <	
							10 %:	
							Xn, N;	
								3-51/53-68
							1 % ≤	
							C < 2,5	
							%: Xn;	
								3-52/53-68
							0,1 %	
							$\leq C < 1$	
							%: Xn;	
							R20/22-3	3_52/53
							0,25	1-34133
							0,23 $\% \le C$	
							/0 ≥ C	

						< 0,1 %: Xn; R20/22-33-52/53
048-004-0	Oal lmium cyanide	208-829-1	1542-83-6	T+; R26/27/28 R32 R33 Xn; R68 N; R50-53	26/27/28- S:	$C \ge 25$ %: T 32-B8-68-50/53 R26/27/28-32-33-50/53-68 3-29-45-60-61 <25 %: T+, N; R26/27/28-32-33-51/53-68 $\le C < 7$ %: T, N; R23/24/25-32-33-51/53-68 1 % $\le C < 2,5$ %: T; R23/24/25-32-33-52/53-68 0,25 % $\le C < 1$ %: Xn; R20/21/22-33-52/53 $0,1$ % $\le C < 0,25$ %: Xn; R20/21/22-33
	0admiuml cadmium fluorosilio	s 2l4d a(0 %2-)	017010-21	-8; R23/25 R33 Xn; R68 N; R50-53	T; N R: 23/25-33- S: (1/2-)22-4	C ≥ 25 %: T, N; 6823025333-50/53-68 10 % ≤ 15:60261 %: T, N; R23/25-33-51/53-68 2,5 % ≤ C< 10 %: Xn, N; R20/22-33-51/53-68 1 % ≤ C < 2,5 %: Xn; R20/22-33-52/53-68 0,25 % ≤ C < 1 %: Xn; R20/22-33-52/53 0,1 % ≤ C < 0,25 %: Xn; R20/22-33-52/53 0,1 % ≤ C < 0,25 %: Xn; R20/22-33-52/53

	1		1	-				
048-006-0 0a2 lmi	um E	232-222-0)7790-79-6	Carc. Cat.	T+; N	$C \ge 25$		
fluorio	le			2; R45	R:	%∷ T		
				Muta.	45-46-60-	64,-1215-26-	48/23/25-50/	53
				Cat. 2;	S:			8/23/25-50/53
					53-45-60-			
				Repr.		≤ C <		
				Cat. 2;		25 %:		
				R60-61		T+, N;		
				T+; R26			0 61 25 26 4	8/23/25-51/53
				-			0-01-23-20-4	0/23/23-31/33
				T;	2/25	$7\% \le C$		
				R25-48/2	3/25	< 10 %:		
				N;		T+, N;		_,,
				R50-53			0-61-22-26-4	8/23/25-51/53
						2,5 %		
						\leq C $<$ 7		
						%: T, N;		
						R45-46-6	0-61-22-23-4	8/20/22-51/53
						1 % ≤		
						C < 2,5		
						%: T;		
							 0_61_22_23_4	8/20/22-52/53
						0,5 %	0-01-22-23-4	0120122-32133
						≤ C <		
						1 %: T;	0 (1 00/00 4	0/00/00 50/50
							0-61-20/22-4	8/20/22-52/53
						0,25		
						% ≤ C		
						< 0,5		
						%: T;		
						R45-46-2	0/22-48/20/2	2-52/53
						$0,1 \% \le$		
						C < 0.25		
						%: T;		
							0/22-48/20/2	2
						0,01 %	0/22 10/20/2	4
						0,01 70 ≤ C <		
						0,1 %:		
						T; R45		
048-007-00a&mi	um	232-223-6	57790-80-9	T;	T; N	C ≥ 25		
iodide			1	R23/25	R:	%: T, N;		
104144				R33			3-50/53-68	
				Xn; R68	S:	10 % ≤	00/23 00	
				N;	(1/2-)22-4			
					(1/2-)22-2			
				R50-53		%:T, N;	2.51/52.60	
							3-51/53-68	
						2,5 %		
						≤ C<		
						10 %:		
						Xn, N;		
						R20/22-3	3-51/53-68	
						1 % ≤		
						C < 2,5		
						2,5	I	

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048-008-00admium E chloride	233-296-710108-64	2; R45 Muta. Cat. 2;	R: 45-46-60- S: 53-45-60-	$ \leq C < 25\% : \\ T+,N; \\ R45-46-60-61-25-26-48/23/25-51/53 \\ 7\% \leq C < 10\% : \\ T+,N; \\ R45-46-60-61-22-26-48/23/25-51/53 \\ 2,5\% \\ \leq C < 7 \\ \%: T, N; \\ R45-46-60-61-22-23-48/20/22-51/53 \\ 1\% \leq C < 2,5 \\ \%: T; \\ R45-46-60-61-22-23-48/20/22-52/53 \\ 0,5\% \\ \leq C < 1\% : T; \\ R45-46-60-61-20/22-48/20/22-52/53 \\ 0,25 \\ \% \leq C < 0,5 \\ \%: T; \\ R45-46-20/22-48/20/22-52/53 \\ 0,1\% \leq C < 0,25 \\ \end{aligned} $
048-009-00a@mium E sulphate	233-331-610124-36	2; R45	R:	%: T; R45-46-20/22-48/20/22 0,01 % ≤ C < 0,1 %: T; R45 C≥ 25 %: T 64-N5-26-48/23/25-50/53 R45-46-60-61-25-26-48/23/25-50/53

				Muta. Cat. 2; R46 Repr. Cat. 2; R60-61 T; R48/23/2: T+; R26 T; R25 N; R50-53	S: 53-45-60-	$\begin{array}{c} 25 \%: \\ T+, N; \\ R45-46-6 \\ 7 \% \\ \leq C < \\ 10\%: T \\ +, N; \\ R45-46-6 \\ 2,5 \% \\ \leq C < 7 \\ \%: T, N; \\ R45-46-6 \\ 1 \% \leq \\ C < 2,5 \\ \%: T; \\ R45-46-6 \\ 0,5 \% \\ \leq C < \\ 1 \%: T; \\ R45-46-6 \\ 0,25 \\ \% \leq C \\ < 0,5 \\ \%: T; \\ R45-46-2 \\ 0,1 \% \leq \\ C < 0,25 \\ \%: T; \end{array}$	0-61-25-26-48/23/25-51/53 0-61-22-26-48/23/25-51/53 0-61-22-23-48/20/22-51/53 0-61-22-23-48/20/22-52/53 0-61-20/22-48/20/22-52/53 0/22-48/20/22
048-010-0 0a4 mium	E	215-147-8	31306-23-6	6Carc. Cat.	T; N	$T; R45$ $C \ge 25$	1
sulphide				2; R45 Muta. Cat. 3; R68 Repp. Cat. 3; R62-63 T; R48/23/25 Xn; R22 R53	R: 45-22-48/ S: 53-45-61	%: T; 284255262246 10 % ≤ C < 25 %: T; R45-22-4 5 % ≤ C < 10 %: T;	8326825362-63-68-53 8/23/25-62-63-68 0/22-62-63-68

050-001-0	ปีเช tetrachlor stannic	ide	231-588-9	97646-78-8	8C; R34 R52-53	C R: 34-52/53	1 %: T; R45-48/20 C ≥ 25 %: C; R34-52/5	
	chloride					S: (1/2-)7/8-	10 % ≤ 26-45-51 %: C; R34 5 % ≤ C < 10 %: Xi; R36/37/38	3
050-005-0	O ri 7 methylt				T+;	T+; N	C ≥ 25	1
	compound with the	as,			R26/27/28 N;	sk: 26/27/28-	%: T 5⊕/ % ®	
	exception				R50-53	S:	R26/27/28	
	of those specified					(1/2-)26-2	27-38⁄4 5-6 ≤ C <	0-61
	elsewhere	}					25 %:	
	in this Annex						T+, N; R26/27/28	R-51/53
	Ailica						0,5 % ≤	5-51/55
							C < 2,5 %: T+;	
							R26/27/28	3-52/53
							0,25 % ≤ C	
							< 0,5	
							%: T; R23/24/25	5-52/53
							0,1 % ≤)-32/33
							C < 0,25 %: T;	
							R23/24/25	5
							0,05 % ≤ C	
							< 0,1	
							%: Xn; R20/21/22	2
050-006-0) 0∹i ≥thyltin				T+;	T+; N	C ≥ 25	1
	compound with the	ds,			R26/27/28 N;	3R: 26/27/28-	%: T	
	exception				N, R50-53	S:	30, 30, R26/27/28	3-50/53
	of those					(1/2-)26-2	2 2-38/4 5-6	0-61
	specified elsewhere	,					≤ C < 25 %:	
	in this						T+, N;	0 51/52
	Annex						$ R26/27/28 0.5 \% \le $	5-31/33
							C < 2,5	

				%: T+; R26/27/2 0,25 % ≤ C < 0,5 %: T; R23/24/2 0,1 % ≤ C < 0,25 %: T; R23/24/2 0,05 % ≤ C < 0,1 %: Xn; R20/21/2	5-52/53 5
wit exc of t spe else in t	npounds, h the eption hose cified ewhere	T; R23/24 N; R50-5	23/24/25 S:	C ≥ 25 %: T, N; -580253324/2 2,5 % ≤ 27G282455-6 %: T, N; R23/24/2 0,5 % ≤ C < 2,5 %: T; R23/24/2 0,25 % ≤ C < 0,5 %: Xn; R20/21/2 0,1 % ≤C < 0,25 %: Xn; R20/21/2	5-51/53 5-52/53 2-52/53
wit excoof t spe else in t	npounds, h the eption hose cified ewhere	T; R25-4; Xn; R3 Xi; R36/3; N; R50-5;	S: (1/2-)35-	2,5 % ≤ 36/34/259-4 %: T, N; R21-25-3 1 % ≤ C < 2,5 %: T;	1 26338948323/25-50/53 5-60-61 6/38-48/23/25-51/53 6/38-48/23/25-52/53

							%: Xn; R22-48/2	0/22-52/53
050-009-0	[1]	entylstanna yldistannox	[1]	[1]	R20/21/22	Xn; N 2R: 20/21/22- S: (2-)26-28	R20/21/22	2-51/53
050-010-0	Oh4 orotrih	exylstanna	n243-547-2	220153-50	- % n; R20/21/22 N; R50-53	Xn; N 2R: 20/21/22- S: (2-)26-28	R20/21/22	2-51/53
050-011-0 X	Oriphenylt compound with the exception of those specified elsewhere in this Annex	ds,			T; R23/24/2; N; R50-53	23/24/25- S:	C ≥ 25 %: T, N; 5025324/2: 2,5 % ≤ 27528245-6 %: T, N; R23/24/2: 1 % ≤ C < 2,5 %:T; R23/24/2: 0,25 % ≤ C < 1 %: Xn; R20/21/2:	0-61 5-51/53 5-52/53

050 012 (0.5	1 4 1 .	015 010	11440 55	137	37 37	a.	1
050-012-0		h & xylstanr				Xn; N	C ≥	1
	[1]	1_1	[1]	[1]	R20/21/22		25 %:	
		yclohexyls				20/21/22-		50/53
	[2]		[2]	[2]	R50-53	S:	R20/21/22	2-50/53
		clohexylst)	(2-)26-28		
	[3]		[3]	[3]			≤ C <	
							25 %:	
							Xn, N;	
							R20/21/22	2-51/53
							1 % ≤	
							C < 2.5	
							%: Xn;	
							R20/21/22	2-52/53
							0,25 %	
							≤ C <	
							1 %:	
							R52/53	
050 012 0	0ti∂ ctyltin	٨			Vi·	Xi	C ≥ 25	1
030-013-0	compound				Xi; R36/37/38		%: Xi;	1
	with the	μ5,			R53		5 B 36/37/3	2 52
					NJ3	S: (2-)61		b-33
	exception					5. (2-)01		
	of those						C < 25	
	specified						%: Xi;	
	elsewhere	Ì					R36/37/3	8
	in this							
	Annex							
051-002-0	@n3 imony		231-601-8	37647-18-9	9C; R34	C; N	C ≥ 25	
	pentachlo	ride			N;	R:	%: C, N;	
					R51-53	34-51/53	R34-51/5	3
						S:	10 % ≤	
						(1/2-)26-4	€ 6125	
							%: C;	
							R34-52/5	3
							5 % ≤	
							C < 10	
							%: Xi;	
							R36/37/38	8-52/53
							2,5 % ≤	22 ,23
							C < 5%:	
							R52/53	
051 002 0)	A			V	3 7 3 .T		1
US1-UU3-(O r O imony				Xn;	Xn; N	C≥	1
	compound	ns,			R20/22	R:	25 %:	
	with the				N;	20/22-51/		1 /50
	exception				R51-53	S: (2-)61		1/53
	of the						2,5 % ≤	
	tetroxide						C < 25	
	$(Sb_2O_4),$						%: Xn;	
	pentoxide						R20/22-5	2/53
	$(Sb_2O_5),$						0,25	
	trisulphid	e					% ≤ C	
	$(Sb_2S_3),$	-					< 2,5	
	(~~2~3),					I	l í	I

pentasul (Sb ₂ S ₅) and those specified elsewher in this Annex					%: Xn; R20/22	
080-002-00a6rgani compour of mercury with the exception of mercuric sulphide and those specified elsewher in this Annex	nds n		T+; R26/27/28 R33 N; R50-53	26/27/28- S:	R26/27/28 R26/27/28 R26/27/28 S=\$ %: T+, N; R26/27/28 C < 2,5 %: T+; R26/27/28 0,5 % S= C < 2 %: T; R23/24/28 0,25 % S C < 0,5 %: Xn;	8-33-51/53 8-33-52/53 5-33-52/53
080-004-00rganic compound of mercury with the exception of those specified elsewher in this Annex	n I		T+; R26/27/28 R33 N; R50-53	26/27/28- S:	R26/27/28 28,36/45-6 ≤ C < 25 %: T+, N; R26/27/28 1 % ≤ C < 2,5 %: T+; R26/27/28 0,5 % ≤ C < 1 %: T;	1 8-33-50/53 0-61 8-33-51/53 8-33-52/53

	Odifinethylr [1] diethylme [2]	ercury	[1]	3593-74-8 [1] 7627-44-1 [2]	R26/27/28	26/27/28- S: (1/2-)13-2	R26/27/28-33-50/53 28,36/45-60-61 < C < 25 %: T+, N; R26/27/28-33-51/53 0,5 % ≤ C < 2,5 %: T+; R26/27/28-33-52/53 0,25 % ≤ C < 0,5 %: T; R23/24/25-33-52/53 0,1 % ≤ C < 0,25 %: T; R23/24/25-33 0,05 % ≤ C < 0,1 %: Xn; R20/21/22-33
082-001-0	compound with the exception of those specified elsewhere in this Annex				Repr. Cat. 1; R61 Repr. Cat. 3; R62 Xn; R20/22 R33 N; R50-53	T; N R: 61-20/22- S: 53-45-60-	C≥25 1 %: T, N; 338662250253333-62-50/53 5 % ≤ 61 < 25 %:T, N; R61-20/22-33-62-51/53 2,5 % ≤ C < 5 %:T, N; R61-20/22-33-62-51/53 1 % ≤ C < 2,5 %:T; R61-20/22-33-52/53

	AE			Repr.	T+; N	0,5 % ≤ C < 1 %:T; R61-33-5 0,25 % ≤ C < 0,5 %: R52/53 C ≥ 25 %: T	2/53	
alkyls				Cat. 1; R61 Repr. Cat. 3; R62 T+; R26/27/28 R33 N; R50-53	S: 53-45-60-	28,N3-62-5 R61-26/2' 61% ≤ C < 25 %:T +, N; R61-26/2' 2,5 % ≤ C < 5 %: T+, N; R61-26/2' 0,5 % ≤ C < 2,5 %: T+; R61-26/2' 0,25 % ≤ C < 0,5 %: T;	7/28-33-62 7/28-33-62 7/28-33-51 7/28-33-52 7/28-33-52	-51/53 /53 /53
601-010-0 0 tBylene		200-815-3	374-85-1	F+; R12 R67	F+ R: 12-67 S: (2-)9-16-3			
601-014-00s6prene (stabilized 2- methyl-1,; - butadiene		201-143-3	378-79-5	F+; R12 Car. Cat. 2; R45 Muta. Cat. 3; R68 R52-53	F+; T R: 45-12-68- S: 53-45-61	52/53		
601-017-00yblohexa	ne	203-806-2	2110-82-7	F; R11	F; Xn; N		4 6	

					Xn; R65 Xi; R38 R67 N; R50-53	11-38-65- S:	67-50/53 25-33-60-6	1-62
601-020-0	0 6 € mazene	E	200-753-7	771-43-2	F; R11 Carc Cat. 1; R45 Muta. Cat. 2; R46 T; R48/23/24 Xn; R65 Xi; R36/38	S: 53-45	36/38-48/2	23/24/25-65
601-021-0	0⊕Buene		203-625-9	0108-88-3	F; R11 Repr. Cat.3; R63 Xn; R48/20-6; Xi; R38 R67	S: (2-)36/37	20-63-65-6 -62-46	4,6
601-025-0	0046sitylen 1,3,5- trimethyll		203-604-4	4108-67-8	R 10 Xi; R37 N; R51-53	Xi; N R: 10-37-51/ S: (2-)61		3
601-027-0	0 2- 6 phenylpro α- methylstr	-	202-705-0)98-83-9	R10 Xi; R36/37 N; R51-53	Xi; N R: 10-36/37- S: (2-)61	C ≥ 25 %: \$\forall i5\text{N}; R36/37-5 2,5 % \leq C < 25 %: R52/53	1/53
601-028-0	0 2- 1 methylsty 2- vinyltolue		210-256-7	7611-15-4	Xn; R20 N; R51-53	Xn; N R: 20-51/53 S: (2-)24-61	C ≥ 25%:Xn, N; R20-51/5: 2,5 % ≤ C < 25 %: R52/53	3

601-032-0 0en zo benzo	[a]pyrene [def]chrysene	200-028-3	550-32-8	Carc. Cat. 2; R45 Muta. Cat. 2; R46 Repr. Cat. 2; R60-61 R43 N; R50-53	R:	2,5 % ≤ %CI ≤ 25 %: T, N; R43-45-4 1 % ≤ C ≤ 2,5 %: T; R43-45-4 0,5 % ≤ C < 1 %: T;	6-51-53-60-61 6-52-53-60-61 2-53-60-61
601-037-0 0 -0 hexan			5110-54-3	Repr. Cat. 3; R62 Xn; R65-48/20 Xi; R38 R67 N; R51-53	S: (2-)9-16-2	2 98% 36/3 C < 25 %: Xn;	0-62-51/53 7-61-62 0-62-52/53
601-041-0 di Benz	<i>[a,h]</i> anthrace	n2:00-181-8	353-70-3	Carc. Cat. 2; R45 N; R50-53	T; N R: 45-50/53 S: 53-45-60-	C ≥ 25 %: T, N; R45-50/5: 2,5 % ≤ C1 < 25 %; T, N; R45-51/5:	

						0,25 % ≤ C < 2,5 %: T; R45-52/5 0,01 % ≤ C < 0,25 %: T; R45	3	
601-048-0	00H0 ysene	205-923-4	1218-01-9	Carc. Cat 2; R45 Muta. Cat. 3; R68 N; R50-53	T; N R: 45-68-50/ S: 53-45-60-			
601-052-0	00aphthalene	202-049-5	591-20-3	Carc. Cat R40 Xn; R22 N; R50-53	3xn; N R: 22-40-50/ S: (2-)36/37			
601-053-0	Memylphenol [1] 4- nonylphenol, branched [2]	[1]	025154-52 [1] 584852-15 [2]	Cat.3;	C; N R: 22-34-62- S: (1/2-)26-3		5-46-60-61	
602-003-0	0di8romomethane	200-824-2	274-95-3	Xn; R20 R52-53	Xn R: 20-52/53 S: (2-)24-61	12,5 %	3	
602-008-0	tetrachloride tetrachloromethane	200-262-8	356-23-5	Carc. Cat 3; R40 T; R23/24/2: R52-53 N; R59	R: 23/24/25- 5848/23	1 % ≤ 6/37245-5 %: T, N; R23/24/2: 0,2 % ≤ C < 1 %: Xn, N;	5945245323- 9-61 5-40-48/23- 2-48/20-59	

602-010-0)0 - 5 - dibromoe	E thane	203-444-5	5106-93-4	Carc. Cat. 2; R45 T; R23/24/2; Xi; R36/37/38 N; R51-53	R: 45-23/24/ 5S: 53-45-61	20 % ≤ C < 25 %: T, N; R45-23/2: 2,5 % ≤ C < 20 %: T, N;		
602-011-0	0-1- dichloroe	thane	200-863-5	575-34-3	F; R11 Xn; R22 Xi; R36/37 R52-53	F; Xn R: 11-22-36/ S: (2-)16-23	C ≥ 25 %: Xn; 3₹2523633 20 % ≤ -61< 25 %: Xn; R22-36/3 12,5 % ≤ C < 20 %: Xn; R22		
602-014-0	00-8,2- trichloroe	thane	201-166-9	979-00-5	Carc. Cat. R40 Xn; R20/21/22 R66	R: 20/21/22-	C≥5 %: Xn; 4R)26621/22	2	
602-015-0	0,3,2,2- tetrachlor	oethane	201-197-8	379-34-5	T+; R26/27 N; R51-53	T+; N R: 26/27-51/ S: (1/2-)38-4	R26/27-5	2/53	

					1 % ≤ C < 2,5 %: T; R23/24 0,1 % ≤ C < 1 %: Xn; R20/21
602-016-00-9,2,2- tetrabromoethane	201-191-5	579-27-6	T+; R26 Xi; R36 R52-53	T+ R: 26-36-52/ S: (1/2-)24-2	C ≥ 25 %: T+; 5\$26-36-52/53 20 % ≤ 27-45261 %: T+; R26-36 7 % ≤ C < 20 %: T+; R26 1 % ≤ C < 7 %: T; R23 0,1 % ≤ C < 1 %: Xn; R20
602-017-0pentachloroethane	200-925-3	76-01-7	Carc. Cat. 3; R40 T; R48/23 N; R51-53	R: 40-48/23- S:	C ≥ 25 %: T, N; SR450348/23-51/53 2,5 % ≤ 6037255-61 %: T; R40-48/23-52/53 1 % ≤ C < 2,5 %: T; R40-48/23 0,2 % ≤ C < 1 %: Xn; R48/20
602-019-0 0- 5 bromopropane n-propyl bromide	203-445-0)106-94-5	F; R11 Rep. Cat. 2; R60 Rep. Cat. 3; R63 Xn; R48/20 Xi; R36/37/38 R67	T; F R: 60-11-36/ 8-48/20-6 S: 53-45	

602-025-0	00-8 -	D	200-864-0	75-35-4	F; R12	F+; Xn	C≥	
	dichloroet vinylidend chloride	hylene	200 001 (Carc. Cat R40 Xn; R20	R : 12-20-40 S:	12,5%:	6
602-026-0	dichloroet [1] cis-dichloroet [2] trans-dichloroet [3]	thylene	[1] 205-859-7 [2]	2540-59-0 [1] 7156-59-2 [2] 2156-60-5 [3]	Xn; R20	S:	C ≥ 25 %: Xn; 5320-52/5 12,5 % 29-61< 25 %: Xn; R20	3
602-029-0 X	06- chloropro allyl chloride	D pene	203-457-0	5107-05-1	F; R11 Carc. Cat R40 Muta. Cat.3; R68 Xn; R20/21/22 Xi; R36/37/33 N; R50	11-20/21/ S: (2-)16-25 248/20	22-36/37/3 -26-36/37-	46-61
602-033-0	0 0h1 oroben	zene	203-628-:	5108-90-7	R10 Xn; R20 N; R51-53	Xn; N R: 10-20-51/ S: (2-)24/25/	533% ≤ C < 25 %:	N;R20-51/53
602-034-0	00-7- dichlorob o- dichlorob		202-425-9	95-50-1	Xn; R22 Xi; R36/37/38 N; R50-53	Xn; N R: 822-36/37/ S: (2-)23-60	R22-36/3 -80 % ≤ C < 25 %: Xn, N;	7/38-50/53 7/38-51/53

							2,5 % < C < 5 %:N; R51/53 0,25 % < C < 2,5 %: R52/53		
602-035-0	0 0-2- dichlorob <i>p</i> - dichlorob		203-400-5	106-46-7	Xi; R36 Carc. Cat. 3; R40 N; R50-53	Xn; N R: 36-40-50/ S: (2-)36/37			
602-036-0	00180 oropre (stabilized 2- chlorobut diene	d)	204-818-0	0126-99-8	F; R11 Carc. Cat. 2; R45 Xn; R20/22-4; Xi; R36/37/38	45-11-20/ S: 53-45 8/20	22-36/37/	38-48/20	
602-039-0	∲e 4ychlor PCB	o biphenyls	215-648-	11336-36-2	R33 N; R50-53	Xn; N R: 33-50/53 S: (2-)35-60	R33-50/5	3	
602-043-0	or γ- BHC γ-1,2,3,4,	5,6- ocyclohex	200-401-2 ane	258-89-9	T; R25 Xn; R20/21-44 R64 N; R50-53	S:	10 % 3 ~4 5<60-6 25 %: Xn, N;	1 2-64-50-53	2-64-50-53

							2,5 %	3	
602-062-0 X	0 0 -2,3- trichlorop	D ropane	202-486-3	196-18-4	Carc. Cat. 2; R45 Repr. Cat. 2; R60 Xn; R20/21/22	R: 45-60-20/ S: 53-45	21/22		
602-073-0 X	00,4- dichlorob ene	E ut-2-	212-121-8	3764-41-0	Carc. Cat. 2; R45 T+; R26 T; R24/25 C; R34 N; R50-53	R:	610 % <pre> <pre> <pre> <pre> 610 % <pre> <pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre>	5-26-34-50 2-26-34-51 2-26-36/37 2-23-36/37 2-23-51/53	2/38-51/53 2/38-51/53

						0,25 % <pre></pre>	2/53
603-006-0pentanol isomers, with the exception fo those specified elsewhere in this Annex		250-378-8	30899-19	-R10 Xn; R20 Xi; R37 R66	Xn R: 10-20-37- S: (2-)46	66	
603-007-0 Q-2 methylbur ol tertpentanol	tan-2-	200-908-9	75-85-4	F; R11 Xn; R20 Xi; R37/38	F; Xn R: 11-20-37/ S: (2-)46	38	
603-029-0 b ia(2-chloroeth ether	yl)	203-870-	1111-44-4	R10 Carc. Cat. R40 T+; R26/27/28	10-26/27/ S:	C ≥ 7 %: T+; 28246027/28 1 % ≤ 27-287-36/3 %: T; R23/24/25 0,1 % ≤ C < 1 %: Xn; R20/21/22	7-45 40
603-030-0 Q -8 aminoetha ethanolan		205-483-3	3141-43-5	Xn; R20/21/22 C; R34	20/21/22- S:	C≥25 %: C; 3420/21/22 10 % ≤ 36/37/239-45 %: C; R34 5 % ≤ C < 10 %: Xi; R36/37/38	5
603-031-0 0-2 -	yethane	203-794-9	110-71-4	Repr. Cat.2; R60	F; T R: 60-61-11-	19-20	

603-054-0	ethylene glycol dimethyl ether EGDME		205-575-3	3142-96-1	Repr. Cat.2; R61 F; R11 R19 Xn; R20 R10 Xi; R36/37/38 R52-53	Xi R: 810-36/37/ S: (2-)61	C≥10 %:Xi; 31836235733	3
603-063-0	02,8- epoxypro ol glycidol oxiranem		209-128-3	3556-52-5	Carc. Cat. 2; R45 Muta. Cat. 3; R68 Repr. Cat. 2; R60 T; R23 Xn; R21/22 Xi; R36/37/38	R: 45-60-21/ S: 53-45	22-23-36/3	37/38-68
603-066-0	epoxy-4-			7106-87-6	T; R23/24/2; Xn; R68		C≥1 %: T; 6823/24/25 0,1 % 24-€5<1 %: Xn; R20/21/22	
603-067-0 X	glycidyl ether 2,3- epoxyprophenyl ether 1,2- epoxy-3- phenoxyp		204-557-2	2122-60-1	Carc. Cat. 2; R45 Muta. Cat. 3; R68 Xn; R20 Xi; R37/38 R43 R52-53	R:	38-43-68-:	52/53
603-070-0	0 0- 6 amino-2- methylpro	panol	204-709-8	3124-68-5	Xi; R36/38 R52-53	Xi R: 36/38-52/ S: (2-)61	C≥25 %: Xi; 5836/38-52 10 % ≤ C < 25 %: Xi; R36/38	2/53
603-074-0	06e8ction product:		500-033-5	25068-38	- & i; R36/38	Xi; N	C ≥ 25 %:	

	bisphenol A- (epichlorhepoxy resin (number average molecular weight \le 700)	nydrin)			R43 N; R51-53	R: 36/38-43- S: (2-)28-37	Xi, N; \$\frac{135338-4}{5\%\leq}\$5 \%\leq\$ \%: Xi; R36/38-4 2,5\%\leq\$C < 5\%: Xi; R4	3-52/53	
603-076-0	ObtΩ-2- yne-1,4- diol 2- butyne-1, diol	D 4-	203-788-6	5110-65-6	T; R23/25 Xn;	C; T R: 21-23/25- S: 2(1/2-)25-2	25%≤C 26-36/37/3	21-23/25-3 8/22-43	-43 6/38-48/22-43
603-095-0		y)ethanol	220-548-6	52807-30-9	Xn; R21 Xi; R36	Xn R: 21-36 S: (2-)26-36	/37-46		
603-105-0	O б ыБап	E	203-727-3	3110-00-9	R19	S: 53-45-61	20/22-38-	48/22-68-5	2/53
604-001-0	carbolic acid	roxybenze ohol		7108-95-2	Muta. Cat.3; R68 T; R23/24/2: Xn; R48/20/2	S: 5(1/2-)24/2	$3\% \leq C$	51428226821 6/37/39-45 2-34-68	

604-009-0	മുള്gallo 1,2,3- trihydroxy		201-762-9	987-66-1	Muta. Cat. 3; R68 Xn; R20/21/22 R52-53	Xn R: 20/21/22- S: 2(2-)36/37-	10 % ≤	2-68-52/53
604-010-0	0edorcinol 1,3- benzened		203-585-2	2108-46-3	Xn; R22 Xi; R36/38 N; R50	Xn; N R: 22-36/38- S: (2-)26-61	Xn; R68 C≥ 25 %: 5\mathcal{0}n, N; R22-36/3	
604-012-0	04-2 chloro-o- cresol 4- chloro-2- methyl phenol		216-381-3	31570-64-5	5T; R23 C; R35 N; R50	T; C; N R: 23-35-50 S: (1/2-)26-3	C ≥ 25 %: T, C, N; R23-35-5 66/8 1/39-4 C < 25 %: C; R20-35 5 % ≤ C < 10 %: C; R20-34 3 % ≤ C < 5 %: Xn; R20-36/3 1 % ≤ C < 3 %: Xi; R36/37/3	5-61 7/38
604-013-0	0 2-8 ,4,6- tetrachlor	ophenol	200-402-8	358-90-2	T; R25 Xi; R36/38 N; R50-53	S:	C≥25 %: T, N; 58025336/3 20 % ≤ 2€372455-6	

							%: T, N; R25-51/53 5 % \(\) C < 20 %: T, N; R25-36/38-51/53 2,5 % \(\) C < 5 %: Xn, N; R22-51/53 0,5 % \(\) C < 2,5 %: Xn; R22-52/53 0,25 % \(\) C < 0,5 %: R52/53
604-014-0	00hBorocres 4- chloro-m- cresol 4- chloro-3- methylpho		200-431-6	559-50-7	Xn; R21/22 Xi; R41 R43 N; R50	Xn; N R: 21/22-41- S: (2-)26-36	R21/22-41-43-50
604-015-0	methyleno (3,4,6- trichlorop hexachlor	henol)	200-733-8	370-30-4	T; R24/25 N; R50-53	S:	C ≥ 25 %: T, N; 5\(\frac{8}{2}\)24/25-50/53 2,5 % ≤ 3\(\frac{7}{2}\)452\(\frac{9}{2}\)60-61 %: T, N; R24/25-51/53 2 % ≤ C < 2,5 %: T; R24/25-52/53 0,25 % ≤ C < 2 %: Xn; R21/22-52/53 0,2 % ≤ C < 0,25

							%: Xn; R21/22	
604-017-0 X	0 2 ,4,5- trichlorop	henol	202-467-8	895-95-4	Xn; R22 Xi; R36/38 N; R50-53	Xn; N R: 22-36/38- S: (2-)26-28	R22-36/3	8-51/53 1/53
604-030-0	ObiSphenol A 4,4'- isopropyl	denediphe	201-245-8 nol	880-05-7	Repr. Cat. 3; R62 Xi; R37-41 R43	Xn R: 37-41-43- S: (2-)26-36		
605-002-0	00,9,5- trioxan trioxymet	hylene	203-812-5	5110-88-3	F;R11 Repr.Cat R63 Xi; R37	F; Xn 3R: 11-37-63 S: (2-)36/37	-46	
605-016-0	@IToxal % ethandial. %		203-474-9	107-22-2	Muta. Cat. 3; R68 Xn; R20 Xi; R36/38 R43	Xn R: 20-36/38- S: (2-)36/37	C≥10 %: Xn; 4826836/33 1 % ≤ C < 10 %: Xn; R43-68	8-43-68
605-020-0	0€a∯role 5- allyl-1,3- benzodion	E cole	202-345-4	194-59-7	Carc. Cat 2; R45 Muta. Cat. 3; R68 Xn; R22	T R: 45-22-68 S: 53-45		

605-022-0	Q lutaral		203-856-5	5111-30-8	T;	T; N	C ≥ 50
	glutaralde	hvde			R23/25	R:	%: T, N;
	1,5-	<i>y</i>			C; R34		4R22B 32 5 034-42/43-50
	pentanedi	al			R42/43	S:	25 % ≤
-	pentanean				N; R50		6 037 <i>5</i> 09-45-61
					11, 1650	(1/2)20 3	%: T;
							R22-23-34-42/43
							10 % \leq
							C < 25
							%: C;
							R20/22-34-42/43
							2 % ≤
							C < 10
							%: Xn;
							R20/22-37/38-41-42/4
							1 % ≤
							C < 2
							%: Xn;
							R36/37/38-42/43
							0,5 %
							≤ C < 1
							%: Xi;
							R36/37/38-43
605-025-0	Ab A oroace	taldehvde	203-472-9	2107-20-0	Carc Cat	T+· N	C ≥ 25
003-023-0	OFMOTOACC	arucityuc	203-472-0	5107-20-0	3; R40	R:	%: T
					T+; R26	24/25-26-	I
						S:	
					T;		R24/25-26-34-40-50
					R24/25	(1/2-)20-2	28@6/3 <u>7</u> /39-45-61
					C; R34		C < 25
					N; R50		%: T+;
							R21/22-26-34-40
							7 % ≤
							C < 10
							%: T+;
							R21/22-26-36/37/38-4
							5 % ≤
							C < 7
							%: T;
							R21/22-23-36/37/38-4
							3 % ≤
							C < 5
							%: T;
							R21/22-23-40
							1 % ≤
							C < 3
							%: T;
							R23-40
							0,1 % ≤
			l l				U,1 /U >
							C < 1 %: Xn; R20

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606-037-00-i4dimefon (ISO) 1-(4- chloropheno dimethyl-1		1	Xn; R22 R43 N; R51-53	Xn; N R: 22-43-51/ S: (2-)24-37			
-(1,2,4- triazol-1- yl)butanone							
606-048-02'-4 anilino-3'- methyl-6'- dipentylamin xanthen)-3- one	406-480-7 nospiro(isobenzoft		R53 9'-	R: 53 S: 61			
607-004-00-i7hloroacet	tic 200-927-2	1	C; R35 N; R50-53	C; N R: 35-50/53 S: (1/2-)26-3	C≥25 %: C, N; R35-50/5 10 % ≤ 86/37/289-4 %: C, N; R35-51/5 5 % ≤ C < 10 %: C, N; R34-51/5 2,5 % ≤ C < 5 %: Xi, N; R36/37/3 1 % ≤ C < 2,5 %: Xi; R36/37/3 8-52/53 0,25 % ≤ C < 1 %: R52/53	5-60-61 3	
607-019-0m9thyl chloroforma	201-187-3		F; R11 Γ+; R26 Xn; R21/22 C; R34	F; T+ R: 11-21/22- S: (1/2-)26-1		7-39-36/37/3	39-45-46-6
607-049-0m2coprop (ISO) [1] and its salts 2-(4- chloro- o-	230-386-8 [1] 202-264-4 [2]	193-65-2 I [2] I	Xn; R22 Xi; R38-41 N; R50-53	Xn; N R: 22-38-41- S: (2-) 13-26-37/	R22-38-4	1-50-53	

	tolyloxy) propionic acid (RS)-2- (4- chloro-o- tolyloxy) propionic acid [1] 2-(4- chloro-2- methylpho acid [2]	enoxy)prop	pionic				Xi, N; R38-41-5 10 % ≤ C < 20 %: Xi, N; R41-50-5 5% ≤ C < 10 %: Xi,N;R36 0,25 % ≤ C < 5 %: N; R50-53 0,025 %. ≤ C < 0,25 %: N; R51-53 0,0025 % ≤ C < 0,025 %: Si, N; R51-53 0,0025 %: Si, N; R51-53 0,0025 %: Si, N; R51-53 0,0025 %: Si, N;	3
1	OMCPB (ISO) 4-(4- chloro- o- tolyloxy) butyric acid		202-365-3	894-81-5	N; R50-53	N R: 50/53 S: 60-61		
]	esylic acid prop-2- enoic acid	D	201-177-9	79-10-7		S:	C≥25 %: C, N; 2R20525022 10 % ≤ 66/37/289-4. %: C; R35 5 % ≤ C < 10 %: C; R34 1 % ≤ C < 5 %: Xi; R36/37/38	5-61
607-064-0	0e hzyl chlorofori	mate	207-925-0)501-53-1	C; R34 N; R50-53	C; N R: 34-50/53 S: (1/2-)26-4	10 % ≤	3

							%: C, N; R34-51/53 5 % \le C < 10 %: Xi, N; R36/37/38-51/53 2,5% \le C < 5 %: N; R51/53 0,25 % \le C < 2,5 %: R52/53
607-072-0	0 2- 8 hydroxye acrylate	D thyl	212-454-9	9818-61-1	T; R24 C; R34 R43 N; R50	S:	C≥25 %: T; \$£24-34-43-50 10 % ≤ \$£392\$5-61 %: T; R24-34-43 5 % ≤ C < 10 %: T; R24-36/38-43 2 % ≤ C < 5 %: T; R24-43 0,2 % ≤ C < 2 %: Xn; R21-43
607-086-0) di4 llyl phthalate		205-016-3	3131-17-9	Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)24/25	R22-50/53
607-091-0	Orifluoroa acid %	e B tic	200-929-3	376-05-1	Xn; R20 C; R35 R52-53	S:	C ≥ 25 %: C; 5820-35-52/53 10 % ≤ 5027-228-45-61

							%: C; R20-35 5 % \le C < 10 %: C; R34 1 % \le C < 5 %: Xi; R36/38	
607-094-0	Peracetic acid %		201-186-8	379-21-0	R10 O; R7 Xn; R20/21/22 C; R35 N; R50	2S:	C≥25 %: C, N; IR220/35/30 10 % ≤ IG-36/37/3 %: C; R20/21/22 5 % ≤ C < 10 %: C; R34 1 % ≤ C < 5 %: Xi, R36/37/38	9-45-61 2-35
607-107-0	0 2- 7 ethylhexy acrylate	D l	203-080-7	7103-11-7	Xi; R37/38 R43	Xi R: 37/38-43 S: (2-)36/37-	-46	
607-113-0 X	0sobutyl methacryl	D ate	202-613-0	97-86-9	R10 Xi; R36/37/38 R43 N; R50	Xi; N R: 310-36/37/ S: (2-)24-37	R36/37/38	
607-116-0	03 6 lohexy acrylate	vD	221-319-3	33066-71-5	5Xi; R37/38 N; R51-53	Xi; N R:37/38-5 S: (2-)61		

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						10 %:
or monoaryl or monoalky esters of acrylic acid with the exception of those specified elsewhere in this Annex	rlaryl			Xi; R36/37/38 N; R51-53	Xi; N 8R: 36/37/38- S: (2-)26-28	R36/37/38-51/53
607-151-0propargite (ISO) 2- (4-tert-butylpher cyclohexy prop-2-ynyl sulphite	noxy)	219-006-1	2312-35-8	R40 T; R23 Xi; R38-41 N; R50-53	R: 23-38-40- S:	$C \ge 25$ %: T, N; $442503340-41-50-53$ 20% $36/57/39-45-60-61$ 25% : Xn, N; R20-38-40-41-50-53 10% $\le C < 20\%$: Xn, N; R20-40-41-50-53 $5\% \le C < 10\%$: Xn, N; R20-40-36-50-53 $3\% \le C < 5\%$: Xn, N; R20-40-50-53 $2.5\% \le C < 3\%$: Xn, N; R40-50-53 $1\% \le C < 2.5\%$: Xn, N; R40-51-53 $0.25\% \le C < 1\%$: N;R51-53 $0.025\% \le C < 1\%$: N;R51-53 $0.025\% \le C < 1\%$

							0,25 %: R52-53	
607-189-0	0ri4 nethyle acid	enediamine	eteOraakteOic	91939-36-2	2Xn; R22 Xi; R41 N; R50-53	Xn; N R: 22-41-50/ S: (2-)22-26	353	
607-244-0	0s2octyl acrylate		249-707-8	329590-42	- X i; R36/37/3; N; R50-53	Xi; N 8R: 36/37/38- S: (2-)26-28	R36/37/38	8-50/53
607-245-0	0æ&- butyl acrylate	D	216-768-7	71663-39-4	Xn;	F; Xn R: 211-20/21/ S: (2-)16-25	20 % ≤	233523\$343-52-53
607-247-0	delecyl methacry	ate	205-570-6	5142-90-5	Xi; 36/37/38 N; R50-53	Xi; N R: 36/37/38- S: (2-)26-28	R36/37/38	8-51/53

							2,50 %: R52/53	
607-249-0 X	methyl-1,	l)bis[oxy(r l)]		242978-66	-Xi; R36/37/38 R43 N; R51-53	Xi; N 8R: 36/37/38- S: (2-)24-37	R36/37/38 -60 % ≤ C < 25 %: Xi;	8-43-51/53 8-43-52/53
	00e4 ylonitr			3107-13-1	Carc. Cat 2; R45 T; R23/24/2: Xi; R37/38-4 R43 N; R51-53	45-11-23/ S: 59-16-53-4	20 % \(\leftilde{9} \) \(\left	
608-006-0	040 тохуг (ISO)	nil	216-882-7	71689-84-3	Repr. Cat. 3; R63	T+; N R: 25-26-43-	C ≥ 25 %: T 63-50/53	

	and its salts 3,5-dibromo-4hydroxyb bromoxyr phenol	enzonitrile			T+; R26 T; R25 R43 N; R50-53		7 % ≤ C < 25 %: T+, N; R22-26-4 5 % ≤ C < 7 %: T, N;	0-53 1-53
608-007-0	Obsynil (ISO) and its salts 4- hydroxy-3 diiodoben		216-881-1	1689-83-4	Repr. Cat. 3; R63 T; R23/25 Xn; R21-48/2; Xi; R36 N; R50-53	S: (1/2-)36/3	20 % 52.45.60-6 25 %: Xn, N; R20/22-36 10 % ≤ C < 20 %: Xn, N;	6-48/22-63-50-53 8/22-63-50-53

						3 % ≤ C < 5 %: Xn, N; R20/22-50-53 2,5 % ≤ C < 3 %: N; R50-53 0,25 % ≤ C < 2,5 %: N; R51-53 0,025 % ≤ C < 0,25 %: R52-53
608-010-0m2thac 2- methyl propen nitrile	-2-	204-817-5	5126-98-7	T;	S:	C≥1 %: T; 2\(\mathbb{R}\)2\(\frac{24}{25}\)425-43 0,2 % 5\(\frac{4}{25}\)2\(\frac{29}{29}\)45 %: Xn; R20/21/22-43
608-014-00Morot (ISO) tetrach	halonil loroisophthal		1897-45-6	Carc. Cat 3; R40 T+; R26 Xi; R41 Xi; R37 R43 N; R50-53	R: 26-37-40- S:	$C \ge 20$ %: T 44, $-M3$, -50 /53 R26-37-40-41-43-50-53 /3 0 /3 0 -45-60-61 $\le C <$ 20 %: T+, N; R26-40-41-43-50-53 7 % $\le C$ < 10 %: T+, N; R26-40-36-43-50-53 5 % $\le C < 7$ %: T, N; R23-40-36-43-50-53 2,5 % $\le C < 5$ %: T, N; R23-40-43-50-53 1 % $\le C < 2,5$ %: T, N; R23-40-43-51-53 0,25 % $\le C < 1$:

							Xn, N; R20-51-53 0,1 % \le C < 0,25 %: Xn; R20-52-53 0,025 % \le C < 0,1 %: R52-53
	Obromoxyr octanoate (ISO) 2,6- dibromo cyanophe octanoate	4- nyl		31689-99-2	Cat. 3; R63 T; R23 Xn; R22 R43 N; R50-53		22-23-43-63-50-53 78-46-63-60-61 < 25 %: Xn, N; R20-43-63-50-53 3 % ≤ C < 5 %: Xn, N; R20-43-50-53 2,5 % ≤ C < 3 %: Xi, N; R43-50-53 1 % ≤ C < 2,5 %: Xi, N; R43-51-53 0,25 % ≤ C < 1 %: N; R51-53 0,025 % ≤ C < 0,25 %: R52-53
608-018-0	Obeynil octanoate (ISO) 4- cyano-2,6 diiodophe octanoate	- enyl	223-375-4	13861-47-(ORepr. Cat. 3; R63 T; R25 Xi; R36 R43 N; R50-53	S:	C≥25 %: T, N; 6\(\frac{82503\(\frac{63}{63} \) 43-63-50-53 20 % 8\(\frac{637}{45} \) 45-60-61 25 %: Xn, N; R22-36-43-63-50-53 5 % ≤ C < 20 %: Xn, N; R22-43-63-50-53 3 % ≤ C < 5 %:

							Xn, N; R22-43-5 2,5 % ≤ C < 3 %: N; R43-50-5 1 % ≤ C < 2,5 %: N; R43-51-5 0,25 % ≤ C < 1 %: N; R51-53 0,025 % ≤ C < 0,25 %: R52-53	3
608-021-0	(diaminor	nethylenea hio)propio	mino)thiaz	276823-93 zol-4-	- X n; R22 R43	Xn R: 22-43 S: (2-)22-24	-37	
609-007-0	dinitrotoli dinitrotoli technical grade [1] dinitrotoli [2]	iene,	[1]	0121-14-2 [1] 125321-14 [2]	2; R45	R: 45-23/24/ S: 53-45-61	25-48/22-6	52-68-51/53
609-023-0	Odiócap (ISO)	E	254-408-0)39300-45	Cat. 2; R61 Xn;	T; N R: 61-20-22- S: 253-45-60-	-38-43-48// -61	22-50/53
609-043-0	θ ι fintozen (ISO) pentachlo	e ronitroben	201-435-(zene	082-68-8	R43 N; R50-53	Xi; N R: 43-50/53 S: (2-)13-24	-37-60-61	

609-049-0 2-8 -	Е	210-106-0	0606-20-2				
dinitroto	luene			2; R45 Muta. Cat. 3;R68 Repr. Cat. 3; R62 T; R23/24/2; Xn; R48/22 R52-53	S: 53-45-61	25-48/22-0	52-68-52/53
609-050-0 Q -3-dinitroto	E luene	210-013-:	5602-01-7	Carc. Cat. 2; R45 Muta. Cat. 3; R68 Repr. Cat. 3; R62 T; R23/24/25 Xn; R48/22 N; R50-53	R: 45-23/24/ S: 53-45-60		52-68-50/53
609-051-00-9- dinitroto	E luene	210-222-	1610-39-9	Carc. Cat. 2; R45 Muta. Cat. 3; R68 Repr. Cat. 3; R62 T; R23/24/2: Xn; R48/22 N; R51-53	R: 45-23/24/ S: 53-45-61	25-48/22-0	52-68-51/53
609-052-0 0,3 -dinitroto	E luene	210-566-2	2618-85-9	Carc. Cat. 2; R45 Muta. Cat. 3; R68 Repr. Cat. 3; R62 T; R23/24/2:	R: 45-23/24/ S: 53-45-61	/25-48/22-0	52-68-52/53

					Xn; R48/22 R52-53			
609-055-0	ወ _ን 5 - dinitrotol	E uene	210-581-4	1619-15-8	Carc. Cat. 2; R45 Muta. Cat. 3; R68 Repr. Cat. 3; R62 T; R23/24/2: Xn; R48/22 N; R51-53	R: 45-23/24/ S:53-45-6		52-68-51/53
609-056-0	dibromo-introethar			969094-18	Carc. Cat. 3; R40 Xn; R22-48/2: C; R35 R43 N; R50-53	2-22-35-4 S: 2(1/2-)23-2	10 % ≤ 26-352 5 6/3 %: C, N;	1/53
610-005-0	0-5 chloro-4- nitrobenz	ene	202-809-6	5100-00-5	Carc. Cat. 3; R40 Mut. Cat. 3; R68 T; R23/24/2: Xn; R48/20/2	R: 23/24/25- S: (1/2-)28-3	40-48/20/2 36/37-45-6	21/22-68-51/53

				N;			
				R51-53			
611-001-001-001-001-001-001-001-001-001-	n E	203-102-5	5103-33-3	Carc. Cat 2; R45 Muta. Cat. 3; R68 Xn; R20/22-4 N; R50-53	R: 45-20/22- S: 53-45-60-	48/22-68 -61	50/53
611-060-0048		413-180-4	1	Xi; R41	Xi		
mixture of: sodium 5-[8-[4- [4-[4- [7-(3,5- dicarboxyhydroxy- disulfona ylamino]hydroxy- triazin-2- yl]-2,5- dimethyly yl]-6- hydroxy- triazin-2- ylamino]hydroxy- disulfona ylazo]- isophthal ammoniu 5-[8-[4- [4-[4- [7-(3,5- dicarboxyhydroxy- disulfona - ylamino]hydroxy- triazin-2- ylamino]- hydroxy- disulfona	tonaphthalo 6- 1,3,5- piperazin-1 1,3,5- 1- 3,6- tonaphthalo ate; m	azo)-8- en-1- en-2- en-1		AI, K41	R:41 S: (2-)22-26	-39	

	hydrowy		I	I	I	I	I	I		
	hydroxy- 3,6-									
	disulfonatona	phthal	en-2-							
	ylazo] -									
	isophthalate;									
	5-[8-[4-									
	[4-[4- [7-(3,5-									
	dicarboxylato	nhenv	lazo)-8-							
	hydroxy-3,6-	piicity	uzo) o							
	disulfonationa	phthal	en-1-							
	ylamino]-6-	•								
	hydroxy-1,3,5	5-								
	triazin-2-									
	yl]-2,5-	: 1								
	dimethylpiper yll-6-	azın-ı	_							
	hydroxy-1,3,5	<u>.</u>								
	triazin-2-	,								
	ylamino]-1									
	-									
	hydroxy-3,6-									
	disulfonaphth	alen-2	_							
	ylazo] -									
	isophthalic acid									
(11,062,6	00risodium		412 500 1	3164058-2	When Cat	т				
011-003-0	[4'-(8-		413-390	3104038-2	2; R45	R: 45				
	acetylamino-3	3 6-			2, 1173	S: 53-45				
	disulfonato-2-					2.00				
	naphthylazo)-									
	(6-									
	benzoylamino)-3-								
	sulfonato-2-									
	naphthylazo)- biphenyl-1,3',	2! 1!!!								
	tetraolato-O, C	3,1 - 7'0"(]]''']conneri	(11)						
612 000		,,,,,,,			G G .	T. N.	G > 25			
612-008-	0 0 nIline		200-539-3	362-53-3	Carc. Cat		$C \ge 25$			
					3; R40 Muta.	R:	%: T, N;	 {	548825024/25-50	68
					Cat. 3;	S:	$10\% \le 1000$, 10, 10, 27, 21, 17, 1 2, 1	5 -1002-3 02-1/23-30	-00
					R68			9-45-46-61	-63	
					T;		%: T;			
					R23/24/2	5-48/23/24		2-40-41-43	-48/23/24/25-68	
					Xi; R41		1 % ≤			
					R43		C < 10			
					N; R50		%: T;	10 42 40	1/22/24/25 69	
							0,2 %	k-4U-43-48 	3/23/24/25-68	
							$ \stackrel{0,2}{\leq} \stackrel{70}{C} $			
	I I		I	I	I	I	1 – 0	I		

		•	4	•	•				
							%: Xn; R48/20/2	1/22	
612-009-0	00alts of aniline	A			Carc. Cat. 3;R40 Muta. Cat. 3; R68 T; R23/24/25 Xi; R41 R43 N; R50	R: 23/24/25- S: (1/2-)26-2	10 % \(\leq\$ 276-362357/3' %: T; R20/21/22 1 % \(\leq\$ C < 10 %: T;	9-45-61-63 2-40-41-43- 2-40-43-48/	548825024/25-50-68 -48/23/24/25-68 /23/24/25-68
612-010-0	00180roanii (with exception of those specified elsewhere in this Annex)				R23/24/25 R33 N;	23/24/25- S:	-33-50/53 36/37-45-6	0-61	
612-022-0	0 0- 3 naphthyla	E nmine	202-080-4	191-59-8	/	R: 45-22-51/ S:	C ≥ 25 %: T, N; /\$345-22-5 2,5 % ≤ C < 25 %: T; R45-52/5: 0,01 % ≤ C < 2,5 %: T; R45		
612-023-(chloride [2] phenylhyd hydrochlo [3]	drazinium drazine	[1] 200-444-7 [2] 248-259-0 [3] 257-622-2 [4]	[1]	2; R45 Muta. Cat. 3; - R 68 T;	R: 45-23/24/ S: 53-45-61		43-48/23/24	./25-68-50
612-025-0 X	00itrotoluic with the	lines,			T; R23/24/25	T; N 5			

	exception of those specified elsewhere in this Annex				R33 N; R51-53	R: 23/24/25- S: (1/2-)28-3	33-51/53 36/37-45-6	1	
612-035-0	0 2- 4 methoxya <i>o</i> - anisidine	E niline	201-963-	190-04-0	Carc. Cat. 2; R45 Muta Cat. 3; R68 T; R23/24/2:	R: 45-23/24/ S: 53-45	25-68		
612-042-0	00 Azidine 1,1'- biphenyl- diamine 4,4'- diaminob biphenyl- ylenedian	4,4'- iphenyl 4,4'-	202-199-1	192-87-5	Carc. Cat. 1; R45 Xn; R22 N; R50-53	R:	C≥25 %: T, N; \$\$45-22-5 2,5 % ≤ 60 < 25 %: T, N; R45-51/5: 0,01 % ≤ C < 2,5 %: T; R45		
612-051-0		E iphenylme edianiline		1101-77-9	Carc. Cat. 2; R45 Muta. Cat. 3; R68 T;	R:	24/25-43-4	48/20/21/22	-68-51/53
					R39/23/24 Xn; R48/20/2 R43 N; R51-53				

							%: Xn; R20/21/22-33
612-056-0	dimethyl-toluidine [1] N,N-dimethyl-toluidine [2] N,N-dimethyl-toluidine [3]	m-	[2]	499-97-8 [1] 5121-72-2 [2] 8609-72-3 [3]	T; R23/24/25 R33 R52-53	23/24/25- S:	C ≥ 25 %: T; 3823223325-33-52-53 5 % ≤ 36/37245-61 %: T; R23/24/25-33 1 % ≤ C < 5 %: Xn; R20/21/22-33
612-059-0	diazaocta	nethylened etetramine	iamin	5112-24-3	Xn; R21 C; R34 R43 R52-53	S:	C ≥ 25 %: C; \$225334-43-52/53 10 % ≤ \$637229-45-61 %: C; R34-43 5 % ≤ C < 10 %: Xi; R36/38-43 1 % ≤ C < 5 %: Xi; R43
612-060-0	triazaunde	ecamethyle enepentam	nediamine	2112-57-2	Xn; R21/22 C; R34 R43 N; R51-53	S:	C ≥ 25 %: C, N; 4825122334-43-51/53 10 % ≤ 36/37/239-45-61 %: C; R34-43-52/53 5 % ≤ C < 10 %: Xi; R36/38-43-52/53 2,5 % ≤ C < 5 %: Xi; R43-52/53 1 % ≤ C < 2,5 %: Xi; R43
612-064-0	tetra-	camethyle		94067-16-7	7C; R34 R43	C; N R: 34-43-50/	C ≥ 25 %: C, N; \$34-43-50/53

	pentacthy	lenehexam	line		N; R50-53	S: (1/2-)26-3	10 % ≤ 6/37/239-4 %: C, N; R34-43-5 5 % ≤ C < 10 %: Xi,N; R36/38-4 2,5 % ≤ C < 5 %: Xi, N; R43-51/5 1 % ≤ C < 2,5 %: Xi; R43-52/5 %: Zi; R43-52/5 %: Zi; R43-52/5 %: Zi; R43-52/5 %: Zi %: R52/53	1/53 3-51/53
612-065-0	pelyethly with the exception of those specified elsewhere in this Annex		ines		Xn; R21/22 C; R34 R43 N; R50-53	S:	C≥25 %: C, N; 482502333. 10 % ≤ 6037239-4 %: C, N; R34-43-5 5 % ≤ C < 10 %: Xi,N; R36/38-4 1 % ≤ C < 2,5 %: Xi; R43-52/5 0,25 % ≤ C < 1 %: R52/53	1/53 3-51/53
612-066-0) d i∂yclohe	xylamine	202-980-7	7101-83-7	Xn; R22 C; R34 N; R50-53	S:	C≥25 %: C, N; 5322-34-5 10 % ≤ 6/37/239-4 %: C, N; R34-51/5 2,5 % ≤ C < 10 %: Xi, N; R36/38-5	5-60-61 3

612-067-0	aminomet	hyl-3,5,5- cyclohexyl	32855-13-2	2Xn; R21/22 C; R34 R43 R52-53	S:	2 % \leq C < 2,5 %: Xi; R36/38-5% 0,25 % \leq C < 2 %: R52/53 C \geq 25 %: 43; \text{R2/532} 25 %: 43; \text{R2/532} 25 %: C; R34-43 5 % \leq C < 10 %: Xi; R36/38-4: 1 % \leq C < 5 %: Xi; R43	-34-43-52/53 5-61
612-077-0	Odi nethylr <i>N</i> - nitrosodir	nethylamir	862-75-9	Carc. Cat. 2; R45 T+; R26 T; R25-48/2: N; R51-53	R:	10 % ≤ C < 25 %: T+; R45-22-2 7 % ≤ C < 10 %: T+; R45-22-2 3 % ≤ C < 7 %: T; R45-22-2 2,5 % ≤ C < 3 %: T;	6-48/25-51/53 6-48/25-52/53 6-48/22-52/53 3-48/22-52/53 8/22-52/53

							0,1 %: T; R45	
612-086-0	Odraitraz (ISO) N,N- bis(2,4- xylylimin methylam	omethyl) ine	251-375-4	433089-61	-Kn; R22-48/22 R43 N; R50-53	22-43-48/ S:		2-50-53
612-087-0	Ogt& izatine		236-855-3	313516-27	Xn; R21/22 Xi;	S:	37/38-41-: 28-36/37/3	50/53 9-38-45-46-60-61-63
612-094-0	chloro-4-		402-190-4 noxy-2-	1	T; R48/25 Xn; R22-48/20 Xi; R41 R43 N; R50-53	0S:	48/20-48/2 36/37/39-4	
612-121-0	Omines, polyethyl HEPA	enepoly-	268-626-9	968131-73	-Xn; R21/22 C; R34 R43 N; R50-53	S:	C ≥ 25 %: C, N; 482502233 10 % ≤ 6637239-4 %: C, N; R34-43-5 5 % ≤ C < 10 %:	

							Xi, N; R36/38-4. 2,5 % \(\leq\) C < 5 %: Xi, N; R43-51/5. 1 % \(\leq\) C < 2,5 %: Xi; R43-52/5. 0,25 % \(\leq\) C < 1 %: R52/53	3
612-136-0	isopropyl- phenyl- <i>p</i> - phenylene		202-969-7	7101-72-4	Xn; R22 R43 N; R50-53	Xn; N R: 22-43-50/ S: (2-)24-37	R22-43-5	3
612-151-0	diaminoto technical product - mixture of [2] and [3] methyl-phenylene [1] 4- methyl-m phenylene diamine [2] 2- methyl-m phenylene diamine	ediamine	[1] 202-453-1 [2]	325376-45 [1] 195-80-7 [2] 9823-40-5 [3]	-&arc. Cat. 2; R45 T; R25 Xn; R20/21 Xi; R36 R43 N; R51-53	R:	25-36-43-	51/53

	[3]						
613-009-0	02,4,6- trichloro- triazine cyanuric chloride	1,3,5-	203-614-9	108-77-0	T+; R26 Xn; R22 C; R34 R43 R14	S:	C ≥ 25 %: T+; R122326-34-43 10 % ≤ 2€36237/39-45-46-63 %: T+; R26-34-43 7 % ≤ C < 10 %:T +;R26-36/37/38-43 5 % ≤ C < 7 %: T; R23-36/37/38-43 1 % ≤ C < 5 %: T; R23-43 0,1 % ≤ C < 1 %: Xn; R20
613-011-0	Oarhitrole (ISO) 1,2,4- triazol-3- ylamine		200-521-:	561-82-5	Repr.Cat. R63 Xn; R48/22 N; R51-53	3Xn; N R: 48/22-63- S: (2-)13-36	
613-033-0	0 2- 6 methylazi propylene		200-878-	775-55-8	F; R11 Care. Cat 2; R45 T+; R26/27/2 Xi; R41 N; R51-53		%: T 24/ 28 -41-51/53 R45-26/27/28-41-51/53

613-040-(Oz4 conazo	ole	262-102-3	360207-31	- 0 Xn; R22	Xn	%: T; R45-23/2 0,1 % ≤ C < 1 %: T; R45-20/2 0,01 % ≤ C < 0,1 %: T; R45	
	dioxolan-	henyl)-1,3- 2-)-1 <i>H</i> -1,2,4				R: 22 S: (2-)46		
613-043-(sulphate (ISO) powder 1-[2- (allyloxy) (2,4- dichloropimidazoli hydrogen sulphate [1] (±)-1-[2- (allyloxy) (2,4- dichloropimidazoli hydrogen sulphate [2]	henyl)]-1 <i>H</i> um ethyl-2- henyl)]-1H um	[1] 281-291-3 [2]	58594-72 [1] 383918-57 [2]	R43 -N; R50-53		53	61
613-048-0	00a8bendaz (ISO) methyl benzimida ylcarbama	azol-2-	234-232-(10605-21	-Muta. Cat. 2; R46 Repr. Cat. 2; R60-61 N; R50-53	T; N R: 46-60-61- S: 53-45-60-		
613-049-0	Denomyl (ISO) methyl 1-		241-775-7	717804-35	-Muta. Cat. 2; R46 Repr.Cat. R60-61		2,5 % ≤	5 035338-43-50-5

	(butylcarty) ylcarbama	bamoyl)bei	hzimidazol	-2-	Xi; R37/38 R43 N; R50-53		1 % ≤ C < 2,5 %: T, N;	3
613-051-0	(ISO) S- ethyl 1-)2212-67-	Carc. Cat 3; R40 Repr. Cat 3; R62 Xn; R20/22 Xn; R48/22 R43 N; R50-53	R:	4 6-60- 61 ≤ C < 25 %: Xn, N;	0-43-48/22-62-50-53 8/22-62-50-53 2-50-53

						%: R52-51	
613-058-0	Ø€2methri (ISO) m- phenoxyb 3-(2,2- dichlorov dimethylo	enzyl	952645-53	-Kn; R20/22 R43 N; R50-53	Xn; N R: 20/22-43- S: (2-) 13-24-36/	C ≥ 25 %:	51
613-075-0	dichloro-5 ethyl-5-	5- idazolidine	789415-87	-D; R8 T; R23 C; R34 Xn; R22 R43 N; R50	O; T; N R: 8-22-23-3 S: (1/2-)8-26	4-43-50 5-36/37/39-	45-61
613-088-0		iazol-3(2 <i>H</i> iazolin-3-	92634-33-5	5Xn; R22 Xi; R38-41 R43 N; R50	Xn; N R: 22-38-41- S: (2-)24-26	R22-38-4	

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		,	,			Y		
613-112-0	10- 5		247-761-7	726530-20		T; N	C ≥ 25	
	octyl-2H-				R23/24	R:	%: T, N;	
	isothiazol	-3-			Xn; R22	22-23/24-		334-43-50/53
	one				C; R34	S:	10 % ≤	
					R43		6037239-45	-60-61
					N;	(1/2)20 5	%: C, N;	00 01
					R50-53		R20/21-34	12 51/52
					K30-33			-43-31/33
							5 % ≤	
							C < 10	
							%:Xn,	
							N;	
								/38-43-51/53
							$3\% \leq C$	
							< 5 %:	
							Xn, N;	
							R20/21-43	-51/53
							2,5 % ≤	
							C < 3 %:	
							Xi, N;	
							R43-51/53	
							0,25	
							% ≤ C	
							< 2,5	
							%: Xi;	
							R43-52/53	
							0,05 %	
							≤ C <	
							0,25 %:	
							Xi; R43	
					_		,	
613-124-0) €e£ propin	norph	266-719-9	967564-91		Xn; N		
	cis-4-[3-				Cat. 3;	R:		
	(p-tert-				R63	22-38-63-	51/53	
	butylphen	ıyl)-2-			Xn; R22	S:		
	methylpro				Xi; R38	(2-)36/37	-46-61	
		norpholine			N;			
		r			R51-53			
613-129-0	0118tamitro	n	255-349-3	341394-05		Xn; N		
	4-				N; R50	R: 22-50		
	amino-3-					S: (2-)61		
	methyl-6-							
	phenyl-1,							
	triazin-5-	_, .						
	one							
	OHE							
613-167-0	On Exture			55965-84	-9 ;	T; N	C ≥ 25	
	of: 5-				R23/24/2		%: T, N;	
	chloro-2-				C; R34			334-43-50/53
	methyl-4-				R43	S:	3 % ≤	
	isothiazol				N;		- 6 6/3 7 539-4	5-60-61
		111-3-			R50-53	(2-)20-20		J-00 - 01
	one				K30-33		%: C, N;	24 42 51/52
	[EC no.	77					K20/21/22	-34-43-51/53
	247-500-	/]						

	and 2-						2,5 %	
	methyl-2	1					$\leq C < 3$	
	isothiazol						%: C, N;	
		-3-						1/52
	one						R34-43-5	1/33
	[EC no.	_					$0,6 \% \le$	
	220-239-6	5]					C < 2,5	
	(3:1)						%: Xi;	
	mixture						R34-43-5	2/53
	of: 5-						0,25	_, 00
	chloro-2-						$\% \le C$	
	methyl-4-						< 0,6	
	isothiazol	ın-3-					%: Xi;	
	one						R33/38-43	3-52/53
	[EC no.						0,06	
	247-500-7	7]					% ≤ C	
	and 2-	J					< 0,25	
	methyl-4-						%: Xi;	
	isothiazol	ın-3-					R36/38-4	3
	one						0,0015	
	[EC no.						$\% \le C <$	
	220-239-6	5]					0,06 %:	
	(3:1)	_					Xi; R43	
	` ′						111, 1110	
613-175-0	Œ_t9 oxicon	azole	406-850-2	2133855-9	8€&arc. Cat.	Xn; N		
	(2RS,3SR))-3-			3; R40	R:		
	(2-				Repr.	40-62-63-	51/53	
	chlorophe	nv1)-2-			Cat. 3;	S:		
	(4-	1191) 2			R62	(2-)36/37	16 61	
		1\				(2-)30/3/	-40-01	
	fluorophe	nyı)-			Repr.			
	[(1 <i>H</i> -				Cat. 3;			
	1,2,4-				R63			
	triazol-1				N;			
	_				R51-53			
	yl)methyl	1						
	oxirane	ı						
	Oxirane							
615-001-0)@na∉thyl		210-866-3	624-83-9	F+; R12	F+; T+		
	isocyanate	•			Repr.	R:		
					Cat. 3;		26-37/38-4	11-42/43-63
					R63	S:	20 3 1130	11 12/15 05
							7/20 26/2	7/39-45-63
					T+; R26	(1/2-)20-2	1/28-30/3	1/39-43-03
					T;			
					R24/25			
					R42/43			
					Xi;			
					R37/38-4	1		
					1037730 1			
615-004-0		A			Xn;	Xn		
	thiocyanic	;			R20/21/22	2R:		
	acid				R32	20/21/22-	32-52/53	
					R52-53	S:	,	
					102 33	(2-)13-61		
						(2-)13-01		

615-006-0		202-039-0		Carc. Cat.		C ≥ 25	
	methyl-m-	[1]	[1]	3; R40	R:	%: T+;	
	phenylene	209-544-5	5584-84-9	T+; R26	26-36/37/	318246034623	#338245 0342/43-52/5
	diisocyanate	[2]	[2]	Xi;	S:	20 % ≤	
	toluene-2.4-	247-722-4	126471-62	- R 36/37/38		36/3/2/455-6	1
	di-			R42/43	5(1/2)25	%: T+;	
		[3]	[3]				7/20 40 42/42
	isocyanate			R52-53			7/38-40-42/43
	[1]					7 % ≤	
	4-					C < 20	
	methyl-m-					%: T+;	
	phenylene					R26-40-4	2/43
							2/43
	diisocyanate					1 % ≤	
	toluene-2,6-					C < 7	
	di-					%: T;	
	isocyanate					R23-40-4	2/43
	[2]					0,1% ≤	
						C < 1	
	<i>m</i> -						0.42
	tolylidene					%:Xn;R2	0-42
	diisocyanate						
	toluene-						
	diisocyanate						
	[3]						
	[3]						
515-008-0	08-5	223-861-6	4098-71-9	T R23	T; N	C ≥ 25	2
	isocyanatomethyl-3,	5.5		Xi;	R:	%: T, N;	_
							571/2/502 A 2 / A 2 5 1 / 5 2
	trimethyleyclohexyl						713\$342/43-51/53
	isocyanate			R42/43	S:	20 % ≤	
	isophorone			N;	(1/2-)26-2	28 5-38 24 5-6	1
	di-			R51-53		%: T;	
	isocyanate						7/38-42/43-52/53
	isocyanacc						1/30-72/73-32/33
						2,5 % ≤	
						C < 20	
						%: T;	
						R23-42/4	3-52/53
						2 % ≤	
						C < 2,5	
						%: T;	
						R23-42/4	3
						0,5 %	
						$\leq C < 2$	
						%: Xn;	
						R20-42/4	3
615-015-0	0_3 7_	204-081	5115-31-1	Xn; R22	Xn; N		
,1J-U1J-U	ν ν, θ, / - 		1-16-611				
	trimethylbicyclo(2,2	,1)nept-2-		N;	R:		
	yl			R50-53	22-50/53		
	thiocyanatoacetate				S:		
	isobornyl				(2-)24/25	60-61	
	thiocyanoacetate				(2)2 1/23	00 01	
	unocyanoacetate						
					V NI	C >	
516-015-0	1916chlor	24()-11()-8	R15972-60	- 6 Cate Cat	Xn. N	(<i>/</i>	
616-015-0		240-110-8	315972-60			C ≥	
616-015-0	(ISO)	240-110-8	§15972-60	3; R40	R:	25 %:	
516-015-0		240-110-8	§15972-60			25 %:	

diethyl-N- (methoxymethyl)ac	etanilide	N; R50-53	S: (2-)36/37	R22-40-43- 56-60 -61 1 % ≤ C < 25 %: Xn, N; R40-43-50-53 0,25 % ≤ C < 1 %: N; R50-53 0,025 % ≤ C < 0,25 %: N; R51-53 0,0025 % ≤ C < 0,025 %: C < 0,025 %: R52-53
616-024-0 Q-6 4,4- dimethyl-2,5- dioxooxazolidin-1 -yl)-2- chloro-5- (2-(2,4- di-tert- pentylphenoxy)buty dimethyl-3- oxovaleranilide	402-260-4 vramido)-4,4-	R53	R: 53 S: 61	
dimethylbenzyl hydroperoxide cumene hydroperoxide	201-254-780-15-	9 O; R7 T; R23 Xn; R21/22-4 C; R34 N; R51-53	8\$20/22	C≥25 %: T, N; 3R34/48/20/324-348/30/22-51/53 10 % ≤ 14-36/57/39-45-50-61 %: C; R20-34-48/20/22-52/53 3 % ≤ C < 10 %: Xn; R20-37/38-41-52/53 2,5 % ≤ C < 3 %: Xi; R36/37-52/53 1 % ≤ C < 2,5 %: Xi; R36/37

nap	2,3,4- rahydro ohthyl dropero		212-230-0)771-29-9	O; R7 Xn; R22 C; R34 N; R50-53	S:	C ≥ 25 %: C, N; (R22-34-5 10 % ≤ 14-26-36/3 %: C, N; R34-51/5; 5 % ≤ C < 10 %: Xi, N; R36/37/38 2,5 % ≤ C < 5 %: N;R51/53 0,25 % ≤ C < 2,5 %: R52/53	37/39-45-60-61 3 3-51/53
fra ace free Wa Oil Ree [Th ren afte ren by cry pro of ace fro ace oil coa Co prii of nap and alk	enaphth ction, enaphth e ash l distillate oil maining er moval a vstallizate enaphth em enaphth from al tar. emposed marily ohthale distylnaphr	ene ene tion ene ene thalenes.]			•©arc. Cat 2; R45	R: 45 S: 53-45		
	al	Н	295-506-3	92061-93	- C arc. Cat 2; R45	T R: 45 S: 53-45		

	Wash						
	Oil						
	Redistilla	t-a					
		le					
	[The						
	residue						
	from the						
	fractional						
	distillation	n					
	of wash						
	oil						
	boiling						
	in the						
	approxim	ate					
	range of						
	270°C to						
	330°C						
	(518°F						
	to						
	626°F).						
	1						
	consists						
	predomin	antly					
	of						
	dinuclear						
	aromatic						
	and						
	heterocyc	lic					
	hydrocarb	ons.]					
		Н	292-605-3	390640-84	- © arc. Cat.		
X	oil,				2; R45	R: 45	
	acenaphth	iene				S: 53-45	
	fraction						
	Wash						
	Oil						
	ſΑ						
	complex						
	combinati	on					
	of						
	hydrocarb	ons					
	produced	70113					
	by the						
	distillation						
		11					
	of coal						
	tar and						
	boiling						
	in the						
	range of						
	Tunge of				l l		I
	approxim	ately					
	approxim	ately					
	approximate 240°C to	ately					
	approximate 240°C to 280°C	ately					
	approximate 240°C to	ately					

	to 536°F). Composed primarily of acenaphth naphthale and alkyl naphthale	ene, ne ne.]					
648-099-(offeosote oil [A complex combination of hydrocarb obtained by the distillation of coal tar. It consists primarily of aromatic hydrocarb and may contain appreciab quantities of tar acids and tar bases. It distills at the approximation approximation of the approx	oons oons de	263-047-8	361789-28	•Carc. Cat. 2; R45	T R: 45 S: 53-45	
648-100-(O-@cosote oil, high- boiling distillate Wash Oil	Н	274-565-9	070321-79	•Carc. Cat. 2; R45	T R:45 S: 53-45	

	[The							
	high-							
	hailina							
	boiling							
	distillation	n						
	fraction							
	obtained							
	from							
	the high							
	temperatu	ire						
	carboniza							
		uon						
	of							
	bituminou	IS						
	coal							
	which is							
	further							
	refined							
	to							
	remove							
	excess							
	crystalline	₽						
	salts. It							
	consists							
	primarily							
	of							
	creosote							
	oil with							
	some							
	of the							
	normal							
	polynucle	ar						
	aromatic							
	salts,							
	which							
	are							
	componer	nts						
	of coal							
	tar							
	distillates							
	removed.							
	It is							
	crystal							
	free at							
	approxim	ately						
	5°C							
	$(41^{\circ}F)$.]							
648-101-0	Markosota	Н	232_287 4	8001 58 0	Carc. Cat.	Т		
040-101-0		11	232-201-J	70001-30-3				
	[The				2; R45	R:45		
	distillate					S: 53-45		
	of coal							
	tar							
	produced							
	by the							
	,	ı	ļ		l	I	I	I

	high temperatu carbonizat of bituminou coal. It consists primarily of aromatic hydrocarb tar acids and tar bases.]	oons,					
648-102-0 X	Extract residues (coal), creosote oil acid Wash Oil Extract Residue [A complex combination of hydrocarb from the base-freed fraction from the distillation of coal tar, boiling in the range of approxima 250°C to 280°C (482°F to 536°F). It consists predomina of biphenyl and	n	310-189-2	1122384-7	7C4arc. Cat 2; R45	T R:45 S:53-45	

	isomeric						
	diphenyln	aphthalene	es.]				
(40, 120, (00.6	TT	274.566	170221 00	C C-4	т	
048-138-0	O -cosote	Н	2/4-300-4	+/0321-80	-Carc. Cat.		
	oil, low-				2; R45	R: 45	
	boiling					S: 53-45	
	distillate						
	Wash						
	Oil						
	[The						
	low-						
	boiling						
	distillation	n					
	fraction						
	obtained						
	from						
	the high						
	temperatu	ire					
	carboniza	tion					
	of						
	bituminou	ıs					
	coal,						
	which is						
	further						
	refined						
	to						
	remove						
	excess						
	crystalline	2					
	salts. It						
	consists						
	primarily						
	of						
	creosote						
	oil with						
	some						
	of the						
	normal						
	polynucle	ar					
	aromatic						
	salts,						
	which						
	are						
	componer	nts					
	of coal						
	tar						
	distillate,						
	removed.						
	It is						
	crystal						
	free at						
	approxim	ately					

	38°C						
	(100°F).]						
649-001-0	Estracts (petroleur light naphtheni distillate solvent		265-102-1	64742-03	- C arc. Cat. 2; R45	T R:45 S: 53-45	
649-002-0	Destracts (petroleur heavy paraffinic distillate solvent	H n)	265-103-7	764742-04	- C arc. Cat. 2; R45	T R: 45 S: 53-45	
649-003-0	Dextracts (petroleur light paraffinic distillate solvent	H n),	265-104-2	264742-05	- € arc. Cat. 2; R45	T R:45 S: 53-45	
649-004-0 X	Extracts (petroleur heavy naphtheni distillate solvent		265-111-0	64742-11	• C arc. Cat. 2; R45	T R: 45 S: 53-45	
649-005-0	Estracts (petroleur light vacuum gas oil solvent	Н n),	295-341-7	791995-78	- C arc. Cat. 2; R45	T R: 45 S: 53-45	
649-006-0	Daydi rocarb C ₂₆₋₅₅ , arom- rich	o dh s	307-753-7	797722-04	- C arc. Cat. 2; R45	T R: 45 S: 53-45	
649-062-0	(petroleur catalytic cracked naphtha depropani overhead, C ₃ -rich acid-free Petroleum gas	zer	270-755-0	068477-73	-6arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	[A complex combination of hydrocarbon obtained from fractionation of catalytic cracked hydrocarbon and treated to remove acidic impurities. It consists of hydrocarbon numbers in the range of C ₂	oons oons					
	through C ₄ ,						
	predomin C ₃ .]	antly					
649-063-0) G -ases	НК	270-756-6	668477-74	- © arc. Cat.	T	
	(petroleur catalytic cracker Petroleum gas [A complex combinati of hydrocarb produced	on oons			1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
	by the distillation of the products from a catalytic cracking process.						

It consists predomin of aliphatic hydrocarb having carbon numbers predomin in the range of C ₁ through C ₆ .]	oons	270 757	169477 75	- € arc. Cat.	Т	
(petroleur catalytic cracker, C ₁₋₅ -rich Petroleum gas [A complex combination of hydrocart produced by the distillation of products from a catalytic cracking rocess. It consists of aliphatic hydrocart having carbon numbers in the range of C ₁ through C ₆ , predomin C ₁	n), ion oons oons	2/0-/3/-	1004/7-73	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	

	through C ₅ .]						
649-065-0		on oons ion on æed	270-758-7	768477-76	-£arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
(40,000	C ₄ .]	11.17	270 760 6	0.60.477.70	&	T	
649-066-0	(petroleur catalytic reformer, C ₁₋₄ -rich Petroleum gas [A complex	ŕ		000477-79	-£arc. Cat. 1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	

1	aomhiat	on	 	ĺ			l	l
	combinati of	on						
	hydrocarb	ong						
	produced	0115						
	by							
	distillation	1						
	of	1						
	products							
	from a							
	catalytic							
	reforming							
	process.	•						
	It							
	consists							
	of							
	hydrocarb	ons						
	having	70115						
	carbon							
	numbers							
	in the							
	range							
	of C ₁							
	through							
	C_6 ,							
	predomina	antly						
	C_1	arrery						
	through							
	through C ₄ .1							
(40,067,6	C ₄ .]	11.17	270 765		6 C 4	T		
649-067-0	C ₄ .]	H K	270-765-5	568477-83	- C arc. Cat.			
649-067-0	C ₄ .] OG-3ses (petroleun		270-765-5	568477-83	1; R45	R: 45-46		
649-067-0	C ₄ .] OG-2ses (petroleun C ₃₋₅		270-765-5	568477-83	1; R45 Muta.			
649-067-0	C ₄ .] OGases (petroleun C ₃₋₅ olefinic-	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C ases (petroleun C ₃₋₅ olefinic- paraffinic	n),	270-765-5	568477-83	1; R45 Muta.	R: 45-46		
649-067-0	C ₄ .] ©:3 ses (petroleum C ₃₋₅ olefinic- paraffinic alkylation	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] ©:3 ses (petroleun C ₃₋₅ olefinic- paraffinic alkylation feed	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C ases (petroleun C ₃₋₅ olefinic- paraffinic alkylation feed Petroleum	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C ases (petroleun C ₃₋₅ olefinic- paraffinic alkylation feed Petroleum gas	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C ases (petroleum C ₃₋₅ olefinic- paraffinic alkylation feed Petroleum gas [A	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C 3-5 olefinic- paraffinic alkylation feed Petroleum gas [A complex	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C 3-5 olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] OGâses (petroleum C ₃₋₅ olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C 3-5 olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] OGases (petroleum C ₃₋₅ olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C 3-5 olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and paraffinic	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C 3-5 olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and paraffinic hydrocarb	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C 3-5 olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and paraffinic hydrocarb having	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] Châses (petroleum C ₃₋₅ olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and paraffinic hydrocarb having carbon	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] C 3-5 olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and paraffinic hydrocarb having	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] OGases (petroleum C ₃₋₅ olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and paraffinic hydrocarb having carbon numbers in the	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] Cases (petroleum C ₃₋₅ olefinic-paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and paraffinic hydrocarb having carbon numbers in the range	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		
649-067-0	C ₄ .] OGases (petroleum C ₃₋₅ olefinic- paraffinic alkylation feed Petroleum gas [A complex combinati of olefinic and paraffinic hydrocarb having carbon numbers in the	n),	270-765-5	568477-83	1; R45 Muta. Cat. 2;	R: 45-46		

	C ₅ which are used as alkylation feed. Ambient temperatu normally exceed the critical temperatu of these combinations.	res					
649-068-0	(petroleur C ₄ -rich Petroleum gas [A complex combination of hydrocarb produced by distillation of products from a catalytic fractionat process. It consists of aliphatic hydrocarb having carbon numbers in the range of C ₃ through C ₅ , predomina C ₄ .]	ion oons oons antly		668477-85	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
649-069-0	64 ses (petroleur	Н К n),	270-768-1	168477-86	- C arc. Cat. 1; R45	T R: 45-46	

	deethaniz overheads Petroleum gas [A complex combination of hydrocarts produced from distillation of the gas and gasoline fractions from the catalytic cracking process. It contains predomine ethane	on oons			Muta. Cat. 2; R46	S: 53-45	
	and ethylene.]						
			•======================================				
649-070-0 X	OGases (petroleur deisobutatower overheads Petroleur gas [A complex combination of hydrocarb produced by the atmosphedistillation of a butane-butylene stream. It consists of aliphatic hydrocarb having	nizer on ons ric n	270-769-7	768477-87	-2 arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	carbon							
	numbers	_						
	predomin	antly						
	in the							
	range							
	of C ₃							
	through							
	C_{4} .							
649-071-0		ΗK	270-772-3	368477-90	- © arc. Cat.			
	(petroleur	n),			1; R45	R: 45-46		
	depropani	zer			Muta.	S: 53-45		
	dry,	-			Cat. 2;			
					R46			
	propene-				N40			
	rich							
	Petroleum	1						
	gas							
	[A							
	complex							
	combinati	on						
	of							
	hydrocarb	ons						
	produced							
	by the							
		_						
	distillation	1						
	of							
	products							
	from the							
	gas and							
	gasoline							
	fractions							
	of a							
	catalytic							
	cracking							
	process.							
	It .							
	consists							
	predomin	antly						
	of	-						
	propylene							
	with							
	some							
	ethane							
	and							
	propane.]							
649-072-0		ΗK	270-773-9	968477-91	- C arc. Cat.			
	(petroleur	n),			1; R45	R: 45-46		
	depropani				Muta.	S: 53-45		
	overheads					D. 55- 1 5		
					Cat. 2;			
	Petroleum	1			R46			
	gas							
	[A							
	complex							
	Julyion						l	

	combinati	on						
	of							
	hydrocarb	ons						
	produced	0115						
	by							
	distillation							
		11						
	of							
	products							
	from the							
	gas and							
	gasoline							
	fractions							
	of a							
	catalytic							
	cracking							
	process.							
	Ît							
	consists							
	of							
	aliphatic							
		ong						
	hydrocarb	JOHS						
	having							
	carbon							
	numbers							
	predomin	antly						
	in the							
	range							
	of C ₂							
	through							
	$C_{4\cdot}$							
649-073-0		ΗK	270-777-0	068477-94				
	(petroleur	n),			1; R45	R: 45-46		
	gas				Muta.	S: 53-45		
	recovery				Cat. 2;			
	plant				R46			
	depropani	zer						
	overheads	5						
	Petroleum							
	gas							
	[A							
	complex							
	combinati	on						
	of	1011						
	hydrocarb	JOHS						
	obtained							
	by							
	fractionat	ion						
	of							
	miscellan							
	hydrocarb	on						
	streams.							
	It							
		1			1	•	1	1

consists predomin of hydrocart having carbon numbers in the range of C ₁ through C ₄ , predomin propane.]	oons	270 778	569477 05	Para Cat	Т	
(petroleun Girbatol unit feed Petroleun gas [A complex combinat of hydrocarl that is used as the feed into the Girbatol unit to remove hydrogen sulfide. It consists of aliphatic hydrocarl having carbon numbers predomin in the range of C2 through C4.]	ion pons antly		568477-95	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
649-075-0 © ases (petroleur isomerize	HK n), d	270-782-8	868477-99	- C arc. Cat. 1; R45	T R: 45-46 S: 53-45	

	naphtha	I		Muta.		
	fractionator,			Cat. 2;		
	C ₄ -rich,			R46		
				K+0		
	hydrogen					
	sulfide-					
	free					
	Petroleum					
	gas					
649-076-0	OGall gas HK	270-802-4	68478-21	Tarc Cat	Т	
047-070-0	(petroleum),	270-002-	00476-21	1; R45	R: 45-46	
	catalytic			Muta.	S: 53-45	
	cracked				3. 33 -4 3	
				Cat. 2;		
	clarified			R46		
	oil and					
	thermal					
	cracked					
	vacuum					
	residue					
	fractionation					
	reflux					
	drum					
	Petroleum					
	gas					
	[A					
	complex					
	combination					
	of					
	hydrocarbons					
	obtained					
	from					
	fractionation					
	of					
	catalytic					
	cracked					
	clarified					
	oil and					
	thermal					
	cracked					
	vacuum					
	residue.					
	It					
	consists					
	predominantly					
	of					
	hydrocarbons					
	having					
	carbon					
	numbers					
	nradominantly					
	predominantly in the					
	in the					
	range					
	of C ₁					

	through						
	C_{6} .						
649-077-0	C ₆ .] Otal gas (petroleur catalytic cracked naphtha stabilizati absorber Petroleum gas [A complex combinati of hydrocarb obtained from the stabilizati of catalytic cracked naphtha. It consists predoming carbon numbers predoming in the	on on ons on antly ons	270-803-0	068478-22	-&arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
	range of C_1 through C_6 .]						
649-078-0	(petroleur catalytic cracker, catalytic reformer and hydrodest combined fractionat Petroleum gas [A complex	ılfurizer er	270-804-6	668478-24	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	combination						
	of						
	hydrocarbons						
	obtained						
	from the						
	fractionation						
	of						
	products						
	from						
	catalytic						
	cracking,						
	catalytic						
	reforming						
	and						
	hydrodesulfurizing						
	processes						
	treated						
	to						
	remove						
	acidic						
	impurities.						
	It						
	consists						
	predominantly						
	of						
	hydrocarbons						
	having						
	cabon						
	numbers						
	predominantly						
	in the						
	range						
	of C ₁						
	through						
	C ₅ .]						
649-079-0	OG H K	270-806-	768478-26	-Carc. Cat.	T		
	(petroleum),			1; R45	R: 45-46		
	catalytic			Muta.	S: 53-45		
	reformed			Cat. 2;			
	naphtha			R46			
	fractionation						
	stabilizer						
	Petroleum						
	gas						
	[A						
	complex						
	combination						
	of						
	hydrocarbons						
	obtained						
	from the						
	fractionation						
	1	1	1	1	1	1	1

	stabilization						
	of						
	catalytic						
	reformed						
	naphtha.						
	It						
	consists						
	predominantly						
	of						
	hydrocarbons						
	having						
	carbon						
	numbers						
	predominantly						
	in the						
	range						
	of C ₁						
	through						
	C_{4} .						
649-080-0	OGAI gas HK	270-813-3	68478-32				
	(petroleum),			1; R45	R: 45-46		
	saturate			Muta.	S: 53-45		
	gas plant			Cat. 2;			
	mixed			R46			
	stream,						
	C ₄ -rich						
	Petroleum						
	gas						
	[A						
	complex						
	combination						
	of						
	hydrocarbons						
	obtained						
	from the						
	fractionation						
	stabilization						
	of						
	straight-						
	run						
	naphtha,						
	distillation						
	tail gas						
	and						
	catalytic						
	reformed						
	naphtha						
	stabilizer						
	tail						
	gas. It						
	consists						
	of						

	hydrocarb having carbon numbers in the range of C ₃ through C ₆ , predomin butane and isobutane	antly					
649-081-0 X	Ofail gas (petroleur saturate gas recovery plant, C ₁₋₂ -rich Petroleum gas [A complex combinate of hydrocart obtained from fractionat of distillate tail gas, straight-run naphtha, catalytic reformed naphtha stabilizer tail gas. It consists predomin of hydrocart having carbon numbers in the range of C ₁	ion oons ion	270-814-0	068478-33	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	through C ₅ , predoming methane and ethane.]	antly					
649-082-((petroleur vacuum residues thermal cracker Petroleum gas [A complex combination of hydrocarbobtained from the thermal cracking of vacuum residues. It consists of hydrocarb having carbon numbers predominin the range of C1 through C5.]	on oons oons		668478-34	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
649-083-0	OH9drocard C ₃₋₄ - rich, petroleum distillate Petroleum gas [A complex combinati of hydrocard	on	270-990-9	968512-91	•Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

produced by distillation and condensate of crude oil. It consists of hydrocard having carbon numbers in the range	n tion					
of C ₃ through						
C ₅ , predomin	antly					
through C ₄ .]						
649-084-0 ©-6 ses	НК	271-000-8	368513-15	-Carc. Cat.	T	
(petroleur full-range straight-run naphtha dehexanis off petroleur gas [A complex combinat of hydrocarl obtained by the fractionat of the full-range straight-run naphtha. It consists	m), zer n ion oons	2/1-000-8	368513-15	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
of hydrocar having	ons					

carbon number predor in the range of C ₂ throug C ₆ .]	ers ninantly					
deproposition of hydrocoproduct from a hydrocoproces It consisting predomotion of hydrocoproces are consisting predomotion of hydrocoproces are consisting predomotion of hydrocoproces are consisting predomotion of hydrocoproces having carbon number of hydrocoproces are consisting predomotion of hydrocoproces having carbon number of hydrocoproces having hydrocoproces having hydrocoproces having hydrocoproces have hydrocoproces have hydrocoproces have hydrocoproces have hydrocoproces have hydrocoproces have hydrocoproces hydrocoproces hydrocoproces have hydrocoproces hydroc	eracking carbon- eum ex nation carbon ced ation cts cracking s. ts minantly carbons g. h ers minantly carbons g. h ers minantly	271-001-3	868513-16	-6arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	hydrogen sulfide.]						
649-086-0		on oons oon	271-002-9	968513-17-		T R: 45-46 S: 53-45	
	Residues (petroleur alkylation splitter, C ₄ -rich Petroleum gas [A complex residuum from the distillation	1	271-010-2	268513-66	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	of							
	streams							
	various							
	refinery							
	operation	s S						
	It							
	consists							
	of							
	hydrocarb	ons						
	having							
	carbon							
	numbers							
	in the							
	range							
	of C ₄							
	through							
	C_5 ,							
	predomin	antly						
	butane							
	and							
	boiling							
	in the							
	range of							
	approxim	atery						
	-11.7°C							
	to							
	27.8°C							
	(11°F to							
	82°F).]							
649-088-0	01 8drocar	bblnK.	271-032-2	268514-31	-Carc. Cat	Т		
	C_{1-4}	,			1; R45	R: 45-46		
	Petroleum	1			Muta.	S: 53-45		
		Ī			Cat. 2;	0.00 10		
	gas				R46			
	[A				K+0			
	complex							
	combinati of	lon						
	hydrocarb							
	provided							
	by							
	thermal							
	cracking							
	and							
	absorber							
	operations	\$						
	and by							
	distillation	n						
	of crude							
	oil. It							
	consists							
	of							
	1	I				I	I	I

hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₄ and boiling in the range of approximately minus 164°C to minus 0.5°C (-263°F to 31°F).]					
649-089- 001 3drocarbHnK,	2/1-038-5	68514-36	- & arc. Cat. 1; R45 Muta.	R: 45-46 S: 53-45	
sweetened Petroleum			Cat. 2;	3. 33 -4 3	
gas [A			R46		
complex combination					
of					
hydrocarbons obtained					
by					
subjecting hydrocarbon					
gases					
to a					
sweetening					
process to					
convert					
mercaptans					
or to remove					
acidic					
impurities.					
It					
consists of					
hydrocarbons					
having					

carbon numbers predominantly in the range of C ₁ through C ₄ and boiling in the range of approximately -164°C to -0.5°C (-263°F to					
31°F).]					
	271 250 /	768527-16	6 5 C 4		
$\begin{array}{c} \text{649-090-0019drocarbHnK}, \\ C_{1\text{-}3} \\ \text{Petroleum} \\ \text{gas} \\ \text{[A]} \\ \text{complex} \\ \text{combination} \\ \text{of} \\ \text{hydrocarbons} \\ \text{having} \\ \text{carbon} \\ \text{numbers} \\ \text{predominantly} \\ \text{in the} \\ \text{range} \\ \text{of } C_1 \\ \text{through} \\ C_3 \text{ and} \\ \text{boiling} \\ \text{in the} \\ \text{range of} \\ \text{approximately} \\ \text{minus} \\ 164^{\circ}\text{C} \\ \text{to minus} \\ 42^{\circ}\text{C} \\ \text{(-263°F)} \\ \text{to} \\ \end{array}$	2/1-23/-	700327-10	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
-44°F).]	271 261 9	60507 10	Coro Cat	т	
649-091-0 014 drocarb HnK , C ₁₋₄	2/1-261-8	868527-19	-carc. Cat. 1: R45	R: 45-46 S: 53-45	

	debutaniz fraction Petroleum gas				Muta. Cat. 2; R46		
640.000	_	** **	251 (24)		5 0 .		
649-092-0 X	OGases (petroleur C ₁₋₅ , wet Petroleum gas [A complex combinati of hydrocarb produced by the distillation of crude oil and/ or the cracking of tower gas oil. It consists of hydrocarb having carbon numbers predominati in the range	on oons n	271-624-0	068602-83	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
	of C ₁						
	through C ₅ .]						
649-093-0	OH\$drocarl C ₂₋₄ Petroleum gas		271-734-9	968606-25	-Varc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
640 004 0	Modraga	ablak	271 725	168606-26	Cara Cat	Т	
649-094-0	0049drocarl C ₃ Petroleum gas	·	2/1-/35-2	1 080U0-20	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
649-095-0	Offises (petroleur alkylation feed		271-737-5	68606-27	• £ arc. Cat. 1; R45	T R: 45-46 S: 53-45	

Petroleum gas [A complex combinati			Muta. Cat. 2; R46		
of hydrocarb produced by the catalytic cracking of gas	oons				
oil. It consists of hydrocarb having carbon	oons				
numbers predoming in the range of C ₃ through	antly				
C ₄ .] 649-096- © Gases	H K 271-74	12-268606-34	- K arc Cat	Т	
(petroleur depropani bottoms fractionati off Petroleum gas [A complex combinati of hydrocarb obtained from the fractionati of depropani bottoms. It consists predominati of butane, isobutane and butadiene	n), zer ion on oons ion zer antly	+2-400000-34	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	

649-097-0	petroleum refinery blend Petroleum gas [A complex combination obtained from various processes. It consists of hydrogen, hydrogen sulfide and hydrocarb having carbon numbers predomina in the range of C ₁ through C ₅ .]	on			Muta. Cat. 2; R46	R: 45-46 S: 53-45	
649-098-0	(petroleur catalytic cracking Petroleum gas [A complex combination of hydrocarb produced by the distillation of the products from a catalytic cracking process. It consists	on oons	272-203-4	168783-64	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

predominan of hydrocarbor having carbon numbers predominan in the range of C ₃ through C ₅ .]	ntly			
(petroleum) C2-4, sweetened Petroleum gas [A complex combination of hydrocarbor obtained by subjecting a petroleum distillate to a sweetening process to convert mercaptans or to remove acidic impurities. It consists predominan of saturated and unsaturated hydrocarbor having carbon numbers predominan in the	n ns	568783-65-&arc 1; R4 Muta Cat. R46	R: 45-46 S: 53-45	

range of C ₂ through C ₄ and boiling in the range of approxim -51°C to -34°C (-60°F to -30°F).]						
(petroleun crude oil fractionat off Petroleun gas [A complex combinat of hydrocarl produced by the fractionat of crude oil. It consists of saturated aliphatic hydrocarl having carbon numbers predomin in the range of C ₁ through C ₅ .]	ion ion oons ion	272-871-7	768918-99	- 0 'arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-101-0 Gases (petroleum dehexaniz off Petroleum gas [A complex	zer	272-872-2	268919-00	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	1 1 .4	ı	i	ı	ı		I	I
	combinati	on						
	of							
	hydrocarb	ons						
	obtained							
	by the							
	fractionati	ion						
	of							
	combined							
	naphtha							
	streams.							
	It							
	consists of							
	saturated							
	aliphatic	.010.0						
	hydrocarb	OOIIS						
	having carbon							
	numbers							
	predomina	antly						
	in the	uiitiy						
	range							
	of C ₁							
	through							
	C ₅ .]							
	C ₃ .]							
649-102-0		НК	272-878-5	668919-05				
649-102-0	(petroleun		272-878-5	568919-05	1; R45	R: 45-46		
649-102-0	(petroleun light		272-878-5	568919-05	1; R45 Muta.			
649-102-0	(petroleun		272-878-5	568919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run		272-878-5	568919-05	1; R45 Muta.	R: 45-46		
649-102-0	(petroleun light straight run gasoline	m),	272-878-5	568919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati	m),	272-878-5	568919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer	m),	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off	n), ion	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum	n), ion	272-878-5	568919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas	n), ion	272-878-5	568919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A	n), ion	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex	n), ion	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati	n), ion	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of	n), ion on	272-878-5	568919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb	n), ion on	272-878-5	568919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained	n), ion on	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained by the	ion on oons	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained by the fractionati	ion on oons	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained by the fractionati of light	ion on oons	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained by the fractionati of light straight-	ion on oons	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained by the fractionati of light straight-run	ion on oons	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained by the fractionati of light straight-run gasoline.	ion on oons	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained by the fractionati of light straight-run gasoline. It	ion on oons	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		
649-102-0	(petroleun light straight run gasoline fractionati stabilizer off Petroleum gas [A complex combinati of hydrocarb obtained by the fractionati of light straight-run gasoline.	ion on oons	272-878-5	668919-05	1; R45 Muta. Cat. 2;	R: 45-46		

saturated aliphatic hydrocarb having carbon numbers predomin in the range of C_1 through C_5 .]						
(petroleur naphtha unifiner desulfuriz stripper off Petroleum gas [A complex combination of hydrocarb produced by a naphtha unifiner desulfuriz process and stripped from the naphtha product. It consists of saturated aliphatic hydrocarb having carbon numbers predomination the range of C1 through C4.]	ation ons cation	272-879-0	068919-06	-£arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

649-104-0 0		НК	272-882-7	768919-09	Carc. Cat.	T	
(p	etroleun	n),			1; R45	R: 45-46	
	raight-				Muta.	S: 53-45	
ru					Cat. 2;		
	aphtha				R46		
	talytic						
	forming						
of	T etroleum						
I							
ga [A							
	mplex						
	mbinati	on					
of							
	drocarb	ons					
	otained						
by	the !						
	ıtalytic						
	forming						
of							
	raight-						
ru	n aphtha						
an							
	actionati	on					
	the						
	tal						
	fluent.						
It							
	onsists						
of							
	ethane,						
	hane,						
an							
pr	opane.]						
649-105- 00 -		ΗK	272-893-7	768919-20			
	etroleun	1),			1; R45	R: 45-46	
	uidized				Muta.	S: 53-45	
	italytic				Cat. 2;		
	acker				R46		
sp	olitter verheads						
	etroleum						
ga	I	•					
	~						
	mplex						
	mbinati	on					
of							
	drocarb	ons					
	oduced						
	the						
fra	actionati	on					

of the charge to the C_3 - C_4 splitter.	
It	
consists predominantly of C ₃ hydrocarbons.]	
649-106-0 G-4ses H K 272-883-268919-10-Carc. Cat T	
(petroleum), straight- run stabilizer off Petroleum gas [A complex combination of hydrocarbons obtained from the liquid from the first tower used in the distillation of crude oil. It consists of saturated aliphatic hydrocarbons having carbon numbers predominantly in the	
$ \begin{array}{c c} range \\ of \ C_1 \\ through \\ C_4. \end{array} $	

649-107-0) Gases	НК	273-169-3	868952-76	Carc. Cat.	T	
X	(petroleur				1;R45	R: 45-46	
	catalytic	()			Muta.	S: 53-45	
	cracked				Cat. 2;		
	naphtha				R46		
	debutaniz	er					
	Petroleum	1					
	gas						
	[A						
	complex						
	combinati	on					
	of						
	hydrocarb	ons					
	obtained						
	from						
	fractionat	ion					
	of						
	catalytic						
	cracked						
	naphtha.						
	It						
	consists						
	of						
	hydrocarb	ons					
	having						
	carbon						
	numbers	41					
	predomin in the	antiy					
	range of C ₁						
	through						
	C ₄ .]						
649-108-0	Ofafil gas	ΗK	273-170-9	68952-77	•Carc. Cat.	T	
	(petroleur	n),			1; R45	R: 45-46	
	catalytic				Muta.	S: 53-45	
	cracked				Cat. 2;		
	distillate				R46		
	and						
	naphtha						
	stabilizer						
	Petroleum	1					
	gas						
	[A						
	complex						
	combinati	on					
	of						
	hydrocarb	ons					
	obtained						
	by the	ion					
	fractionat	1011					
	of						

649-109-0	catalytic cracked naphtha and distillate. It consists predoming of hydrocarb having carbon numbers predoming in the range of C ₁ through C ₄ .]	oons	272 175	668952-81			
	(petroleur thermal- cracked distillate, gas oil				1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
	and naphtha absorber petroleum gas	ı					
	[A complex combination of						
	hydrocarb obtained from the separation of						
	thermal- cracked distillates naphtha	,					
	and gas oil. It consists pedromina of	antly					
	hydrocarb having carbon numbers	oons					

predomin in the range of C_1 through C_6 .	antly					
649-110-0 (Tail gas (petroleum thermal cracked hydrocart fractionat stabilizer, petroleum coking Petroleum gas [A complex combination of hydrocart obtained from the fractionat stabilization of thermal cracked hydrocart from petroleum coking process. It consists of hydrocart having carbon numbers predomin in the range of C1 through C6-]	oon ion ion oons ion oons	273-176-1	68952-82	-varc. Cat. 1: R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-111-0@ases (petroleur light steam-	HK n,	273-265-5	568955-28	- £ arc. Cat. 1; R45	T R: 45-46 S: 53-45	

cracked, butadiene conc. Petroleum gas [A complex combination of hydrocarb produced by the distillation of products from a thermal cracking process, It consists of hydrocarb having a carbon number predomin of C ₄]	on oons			Muta. Cat. 2; R46		
649-112-0	on oons	273-270-2	268955-34	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

naphtha and the fractionat of the total effluent. It consists of saturated aliphatic hydrocart having carbon numbers predomin	ons					
in the range of C ₂ through C ₄ .]	Ž					
649-113-0013 drocar C ₄ Petroleum gas		289-339-	587741-01	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-114-0 % R kanes, C ₁₋₄ , C ₃ - rich Petroleum gas		292-456-4	190622-55	Parc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-115-0 Gases (petroleur steam-cracker C3-rich Petroleum gas [A complex combination of hydrocard produced by the distillation of products from a steam cracking	ion oons	295-404-9	992045-22	Parc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

process. It consists predominantly of propylene with some propane and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0019drocarbitnK, C4. steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4., predominantly 1-butene and 2-								
consists predominantly of propylene with some propane and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-00fydrocarbinK, C4, steam- cracker distillate Petroleum gas IA complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-buttene								
predominantly of propylene with some propane and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0049drocarbthK, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene		Īt						
of propylene with some propane and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0019drocarbtinK, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-buttene								
propylene with some propane and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0019drocarbtink. C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene			antly					
with some propane and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0019drocarbtink, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
some propane and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-00i9drocarbinK, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-buttene								
propane and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0049drocarbbhK, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
and boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0049drocarbbnK, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
boils in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0049drocarbbhK, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
in the range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0049drocarblink, C4, steam- cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-buttene								
range of approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0049drocarbtink, C4, steam- cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
approximately -70°C to 0°C (-94°F to 32°F).] 649-116-0049drocarbtink, C4, steam- cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
-70°C to 0°C (-94°F to 32°F).] 649-116-0049drocarbHnK, C4, steam- cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene		annravim	otoly,					
to 0°C (-94°F to 32°F).] 649-116-0049drocarbtinK, C4, steam- cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene			atery					
(-94°F to 32°F).] 649-116-0049drocarbtink, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
32°F).] 649-116-0019drocarbHnK, C4, steam-cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
649-116-0049drocarbtinK, C4, steam-cracker distillate Petroleum gas [A combination of hydrocarbons produced by the distillation of fa steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene		(-94°F to						
649-116-0049drocarbkink, C4, steam-cracker distillate Petroleum gas [A combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene		32°F).]						
steam- cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene	640 116 0		o Palva K	205 405 /	102045 22	Fore Cat	т	
steam- cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene	047-110-0		DUITIS,	293 -4 03-4	+32043-23			
cracker distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
distillate Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene							5. 55 15	
Petroleum gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
gas [A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene			1					
[A complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
complex combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene		[A						
combination of hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
hydrocarbons produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene			on					
produced by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
by the distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene			ons					
distillation of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
of the products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
products of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene			n					
of a steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C4, predominantly 1-butene								
steam cracking process. It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
cracking process. It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
process. It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
It consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
consists predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
predominantly of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
of hydrocarbons having a carbon number of C ₄ , predominantly 1-butene			antly					
hydrocarbons having a carbon number of C ₄ , predominantly 1-butene								
having a carbon number of C ₄ , predominantly 1-butene			ons					
a carbon number of C ₄ , predominantly 1-butene		having						
of C ₄ , predominantly 1-butene		a carbon						
predominantly 1-butene								
1-butene		of C_4 ,						
		predomin	antly					
and 2-								
		and 2-						

butene, containing also butane and isobutene and boiling in the range of approximations 12°C to 5°C (10.4°F to 41°F).]						
649-117-0 Pet roleum	HKS	295_463 (092045_80	ΔT+· R12	F+; T	
gases, liquefied, sweetened C4 fraction Petroleum gas [A complex combination of hydrocarb obtained by subjecting a liquified petroleum gas mix to a sweetenin process to oxidize mercaptar or to remove acidic impurities It consists	d, on oons g	293-403-(092045-80	-E+; R12 Carc. Cat. 1; R45 Muta. Cat. 2; R46	R: 12-45-46 S: 53-45	

saturated and unsaturated				
hydrocarbons	.]			
649-119-0 Caffinates H I (petroleum), steam-cracked C4 fraction cuprous ammonium acetate extn., C3-5 and C3-5 unsatd., butadiene-free Petroleum gas	307-769-4	497722-19	T R: 45-46 S: 53-45	
(petroleum), amine system feed Refinery gas [The feed gas to the amine system for removal of hydrogen sulfide. It consists of hydrogen. Carbon monoxide, carbon dioxide, hydrogen sulfide and aliphatic hydrocarbons		168477-65	T R: 45-46 S: 53-45	

having carbon number predom in the range of C_1 through C_5 may also be present	inantly					
(petrole benzend unit hydrode off Refiner gas [Off gases produce by the benzend unit. It consists primari of hydroge Carbon monoxi and hydroca having carbon number predom in the range of C ₁ through C ₆ , including benzend may also be present.	esulfurizer y ed el sly en. de arbons s inantly	270-747-7	768477-66	-Varc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-122-0 G -dses (petrole benzend unit	H K	270-748-2	268477-67	- C arc. Cat. 1; R45	T R: 45-46 S: 53-45	

recycle hydrogorich Refiner gas [A comple combin of hydroca obtaine by recycling the gases of the benzend unit. It consists primari of hydrogowith various small amount of carbon monoxi and hydroca having carbon number in the range of C1 through C6-]	x ation arbons d sy en			Muta. Cat. 2; R46		
649-123-06-ases (petrole blend oil, hydrogenitrogen rich Refiner gas [A comple combin of	en- 1- y	270-749-8	s684 ⁻ /7-68	-varc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	hydrocarb	ons					
	obtained	0115					
	by						
	distillation	n					
	of a						
	blend						
	oil. It						
	consists						
	primarily						
	of						
	hydrogen						
	and						
	nitrogen						
	with						
	various						
	small						
	amounts						
	of						
	carbon						
	monoxide						
	carbon	,					
	dioxide,						
	and						
	aliphatic						
	hydrocarb	ons					
	having						
	carbon						
	numbers						
	predomin	antly					
	in the						
	range						
	of C ₁						
	through						
	C_{5} .						
649-124-0	1 2 3 3 3	НК	270 750 2	060477 77	Cara Cat	Т	
049-124-0			2/0-/39-2	2004//-//	- C arc. Cat.		
	(petroleur	n),			1; R45	R: 45-46	
	catalytic				Muta.	S: 53-45	
	reformed				Cat. 2;		
	naphtha				R46		
	stripper						
	overheads	•					
	Refinery						
	gas						
	[A						
	complex						
	combinati	on					
		OH					
	of						
	hydrocarb	ons					
	obtained						
	from						
	stabilizati	on					
	of						
	. '					•	

catalytic reformed naphtha. Its consists of hydrogen and saturated hydrocarb having carbon numbers predomina in the range of C ₁ through C ₄ .]	untly					
(petroleum C ₆₋₈ catalytic reformer recycle Refinery gas [A complex combination of hydrocarbe produced by distillation of products from catalytic reforming of C ₆ -C ₈ feed and recycled to conserve hydrogen. It consists primarily of hydrogen. It may	on ons	270-761-3	368477-80	-&arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

ı	also	I					l	
	contain							
	various							
	small							
	amounts							
	of							
	carbon							
	monoxide							
	carbon	,						
	dioxide,							
	nitrogen,							
	and							
	hydrocarb	ons						
	having	70113						
	carbon							
	numbers							
	predomin	antly						
	in the	untry						
	range							
	of C ₁							
	through							
	C_{6} .							
(40.12(.0		НК	270.762.6	0.60477 01	C C-4	Т		
649-126-0			2/0-/62-9	9084//-81	- 6 arc. Cat.	R: 45-46		
	(petroleur	11),			1; R45 Muta.	S: 53-45		
	C_{6-8}					5. 33-43		
	catalytic				Cat. 2; R46			
	reformer Refinery				IX 4 0			
	-							
	gas [A							
	complex							
	combinati	ion						
	of							
	hydrocarb	ons						
	produced	Johns						
	by							
	distillation	n						
	of							
	products							
	from							
	catalytic							
	reforming	Į						
	of C_6 - C_8							
	feed. It							
	consists							
	of							
	hydrocarb	ons						
	having							
	carbon							
	numbers							
	in the							
	range							
	,		· ·		· ·			

	of C ₁						
	through						
	C ₅ and						
		3					
	hydrogen.	.]					
649-127-0	10:9 ses	НК	270-763-4	168477-82	Tarc Cat	Т	
019127	(petroleur		270 703	100177 02	1; R45	R: 45-46	
	C ₆₋₈	11),			Muta.	S: 53-45	
						3. 33-43	
	catalytic				Cat. 2;		
	reformer				R46		
	recycle,						
	hydrogen-	-					
	rich						
	Refinery						
	gas						
649-128-0		ΗK	270-766-0	68477-84			
	(petroleur	n),			1; R45	R: 45-46	
	C_2 -				Muta.	S: 53-45	
	return				Cat. 2;		
	stream				R46		
	Refinery						
	gas						
	[A						
	complex						
	combinati	on					
	of	011					
		ana					
	hydrocarb	OHS					
	obtained						
	by the						
	extraction						
	of						
	hydrogen						
	from						
	a gas						
	stream						
	which						
	consists						
	primarily						
	of						
	hydrogen						
	with						
	small						
	amounts						
	of						
	nitrogen,						
	carbon						
	monoxide						
	methane,	,					
	ethane,						
	and						
	ethylene.						
	It						

	contains	I					
	predomin	antly					
	hydrocart	antity					
		JOHS					
	such as						
	methane,						
	ethane,						
	and						
	ethylene						
	with						
	small						
	amounts						
	of						
	hydrogen	,					
	nitrogen						
	and						
	carbon						
	monoxide	1					
		-					
649-129-0		ΗK	270-774-4	168477-92	-© arc. Cat.	T	
X	(petroleui	n),			1; R45	R: 45-46	
	dry sour,	,,			Muta.	S: 53-45	
	gas-				Cat. 2;		
	concn				R46		
	unit-off				K40		
	Refinery						
	gas						
	[The						
	complex						
	combinat	ion					
	of dry						
	gases						
	from						
	a gas						
	concentra	tion					
	unit. It						
	consists						
	of						
	hydrogen						
	hydrogen						
	sulfide						
	and						
]					
	hydrocarb	ons					
	having						
	carbon						
	numbers						
	predomin	antly					
	in the						
	range						
	of C ₁						
	through						
	$C_{3.}$						
649-130-0	055cec	НК	270-776	568477-93	Carc Cat	Т	
U 1 ₹ 1 3 U - (2/0-//0-	700 1 // -7 3	1; R45	R: 45-46	
	(petroleui	μ),			1, 1143	IX. 43-40	

gas concn. reabsorbe distn. Refinery gas [A complex combination of hydrocart produced by distillation of products from combined gas streams in a gas concentra reabsorbe It consists predomin of hydrogen carbon monoxide carbon dioxide, nitrogen, hydrogen sulfide and hydrocart having	tion r.			Muta. Cat. 2; R46	S: 53-45	
hydrocarb having carbon numbers in the range of C ₁ through	ons					
C ₃ .] 649-131-060ses (petroleur hydrogen absorber off Refinery gas		270-779-1	68477-96	Garc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	[A						
	complex						
	combinati	on					
	obtained	OII					
	- by						
	absorbing						
	hydrogen						
	from a						
	hydrogen						
	rich						
	stream.						
	It						
	consists						
	of						
	hydrogen,						
	carbon						
	monoxide	,					
	nitrogen,						
	and						
	methane						
	with						
	small						
	amounts						
	of C ₂						
		one 1					
	hydrocarb	ons.j					
649-132-0	OG asses	ΗK	270-780-7	768477-97	- € arc. Cat.	T	
	(petroleun	n).			1; R45	R: 45-46	
	hydrogen-				Muta.	S: 53-45	
	rich				Cat. 2;	5.00 .0	
	Refinery				R46		
	- 1				11.40		
	gas						
	[A						
	complex						
	combinati	on					
	separated						
	as a gas						
	from						
	hydrocarb	on					
	gases by						
	chilling.						
	chilling. It						
	It						
	It consists						
	It consists primarily						
	It consists primarily of						
	It consists primarily of hydrogen						
	It consists primarily of hydrogen with						
	It consists primarily of hydrogen with various						
	It consists primarily of hydrogen with various small						
	It consists primarily of hydrogen with various						
	It consists primarily of hydrogen with various small						
	It consists primarily of hydrogen with various small amounts of						
	It consists primarily of hydrogen with various small amounts						

	nitrogen,						
	methane, and C ₂						
	hydrocarb	ons.1					
649-133-0		H K	270 791 2	060477 00	-Carc. Cat.	т	
049-133-0	(petroleur		2/0-/61-2	2004//-90	1; R45	R: 45-46	
	hydrotrea				Muta.	S: 53-45	
	blend oil				Cat. 2;		
	recycle,				R46		
	hydrogen- nitrogen-	_					
	rich						
	Refinery						
	gas						
	[A						
	complex combinati	ion					
	obtained						
	from						
	recycled						
	hydrotrea blend	tea					
	oil. It						
	consists						
	primarily						
	of hydrogen						
	and						
	nitrogen						
	with						
	various small						
	amounts						
	of						
	carbon						
	monoxide carbon	,					
	dioxide						
	and						
	hydrocarb	ons					
	having carbon						
	numbers						
	predomin	antly					
	in the						
	range of C ₁						
	through						
	C ₅ .]						
649-134-0		НК	270-783-3	868478-00	-£ arc. Cat.	Т	
017 101	(petroleur		270 703 3	. 30 170 00		R: 45-46	

recycle, hydroge rich Refinery gas [A complex combina obtained from recycled reactor gases. It consists primarily of hydroge with various small amounts of carbon monoxid carbon dioxide, nitrogen hydroge sulfide, and saturated aliphatic hydroca having carbon numbers in the range of C1 through C5.]	tition de, n de, n de, n			Muta. Cat. 2; R46	S: 53-45	
649-135-0	n-	2/0-/84-9	9684/8 - 01	-& arc. Cat. 1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	

	combinati	on					
	obtained	OII					
	from the						
	reformers.	•					
	It						
	consists						
	primarily						
	of						
	hydrogen with						
	various						
	small						
	amounts						
	of						
	carbon						
	monoxide						
	and						
	aliphatic						
	hydrocarb	ons					
	having carbon						
	numbers						
	predomina	antly					
	in the	•					
	range						
	of C ₁						
	through						
	C ₅ .]						
649-136-0		ΗK	270-785-4	168478-02	-€arc. Cat.		
	(petroleun	n),			1; R45	R: 45-46	
	reforming				Muta.	S: 53-45	
		or			Cat. 2;		I .
	hydrotreat Refinery	ter					
	Refinery	ter			R46		
	Refinery gas	ter					
	Refinery gas [A complex						
	Refinery gas [A complex combinati						
	Refinery gas [A complex combination obtained]						
	Refinery gas [A complex combinati obtained from the	on					
	Refinery gas [A complex combinati obtained from the reforming	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat process.	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat process. It consists primarily	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat process. It consists primarily of	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat process. It consists primarily of hydrogen,	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat process. It consists primarily of hydrogen, methane,	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat process. It consists primarily of hydrogen, methane, and	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat process. It consists primarily of hydrogen, methane, and ethane	on					
	Refinery gas [A complex combination obtained from the reforming hydrotreat process. It consists primarily of hydrogen, methane, and	on					

	and aliphatic hydrocarb having carbon numbers predoming in the range of C ₃ thorugh C ₅ .]	antly					
649-137-0		ΗK	270-787-5	68478-03			
047-13/-0	(petroleur reforming hydrotrear hydrogen-methane-rich Refinery gas [A complex combination obtained from the reforming hydrotrear process. It consists primarily of hydrogen and methane with various small amounts of carbon monoxide carbon dioxide,	n), ter, tering	270-787-3	700470-03	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
	nitrogen and saturated aliphatic hydrocarb having carbon	oons					

	numbers predoming in the range of C ₂ through C ₅ .]	antly					
649-138-0	0 G-9 ses	НК	270-788-0	068478-04	-Carc. Cat	T	
649-138-0	(petroleur reforming hydrotrea make- up, hydrogen- rich Refinery gas [A complex combination obtained from the reforming hydrotrea process. It consists primarily of hydrogen with various small amounts of carbon monoxide and aliphatic hydrocarb having carbon numbers predoming in the	on ting	270-788-0	68478-04	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
	range of C_1 through C_5 .]						
649-139-0		Н К n),	270-789-6	668478-05	- C arc. Cat. 1; R45	T R: 45-46	

district Refiners and the sufficient of the suff	king n. nery plex bination luced llation lucts n a mal king ess. sists rogen de, on oxide, on oxide, on ide rocarbons ng on bers ominantly ee ee l ugh			Muta. Cat. 2; R46	S: 53-45	
cata crac refra abso Refri gas [A com com of hydri	roleum), lytic	270-805-1	68478-25	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	from refraction of products from a catalytic cracking process. It consists of hydrogen and hydrocarbhaving carbon numbers predomin in the range of C ₁ through C ₃ .]	ons					
649-141-0	(petroleur catalytic reformed naphtha separator Refinery gas [A complex combination of hydrocard obtained from the catalytic reforming of straight run naphtha.	ion oons	270-807-2	268478-27	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
	It consists of hydrogen and hydrocarbhaving carbon						

number predoming the range of C ₁ throug C ₆ .]	minantly					
cataly reform naphtl stabilist Refine gas [A comple combine of hydro obtain from the stabilist of cataly reform naphtl It consist of hydro and hydro having carbot number the stabilist of cataly reform naphtl It consist of hydro and hydro having carbot number the stabilist of hydro and hydro having carbot number the stabilist of hydro hydro having carbot number the stabilist of hydro hydro having carbot number the stabilist of hydro hydro hydro hydro having carbot number the stabilist of hydro hydr	leum), tic ned na zer ery lex nation carbons ned he zation tic ned na. ets gen carbons g n ers minantly	270-808-8	868478-28	€arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
cracke distilla	leum), ed ate treater ator	270-809-3	368478-29	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	[A							
	complex							
	combinati	ion						
		1011						
	of							
	hydrocart	ons						
	obtained							
	by							
	treating							
	cracked							
	distillates							
	with							
	hydrogen							
	in the							
	presence							
	of a							
	catalyst.							
	It							
	consists							
	of							
	hydrogen							
	and							
	saturated							
	aliphatic							
	hydrocarb	ons						
	having	1						
	carbon							
	numbers	_						
	predomin	antly						
	in the							
	range							
	of \tilde{C}_1							
	through							
	C_{5} .							
649-144-0	மூரி எவ	НК	270-810-9	068478-30	-Carc. Cat.	Т		
047-144-0			270-810-3	700470-30				
	(petroleur				1; R45	R: 45-46		
	hydrodesi	ulfurized			Muta.	S: 53-45		
	straight-				Cat. 2;			
	run				R46			
	naphtha							
	separator							
	Refinery							
	gas							
	[A							
	complex							
	combinati	ion						
	of							
	hydrocart	ons						
		10115						
	obtained							
	from							
		ulfurization	h					
	of							
	straight-							
	0	I	I	I	I	I	I	I

	run						
	naphtha.						
	It						
	consists						
	of						
	hydrogen						
	and						
	saturated						
	aliphatic						
	hydrocart	ons					
	having						
	carbon						
	numbers						
	predomin	antlv					
	in the						
	range						
	of C ₁						
	through						
	C_{6} .						
649-145-0		ΗK	270-999-8	868513-14	-€arc. Cat.		
	(petroleur	n),			1; R45	R: 45-46	
	catalytic				Muta.	S: 53-45	
	reformed				Cat. 2;		
	straight-				R46		
	run						
	naphtha						
	stabilizer						
	overheads	•					
	Refinery						
	gas						
	[A						
	complex						
	combinati	on					
	of						
	hydrocart	ons					
	obtained						
	from the						
	catalytic						
	reforming						
	of						
	straight-						
	run						
	naphtha						
	followed						
	by						
	fractionat	ion					
	of the						
	total						
	effluent.						
	It						
	consists						
	of						

hydroge methane ethane and propane	,					
649-146-0 Gases (petrole reforme effluent high-pressure flash drum of Refiner gas [A complex combina produce by the high-pressure flashing of the effluent from the reforming reactor. It consists primaril of hydroge with various small amounts of methane, and propane	H K (mm), (mm), (mm), (mm)	271-003-4	468513-18	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-147-0 G-Asses (petrole reforme effluent low-pressure flash drum of Refinery gas		271-005-:	568513-19	- © arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	[A						
	complex						
	combinati	on					
	produced						
	by low-						
	pressure						
	flashing						
	of the						
	effluent						
	from the						
	reforming						
	reactor.						
	It						
	consists						
	primarily						
	of						
	hydrogen						
	with						
	various						
	small						
	amounts						
	of						
	methane,						
	ethane,						
	and						
	propane.]						
649-148-0	OG-3ses	ΗK	271-258-1	68527-15	-Carc. Cat.		
649-148-0	0 6-3 ses (petroleun		271-258-1	68527-15	1; R45	R: 45-46	
649-148-0	063ses (petroleun oil		271-258-1	168527-15	1; R45 Muta.		
649-148-0	06∙3ses (petroleun oil refinery		271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	OGases (petroleun oil refinery gas		271-258-1	68527-15	1; R45 Muta.	R: 45-46	
649-148-0	06∙3ses (petroleun oil refinery		271-258-1	68527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	OGases (petroleun oil refinery gas		271-258-1	68527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	OGases (petroleun oil refinery gas distn. off Refinery		271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	Ocases (petroleun oil refinery gas distn. off Refinery gas		271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	(petroleun oil refinery gas distn. off Refinery gas [A		271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	(petroleun oil refinery gas distn. off Refinery gas [A complex	n),	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	(petroleun oil refinery gas distn. off Refinery gas [A complex combinati	n),	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	(petroleun oil refinery gas distn. off Refinery gas [A complex combinati separated	n),	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	(petroleun oil refinery gas distn. off Refinery gas [A complex combinati separated by	n), on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of ases (petroleun oil refinery gas distn. off Refinery gas [A complex combinati separated by distillation	n), on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	Operated the complex combination of a gas	n), on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of a gas stream	n), on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of a gas stream containing	n), on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of a gas stream containing hydrogen,	n), on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of a gas stream containing hydrogen, carbon	on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of a gas stream containing hydrogen, carbon monoxide	on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of ases (petroleun oil refinery gas distn. off Refinery gas [A complex combinati separated by distillation of a gas stream containing hydrogen, carbon monoxide carbon	on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of a gas stream containing hydrogen, carbon monoxide carbon dil refinery gas distn. off Refinery gas [A complex combinati separated by distillation of a gas stream containing hydrogen, carbon monoxide carbon dioxide	on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of ases (petroleun oil refinery gas distn. off Refinery gas [A complex combinati separated by distillation of a gas stream containing hydrogen, carbon monoxide carbon dioxide and	on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of ases (petroleun oil refinery gas distn. off Refinery gas [A complex combinati separated by distillation of a gas stream containing hydrogen, carbon monoxide carbon dioxide and hydrocarb	on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	
649-148-0	of ases (petroleun oil refinery gas distn. off Refinery gas [A complex combinati separated by distillation of a gas stream containing hydrogen, carbon monoxide carbon dioxide and	on	271-258-1	168527-15	1; R45 Muta. Cat. 2;	R: 45-46	

	numbers						
	in the						
	range						
	of C ₁						
	through						
	C_6 or						
	obtained						
	by						
	cracking						
	ethane						
	and						
	propane.						
	It						
	consists						
	of						
	hydrocarb	one					
	having	70115					
	carbon						
	numbers						
	predomin	ontly					
	in the	anny					
	range						
	of C ₁						
	through						
	C_2 ,						
	hydrogen,	}					
	nitrogen,						
	and						
	carbon	_					
	monoxide	.]					
649-149-0	0 0. 9ses	НК	271-623-4	68602-82	-Carc Cat	Т	
015 115 ((petroleur		271 023 3	700002 02	1; R45	R: 45-46	
	benzene	,			Muta.	S: 53-45	
	unit				Cat. 2;	0. 55 15	
	hydrotrea	ter			R46		
	depentani				10		
	overheads	2.01					
	Refinery						
	gas						
	[A						
	complex						
	combinati	ion					
	produced	1011					
	by						
	treating						
	the feed						
	from the						
	benzene						
	unit with						
	hydrogen in the						
	presence						
	presence						

	of a	1	1	1		1	
	catalyst						
	followed						
	by						
	depentanizing.						
	It						
	consists						
	primarily						
	of						
	hydrogen,						
	ethane						
	and						
	propane						
	with						
	various						
	small						
	amounts						
	of						
	nitrogen,						
	carbon						
	monoxide,						
	carbon						
	dioxide						
	and						
	hydrocarbons						
	having						
	carbon						
	numbers						
	predominantly						
	in the						
	range						
	of C ₁						
	through						
	C ₆ . It						
	may						
	contain						
	trace						
	amounts						
	of						
	benzene.]						
649-150-0		271-625-6	68602-84				
	(petroleum),			1; R45	R: 45-46		
	secondary			Muta.	S: 53-45		
	absorber			Cat. 2;			
	off,			R46			
	fluidized						
	catalytic						
	cracker						
	overheads						
	fractionator						
	Refinery						
	gas						

	[A					
	complex					
	combination					
	produced					
	by the					
	fractionation					
	of the					
	overhead					
	products					
	from the					
	catalytic					
	cracking					
	process					
	in the					
	fluidized					
	catalytic					
	cracker.					
	It					
	consists					
	of					
	hydrogen,					
	nitrogen,					
	and					
	hydrocarbons					
	having					
	carbon					
	numbers					
	predominantly					
	in the					
	range					
	of C ₁					
	through					
	C_{3} .					
640 151 (P etroleum H K	271 750	668607-11	Cor Cot	Т	
		2/1-/30-	000007-11			
X	products,			1; R45	R: 45-46	
	refinery			Muta.	S: 53-45	
	gases			Cat. 2;		
	Refinery			R46		
	gas					
	[A					
	complex					
	combination					
	which					
	consists					
	primarily					
	of					
	hydrogen					
	with					
	various					
	small					
	amounts					
	of					

	methane, ethane, and						
	propane. _]						
649-152-0	propane.] OGAses (petroleur hydrocrace low-pressure separator Refinery gas [A complex combinate obtained by the liquid-vapor separation of the hydrocrace process reactor effluent. It consists predomin of hydrogen and saturated hydrocare having carbon numbers	H K n), sking on sking antly	272-182-3	168783-06	-£arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
	predomin in the range of C ₁ through C ₃ .]	antly					
649-153-0	(petroleur refinery Refinery gas [A complex combinati		272-338-9	68814-67	-£arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	obtained							
	from							
	various							
	petroleum	l						
	refining							
	operations	2						
	It							
	consists							
	of							
	hydrogen							
	and							
	hydrocarb	ons						
	having							
	carbon							
	numbers							
	predomin	antly						
	in the							
	range							
	of C ₁							
	through							
	C ₃ .]							
						_		
649-154-0		ΗK	272-343-6	668814-90	-€arc. Cat.			
	(petroleur				1; R45	R: 45-46		
	platforme	r			Muta.	S: 53-45		
	products				Cat. 2;			
	separator				R46			
	off							
	Refinery							
	gas							
	[A							
	complex							
	combinati	on						
	obtained							
	from the							
	chemical							
	reforming							
	of	1						
	naphthene	• • •						
	to	23						
	aromatics							
	It	•						
	consists							
	of							
	hydrogen							
	and							
	saturated							
	aliphatic							
	hydrocarb	ons						
	having							
	carbon							
	numbers							
	predomin	antly						
					•		•	•

	in the						
	range						
	of C ₂						
	through						
	C ₄ .]						
	C4.j						
649-155-0) Calses	ΗK	272-775-5	68911-58	-Carc. Cat.	T	
	(petroleur	n),			1; R45	R: 45-46	
	hydrotrea	ted			Muta.	S: 53-45	
	sour				Cat. 2;		
	kerosine				R46		
	depentani	zer					
	stabilizer						
	off						
	Refinery						
	gas						
	[The						
	complex						
	combinati	on					
	obtained	011					
	from the						
	depentani	7er					
	stabilizati						
	of	OII					
	hydrotrea	ted					
	kerosine.	ica					
	It						
	consists						
	primarily						
	of						
	hydrogen,						
	methane,	•					
	ethane,						
	and						
	propane						
	with						
	various						
	small						
	amounts						
	of						
	nitrogen,						
	hydrogen						
	sulfide,						
	carbon						
	monoxide						
	and						
	hydrocarb	ons					
	having						
	carbon						
	numbers						
	predomin	antly					
	in the						
	range						
	iuiige		I				l

	CO	I	I	l	I	İ	I	
	of C ₄							
	through							
	$C_{5.}$							
649-156-0		ΗK	272-776-0	068911-59				
	(petroleur	n),			1; R45	R: 45-46		
	hydrotrea	ted			Muta.	S: 53-45		
	sour				Cat. 2;			
	kerosine				R46			
	flash							
	drum							
	Refinery							
	gas							
	[A							
	complex							
	combinati	on						
	obtained							
	from							
	the flash							
	drum of							
	the unit							
	treating							
	sour							
	kerosine							
	with							
	hydrogen							
	in the							
	presence							
	of a							
	catalyst.							
	It							
	consists							
	primarily							
	of							
	hydrogen							
	and							
	methane							
	with							
	various							
	small							
	amounts							
	of							
	nitrogen,							
	carbon							
	monoxide	<u>.</u>						
	and	2						
	hydro-							
	carbons							
	having							
	carbon							
	numbers							
	predomin	antly						
	Prodomini	m,	l	I	l	l	l	

	in the	I	<u> </u>	<u> </u>	l		l	
	range							
	of C ₂							
	through							
	C_{5} .]							
649-157-0) G -2ses	НК	272-873-8	868919-01	-Carc. Cat.	Т		
	(petroleur	n),			1; R45	R: 45-46		
	distillate				Muta.	S: 53-45		
	unifiner				Cat. 2;			
	desulfuriz	ation			R46			
	stripper off							
	Refinery							
	gas							
	[A							
	complex							
	combinati	lon						
	stripped from the							
	liquid							
	product							
	of the							
	unifiner desulfuriz							
	process.	ation						
	It							
	consists							
	of							
	hydrogen sulfide,							
	methane,							
	ethane,							
	and							
	propane.]							
649-158-0) G 8 ses	НК	272-874-3	868919-02	-Carc. Cat.	Т		
	(petroleur	n),			1; R45	R: 45-46		
	fluidized				Muta.	S: 53-45		
	catalytic cracker				Cat. 2; R46			
	fractionat	ion			1040			
	off							
	Refinery							
	gas							
	[A complex							
	combinati	ion						
	produced							
	by the							
	fractionat	ion						
	of the							
	overhead product							
	Product				I			

	of the fluidized catalytic cracking process. It consists of hydrogen hydrogen sulfide, nitrogen, and hydrocarb having carbon numbers predoming in the range of C ₁ through C ₅ .]	oons					
649-159-0	(petroleur fluidized catalytic cracker scrubbing secondary absorber off Refinery gas [A complex		272-875-9	968919-03	- £ arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
	combination produced by scrubbing the overhead gas from the fluidized catalytic cracker. It consists of hydrogen, nitrogen,						

	methane,							
	ethane							
	and							
	propane.]							
649-160-0		ΗK	272-876-4	168919-04	-C arc. Cat.			
	(petroleur	n),			1; R45	R: 45-46		
	heavy				Muta.	S: 53-45		
	distillate				Cat. 2;			
	hydrotrea	ter			R46			
	desulfuriz				_			
	stripper							
	off							
	Refinery							
	gas							
	[A							
	complex combinati	0.10						
		On						
	stripped							
	from the							
	liquid							
	product							
	of the							
	heavy							
	distillate							
	hydrotrea							
	desulfuriz	ation						
	process.							
	It							
	consists							
	of							
	hydrogen,	,						
	hydrogen							
	sulfide,							
	and							
	saturated							
	aliphatic							
	hydrocarb	ons						
	having	0110						
	carbon							
	numbers							
	predomin	antly						
	in the	milti y						
	range							
	of C ₁							
	through							
	C_{5} .							
	C ₅ .]							
649-161-0	OG-alses	НК	272-880-6	68919-07	- C arc. Cat.	Т		
	(petroleur				1; R45	R: 45-46		
	platforme				Muta.	S: 53-45		
	stabilizer				Cat. 2;			
	off, light				R46			
'	,		II .	I	-		I	I

	ends						
	fractionat	ion					
	Refinery						
	gas						
	[A						
	complex						
	combinati	on					
	obtained	1011					
	by the						
	fractionat	ion					
	of the						
	light						
	ends						
	of the						
	platinum						
	reactors						
	of the						
	plattforme	er					
	unit. It						
	consists						
	of						
	hydrogen,	,					
	methane,						
	ethane						
	and						
	propane.]						
	propune.j						
640 162 (272 991 1	69010 09	Cara Cat	Т	
649-162-0) G ases	НК	272-881-1	168919-08			
649-162-0 X) G ases (petroleur	НК	272-881-1	168919-08	1; R45	R: 45-46	
	Ofases (petroleur preflash	НК	272-881-1	168919-08	1; R45 Muta.		
	OG-ases (petroleur preflash tower	НК	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	Offases (petroleur preflash tower off,	НК	272-881-1	68919-08	1; R45 Muta.	R: 45-46	
	OGases (petroleur preflash tower off, crude	НК	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn.	НК	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery	НК	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	(petroleur preflash tower off, crude distn. Refinery gas	НК	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OFases (petroleur preflash tower off, crude distn. Refinery gas [A	НК	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex	Н К n),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati	H Kn),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced	H K n),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from	H K n),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first	H K n),	272-881-1	68919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower	H K n),	272-881-1	68919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used	H K n),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the	H Kn),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the distillation	H Kn),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the distillation of crude	H Kn),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the distillation of crude oil. It	H Kn),	272-881-1	68919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the distillation of crude oil. It consists	H Kn),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the distillation of crude oil. It consists of	H Kn),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the distillation of crude oil. It consists of nitrogen	H Kn),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the distillation of crude oil. It consists of nitrogen and	H Kn),	272-881-1	168919-08	1; R45 Muta. Cat. 2;	R: 45-46	
	OGases (petroleur preflash tower off, crude distn. Refinery gas [A complex combinati produced from the first tower used in the distillation of crude oil. It consists of nitrogen	H Kn),	272-881-1	68919-08	1; R45 Muta. Cat. 2;	R: 45-46	

aliphati hydroca having carbon number predom in the range of C_1 through C_5 .]	rbons s inantly					
649-163- Gases (petrole tar stripper off Refiner gas [A complet combin obtained by the fraction of reduced crude oil. It consists of hydroge and hydroca having carbon number predom in the range of C ₁ through C ₄ .]	ntion I ation rbons	272-884-8	868919-11	-varc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-164-0 Gases (petrole unifiner stripper off Refiner gas [A combin	y	272-885-2	68919-12	•Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

of hydroger and methane obtained by fractiona of the products from the unifiner unit.]	tion				
naphtha separator Refinery gas [A complex combinat of hydrocar obtained from the	ulfurized ion bons ulfurization	568952-79	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-166-0 Tail gas (petroleu straight- run naphtha hydrodes Refinery gas [A complex combinar obtained from the	H K m), ulfurizer	068952-80	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	of							
	straight-							
	run							
	naphtha:							
	It							
	consists							
	of							
	hydrogen							
	and							
	hydrocarb	ons						
	having							
	carbon							
	numbers							
	predomin	antly						
	in the							
	range							
	of \tilde{C}_1							
	through							
	$C_{5.}$							
649-167-0		ΗK	273-269-7	768955-33				
	(petroleur	n),			1; R45	R: 45-46		
	sponge				Muta.	S: 53-45		
	absorber				Cat. 2;			
	off,				R46			
	fluidized							
	catalytic							
	cracker							
	and							
	gas oil							
	desulfuriz	er						
	overhead							
	fractionat	ion						
	Refinery	1011						
	gas							
	[A							
	complex							
	combinati	ion						
	obtained							
	by the							
	fractionat	ion						
	of							
	products							
	from the							
	fluidized							
	catalytic							
	cracker							
	and							
	gas oil							
	desulfuriz	er.						
	It							
	consists							
	of							
		ı		I			I	I

hydroger and hydrocar having carbon numbers predomin in the range of C_1 through C_4 .]	bons					
(petroleu crude distn. and catalytic cracking Refinery gas [A complex combina produced by crude distillatio and catalytic cracking processe It consists of hydroger hydroger hydroger sulfide, nitrogen, carbon monoxid and paraffinic and olefinic hydrocar having carbon numbers predomit in the range of C1	tion l on s.	273-563-5	568989-88	-&arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	through C_{6} .						
649-169-0	(petroleur gas oil diethanola scrubber off Refinery gas [A complex combination produced by desulfurizof gas oils with diethanola It consists predominof hydrogen sulfide, hydrogen and aliphatic hydrocarb having carbon numbers in the range of C1 through C5.]	amine amine.	295-397-2	292045-15	© arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-170-0	(petroleur gas oil	alfurization		392045-16	•Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	liquid						
	phase						
	from the						
	effluent						
	from the						
	1 1	4:					
	hydrogena	ation					
	reaction.						
	It						
	consists						
	predomin	antly					
	of						
	hydrogen,	,					
	hydrogen						
	sulfide						
	and						
	aliphatic						
	hydrocarb	ons					
	having	70115					
	carbon						
	numbers						
		41					
	predomin	antiy					
	in the						
	range						
	of C ₁						
	through						
	$C_{3.}$						
640 171 (1 1 1 1 1 1 1 1 1 1	II I/	205 200 2	202045 17	Como Cot	т	
649-171-0		H K	295-399-3	92045-17	-Carc.Cat.		
649-171-0	(petroleur		295-399-3	392045-17	1; R45	R: 45-46	
649-171-0	(petroleur gas oil	n),		92045-17	1; R45 Muta.		
649-171-0	(petroleur gas oil hydrodesi			392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodesu purge	n),		392045-17	1; R45 Muta.	R: 45-46	
649-171-0	(petroleur gas oil hydrodesi	n),		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodesu purge Refinery gas	n),		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodesu purge Refinery	n),		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodesu purge Refinery gas	n),		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodesu purge Refinery gas [A complex combinati	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodesu purge Refinery gas [A complex combinati of gases	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combinati of gases obtained	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combinati of gases obtained from the	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combinati of gases obtained from the reformer	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combinati of gases obtained from the reformer and	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges	n), ılfurizatior		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the hydrogene	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the hydrogenareactor.	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the hydrogener reactor. It	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the hydrogener reactor. It consists	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the hydrogenareactor. It consists predomination of gases obtained from the purges from the hydrogenareactor.	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the hydrogena reactor. It consists predomination of	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the hydrogenareactor. It consists predomination of gases obtained from the purges from the hydrogenareactor.	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	
649-171-0	(petroleur gas oil hydrodest purge Refinery gas [A complex combination of gases obtained from the reformer and from the purges from the hydrogena reactor. It consists predomination of	n), ılfurization on		392045-17	1; R45 Muta. Cat. 2;	R: 45-46	

(petroleun hydrogena effluent flash drum off Refinery gas [A complex combinati of gases obtained from flash of the effluents after the hydrogena reaction. It consists predomina of hydrogen and	H K n), ator	295400-7	92045-18	-6arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
aliphatic hydrocarb having carbon numbers predomina in the range of C ₁ through C ₆ .]						
	H K	295-401-2	292045-19	- C arc. Cat. 1; R45	T R: 45-46	

naphtha steam cracking high-pressure residual Refinery gas [IA complex combination obtained as a mixture of the non-condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and operations having carbon numbers predominantly in the range of C1					
cracking high- pressure residual Refinery gas [A complex combination obtained as a mixture of the non-condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range				S: 53-45	
high- pressure residual Refinery gas [A complex combination obtained as a mixture of the non- condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range					
high- pressure residual Refinery gas [A complex combination obtained as a mixture of the non- condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range	cracking		R46		
pressure residual Refinery gas [A complex combination obtained as a mixture of the non-condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
residual Refinery gas [A complex combination obtained as a mixture of the non- condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the lange					
Refinery gas [A complex combination obtained as a mixture of the non-condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
gas [A complex combination obtained as a mixture of the non- condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and olefinic hydrocarbons having carbon numbers predominantly in the range					
[A complex combination obtained as a mixture of the non-condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range	- 1				
complex combination obtained as a mixture of the non- condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbon having carbon numbers predominantly in the range					
combination obtained as a mixture of the non- condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
obtained as a mixture of the non-condensable portions from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range	combination				
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from the product of a naphtha steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
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steam cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
cracking process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
process as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
as well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
well as residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range	_				
residual gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
gases obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
obtained during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
during the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range	obtained				
the preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
preparation of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
of subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
subsequent products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
products. It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
It consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
consists predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
predominantly of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
of hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
hydrogen and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range	of				
and paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
paraffinic and olefinic hydrocarbons having carbon numbers predominantly in the range					
and olefinic hydrocarbons having carbon numbers predominantly in the range					
olefinic hydrocarbons having carbon numbers predominantly in the range					
hydrocarbons having carbon numbers predominantly in the range					
having carbon numbers predominantly in the range					
carbon numbers predominantly in the range	having				
numbers predominantly in the range					
predominantly in the range					
in the range					
range	predominantly				
01 C1					
	or C ₁				

through C_5 with which natural gas may also be mixed.]						
(petroleu residue visbaking off Refinery gas [A complex combinate obtained from viscosity reduction of residues in a furnace. It consists predomin of hydroger sulfide and paraffinical and olefinical hydrocar having carbon numbers predomin in the range of C1 through C5.]	ion	295-402-8	892045-20	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-177-0 6 dses (petroleu C ₃₋₄ Petroleur gas		268-629-3	568131-75	- Q arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	F A	I	I I		I		I	I
	[A							
	complex							
	combinati	on						
	of							
	hydrocarb	ons						
	produced							
	by							
	distillation	<u></u>						
		11						
	of							
	products							
	from the							
	cracking							
	of crude							
	oil. It							
	consists							
	of							
	hydrocarb	none						
	havina	10115						
	having							
	carbon							
	numbers							
	in the							
	range							
	of C ₃							
	through							
	C_4							
		41						
	predomin	antiy						
	of							
	propane							
	and							
	propylene	,						
	and							
	boiling							
	in the							
	range of							
	approxim	ntaly						
		attry						
	-51°C							
	to -1°C							
	(-60 °F							
	to 30							
	°F.)]							
649-178-0	∭வி எவ	НК	260 617 3	268307-98	Tare Cat	Т		
047-170-0	(petroleur		207-017-2	200307-70	1; R45	R: 45-46		
	petioleur	11),				S: 53-45		
	catalytic				Muta.	5. 33-43		
	cracked				Cat. 2;			
	distillate				R46			
	and							
	catalytic							
	cracked							
	naphtha							
	fractionat	ion						
	absorber							
	40501001							

	Petroleum	h					
	gas						
	[The						
	complex						
	combinati	ion					
	of						
	hydrocarb	ons					
	from the						
	distillation	n					
	of the						
	products						
	from						
	catalytic						
	cracked						
	distillates						
	and						
	catalytic cracked						
	naphtha. It						
	consists						
	predomin	antly					
	of	untry					
	hydrocarb	ons					
	having	0115					
	carbon						
	numbers						
	in the						
	range						
	of \tilde{C}_1						
	through						
	C_{4} .						
(40, 170, (11.17	2(0, (10, (0.60207.00	7 0 0 4	Т	
649-179-0		HK	269-618-8	368307-99	-Carc. Cat.		
	(petroleur	n),			1; R45	R: 45-46	
	catalytic				Muta.	S: 53-45	
	polymn.				Cat. 2; R46		
	naphtha fractionat	ion			N40		
	stabilizer	1011					
	Petroleum						
	gas	1					
	[A						
	complex						
	combinati	ion					
	of						
	hydrocarb	ons					
	from the						
	fractionat	ion					
	stabilizati						
	products						
	from						
	polymeriz	ation					

It co proof hy ha ca nu in	onsists redominated drocarb ving urbon umbers the						
of	nge CC ₁ rough						
ca rei na fra sta hy su fre Pe ga [A co co of hy ob fro fra sta of ca rei na an fro wh hy su ha rei by an tre It	etroleum ttalytic formed uphtha actionati abilizer, /drogen ulfide- ee etroleum as omplex ombinati otained om actionati abilizatio formed uphtha ad om hich /drogen ulfide as been moved	on on ons	269-619-3	368308-00	•Øarc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

predomin of hydrocard having carbon numbers predomin in the range of C_1 through C_4 .]	oons					
(petroleur cracked distillate hydrotrea stripper Petroleun gas [A complex combinat of hydrocard obtained by treating thermal cracked distillates with hydrogen in the presence of a catalyst. It consists predomin of saturated hydrocard having carbon numbers predomin in the range of C ₁ through C ₆ .]	ion oons antly	269-620-9	₹68308-01	-©arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

649-182-0	Mrdl gog	НК	260 630 3	269209 10	-Carc. Cat.	Т	
047-102-	(petroleur		209-030	000000-10	1; R45	R: 45-46	
	ctroight	11),			Muta.	S: 53-45	
	straight-					5. 33-43	
	run				Cat. 2;		
	distillate	10 .			R46		
	hydrodesi						
	hydrogen						
	sulfide-						
	free						
	Petroleun	1					
	gas						
	[A						
	complex						
	combinati	ion					
	of						
	hydrocart	ons					
	obtained						
	from						
	catalytic						
	hydrodesi	ulfurization	1				
	of						
	straight						
	run						
	distillates						
	and						
	from						
	which						
	hydrogen						
	sulfide						
	has been						
	removed						
	by						
	amine						
	treatment	†					
	It						
	consists	41					
	predomin	antiy					
	of						
	hydrocarb	ons					
	having						
	carbon						
	numbers						
	predomin	antly					
	in the						
	range						
	of C ₁						
	through						
	C_{4} .						
			• • • • • • •			_	
649-183-	Ju al l gas	ΗK	269-623-5	68308-03	-Carc. Cat.		
	(petroleur	n),			1; R45	R: 45-46	

1	• 1 1	1		Í	Nr.	0 52 45	į i	
	gas oil				Muta.	S: 53-45		
•	catalytic				Cat. 2;			
•	cracking				R46			
	absorber							
	Petroleum							
	gas							
	[A							
	complex							
	combination							
I	of							
	hydrocarbons	5						
	obtained							
	from the							
	distillation							
I	of							
	products							
	from the							
	catalytic							
	cracking							
	of gas							
	oil. It							
	consists							
1	predominantl	У						
	of							
	hydrocarbons	5						
1	having							
	carbon							
	numbers							
	predominantl in the	У						
	range							
	of C ₁							
	through							
1	C ₅ .]							
649-184-00	Tail gas H	K	269-624-0	068308-04	-Carc. Cat.	Т		
	(petroleum),				1; R45	R: 45-46		
	gas				Muta.	S: 53-45		
	recovery				Cat. 2;			
	plant				R46			
	Petroleum							
1	gas							
	[A							
	complex							
	combination							
	of							
	hydrocarbons	;						
	from the							
	distillation							
	of							
1	products							
	from							
-	miscellaneou	S						

	hydrocarb	on					
	streams.						
	It						
	consists predomin	ontly					
	of	anny					
	hydrocarb	ons					
	having	70115					
	carbon						
	numbers						
	predomin	antly					
	in the						
	range						
	of C_1						
	through						
	C ₅ .]						
649-185-0		НК	269-625-6	68308-05	- € arc. Cat.		
	(petroleur	n),			1; R45	R: 45-46	
	gas				Muta.	S: 53-45	
	recovery plant				Cat. 2; R46		
	deethaniz	er			1040		
	Petroleum						
	gas						
	[A						
	complex						
	combinati	ion					
	of hydrocarb	one					
	from the	JU115					
	distillation	n					
	of						
	products						
	from						
	miscellan						
	hydrocarb	on					
	streams. It						
	consists						
	of						
	hydrocarb	on					
	having						
	carbon						
	numbers	41					
	predomin in the	antly					
	range						
	of C ₁						
	through						
	C_{4} .						
	-						

649-186- 00 -01 ga	ıs	НК	269-626-1	168308-06	-Carc. Cat.	Т		
(petro					1; R45	R: 45-46		
		ılfurized			Muta.	S: 53-45		
distilla					Cat. 2;			
and					R46			
	lesi	ılfurized						
naphth								
fractio	nat	or						
acid-fi		, 						
Petrol		1						
gas	Cull							
[A								
compl	ex							
combi		ion						
of	iiat							
hydro	ark	none						
obtain		70113						
from	cu							
fraction	nat	ion						
of	mai	1011						
	laci	ulfurized						
naphth		umumzcu						
and	ıa							
distilla	ıta.							
hydro		non.						
stream		JO11						
	IS							
and	1							
treated	l							
to	_							
remov	e							
acidic	4:							
impur	mes	3 .						
It .	,							
consis		41						
predoi	nın	antiy						
of	1							
hydro		ons						
having								
carbor								
numbe		41						
predoi	nın	antly						
in the								
range								
of C ₁	_							
throug	,h							
$C_{5.}$								
649-187-0 0 a6l ga	15	НК	269-627-2	768308-07	- 6 arc. Cat.	Т		
(petro			207-021-	00200-07	1; R45	R: 45-46		
		ulfurized			Muta.	S: 53-45		
vacuu		arrurizeu			Cat. 2;	IJ. IJ . ¬ IJ		
gas oil					R46			
strippe					10			
Strippe	٠٠,	I	I				I	1

	hydrogen						
	sulfide-						
	free						
	Petroleum	1					
	gas						
	[A						
	complex						
	combinati	on					
	of						
	hydrocarb	ons					
	obtained						
	from						
	stripping						
	stabilizati	on					
	of						
	catalytic						
	Latarytic	16					
	hydrodesi	murizea					
	vacuum						
	gas oil						
	and						
	from						
	which						
	hydrogen						
	sulfide						
	has been						
	removed						
	by						
	amine						
	treatment						
	It						
	consists						
		41					
	predomin	antiy					
	of						
	hydrocarb	ons					
	having						
	carbon						
	numbers						
	predomin	antly					
	in the	uniny					
	range						
	of C ₁						
	through						
	C_{6} .						
649-188-0)Calil gas	ΗK	269-629-8	868308-09			
	(petroleur	n),			1; R45	R: 45-46	
	light	**			Muta.	S: 53-45	
	straight-				Cat. 2;		
	run				R46		
					1340		
	naphtha						
	stabilizer,						
	hydrogen						
		'		'			

free petroleum gas [A complex combination of hydrocarbons obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₃ -] 649-189-00al gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum gas		sulfide-							
petroleum gas [A complex combination of hydrocartons obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocartons having carbon numbers predominantly in the range of C ₁ through C ₅ -] 649-189-00all gas (petroleum), propane- propylene alkylation feed prep deethanizer Petroleum									
gas [A complex combination of hydrocarbons obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00al gas H K (petroleum), propane- propylene alkylation feed prep deethanizer Petroleum									
[A complex combination of hydrocarbons obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00tail gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		_							
combination of hydrocarbons obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₃₋₁ 649-189-00al gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		[A							
combination of hydrocarbons obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₃ -1 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		complex							
of hydrocarbons obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅₋₁ 649-189-00@ gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum			on						
hydrocarbons obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅₋₁ 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
obtained from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C1 through C5-1 649-189-0@all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum			ons						
from fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₃₋] 649-189-0@all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
fractionation stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅₋] 649-189-00ail gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
stabilization of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00all gas H K (petroleum), propane- propylene alkylation feed prep deethanizer Petroleum			ion						
of light straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅₋ .] 649-189-00ail gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
straight run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00al gas (petroleum), propane- propylene alkylation feed prep deethanizer Petroleum									
run naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00all gas H K (petroleum), propane- propylene alkylation feed prep deethanizer Petroleum		straight							
naphtha and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-0tail gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		_							
and from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00all gas H K (petroleum), propanepropylene alkylation feed prep deethanizer Petroleum									
from which hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅₋ .] 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
hydrogen sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C1 through C5.] 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C1 through C5.] 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		which							
sulfide has been removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C1 through C5.] 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		hydrogen							
removed by amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C1 through C5:] 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
by amine treatment It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-0tal gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		has been							
amine treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		removed							
treatment. It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-0@all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		by							
It consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum		amine							
consists predominantly of hydrocarbons having carbon numbers predominantly in the range of C_1 through C_5 .] 649-189-00all gas HK (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
predominantly of hydrocarbons having carbon numbers predominantly in the range of C_1 through C_5 .] $649-189-00$ $(petroleum), propane-propylene alkylation feed prep deethanizer Petroleum)$									
of hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00all gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
hydrocarbons having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-00al gas (petroleum), propane- propylene alkylation feed prep deethanizer Petroleum		predomin	antly						
having carbon numbers predominantly in the range of C ₁ through C ₅ .] 649-189-0 Fall gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
$\begin{array}{c} \text{carbon} \\ \text{numbers} \\ \text{predominantly} \\ \text{in the} \\ \text{range} \\ \text{of } C_1 \\ \text{through} \\ C_5. \end{array} \\ \hline \textbf{649-189-00all gas} \\ \textbf{K} \\ \textbf{(petroleum)}, \\ \textbf{propane-} \\ \textbf{propylene} \\ \textbf{alkylation} \\ \textbf{feed} \\ \textbf{prep} \\ \textbf{deethanizer} \\ \textbf{Petroleum} \\ \hline \end{array}$			ons						
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$									
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$									
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			_						
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			antly						
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$									
through C_{5} .] 649-189-0 \mathbf{T} gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
C ₅ .] C ₆ 49-189-00al gas H K 269-631-968308-11-2arc. Cat T 1; R45 R: 45-46 Muta. S: 53-45 Cat. 2; R46 prep deethanizer Petroleum R46 Prep Cat. 2 R46 R46									
649-189-00al gas H K (petroleum), propane-propylene alkylation feed prep deethanizer Petroleum									
(petroleum), propane- propylene alkylation feed prep deethanizer Petroleum		$[C_5.]$							
(petroleum), propane- propylene alkylation feed prep deethanizer Petroleum	649-189-0) (Tái l gas	НК	269-631-9	968308-11	- C arc. Cat	Т		
propane- propylene alkylation feed prep deethanizer Petroleum									
propylene alkylation feed prep deethanizer Petroleum			/"						
alkylation feed prep deethanizer Petroleum			,						
feed prep deethanizer Petroleum									
prep deethanizer Petroleum									
deethanizer Petroleum									
Petroleum			er						
			1					1	1

	[A						
	complex						
	combinati	on					
	of	OII					
	I I						
	hydrocarb	ons					
	obtained						
	from the						
	distillation	1					
	of the						
	reaction						
	I I						
	products						
	of						
	propane						
	with						
	propylene						
	It						
	consists						
	of						
	I I	0.00					
	hydrocarb	OUIS					
	having						
	carbon						
	numbers						
	predomina	antly					
	in the	J					
	range						
	of C ₁						
	through						
	$C_{4.}$						
640 100 (лга1	ΗK	260 622	160200 12	Tama Cat	Т	
649-190-0			209-032-4	168308-12			
	(petroleun	n),			1; R45	R: 45-46	
	vacuum				Muta.	S: 53-45	
	gas oil				Cat. 2;		
	hydrodesu	ılfurizer,			R46		
	hydrogen						
	10.1						
	sulfide-						
	sulfide-						
	free						
	free Petroleum						
	free Petroleum gas						
	free Petroleum gas [A						
	free Petroleum gas [A complex	ı					
	free Petroleum gas [A complex combination	ı					
	free Petroleum gas [A complex	ı					
	free Petroleum gas [A complex combination of	on					
	free Petroleum gas [A complex combination of hydrocarb	on					
	free Petroleum gas [A complex combination of hydrocarb obtained	on					
	free Petroleum gas [A complex combination of hydrocarb obtained from	on					
	free Petroleum gas [A complex combination of hydrocarb obtained from catalytic	on					
	free Petroleum gas [A complex combination of hydrocarb obtained from catalytic hydrodesu	on	1				
	free Petroleum gas [A complex combination of hydrocarb obtained from catalytic hydrodesu of	on	1				
	free Petroleum gas [A complex combination of hydrocarb obtained from catalytic hydrodesu of vacuum	on	1				
	free Petroleum gas [A complex combination of hydrocarb obtained from catalytic hydrodesu of	on	1				
	free Petroleum gas [A complex combination of hydrocarb obtained from catalytic hydrodesu of vacuum	on	1				
	free Petroleum gas [A complex combination of hydrocarb obtained from catalytic hydrodesu of vacuum gas oil	on	1				

which hydrog sulfide has bee remove by amine treatment of hydroca having carbon number predom in the range of C_1 through C_6 .	nt d inantly arbons s inantly					
649-191-0 6 ses	H K	270-071-2	268409-99			
(petrole catalytic cracked overhead Petrole gas [A comple combin of hydrocaproduct by the distillat of product from the catalytic crackin process It consists of hydrocaproduct from the catalytic crackin process It consists of hydrocaproduct from the catalytic crackin process It consists of hydrocaproduct from the catalytic crackin process It consists of hydrocaproduct from the catalytic crackin process It consists of hydrocaproduct from the catalytic crackin process It consists of hydrocaproduct from the catalytic crackin process It consists of hydrocaproduct from the catalytic crackin process It consists of hydrocaproduct from the catalytic crackin process It consists of hydrocaproduct from the catalytic cracking product from the catalytic cracking product from the catalytic cracking product from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It consists of hydrocaproduct from the catalytic cracking process It can be catal	ds um x ation arbons ed c c g			1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	

	range of C ₃ through C ₅ and boiling in the range of approxim-48°C to 32°C (-54°F to 90°F).]	ately					
649-193-0	OA-Ikanes, C ₁₋₂ Petroleum gas	H K	270-651-	568475-57	•Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-194-0	Akanes, C ₂₋₃ Petroleum gas	H K	270-652-0	068475-58	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-195-0 X	Alkanes, C ₃₋₄ petroleum gas	нк	270-653-0	668475-59	•£arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-196-0	Alk anes, C ₄₋₅ Petroleum gas	H K	270-654-	168475-60	-Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-197-0	gases Petroleum gas [A combinati of light gases. It consists predomin of hydrogen and/ or low molecular weight hydrocarb	on antly	270-667-2	268476-26	-6arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

649-198- OF teel	НК	270-670-9	68476-29	• © arc. Cat	Т		
gases,				1; R45	R: 45-46		
crude				Muta.	S: 53-45		
oil of				Cat. 2;			
distillates	3			R46			
Petroleun	n						
gas							
[A							
complex							
combinat	ion						
of light							
gases							
produced							
by							
distillatio	n						
of crude							
oil							
and by catalytic							
reforming	T						
of	3						
naphtha.							
It							
consists							
of							
hydrogen							
and							
hydrocarl	ons						
having							
carbon							
numbers							
predomin	antly						
in the							
range							
of C ₁							
through							
C ₄ and boiling							
in the							
range of							
approxim	ately						
-217°C							
to -12							
°C(-423							
°F to 10							
^o F).]							
649-199-0 0 1ydrocar	bblnK,	270-681-9	68476-40	- € arc. Cat.	Т		
C_{3-4}				1; R45	R: 45-46		
Petroleun	n			Muta.	S: 53-45		
gas				Cat. 2;			
				R46			
	1	1				L	I

	1	1				
649-200- OH drocar C ₄₋₅ Petroleum gas	1			-Carc. Cat. 1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
649-201- OFF Odrocar C ₂₋₄ , C ₃ - rich Petroleun gas		270-689-2	268476-49	- C arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
649-202-0 Petroleum gases, liquefied Petroleum gas [A complex combinat of hydrocarl produced by the distillatio of crude oil. It consists of hydrocarl having carbon numbers predomin in the range of C ₃ through C ₇ and boiling in the range of approxim -40 °C to 80 °C (-40 °F to 176 °F).]	ion oons n	270-704-2	268476-85	-F+; R12 Carc. Cat. 1; R45 Muta. Cat. 2; R46	F+; T R: 12-45-46 S: 53-45	
649-203-0 Pet roleum gases, liquefied, sweetene		270-705-8	868476-86	- 8 +; R12 Carc. Cat. 1; R45	F+; T R: 12-45-46 S: 45-53	

	i i	1	la.c	ı	1	
Petroleum			Muta.			
gas			Cat. 2;			
[A			R46			
complex						
combinatio	n					
of						
hydrocarbo	ons					
obtained						
by						
subjecting						
liquefied						
petroleum						
gas mix						
to a						
sweetening	5					
process						
to						
convert						
mercaptans	3					
or to						
remove						
acidic						
impurities.						
It						
consists						
of						
hydrocarbo	ons					
having						
carbon						
numbers						
predomina	ntly					
in the						
range						
of C ₃						
through						
C ₇ and						
boiling						
in the						
range of						
approximat	tely					
-40 °C						
to 80 °C						
(-40°Ftol76	5°F) 1					
		0-724-168477-3				
(petroleum	a),		1; R45	R: 45-46		
C_{3-4} ,			Muta.	S: 53-45		
isobutane-			Cat. 2;			
rich			R46			
Petroleum						
gas						

[A complex combination of hydrocarbons from the distillation of			
saturated and unsaturated hydrocarbons			
usually ranging in carbon			
numbers from C ₃ through			
C ₆ , predominantly butane and			
isobutane. It consists of			
saturated and unsaturated			
hydrocarbons having carbon numbers			
in the range of C ₃ through			
C ₄ , predominantly isobutane.]			
649-205-0 Distillates H K (petroleum), C ₃₋₆ , piperylene- rich Petroleum gas	Carc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
[A complex combination of			

hydrocarh from the distillation of saturated and unsaturate aliphatic hydrocarh usually ranging in the carbon numbers C ₃ through C ₆ . It consists of saturated and unsaturate hydrocarh having carbon numbers	n ed eons					
having carbon	antly es.] H K n), s n oons	270-750-3	368477-69	- C arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	
of the butane stream.						

	consists of aliphatic hydrocarb having carbon numbers predoming in the range of C ₃ through C ₄ .]	antly					
649-207-0	(petroleur C ₂ . Petroleum gas [A complex combination of hydrocarb produced by the distillation of products from a catalytic fractionat process. It contains predomination ethane, ethylene, propane, and propylene	on oons n ion antly		968477-70	1; R45 Muta. Cat. 2; R46	R: 45-46 S: 53-45	
649-208-0	(petroleur catalytic-cracked gas oil depropani bottoms, C ₄ -rich acid-free Petroleum gas	zer	270-752-4	468477-71 [,]	€arc. Cat. 1; R45 Muta. Cat. 2; R46	T R: 45-46 S: 53-45	

	[A						
	complex						
	combinati	on					
	of						
	hydrocarb	one					
	obtained	70115					
	from						
	fractionat	ion					
	of						
	catalytic						
	cracked						
	gas oil						
	hydrocarb	on					
	stream	011					
	and						
	treated						
	to						
	remove						
	hydrogen						
	sulfide						
	and						
	other						
	acidic						
	componer	nts					
	It	its.					
	consists						
	of						
	hydrocarb	ons					
	having						
	carbon						
	numbers						
	in the						
	range						
	of \tilde{C}_3						
	through						
	C_5 ,						
	predomin	antle:					
		antiy					
	$C_{4.}$						
649-209-0) ();4 ses	НК	270-754-4	568477-72	-Carc. Cat.	Т	
0.5 205	(petroleur		_, 0 , 0 , 0		1; R45	R: 45-46	
		11),			Muta.	S: 53-45	
	catalytic-					s. 33 - 43	
	cracked				Cat. 2;		
	naphtha				R46		
	debutaniz	er					
	bottoms,						
	C ₃₋₅ -rich						
	Petroleum	ì					
	gas						
	[A						
	complex						
	combinati	on					
		OH					
	of						

hydrocarbons obtained from the stabilization of catalytic cracked naphtha. It consists of aliphatic hydrocarbons having carbon numbers predominantly in the range of C ₃ through C ₅ .]	269-628-268308-08-Varc. Ca		
X (petroleum), isomerized	1; R45 Muta.	R: 45-46 S: 53-45	
naphtha fractionation	Cat. 2; R46		
stabilizer Petroleum			
gas [A			
complex			
combination of			
hydrocarbons			
obtained from the			
fractionation			
stabilization products			
from			
isomerized			
naphtha. It			
consists			
predominantly of			
hydrocarbons			
having			
carbon			
predominantly			

in the range of C ₁ through C ₄ .]							
diesel Gasoil - unspecific [A complex combinat of hydrocart produced by the distillatio of crude oil. It consists of hydrocart having carbon numbers predomin in the range of C ₉ through C ₂₀ and boiling in the range of approxim 163°C to 357°C (325 °F to 675 °F).]	oons oons antly	269-822-	768334-30	-Carc. Cat. 3; R40	R: 40 S: (2-)36/37		
005-009-0@frabutyl	ammoniun enylborate		1120307-0	N;	Xi; N R:		
	-			R50-53	43-50/53 S: (2-)24-37	-56-61	
005-010-0049N- dimethyla tetrakis(p	ınilinium entafluoroj			0@arc.Cat.: R40 Xn; R22	R:	-41	

		Xi; R38-41	S: (2-)22-26	-36/37/39
005-012-0diethyl {4- X	exa-2,5-	14-54R743 N; R50-53	Xi; N R: 43-50/53 S: (2-)24-37	-60-61
011-007-0 propoxycarbazone-sodium		N; R50-53	N R: 50/53 S: 60-61	C ≥ 2,5 %: N; R50/53 0,25 % ≤ C < 2,5 %: N; R51/53 0,025 ≤ C < 0,25 %: R52/53
013-009-0@dium((n- X butyl)x(ethyl)y-1,5- dihydro)aluminate) x = 0.5 y = 1.5	418-720-2	F; R11 R14/15 R 17 Xn; R20 C; R35	F; C R: 11-14/15- S: (1/2-)6-10	17-20-35 5-26-30-36/37/39-43-45
014-026-0di6hloro- (3-(3- chloro-4- fluorophenyl)propy	407-180-3 l)methylsilane	C; R35	C R:35 S: (1/2-)26-3	36/37/39-45
014-027-0 0h0 oro(3- (3- chloro-4- fluorophenyl)propy	410-270-5	C; R35	C R:35 S: (1/2-)8-26	5-28-36/37/39-45
014-028-0 0-6 3-(1- oxoprop-2- eny)l-1- oxypropyl]dimethor ω-[3(1- oxoprop-2- enyl)-1- oxypropyl]dimethor poly(dimetliylsiloxa	xysilyl	R 43	Xi R:43 S: (2-)24-37	
014-029-0 0 ,D'- (ethenylmethylsilylmethylpentan-2- one)oxime]	421-870-1 ene)di[(4-	Repr. Cat. 3; R62 Xn; R22-48/2	Xn R: 22-48/22- S: 2(2-)36/37	

014-030-0	η)-1H- inden-1-	lsilylene)b imethyl]ha	i 40(2-,06 0) 3 (fnium	à.]78a7-390-0	8 10 ; R28	T+ R: 28 S: (1/2-)6-22	2-28-36/37	-45
014-031-0	06i3(1- methyleth dimethox		421-540-7	718230-61	-R10 Xi; R38 R43 R 52-53	Xi R: 10-38-43- S: (2-)24-37		
014-032-0)di8yclope	ntyldimeth	(40)4si3 210e8	3126990-3	5X1; R38-41 N; R50-53	Xi; N R: 38-41-50/ S: (2-)26-37	/53 /3,9-60-61	
015-180-0	(R*,S*)]- [[2- methyl-1- (1- oxopropo (4-	xy)propox yl)phosph	yl-	3137590-3	2Xi; R41 R 43 R 52-53	Xi R: 41-43-52/ S: (2-)24-26		
015-181-0	Op hosphine		232-260-8	37803-51-2	2F+; R12 R17 T+; R26 C; R34 N; R50	F+; T+; N R: 12-17-26- S: (1/2-)28-3	-34-50 36/37-45-6	1-63
015-184-0	glyphosat with the exception of those specified elsewhere in this Annex				N; R51-53	N R: 51/53 S: 61		
015-186-0	049orpyrif methyl	os-	227-011-5	55598-13-()R43 N; R50-53	Xi; N R: 43-50/53 S: (2-)36/37	0,0025	3

		0,00025 % ≤C < 0,0025 %: N; R51-53 0,000025 % ≤C < 0,00025 %: R52-53
o15-187-00-4 mixture of: tetrasodium(((2- hydroxyethyl)i mino)bis(methylen e))bisphosphonate, N-oxide; trisodium ((tetrahydro-2- hydroxy-4H-1,4,2- oxazaphosphorin-4- yl)- methyl)phosphonate N-oxide, P-oxide		R:
015-189-0 pl fenyl bis(2,4,6- trimethylbenzoyl)- phosphine oxide	423-340-\$162881-26 R7 43 R53	
016-086-0 assodium 10- amino-6,13- dichloro-3- (3-(4- (2,5- disulfonatoanilino)- fluoro-1,3,5- triazin-2- ylamino)prop-3- ylamino)-5,12- dioxa-7,14- diazapentacene-4,11 disulfonate		R41 Xi R:41 S: (2-)22-26-39
016-087-04-3 mixture of: thiobis(4,1- phenylene)-	403-490-874227-35- X i; R43 N; R50	R: 36-43-50/53

016-088-1	S,S,S',S'- tetraphenyl bishexafluo diphenyl(4- phenylthion hexafluoron propylene carbonate	orophosph - ohenyl)su phosphate	lfonium	771297-11	-R 52-53	R: 52/53		
010-088-0	(diethylami dimethanes acid	ino)pheny				S: 61		
016-089-6	mixture of esters of 5,5',6,6',7,7 hexahydrox tetramethyl spirobiinda and 2- diazo-1,2- dihydro-1- oxo-5- sulfonaphth	"_ xy-3,3,3',3 -1,1'- n	413-840-3 3'-		E; R2 F; R11 R 53	E R: 2-11-53 S: (2-)33-35	-40-61	
016-090-0 X	00- methyl- N- (methylsulf			14653-91 namide	- % n; R22 Xi; R37-41	Xn R: 22-37-41 S: (2-)26-39		
016-091-0	tert- alkyl ammonium 1- amino-9,10 dihydro-9,1 dioxo-4- (2,4,6- trimethylan anthracen-2 sulfonate)- ()- nilino)-	414-110-5		Xi; R41 N; R50-53	Xi; N R: 41-50/53 S: (2-)26-39	-60-61	
016-093-0	mixture of: 4-(7- hydroxy-2, trimethyl-2 chromanyl) yl-tris(6- diazo-5,6-	4,4- -		1140698-9	6F0 R11 Carc. Cat. 3; R40	F; Xn R: 11-40 S: (2-)7-36/3	3 7	

dihydro-5- oxonaphthalen-1- sulfonate) 4-(7- hydroxy-2,4,4- trimethyl-2- chromanyl)resorcing diazo-5,6- dihydro- 5- oxonaphthalen-1- sulfonate)	olbis(6-					
016-095-0 % -7	417-980-4	1	F; R11	F; Xn		
mixture of:			Carc.Cat.: R40	BR: 11-40 S:		
reaction			K40	(2-)7-36/3	 7	
product				(2), 30,	[
of 4,4'-						
methylenebis[2-						
hydroxybenzyl)-3,6						
dimethylphenol]						
and 6-						
diazo-5,6						
dihydro-5						
oxo- naphthalenesulfona	te					
(1:2)						
Reaction						
product						
of 4,4'- methylenebis						
[2-(4-						
hydroxybenzyl)-3,6	+					
dimethylphenol]						
and 6-						
diazo-5,6- dihydro-5-						
oxo-						
naphthalenesulfona	te					
(1:3)						
016-096-0th2fensulfuron-		79277-27		N		
methyl			R50-53	R: 50/53		
				S: 60-61		
017-015-0023		161807-67		C		
(aminomethyl)phen hydrochloride	yıjacetyich	ioriae	C; R35 R43	R: 22-35-43		
nydrocinoriac			1173	S:		
					6/37/39-4	5
017-016-0m9thyltriphenylpho	-41 0- 400	1021 15 0	37	37 37		
	\$ PH1 (8H4HUM)-2	61031-13-6	SXn;	Xn; N		

017-017-0	00Z4-13- docosenyl- N,N- bis(2- hydroxyeth N- methyl- ammonium chloride	nyl)-	426-210 - €	5120086-5	Xi; R38-41 N; R51-53 8CQ) R34 N; R50-53	C; N R: 34-50/53 S:	41-51/53 -36/37/39- /37/39-45-	
017-018-0 X	004,N,N- trimethyl-2 bis(stearoy chloride		405-660-7		N;R51-53	N R: 51/53 S: 61		
017-019-0	00R5-1,2,3,4 tetrahydro- dimethoxy- veratryliso- hydrochlor	6,7- -1- quinoline	415-110-8	354417-53	-Xn; R22 R52-53	Xn R: 22-52/53 S: (2-)22-61		
017-020-0	Othyl propoxy aluminium chloride		421-790-7	7	C; R35 F; R14/15	C; F R: 14/15-35 S: (1/2-)16-2	23-26-30-3	6/37/39-43-45
017-021-0	Ochenamid dimethyl- (dihydroxy ammonium chloride	propyl)	423-420-1	136920-1	0X0; R41 R43 N; R50-53	Xi; N R: 41-43-50/ S: (2-)26-36	/53 /37/39-60-	61
020-003-0	mixture of: dicalcium (bis(2-hydroxy-5-tetra-propenylph methylami ne)dihydro tri-calcium (tris(2-hydroxy-5-tetra-propenylph ne)tri-hydroxide	nenylmeth xide			Xi; R36/38 R43	Xi R: 36/38-43 S: (2-)24-26	-37	

	1E	1	ſ		1			I
	poly[calci	um						
	((2-	<u> </u>						
	hydroxy-5)-						
	tetra-							
	propenyl-							
		thyl)methy	/lamine)hy	dr				
	oxide]							
024-019-0	0 049 in		419-230-	i	R 43	Xi		
	componer	nt:			R52-53	R:		
	acetoaceti					43-52/53		
	acid					S:		
	anilide/3-					(2-)22-24	-37-61	
	amino-1-					()		
	hydroxyb	enzene						
	(ATAN-							
	MAP):							
	trisodium							
	{6-[(2 or							
	3 or 4)-							
	amino-							
	(4 or 5							
	or 6)-							
	hydroxyp	henylazo]-	5'-					
	(phenylsu	lfamoyl)-3	-					
		naphthalen	e-2-					
	azobenzei	ne-1,2'-						
	diolato}-							
	{6"-[1-							
	(phenylca	rbamoyl)e	thylazo]-5	"'-				
		lfamoyl)-3						
		naphthalen	e-2"-					
	azobenzei	he-1",2"'-						
	diolato]							
	chromate							
	(III)							
	by-							
	product							
	1: acetoaceti							
	acid							
	anilide /							
	acetoaceti							
	acid							
	anilide							
	(ATAN-							
	ATAN):							
	trisodium							
	bis {6-							
	[1-							
		rbamoyl)e	thylazo]-5	_				
	(phenylsu							
	-							
	'			'	'	'		

	sulfonatonaphthaler	ne-2-					
	azobenzene-1,2'- diolatojchromate (III)						
	by- product 2: 3- amino-1- hydroxybenzene /3- amino-1- hydroxybenzene (MAP- MAP): trisodium bis {6- [(2 or 3 or 4)- amino- (4 or 5 or 6)- hydroxyphenylazo]- (phenylsulfamoyl)- sulfonatonaphthaler azobenzene-1,2'- diolato} chromate (III)	-5'- 3-					
024-020-0	Ocidodium bis[(3'- nitro-5'-	418-220-4	1	R43 R52-53	Xi R: 43-52/53		
	sulfonato(6- amino-2- [4-(2-				S: (2-)22-24	-37-61	
	hydroxy-1- naphtylazo)phenyls pyrimidin-5- azo)benzene-2',4- diolato)]chromate(I		no]				
025-005-0	mixture	417-660-4	4	N; R50-53	N R: 50/53		
	of: tri- sodium [29H,31H- phthalocyanine- C,C,C-				S: 60-61		
	trisulfonato (6-)- N29,N30,N31,N32] manganate						
	(3-) tetrasodium [29H,31H-						

	manganat (3-) pentasodi [29H,31H phthalocy C,C,C,C,6 pentasulfo (6-)-	nato N31,N32], e um f- anine- C- onato N31,N32]						
029-012-0	0 9o4 lium		407-340-2	2124719-2	4 Xi ; R41	Xi		
	((N-(3- trimethyla	ammoniopi	opyl)sulfa	moyl)meth	nylsulfonat	R: 41 Sphthalocy (2-)26-39		opper(II)
029-013-0 X	triazin-2- ylamino)- oxido-5- sulfonato	onyl)ethox	y)ethylam benzylider	,	R52-53	Xi R: 41-52/53 S: (2-)24-37	-61	
030-011-0		hosphate)	231-944-3	37779-90-0	N; R50-53	N R: 50/53 S: 60-61		
030-013-0	ป๋ย่ากี่c oxide		215-222-5	51314-13-2	2N; R50-53	N R: 50/53 S: 60-61		
034-003-0) ଡ଼ି ଶାum selenite		233-267-9	910102-18	T; R28 T; R23 R31 R43 N; R51-53	T+; N R: 23-28-31- S: (1/2-)28-3	43-51/53 36/37-45-6	1
053-005-0	methyleth (4- methylph	yl)phenyl) enyl)iodon entafluorop	- ium	3178233-7. ate	2Xh; R21/22-4 N; R50-53	21/22-48/ S:	22-50/53 /37-60-61	

601-056-0	mixture of isomers of: methyldiphenylmetl dimethyldiphenylmet		Xi; R38 N; R50-53	Xi; N R: 38-50/53 S: (2-)37-60	-61
601-057-0 X	dodecyl- [3-(4- dimethylamino)benz propyl]dimethylami tosylate		IXI; R41 R43 N; R50-53	Xi; N R: 41-43-50/ S: (2-)24-26	/53 -37/39-60-61
601-058-0)disL- para- menthene	417-870-6	Xi; R38 R 43 N; R50-53	Xi; N R: 38-43-50/ S: (2-)23-24	-37-60-61
601-059-0	00a0thyl 2- benzylidene-3- oxobutyrate	420-940-915768-07	7-Xi; R36/38 N; R51-53	Xi; N R: 36/38-51/ S: (2-)26-37	
601-060-0	bis[4- fluoro-6- {4- sulfo-5- (2-(4- sulfonaphtalene-3- ylazo)-1- hydroxy-3,6- disulfo-8- aminonaphthalene-7 ylazo)phenylamino} triazin-2ylaminoleth sodium, y- potassium salts x = 7,755 y = 0,245	-1,3,5-	9R143	Xi R:43 S: (2-)22-24	-37
601-061-0	Octhyl-1,2- ethanediyl) [-2- [[[(2- hydroxyethyl)methy propyl]ω- (nonylphenoxy)poly		C; R34 R 43 N; R51-53	C; N R: 34-43-51/ S: (1/2-)26-2	/53 28-36/37/39-45-61

(methyl-lethanediy							
601-062-00-7 mixture of: branched triacontar branched dotriacon branched tetratriaco branched hexatriac	tane	417-030-9	0151006-5	9R653	R: 53 S: 61		
601-063-00-2 mixture of isomers of branched tetracosar	ne		2151006-6	R53	Xn R: 20-53 S: (2-)61		
601-064-0 6 r 8 nched hexatriac		417-070-7	7151006-6	2R153	R: 53 S: 61		
dioxane-3 norcarano (1'a,3'β,6	hylspiro(1, 5,2'- 2) α)-2,2,3',7 hylspiro(1,2'-	',7'-		Xn; R48/22 Xi; R41 N; R51-53	Xn; N R: 41-48/22- S: (2-)22-26	51/53 -37/39-61	
601-066-0 0-0 4- (trans-4- heptylcyc	lohexyl)ph		278531-60 ne	- R 43 R53	Xi R: 43-53 S: (2-)24-37	-61	
601-067-0 tri thyl arsenate		427-700-2	215606-95	-&arc. Cat. 1; R45 T; R23/25 N; R50-53	T; N R: 45-23/25- S: 53-45-60-		
601-068-0 0 -2- X diacetoxy ene	but-3-	421-720-5	518085-02	- X n; R22	Xn R: 22 S: (2-)		

601-069-0 Q -5 ethyl-1 (2-(1,3) dioxan pyridin bromid	- yl)ethyl)- iium	422-680-		R52-53	R: 52/53 S: 61		
601-071-0 0- 6 dimeth nitrobe	oxymethyl-2- nzene		20627-73	- R 43 N; R51-53	Xi; N R: 43-51/53 S: (2-)24-37	-61	
601-073-0 0- 7 bromodifluor	-3,5- obenzene	416-710-2	2461-96-1	R10 Xn; R22-48/22 Xi; R38 R43 N; R50-53	S:	-43-48/22- /37-60-61	50/53
en-1- yl)-1- methyl oxabic: 1-(2,2,- trimeth en-1- yl)-5- methyl oxabic: spiro[c en-1-yl [(4,5,6,- tetrahy tetrame [2H]cy spiro[c en-1-yl [4,5,6,- tetrahy tetrame [2H]cy	ylcyclopent- -2- yclo[2.2.2]oc 3- yclo[3.2.1]oc yclohex-3- -6- dro-3,6',6',6'a ethyl)-1,3'(3'a clopenta[b]ft yclohex-3- -6- dro-4,6',6',6'a ethyl)-1,3'(3'a clopenta[b]]ft	tane 3- tane tane H- H)- tran]		Xi; R36/38 N; R51-53	Xi; N R: 36/38-51/ S: (2-)26-37		
<i>p</i> -	E lorotoluene penzotrichlor		15216-25-1	Carc. Cat. 2; R45 Repr. Cat. 3; R62	R:	37/38-48/2	23-62

					T; R48/23 Xn; R21/22 Xi; R37/38			
602-094-0	di∌ henyle octabrom derivate		251-087-9	32536-52	-Repr. Cat. 2; R61 Repr. Cat. 3; R62	T R: 61-62 S: 53-45		
602-096-0	hálachite green hydrochlo [1] malachite green oxalate [2]	oride	[1]	3569-64-2 [1] 718015-76 [2]	Xi; R41	Xn; N R: 22-41-63- S: (2-)26-36	-50/53 /37-39-46-	60-61
602-097-0	bromo-9- (4,4,5,5,5	- ropentylthi		148757-8	9R\$13 N; R50-53	Xi; N R: 43-50/53 S: (2-)24-37	-60-61	
603-167-0	00,3',5,5'- tetra- tert- butylbiph diol	enyl-2,2'-	407-920-5	6390-69-8	BR 53	R: 53 S: 61		
603-168-0		loxy)propa		270445-33	- % i; R41 R 52-53	Xi R: 41-52/53 S: (2-)26-39	-61	
603-169-0	0(+4-)- trans-4- (4- fluorophe hydroxym N- methylpip	ethyl-	415-550-0	0109887-5	3Xh; R22 Xi; R41 N; R51-53	Xn; N R: 22-41-51/ S: (2-)22-26		
603-170-0 X	mixture of: 2- methyl-1- (6-	yclo[2.2.1		867739-11	-Ki; R36 N; R51-53	Xi; N R: 36-51/53 S: (2-)26-61		

	en-2- yl)pent-1- en-3-ol 2- methyl-1- (1- methylbicycloL2.2.1 en-2-yl)- pent-1- en-3-ol 2- methyl-1- (5- methylbicyclo[2.2.1 en-2- yl)pent-1- en-3-ol						
603-171-0	6- 5 thiazolylmethanol	414-780-9	938585-74	- % i; R41 R 52-53	Xi R: 41-52/53 S: (2-)26-39	-61	
603-172-0	00000000000000000000000000000000000000	415-180-1		Xn; R22 Xi; R41 N; R51-53	Xn; N R: 22-41-51/ S: (2-)22-26		
603-173-0	dimethyl-3,5,8- trioxabicyclo[5.1.0] octane	421-750-9	957280-22	- X i; R36 R 43	Xi R: 36-43 S: (2-)26-36	/37	
603-174-0	04-1 cyclohexyl-2- methyl-2- butanol	420-630-3	883926-73	- X i; R41 N; R51-53	Xi; N R: 41-51/53 S: (2-)26-39	-61	
603-175-0	hexyloxyethoxy)eth: DEGHE diethylene glycol monohexyl ether 3,6- dioxa-1- dodecanol		3112-59-4	Xn; R21 Xi; R41	Xn R: 21-41 S: (2-)26-36	/37-46	

603-176-0	hexyl carbitol 3,6- dioxadodo ol	ecan-1-	203-977-3	3112-49-2	R19	T		
	bis(2-				Repr. Cat.2; R61 Repr. Cat.3; R62	R: 61-19-62 S: 53-45		
603-177-(ethoxypro ol 2PG1EE 1- ethoxy-2- propanol propylene glycol monoethy ether [1] 2- ethoxy-1- methyleth acetate 2PG1EEA [2]	yl	[1]	51569-02-4 [1] 954839-24 [2]	R67	R: 10-67 S: (2-)24		
603-178-0	02-3 hexyloxydethylene glycol monohex ether n- hexylglyc	yl	203-951-1	1112-25-4	Xn R21/22 C; R34	C R: 21/22-34 S: (1/2-)26-3	36/37/39-4.	5
603-179-0	0erg ocalcit Vitamin D2	≘00 -014-9	950-14-6	T+; R26 T; R24/25-4	T+ R: 8 22 \$25-26- S: (1/2-)28-3			
603-180-0	00elecalcif Vitamin D3	erol	200-673-2	267-97-0	T+; R26 T; R24/25-4	T+ R: 8 22 \$25-26- S: (1/2-)28-3		

(02.101.00	216 652 1621 01	ID D44	D 77		
603-181-00ert- X butyl methyl ether MTBE 2- methoxy-2- methylpropane	216-653-11634-04-	4F; R11 Xi; R38	F; Xi R: 11-38 S: (2-)9-16-2	24	
603-183-0 Q-Q 2-(2-butoxyethoxy)ethor TEGBE triethylene glycol monobutyl ether butoxytriethylene glycol	205-592-6143-22-6 exy]ethanol	Xi; R41	Xi R: 41 S: (2-)26-39	C≥30 %:Xi; R41 - 26 % ≤ C < 30 %: Xi; R36	
603-184-0 Q -6 (hydroxymethyl)-2 [[2- hydroxy-3- (isooctadecyloxy)p	416-380-1146925-8 propoxy]methyl]-1,3-	R50-53	N R: 50/53 S: 60-61		
603-185-0 Q,4 - dichloro-3- ethyl-6- nitrophenol	420-740-199817-36	5-4; R25 Xi; R41 R43 N; R50-53	T; N R: 25-41-43- S: (1/2-)26-3	-50/53 36/37/39-4	5-60-61
603-186-0 t -áns- (5RS,6SR)-6- amino-2,2- dimethyl-1,3- dioxepan-5- ol	419-050-379944-37	7- R 43	Xi R: 43 S: (2-)22-24	/25-26-37	
603-187-0 Q-Q (4,6-bis(4-(2-(1-methylpyridinium-yl)vinyl)phenylam triazin-2-yl)(2-hydroxyethyl)amir dichloride	ino)-1,3j5-	77NG R50-53	N R: 50/53 S: 60-61		
603-189-00-3 mixture of complexes of:	405-250-8	N; R51-53	N R: 51/53 S: 61		

	titanium, 2,2'- oxydietha ammonium lactate, nitrilotris(propanol) and ethylene glycol	n						
603-191-0	bis(2,4-dimethylp triazin-2-yl)-5- (3-((2-ethylhexy hydroxypi	l)oxy)-2- ropoxy)pho	,5- enol	\$137658-7 ⁻		R: 53 S: 61		
603-195-0	02-64-(4- methoxyp phenyl-1,3 triazin-2- yl]- phenol		430-810-3	3154825-6	2R452-53	R: 52/53 S: 61		
603-196-0	0 2-(7- ethyl-1H- indol-3- yl)ethanol		431-020-1	41340-36	-Xn; 22-48/22 N; R51-53	Xn; N R: 22-48/22- S: (2-)36/37/		
603-197-0	00-74- chlorophe dimethyl- (1,2,4- triazol-1- ylmethyl) ol	3-	403-640-2	2107534-9	6Repr.Cat.: R63 Xn; R22 N; R51-53	3¾n; N R: 22-51/53- S: (2-)22-36		
603-199-0	0€ŧ8 xazol			153233-9	INI; R50-53	N R: 50/53 S: 60-61	C≥0.25 %: N; R50/53 0.025 % ≤ C < 0.25 %:N; R51/53 0.0025 % ≤ C < 0.025 %: R52/53	

	, ,				
604-065-	≬ 0, 4',4"-	407-460-5111850)-25 N 0,	N	
	(1-		R51-53	R: 51/53	
	methylpropan-1-			S: 61	
	yl-3-				
	ylidene)tris(2-				
	cyclohexyl-5-				
	methylphenol)				
604-066-	0 4 -7	414-550-8	N;	N	
001 000	mixture	111 330 0	R50-53	R: 50/53	
	of:		1630 33	S: 60-61	
	phenol,			5. 00 01	
	6-(1,1-				
	dimethylethyl)-4-				
	tetrapropyl-2-				
	[(2-				
	hydroxy-5-				
	tetra-				
	propylphenyl)methy	<i>i</i> 1			
	(C41-				
	compound)				
	and				
	methane,				
	2,2'-				
	bis[6-				
	(1,1-				
	dimethyl-				
	ethyl)-1-				
	hydroxy-4-				
	tetrapropyl-				
	phenyl)]-				
	(C45-				
	compound)				
	2,6-				
	bis(1,1-				
	dimethylethyl)-4-				
	tetra-				
	propyl-				
	phenol				
	and 2-				
	(1,1-				
	dimethylethyl)-4-				
	tetrapropyl-				
	phenol				
	2,6-				
	bis[(6-				
	(1,1-				
	dimethylethyl)-1-				
	hydroxy-4-				
	tetrapropylphenyl)n	netnyij-4-			
	(tetrapropyl)phenol				
	and 2-				
	[(6-(1,1-				

hydr tetra [1- hydr tetra	ethylethyl)-1- oxy-4- propylphenylm oxy-4- propylphenyl)n propyl)phenol						
dode form oligo with dode phen and 2 form oligo with dode phen and 2 amin amin are 2)	cylphenol] aldehyde, omer 4- cyl ooethanol(n aldehyde, omer 4- cyl oothanol(n	414-520-4		Xi; R38-41 N; R50-53	Xi; N R: 38-41-50/ S: (2-)26-37/		
meth 1]am hydr	-4-		599095-19	- % n; R20/22 R 43	Xn R: 20/22-43 S: (2-)24-26	-37	
tert-	ylpropyl)-4-	421-740-4	451390-14	- © ; R34 N; R51-53	C; N R: 34-51/53 S: (1/2-)26-3	36/37/39-4	5-61
604-070-00-i@lo 2,4,4 trich hydr dipho ether	'- loro-2'- oxy- enyl-	222-182-2	23380-34-:	5Xi; R36/38 N; R50-53	Xi; N R: 36/38-50/ S: 26-39-46-	R36/38-5	0/53

5- chloro-2- (2,4- dichlorophen	noxy)phenol		%: N; R50/53 0,025 % ≤ C < 0,25 %:N; R51/53 0,0025 % ≤ C < 0,025 %: R52/53
mixture of: 2,2- dimethoxyetl (this component is considered to be anhydrous in terms of identity, structure and composition. However, 2,2- dimethoxyetl will exist in a hydrated form. 60% anhydrous is equivalent to 70.4% hydrate) water(Includ free water and water in hydrated 2,2- dimethoxyetl	hanal	R43 Xi R: 4: S: (2-)2	3 24-37

		1	Υ	r	1		
606-062-0	the the desired			R61 Xi; R41 R 52-53	R: 61-41-52/ S: 53-45-61	753	
606-063-0	OE6-3- (2- chlorophenyl)-2- (4- fluorophenyl)proper		5112704-5	1 X 5; R36 R 43	Xi R: 36-43 S: (2-)24-26	-37	
606-064-0	Opregn-5- ene-3,20- dione bis(ethylene ketal)	407-450-0	7093-55-2	2R 53	R: 53 S: 61		
606-065-0	0 0- 74- morpholinophenyl)t one	413-790-0 outan-1-)	N; R51-53	N R: 51/53 S: 61		
606-066-0	0F2-5[(4- chlorophenyl)methy dimethylcyclopenta	lene]-2,2-	131984-2	1N9, R51-53	N R: 51/53 S:61		
606-067-0	mixture of: 1- (2,3,6,7,8,9- hexahydro-1,1- dimethyl-1H- benz(g)inden-4- yl)ethanone 1- (2,3,5,6,7,8- hexahydro-1,1- dimethyl-1H- benz(f)inden-4- yl)ethanone 1- (2,3,6,7,8,9- hexahydro-1,1- dimethyl-1H- benz(g)inden-5- yl)ethanone 1- (2,3,6,7,8,9- hexahydro-3,3- dimethyl-1H- benz(g)inden-5- yl)ethanone		896792-67	R50-53	N R: 50/53 S: 60-61		
606-068-0	Q - 3 ,11- trimethyl-13-	415-770-7	71638-05-1	7Xn; R48/22	Xn		

606-069-0	(2,6,6- trimethylcyclohex-1 en-1- yl)tridecahexaen-2,4 al Oppro[1,3- dioxolane-2,5'- (4',4',8',8'- tetramethyl-		154171-7	R 52-53	R: 43-48/22- S: (2-)22-36 N R: 51/53 S: 24-61	
	hexahydro-3',9'- methanonaphthalen)]				
606-070-(butyryl-2,4,6- trimethylphenyl)-2- [1- (ethoxyimino)propy hydroxycyclohex-2- en-1-one	414-790-3 l]-3-	138164-1	R62-63	8Xn; N R: 22-38-62- S: (2-)22-36	
606-071-0 X	spiro(5,5- dimethyl-1,3- dioxan-2- yl)androsta-1,4- diene-3- one	421-050-3	13258-43	-N; R50-53	N R: 50/53 S: 22-60-61	
606-072-0	00-5 acetyl-1- phenyl- pyrrolidine-2,4- dione	421-600-2	719-86-8	Xn; R48/22 N; R51-53	Xn; N R: 48/22-51/ S: (2-)22-36	
606-073-0	0 0,0'- bis(dimethylamino) Michler's ketone	202-027-5 benzopheno		Carc.Cat.2 R45 Muta.Cat. R68 Xi; R41	R:	
606-075-0	0-1 benzyl-5- ethoxyimidazolidino dione	417-340-4	65855-02	- X n; R22	Xn R: 22 S: (2-)22	
606-076-0	00-7(2- quinolinyl- carbonyl)oxy)-2,5- pyrrolidinedione	418-630-3	136465-9	9Xi; R41 R43	Xi R: 41-43 S: (2-)24-26	-37/39
606-077-0	hexyl-4- [(R)-2- hydroxytridecyl]-2- oxetanone	418-650-2	104872-0	6AQ R50-53	N R: 50/53 S: 60-61	

606-078-0	00-8 octylazep one	in-2-	420-040-0	59227-88	-©; R34 R 43 N; R51-53	C; N R: 34-43-51/ S: (1/2-)26-3	53 86/37/39-45	5-61
606-079-0	butyl-	sothiazol-3	420-590-	7	C; R34 R43 N; R50-53	C; N R: 34-43-50/ S: (1/2-)26-3	53 86/37/39-45	5-60-61
606-080-0	Product of: 3- hydroxy-: di-tert-butylbenz one with o-xylene		417-100-9		R 53	R: 53 S: 61		
606-081-0	0(3\$, 5α, 6β)-3- (acetyloxy bromo-6- hydroxy- androstan one		419-790-7	74229-69-()R43 R52-53	Xi R: 43-52/53 S: (2-)22-36	/37-61	
606-082-0 X-	mixture of: butan-2- one oxime syn- O,O'- di(butan-2)	2- ethoxysilan	406-930- ²	796-29-7	T; R48/22 R43 R52-53	T R: 43-48/25- S: (1/2-)25-3		
606-083-0	chloro-5- sec-	lhydroquin	407-750-3		Xi; R36/38 R43 R52-53	Xi R: 36/38-43- S: (2-)24-26		
606-084-0	nethoxy- benzofura phenyl-1, propaned	inyl)-3- 3-	414-540-3	3484-33-3	N; R50-53	N R: 50/53 S: 60-61		
606-085-0	001 R ,4S)-2 azabicycl		418-530-	79200-56	- X n; R22 Xi; R41 R43	Xn R: 22-41-43		

	hept-5- en-3-one					S: (2-)24-26	-37/39	
606-086-0	00-(3,3- dimethyle en-1-one	yclohexyl)		356973-87	-M; R51-53	N R: 51/53 S: 61		
606-087-0	06-7 ethyl-5- fluoro-4(3 pyrimidon		422-460-5	5137234-8	7Xh; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)60-61		
606-088-0	02-2,4,7- tetramethy octen-3- one	yl-6-	422-520-0)74338-72	- % i; R38 N; R51-53	Xi; N R: 38-51/53 S: (2-)37-61		
606-089-0	mixture of: 1,4- diamino-2 chloro-3- phenoxyar 1,4- diamino-2 bis- phenoxyar	nthraquino	ne	212223-77	- R 53	R: 53 S: 61		
606-091-0	06-9 chloro-5- (2- chloroethy dihydroind one		421-320-0)118289-5.	5N; R50-53	N R: 50/53 S: 60-61		
606-092-0	mixture of: (E)- oxacycloh en-2-one (E)- oxacycloh en-2-one a) (Z)- oxacycloh (12)- en-2- one and b) (Z)- oxacycloh (13)- en-2-one	exadec-13 exadec-	-	3111879-80	0№2 R50-53	N R: 50/53 S: 60-61		

stearate sodium [bis(2- (stearoylor methylsulf sodium [bis(2- hydroxyet N,N- bis(2-	hyl)stearamido]ethy xy)ethyl]amino]	lfonate	Xi;	R: 52/53 S: 61		
mixture of: ammonium bis(hexylo ammonium hexyloxyc octyloxyca ammonium hexyloxyc	n-1,2- exycarbonyl)ethanesi n-1- arbonyl-2- arbonylethanesirlfon n-2-	ulfonate	R38-41 R 52-53	R: 38-41-52/ S: (2-)26-37		
607-381-00 A Red triesters of 2,2-bis(hydrox with C7-alkanoic acids and 2-ethylhexar acid	413-710-4 xymethyl)butanol noic	4	R 53	R:53 S: 61		
607-382-0 2 -8(4-amino-2-nitropheny acid	411-260-3 v1)amino)benzoic	3117907-4	3Xi; R41 R 43 R 52-53	Xi R: 41-43-52/ S: (2-)24-26		
607-383-00-9 mixture of: 2,2,6,6- tetramethy yl- hexadecan	dpiperidin-4-	886403-32-	- % i; R41 R 43 N; R50-53	Xi; N R: 41-43-50/ S: (2-)24-26	53 -37/39-60-	61

2,2,6,6- tetrameth yl- octadecar	ylpiperidin oate	-4-				
branched and linear alkyl 3,5-bis(1,1-dimethyle	ethyl)-4- enzeneproj	panoate	2171090-9	3R053	R: 53 S: 61	
methylpy methylsu	enyl)ethenyridinium		5125229-7	4№5 R51-53	N R: 51/53 S: 61	
607-386-0 % 5 mixture		412-580-6	5174591-5	1 X 6; R38-41	Xi; N	

	of: tetradecar acid (42.5-47.5 poly(1-7) esters of tetradecar acid (52.5-57.5	5%) lactate			R 43 N; R50-53	R: 38-41-43- S: (2-)24-26	-50/53 -37/39-60-	61
607-387-0	mixture of: dodecano acid (35-40%) poly(1-7)! esters of dodecano acid (60-65%)	actate	412-590-0)58856-63	-&i R38-41 R 43 N; R50-53	Xi; N R: 38-41-43- S: (2-)24-26	-50/53 -37/39-60-	61
607-388-0	04-6 ethylamin nitrobenzo acid		412-090-2	22788-74-1	Xn; R22 R 43 R 52-53	Xn R: 22-43-52/ S: (2-)22-24		
607-389-0	Orisodium N,N- bis(carbox amino-2- hydroxyp	xymethyl)- ropionate		1119710-90	5X2n; R22	Xn R: 22 S: (2-)22		
607-390-0	00-2,3,4- tetrahydro nitro- quinoxalii		414-270-6	541959-35	-Xn; R22 N; R51-53	Xn; N R: 22-51/53 S: (2-)22-61		
607-391-0	0di2nethylo dicarboxy		n 4 14,-1240-2	26914-71-2	2R 52-53	R: 52/53 S: 61		
607-392-0	phenoxye 4-((5- cyano-1,6 dihydro-2 hydroxy-1 dimethyl- oxo-3- pyridinyl)	- - !,4-	ate	88938-37		R: 53 S: 61		
607-393-0	00-&cis-1- propenyl) amino-8-	-7-	415-750-8	3106447-4	4R343	Xi R: 43		

607-394-0	oxo-5- thia-1- azabicyclo[4.2.0]oct ene-2- carboxylic acid 06-9 methylpyrazine-2- carboxylic acid		95521-55-1	lXi; R41	S: (2-)22-24 Xi R: 41 S: (2-)26-39		
607-395-0	mixture of: sodium 1- tridecyl-4- allyl-(2 or 3)- sulfobutanedioate sodium 1- dodecyl-4allyl- (2 or 3)- sulfobutanedioate	410-230-	7	C; R34 R 43 N; R51-53	C; N R: 34-43-51/ S: (1/2-)26-3	53 36/37/39-4	5-61
607-396-0 X	Obis (1,2,2,6,6- pentamethyl-4- piperidinyl) 2-(4- methoxybenzylidene		4147783-6	9A5 R50-53	N R:50/53 S: 22-60-61		
607-397-0	mixture of: Ca salicylates (branched C 10-14 and C18-30 alkylated) Ca phenates (branched C10-14 and C18-30 alkylated) Ca ca phenates (branched C10-14 and C18-30 alkylated) Ca sulfurized phenates (branched C10-14 and	415-930-6	5	R 43	Xi R: 43 S: (2-)36/37		

	C18-30							
	alkylated)							
607-398-0	N-(5- chloro-3- (4- (diethylar methylpho methyl-6- oxo-1,4-	enylimino)	-4-	5125630-9	1Ng R50-53	N R: 50/53 S: 60-61		
607-399-0	02-8- dimethyl 3- methyl-3- butenyl propanoat	e	415-610-6	5104468-2	1 X 5; R38 R52-53	Xi R: 38-52/53 S: (2-)37-61		
607-400-0 X	3-	amino)thic		132750-89 thio]propa	R50-53	N R: 50/53 S: 60-61		
607-401-0	hydroxy-5 oxo-3- cyclohexe carboxyla	ene-1-	414-450-4	188805-65	-&i R38-41 R 43	Xi R: 38-41-43 S: (2-)24-26	-37/39	
607-402-0	Mathyl N- (phenoxyo L- valinate	carbonyl)-	414-500-5	5153441-7	7R152-53	R: 52/53 S: 61		
607-403-0	mixture of: bis(1S,2S (1- benzyl-4- tert- butoxycar hydroxy-5	boxamido)	Xn; R48/22 Xi; R41 N; R50-53	Xn; N R: 41-48/22- S: (2-)22-26	50/53 -36/39-60-	61
607-404-0	mixture of: ((Z)-3,7- dimethyl-	2,6-	415-190-4	1	R 43	Xi R: 43 S: (2-)24-37		

	octadienyl)oxycarbacid di- ((E)-3,7- dimethyl-2,6- octadienyl) butandioate di- ((Z)-3,7- dimethyl-2,6- octadienyl) butandioate (Z)-3,7- dimethyl-2,6- octadienyl butandioate ((E)-3,7- dimethyl-2,6- octadienyl butandioate ((E)-3,7- dimethyl-2,6- octadienyl)oxycarbacid						
607-405-0	0 2- 7 hexyldecyl p- hydroxybenzoate	415-380-7	148348-1	2N3 R51-53	N R: 51/53 S: 61		
607-406-0	θ assium 2,5- dichlorobenzoate	415-700-5		Xn; R22 Xi; R41	Xn R: 22-41 S: (2-)26-39		
607-407-0	OttByl 2- carboxy-3- (2- thienyl)propionate	415-680-8	143468-9	6 X 6; R38-41 R 43	Xi R: 38-41-43 S: (2-)24-26	-37/39	
607-408-0	ဖုံ eitassium N-(4- fluorophenyl)glyci	415-710-1 nate		Xn; R48/22 Xi; R41 R 43 R 52-53	Xn R: 41-43-48/ S: (2-)22-26	22-52/53	61
607-409-0	mixture of: (3R)- [1S- (1α,2α,6β- ((2S)-2- methyl-1- oxo- butoxy)-8a.gamma dimethyl-1- naphthalene]-3,5-	415-840-7	-2,6-	R 43 R 52-53	Xi R: 43-52/53 S: (2-)36/37	-61	

	dihydroxy acid inert biomass from Aspergillu terreus	rheptanoic						
607-410-0	(dimethyl (hexadec- enyl)butar and/or mono[2-	nedioate amino)ethy 2-		drogen-2-	Xi; R38-41 R 43 N; R50-53	Xi; N R: 38-41-43- S: (2-)24-26	-50/53 -37/39-60-	61
607-411-0 X	0xiranemod4- methylber sulfonate, (S)		417-210-7	770987-78	R45	2Ţ; N R: 3ŧ5-41-43- S: 53-45-61	51/53	
607-412-0	2-(1-	ohexyl)aco		1133481-1	0X4n; R22-48/22 R 52-53	Xn 2R: 22-48/22- S: (2-)36/37		
607-413-0	Otans-4- phenyl- L- proline		416-020-1	196314-26	- R epr.Cat R62 R 43	3Xn R: 43-62 S: (2-)22-36	/37	
607-414-0	ethylhexy (1,3,5- triazine-2			188122-99	-R 53	R: 53 S: 61		
607-415-0	(methyl methacryl co- (butylmet co-(4-acryloxyb	hacrylate)- utyl- ylalpha.,, enzyl			F; R11 R 43	F; Xi R: 11-43 S: (2-)24-37	-43	

co- (maleicanhydride)					
607-416-0 4-7 2- carboxymethylthio) hydroxy-5- isobutyloxycarbony N-(3- dodecyloxypropyl)- naphthamide	lamino-		N; R50-53	N R: 50/53 S: 60-61	
607-418-0 Q -8 ethylhexyl 4- aminobenzoate	420-170-32	26218-04-	·N; R50-53	N R: 50/53 S: 60-61	
607-419-0@33 carboxymethyl-5- (2-(3- ethyl-3H- benzothiazol-2- ylidene)-1- methyl- ethylidene)-4,4'- dioxo-2'- thioxo- (2,5')bithiazolidinyl yl)- acetic acid	422-240-91	166596-6	3 X 5; R41 R 43	Xi R: 41-43 S: (2-)26-36	/37/39
607-420-0 0₃9 - bis(hydroxymethyll acid	424-090-11 butanoic	10097-02-	Æi; R41 R52-53	Xi R: 41-52/53 S: (2-)26-39	-61
607-421-00ypermethrin cis/ trans +/- 40/60 (RS)-α- cyano-3- phenoxybenzyl (1RS,3RS;1RS,3SR) (2,2- dichlorovinyl)-2,2- dimethylcyclopropa			&n R20/22 Xi; R37 N; R50-53	Xn; N R: 20/22-37- S: (2-)24-36	50/53 /37/39-60-61
607-422-00- X cypermethrin	257-842-96	67375-30	8; R25 Xn; R48/22 Xi; R37 N; R50-53	T; N R: 25-37-48/ S: (2-)36/37/	22-50/53 /39-45-60-61

607-423-00sters of mecoprop and of mecoprop-P			Xn; R22 R43 N; R50-53	Xn; N R: 22-43-50/ S: (2-)13-36		
(SO) (E,E) - α - methoxyimino $\{2$ - $[[[[1-[3-(trifluoromethylelamino]oxy] methyl]benzer acid methyl ester$	- yl)phenyl]ethyli	141517-2 den	IR743 N; R50-53	Xi; N R: 43-50/53 S: (2-)24-37	-46-60-61	
607-425-0ta6talaxyl (ISO) methyl-N- (2,6- dimethylpheny (methoxyacety DL- alaninate	v1)- <i>N</i> -	757837-19	-Kn; R22 R43 R52-53	Xn R: 22-43-52/ S: (2-)13-24	-37-46-61	
607-426-00,2- benzenedicarb acid, dipentylester, branched and linear [1] n- pentyl- isopentylphtha [2] di-n- pentyl phthalate [3] diisopentylpht [4]	oxylic [1] -[2] 205-017- [3] 210-088- [4]	284777-06 [1] -[2] 9131-18-0 [3] 4605-50-5 [4]	- R epr.Cat.: R60-61 N; R50	2Ţ; N R: 60-61-50 S: 53-45-61		
607-427-06+8moxynil heptanoate (ISO) 2,6- dibromo-4- cyanophenyl heptanoate	260-300-	456634-95	Repr.Cat3 R63 Xn; R20/22 R43 N; R50-53	R: 20/22-43- S:	63-50/53 -46-60-61	

607-430- 0BB P benzyl butyl phtalate		201-622-7	785-68-7	Repr.Cat R61 Repr.Cat R62 N; R50-53	2Ț; N R: 361-62-50/ S: 53-45-60-	
607-431-0 prallethrin ETOC 2- methyl-4- oxo-3- (prop-2- ynyl)cyclo en-1- yl 2,2- dimethyl- (2- methylpro enyl)cyclo	opent-2-		923031-36	-¶; R23 Xn; R22 N; R50-53	T; N R: 22-23-50/ S: (1/2-)45-6	
metolachl mixture of (S)-2-chloro-N-ethyl-6-methyl-phenyl)-N (2-methoxy-metolachl (R)-2-chloro-N-(2-ethyl-6-methyl-phenyl)-N (2-methoxy-methyl-ethyl)-acetamide (2-methoxy-methyl-ethyl)-acetamide (0-20%) [2]	or 2- 1- 3- 1- 1- 1-	-[1] -[2]	87392-12 [1] 178961-2 [2]	N;	Xi; N R: 43-50/53 S: (2-)24-37	-60-61
607-433-00ypermetl	nrin	257-842-9	952315-07	- % n; R22	Xn; N	

trans +/- 80/20 (RS)-α- cyano-3- phenoxybenzyl (1RS; 3RS; 1RS, 3SR)-3- (2,2- dichlorovinyl)-2,2- dimethylcyclopropa	necarboxylate	Xi; R37/38 R43 N; R50-53	R: 22-37/38-43-50/53 S: (2-)36/37/39-60-61
607-434-0th6coprop-P[1] and its salts (R)-2-(4-chloro-2-methylphenoxy)propacid	240-539-016484-77	- % n; R22 Xi; R41 N; R51-53	Xn; N R: 22-41-51/53 S: (2-)13-26-37/39-46-61
isopropyl-5R-methyl-1R-cyclohexyl 2,2-dihydroxyacetate	416-810-6111969-6	4X3n; R48/22 Xi; R41 N; R51-53	Xn; N R: 41-48/22-51/53 S: (2-)22-26-36/39-61
607-436-0 2 -6 hydroxy-3- (2- ethyl-4- methylimidazoyl)pro neodecanoate	417-350-9 opyl	Xi; R38-41 N; R50-53	Xi; N R: 38-41-50/53 S: (2-)26-28-37/39-60-61
607-437-0 0-(4- aminophenyl)-2- cyano-2- propenoic acid	417-480-6	R43	Xi R: 43 S: (2-)22-24-37
607-438-0mathyl-2- [(aminosulfonyl)me	419-010-5 thyl]benzoate	Xn; R22 Xi; R36	Xn R: 22-36 S: (2-)22-26
607-439-00n2thyl tetrahydro-2-furancarboxylate	420-670-137443-42	- % i; R41	Xi R: 41 S: (2-)26-39
607-440-0018thyl 2- aminosulfonyl-6-	421-220-7144740-5	9R043 N; R51-53	Xi; N R: 43-51/53

(trifluoromethyl)py carboxylate	ridine-3-	S: (2-)22-24-37-61
607-441-00-\(\beta \)3-(2- dodecyloxy-5- methylphenylcarba hydroxy-1- naphthylthiojpropio acid		R: 53 S: 57-61
607-442-0 be nzyl [hydroxy- (4- phenylbutyl)phospl acetate	416-050-587460-09- K i; R41	Xi R: 41 S: (2-)26-36/39
607-443-0 bis (2,4-di-tert-butyl-6-methylphenyl)ethyl	416-140-4145650-60R853	R: 53 S: 61
607-444-00. X mixture of: cis-1,4- dimethylcyclohexy dibenzoate trans-1,4- dimethylcyclohexy dibenzoate		R: 53 S: 61
607-445-0 0 -6n (III) tris(4-methylbenzenesulf	420-960-877214-82- X i; R41	Xi R: 41 S: (2-)24-26-39
607-446-0@thyl 2-[4-(2- chloro-4- nitrophenylazo)-3- (1- oxopropyl)amino]p	416-240-8155522-12R643 R 53	Xi R: 43-53 S: (2-)22-24-37-61
607-447-06edium 4-[4-(4- hydroxyphenylazo) nitrobenzenesulfon		Xi R: 43-52/53 S: (2-)22-24-37-61
607-448-0 Q , 3 ,5,6-tetrafluorobenzoic acid	416-800-1652-18-6 Xi; R38-41	Xi R: 38-41 S: (2-)22-26-37/39
607-449-0 % -7 mixture	417-080-1 E; R2 R43	E; Xi; N

	of:				N;	R:		
	4,4',4"-				R50-53	2-43-50/5	3	
	[(2,4,6-					S:		
		,5(2H,4H,	6H)-			(2-)24-35	-37-60-61	
	triazine-1							
	triyl)tris[r	nethylene(3,5,5-					
	trimethyl-	3,1-						
	cyclohexa	nediyl)imi	nocarbony	loxy-2,1-				
		l(ethyl)am			niumtri[bi	s(2-		
		pyl)naphtl			_			
	4,4',4",4"							
	[[5,5'-							
		bis[imino(1 5 5-					
	trimethyl-		1,0,0					
		nediyl)me	thylenell-2	2.4.6-				
		,5(2H,4H,		2, 1,0				
	triazine-1		011)					
		trakis[metl	nylene(3.5	5_				
	trimethyl-		1 y 10110(3,3,	,5-				
		nediyl)imi	noorbon	dowy 2.1				
					liazaniumt	atra [hia()		
		l(ethyl)am			nazoniumi	etra[bis(2-		
	metnyipro	pyl)naphtl	naienesuiic	onatej				
607-450-0	00-2		419-040-9	89604-92	-R 53	R: 53		
		penzothiaz				S: 61		
	(Z)- $(2$ -							
	aminothia	zol-4-						
	yl)-2-	201 .						
	(tert-							
	butoxycar	bonyl)						
		yiminoace	tate					
		ymmodec						
607-451-0			417-640-3	161935-1		Xi		
	amino-5-				R43	R: 41-43		
	hydroxy-3	3-				S:		
	(4-(2-					(2-)22-24	-26-37/39	
	sulfoxyetl	hylsulfony	l)phenylaz	o)-2,7-				
	disulfona	pht-6-						
	ylazo]-6-							
	[3-(4-							
	amino-5-							
	hydroxy-3	3-						
	(4-(2-							
		nylsulfony	l)phenylaz	o)-2,7-				
	disulfona		/1 / /					
		nylcarbony	vlaminoibe	nz				
	enesulfon		,					
	acid,	-						
	sodium							
	salt							
607-453-0			418-100-1	172964-1		Xi		
	benzyl-2,				R 53	R: 43-53		
	dihydroxy	y-4-						
			'	'	'	•		

	aza- heptylene bis(2,2- dimethylo	octanoate)				S: (2-)24-37	-61	
607-454-0	mixture of: trans-2-(1-methyleth dioxane-5 carboxylia acid; cis-2-(1-methyleth dioxane-5 carboxylia acid	yl)-1,3-	418-170-3		Xi; R41 R52-53	Xi R: 41-52/53 S: (2-)25-26	-39-61	
607-455-(X	amino-4- (3-[4- chloro-6- (2,5-	ino)-		3172890-9	3R643	Xi R: 43 S: (2-)22-24	-37	
607-456-(00-5 amino-4- chloroben acid, hexadecyl		419-700-6	6143269-7	₽Ŗ R51-53	N R: 51/53 S: 61		
607-457-(dihydroge 1,1"- dihydroxy [p- phenylbis {6- [4-(2- aminoethy yl]}-1,3,5 triazine-4 diyl- imino)]bis	en 7-8,8"- (imino- yl)piperazi - ,2-		172277-9	7X3; R41 N; R51-53	Xi; N R: 41-51/53 S: (2-)26-39	-61	

	azonaphthalene-1',3	16-			
	trisulfonate)]			
607-458-	00-6	420-850-1	N;	N	—
	mixture		R51-53	R: 51/53	
	of: 2-			S: 61	
	ethyl- [2,6-				
	dibromo-4-				
	[1-[3,5-				
	dibromo-4-				
	(2-				
	hydroxyethoxy)phe	nyl]-1-			
	methylethyl]phenox	y]propenoate			
	2,2'-				
	diethyl-				
	[4,4'- bis(2,6-				
	dibromophenoxy)-1				
	methylethylidene]				
	dipropenoate				
	2,2'-[(1-				
	methylethylidene)bi	is[[2,6-			
	dibromo-4,1-	122			
	phenylene)oxy]etha	nol]]			
607-459-	00sdpentyl	418-930-4	R 53	R: 53	
	4-{2-[5-			S: 61	
	cyano-1,2,3,6-				
	tetrahydro-1-				
	isopropoxyethoxy-				
	carbonylmethyl)-4-				
	methyl-2,6-				
	dioxo-3-				
	pyridylidene]hydraz	ino}benzoate			
607-460-	00- 7	418-990-1	Xn;	Xn; N	_
	tridecyloxy-		R48/22	R:	
	propyl-		Xi;	36/38-48/22-50/53	
	ammonium		R36/38	S:	
	9-		N;	(2-)23-26-37/39-60-61	
	octadecenoate		R50-53		
607-461-		421-160-1	R 52-53	R: 52/53	
	mixture			S: 61	
	of:				
	pentasodium 2-{4-{3-				
	methyl-4-				
	[6-				
	sulfonato-4-				
	(2-				
	sulfonato-				
	phenylazo)-				

naphthalen-1- ylazo]- phenylamino}-6- [3-(2- sulfato- ethanesulfonyl)- phenylamino]-1,3,5- triazin-2- ylamino}- benzene-1,4- disulfonate pentasodium 2-{4-{3- methyl-4- [7- sulfonato- phenylazo)- naphthalen-1- ylazo]- phenylamino}-6- [3-(2- sulfato- ethanesulfonyl)- phenylamino]-1,3,5- triazin-2- ylamino]- benzene-1,4- disulfonate					
mixture of: 1- hexyl acetate 2- methyl-1- pentyl acetate; 4- methyl-1- pentyl acetate dinear and branched	421-230-3	188230-35	-N; R51-53	N R: 51/53 S: 61	

	C6-alkyl acetates					
607-463-0	00-3 (phenothiazin-10- yl)propionic acid	421-260-53		N; R51-53	N R: 51/53 S: 24/25-61	
607-464-0	mixture of: 7- chloro-1- ethyl-6- fluoro-1,4- dihydro-4- oxo- quinoline-3- carboxylic acid 5- chloro-1- ethyl-6- fluoro-1,4- dihydro-4- oxo- quinoline-3- carboxylic acid	421-280-46	8077-26-	R 52-53	R: 52/53 S: 61	
607-465-(hydroxyethyl)amm 7-{4- [4-(2- cyanoamino-4- hydroxy-6- oxidopyrimidin-5- ylazo)benzamido]- ethoxy- phenylazo}naphtha disulfonate	2-		R 52-53	R: 52/53 S: 61	
607-466-0 X	mixture of: phenyl 1-(1-[2- chloro-5- (hexadecyloxycarb dimethyl-2- oxobutyl)-1H-2,3,3 tetrahydrobenzotria carboxylate phenyl 2-(1-(2- chloro-5-	sa,7a-		N; R51-53	N R: 51/53 S: 37/39-61	

dimethyl-2- oxobutyl)-1H-2,3, tetrahydrobenzotr carboxylate phenyl 3-(1-(2- chloro-5-	bonyl)phenylcarbamo				
607-467-00,5,3,3- tetrabutyl-1,3- ditinoxydicaprylat	419-430-956533-0 re	00-Xn; R21/22-4 C; R34 N; R50-53	21/22-34- S:	48/22-50/: 36/37/39-4	
mixture of: monosodium 4-((4-(5- sulfonato-2- methoxyphenylan chloro-1,3,5- triazine-2- yl)amino)-2- ((1,4- dimethyl-6- oxido-2- oxo-5- sulfonatomethyl-1 dihydropyridine-3 yl)azo)benzenesul disodium 4-((4-(5- sulfonato-2- methoxyphenylan chloro-1,3,5- triazine-2- yl)amino)-2- ((1,4- dimethyl-6- oxido-2- oxo-5- sulfonatomethyl-1 dihydropyridine-3 yl)azo)benzenesul	,2- - fonate nino)-6-	R43	Xi R: 43 S: (2-)22-24	-37	

trisodium 4-((4-(5- sulfonato-2- methoxyphenylamin chloro-1,3,5- triazine-2- yl)amino)-2- ((1,4- dimethyl-6- oxido-2- oxo-5- sulfonatomethyl-1,2 dihydropyridine-3- yl)azo)benzenesulfo tetrasodium 4-((4-(5- sulfonato-2- methoxyphenylamin chloro-1,3,5- triazine-2- yl)amino)-2- ((1,4- dimethyl-6- oxido-2- oxo-5- sulfonatomethyl-1,2 dihydropyridine-3- yl)azo)benzenesulfo	- nate o)-6-					
disodium 7-((4,6- bis(3- diethylaminopropyla triazine-2- yl)amino)-4- hydroxy-3- (4-(4- sulfonatophenylazo) naphthalene sulfonate	nmino)-1,3		SR\$2-53	R: 52/53 S: 61		
potassium sodium 6,13- dichloro-3,10- bis {2- [4-[3-(2- hydroxysulphonylox (2,5- disulfonatophenylan triazin-2- ylamino]ethylamino [1,4]oxazino[2,3-	nino)-1,3,5	lfonyl)phe -	Xi; R41 R52-53 nylamino]-	Xi R: 41-52/53 S: (2-)39-22	-26-61	

b]phenox disulfona	azine-4,11- te						
607-472-0@r@moniu iron(III) trimethyl hemihydn	enediamine		8111687-30 e	6Mg R51-53	N R: 51/53 S: 61		
607-474-0043(4-(4-dimethylayl)-3-methyl-5-oxo-2-pyrazolinyl)benzoi	minobenzy		4117573-8 ⁶	9R453	R: 53 S: 61		
-1,3,5- triazin-2- ylamino] ureidopho trisulfona tetrasodiu 7-(4-[4- chloro-6- [methyl- (4- sulfonato triazin-2- ylamino]	phenyl)ami 2- enylazo)nap te m phenyl)ami 2- enylazo)nap	no] ohthalene- no]-1,3,5-		8 R6 43	Xi R: 43 S: (2-)22-24	-37	
607-476- 0tris odium N,N- bis(carbo β- alanine	xymethyl)-	414-070-9	9129050-6	2€0) R34 R52-53	C R: 34-52/53 S: (1/2-)26-3	36/37/39-4	5-61
607-478-0 the frameth hydrogen phthalate		क्षी 6-900-ई	579723-02	-T; R25 Xn; R48/22 N; R50	T; N R: 25-48/22- S: (1/2-)25-3		

607-479-0hexadecyl 4- chloro-3- [2-(5,5- dimethyl-2,4- dioxo-1,3- oxazolidin-3- yl)-4,4- dimethyl-3- oxopentamido]benze	418-550-9168689-4	9R453	R: 53 S: 61	
benzenedicarboxylicacid di- C7-11- branched and linear alkylesters		Cat. 2; R61 Repr. Cat. 3; R62	T R: 61-62 S: 53-45	
mixture of: disodium 4-(3- ethoxycarbonyl-4- (5-(3- ethoxycarbonyl-5- hydroxy-1- (4- sulfonatophenyl)pyr yl)penta-2,4- dienylidene)-4,5- dihydro-5- oxopyrazol-1- yl)benzenesulfonate trisodium 4-(3- ethoxycarbonyl-4- (5-(3- ethoxycarbonyl-5- oxido-1- (4- sulfonatophenyl)pyr yl)penta-2,4- dienylidene)-4,5- dihydro-5- oxopyrazol-1- yl)benzenesulfonate	azol-4-	Repr.Cat.: R61 R52-53	R: 61-52/53 S: 53-45-61	
607-488-0 0t hyl (2-	414-210-9147379-3	8N2,	N	
X acetylamino-5- fluoro-4- isothiocyanatopheno	oxy)acetate	R50-53	R: 50/53 S: 60-61	

	mixture of: 2- ethylhexy linolenate linoleate and oleate 2- ethylhexy epoxyolea 2- ethylhexy diepoxylia 2- ethylhexy triepoxylia	l ate l noleate	414-890-7	771302-79	·R43	Xi R: 43 S: (2-)24-37		
	N-methyl glycinate		415-060-7	7	Xi; R41 R43	Xi R: 41-43 S: (2-)24-26	-37/39	
	(3',3'-dimethyl-	l)ethoxy)-		5141773-7	BNI; R51-53	N R: 51/53 S: 61		
	(3aR,4R,7 methyl-4- (1S,2R,3-	propyl)-3a H- 4- -6-		378850-37	% i; R41	Xi R: 41 S: (2-)26-39		
607-494-0		l)octylpho)52894-02-	-N; R50-53	N R: 50/53 S: 60-61		
	4- sulfophen ((1-	yl-6-)amino)hex		5168151-9	2R643	Xi R: 43 S: (2-)24-37		

						T		
607-496-0	methylend di-tert- butyl- phenyl)-2 ethylhexy phosphite	- 1		3126050-5		R:53 S: 61		
607-497-0	00e9ium oxide isostearate	e	419-760-3	3	R53	R: 53 S: 61		
607-498-0	dimethyl-	2,6- lhexadecar		33681-73-0)Xi; R38 R53	Xi R: 38-53 S: (2-)37-61		
607-499-0 X	1,2- ethanediy bis(2-	thyl)ammo	·		Xi; R41 R43 N; R51-53	Xi; N R: 41-43-51/ S: (2-)24-26	-37/39-61	
607-500-0	2,2,bis[(5 tetrapropy		421-670-4 anoate	1	Xi; R38 N; R50-53	Xi; N R: 38-50/53 S: (2-)37-60	-61	
607-501-0	mixture of:	hiophosph	421-820-9 ate)	R53	R: 53 S: 61		
607-502-0	benzyl- N,N,N- tributyl)ar 4-	nmonium enzenesulfo	422-200-()	C; R34 Xn; R22 N; R51-53	C; N R: 22-34-51/ S: (1/2-)26-3	53 36/37/39-45	5-61
607-503-0 X	00,4,6- tri-n- propyl-2,4 trioxo-1,3	1,6-	422-210-5	568957-94	- € ; R34	C R: 34 S: (1/2-)26-3	36/37/39-45	5

607-505-0 pentasodium	422-930-117159	9-84R52-53	R: 52/53		
7-(4-			S: 22-61		
(4-(5-					
amino-4-					
sulfonato-2-					
(4-((2-					
(sulfonato-					
ethoxy)sulfonyl)ph	envlazo)nhenvlam	ino)-6-			
chloro-1,3,5-	ony razo) priony ram				
triazin-2-					
yl)amino-2-					
ureidophenylazo)na	anhtalana 136				
trisulfonate	apintaiene-1,5,0-				
uisuiionate					
607-506- 0% -6	422-970-\$13624	·8-04 N 9	N		
mixture		R51-53	R:51/53		
of:			S: 22-61		
strontium					
(4-					
chloro-2-					
((4,5-					
dihydro-3-					
methyl-5-					
oxo-1-					
(3-	TT				
sulfonatophenyl)-1	П-				
pyrazol-4-					
yl)azo)-5-					
methyl)benzenesul	tonate				
disodium					
(4-					
chloro-2-					
((4,5-					
dihydro-3-					
methyl-5-					
oxo-1-					
(3-					
sulfonatophenyl)-1	H-				
pyrazol-4					
yl)azo)-5-					
methyl)benzenesul	fonate				
- '					
607-507-0poltassium, sodium	422-980-218702	.6-95 X5 ; R41	Xi		
2,4-			R: 41		
diamino-3-			S:		
[4-(2-			(2-)22-26	-39	
sulfonatoethoxysul	fonyl)phenylazo]-:	5-			
[4-(2-					
sulfonatoethoxysul	fonyl)-2-				
sulfonatophenylazo					
benzenesulfonate	1				
	1		<u> </u>		
607-508-0 d i3odium	423-110-4	Xi; R41	Xi		
3,3'-			R: 41		

[iminobistsulfonyl-4 phenylene- (5- hydroxy-3- methylpyrazole-1,4- diyl)azo-4,1- phenylenesulfonylin (4- amino-6-				S: (2-)22-26	-39	
hydroxypyrimidine- diyl)azo-4,1- phenylenesulfonylin amino-6- hydroxypyrimidine- diyl)azo]bis(benzene	nino(4- 2,5-]				
607-512-0 0 -9odium 2,4- diamino-3,5- bis- [4-(2- sulfonatoethoxy)sulf)182926-4 ylazo]benz		R: 52/53 S: 22-61		
mixture of: Trisodium 4- benzoylamino-6- (6- ethenesulfonyl-1 -sulfato- naphthalen-2- ylazo)-5- hydroxynaphthalene disulfonate 5- (benzoylamino)-4- hydroxy-3- ((1- sulfo-6- ((2- (sulfooxy)ethyl)sulfa naphtyl)azo)naphtha disulfonic acid sodium salt 5- (benzoylamino)-4- hydroxy-3- ((1- sulfo-6- ((2- (sulfo-6- ((2-) (sulfo-6- ((2-) (sulfo-6- ((2-) ((2-) (sulfo-6- ((2-) ((2-)	onyl)-2-	3	Xi; R41 R43 R52-53	Xi R: 41-43-52/ S: 22-26-36/		

	(sulfooxy)e naphtyl)azo disulfonic acid	ethyl)sulfo o)naphtha	onyl)-2- lene-2,7-					
607-515-0	mixture of: disodium hexyldiphe ether disulphonal disodium dihexyldiph ether disulphonal	te henyl	429-650-7	7147732-6	0XI; R36 N; R51-53	Xi; N R: 36-51/53 S: (2-)26-61		
607-516-0	bis(trifluord S,S'- bis-L- homocystei	• ,	429-670-6	5105996-5	4Xli; R41 R43	Xi R: 41-43 S: (2-)24-26	-37/39	
607-517-0	0(S6-α- (acetylthio) acid)benzenep		076932-17	-Xn; R22 Xi; R41 R43	Xn R: 22-41-43 S: (2-)22-26	-36/37/39	
607-526-0	00astap 1,3- bis(carbam (dimethylai			15263-53	-N; R50-53	N R: 50/53 S: 60-61		
607-527-0	mixture of: 1- (1'H,1'H,2'I tridecafluor (1"H,1"H,2 tridecafluor 1- (1'H.1'H.2'I tridecafluor (1"H,1"H,2'I tridecafluor (1"H,1"H,2'I tridecafluor (1"H,1"H,2'I tridecafluor (1"H,1"H,2'I tridecafluor (1"H,1"H,2'I tridecafluor (1"H,1"H,2'I tridecafluor	rooctyl)-1 "H,2"H- rooctyl)dd H.2'H- rooctyl)-1 "H,2"H- norodecyl H,2'H- rooctyl)-1 "H,2"H-	odecanedic 2-)dodecane 2- ecyl)dodec	oate	Xn; R48/22	Xn R: 48/22 S: (2-)36		

	pentacosa 1- (1'H,1'H,2 heptadeca (1"H,1'H, heptadeca 1- (1'H,1H,2	fluorodecy 2"H,2"H- fluorodecy	/l)-12- /l)dodecan		e			
	(1"H,1"H	,2"H,2"H- fluorodode		anedioate				
608-031-0	02-7 benzyl-2- methyl-3- butenitrile		407-870-4	197384-48	- % n; R22 R 52-53	Xn R: 22-52/53 S: (2-)61		
608-033-0	butyl-3- (2- chloro-4-			375511-91	- R 43 R 52-53	Xi R: 43-52/53 S: (2-)24-37	-61	
608-034-0	0663 orfena 4- bromo-2- (4- chlorophe ethoxyme trifluorom carbonitri	nyl)-1- thyl-5- iethylpyrro	ble-3-	122453-7	310 R23 Xn; R22 N; R50-53	T; N R: 22-23-50/ S: (1/2-)13-3	53 36/37-45-6	0-61
608-035-0	0(+9-)- α-[(2- acetyl-5- methylpho amino]-2, dichlorob aceto- nitrile	6-	419-290-9		R43 R53	Xi R: 43-53 S: (2-)24-37	-61	
608-036-0	[2-(4-	nyl)vinyl]ţ		379026-02 yl)benzoni		R: 53 S: 61		
608-037-0 X				3124071-4		N R:50/53 S: 60-61		

608-038-0		nnitrile nnitrile	422-580-8	375490-39		Xn; N		
	trimethyl- phenyl- butane- nitrile	4-			N; R51-53	R: 22-51/53 S: (2-)61		
608-039-0		kanenitrile	423-460-8	33508-98-3	3Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)23-60	-61	
608-040-0	dithiobis(amino-1- (2,6- dichloro-4	1- nethyl)phe 3-		130755-4	6N3 R50-53	N R: 50/53 S: 60-61		
608-041-0	butyl-4- oxo-1,3-	-2-		4138401-2·	₽8 R50-53	N R: 50/53 S: 60-61		
608-043-0		xy)propane		5142653-6	1T0 R23 Xn; R22 N; R50-53	T; N R: 22-23-50/ S: (1/2-)13-3	53 86/37-45-6	0-61
609-064-0 X	Mesotrion 2-[4- (methylsu nitrobenze cyclohexa	lfonyl)-2- oyl]-1,3-		104206-8	21×8, R50-53	N R: 50/53 S: 60-61		
609-066-0	sodium 3- amino-10 [4-(10- amino-6,1 dichloro-4	3-)154212-5		Xn 2P68/20/21 20/21/22- S: (2-)36/37	/22 68/20/21/2	2

	[1,4]oxazino[2,3-b]phenoxazine-3-ylamino]-6- [methyl(2-sulfonato-ethyl)amino]-1,3,5-triazin-2-ylamino]-6,13-dichlorobenzo[5,6] [1,4]oxazino[2,3-b]phenoxazine-4,11-disulfonate						
609-067-0	Declium and potassium 4-(3- aminopropylamino) bis[3-(4- methoxy-2- sulfophenylazo)-4- hydroxy-2- sulfo-7- naphthylamino]-1,3 triazine	-2,6-	5156769-9	7R043	Xi R: 43 S: (2-)22-24	-37	
609-068-0	Mailsk xylene 5-tert- butyl-2,4,6- trinitro-m- xylene	201-329-4	481-15-2	Carc. Cat. 3; R40 E; R2 N; R50-53	E; Xn; N R: 2-40-50/5 S: (2-)36/37-		
609-070-0	dichloro-2- (1,1,2,3,3,3- hexafluoropropoxy) nitrobenzene		4130841-2	3Xn; R22 R 43 N; R50-53	Xn; N R: 22-43-50/ S: (2-)36/37/		
609-071-0	mixture of: 2- methylsulfanyl-4,6- bis-(2- hydroxy-4- methoxy- phenyl)-1,3,5- triazine 2-(4,6- bis- methylsulfanyl-1,3,5- triazin-2- yl)-5-		3156137-3	3R643	Xi R: 43 S: (2-)22-24	-37	

	methoxy- phenol						
611-099-0	OnOethylenebis(4,1- phenylenazo(1- (3- (dimethylamino)pro dihydro-6- hydroxy-4- methyl-2- oxopyridine-5,3- diyl)))-1,1'- dipyridinium dichloride dihydrochloride	401-500-5 pyl)-1,2-		Carc.Cat R45 N; R51-53	2T; N R: 45-51/53 S: 53-45-61		
611-100-(opotassium sodium 3,3'- (3(or4)- methyl-1,2- phenylenebis(imino chloro)-1,3,5- triazirie-4,2- diylimino(2- acetamido-5- methoxy)-4,1- phenylenazo)dinaph disulfonate			3 X7 ; R41	Xi R: 41 S: (2-)26-39		
611-101-0 X	chloro-3- cyano-5- formyl-2- thienyl)azo-5'- diethylaminoacetani	405-200-51	104366-2	5R843	Xi R: 43 S: (2-)22-24	-37	
611-103-0	00-i0odium (1-(3- carboxylato-2- oxido-5- sulfonatophenylazo) hydroxy-7- sulfonatonaphthalen amido)nickel(II)			Xi; R41 R 43 N; R51-53	Xi; N R: 41-43-51/ S: (2-)24-26		
611-104-0	mixture of: trisodium (2,4(or 2,6 or 4,6)- bis(3,5- dinitro-2-	406-870-1		R 43 N; R51-53	Xi; N R: 43-51/53 S: (2-)24-37	-61	

oxidophenylazo)-5-		
hydroxyphenolato)		
(2(or 4or		
6)-(3,5-		
dinitro-2-		
oxidophenylazo)-5-		
hydroxy-4(or		
2or 6)-		
(4-(4-		
nitro-2-		
sulfonatoanilino)phenylazo)phenolato)fe	rrate(1-)	
trisodium		
bis(2,4(or		
2,6 or		
4,6)-		
bis(3,5-		
dinitro-2-		
oxidophenylazo)-5-		
hydroxyphenolato)ferrate(1-)		
trisodium		
(2,4(or		
2,6 or		
4,6)-		
bis(3,5-		
dinitro-2-		
oxidophenylazo)-5-		
hydroxyphenolato)		
(2(or 4		
or 6)-		
(3,5-		
dinitro-2-		
oxidophenylazo)-5-		
hydroxy-4(or		
2 or		
6)-(4-		
nitro-2-		
sulfonatophenylazo)phenolato)ferrate(1-)		
trisodium		
(2,4(or		
2,6 or		
4,6)-		
bis(3,5-		
dinitro-2-		
oxidophenylazo)-5-		
hydroxyphenolato)		
(2(or 4		
or 6)-		
(3,5-		
dinitro-2-		
oxidophenylazo)-5-		
hydroxy-4(or		
2 or		
·		

6)-(3- sulfonatophenylazo) disodium 3,3'- (2,4- dihydroxy-1,3(or 1,5 or 3,5)- phenylenediazo)dib	phenolato)ferrate(1-				
611-105-0@odlium 4-(4- chloro-6- (N- ethylanilino)-1,3,5- triazin-2- ylamino)-2- (1-(2- chlorophenyl)-5- hydroxy-3- methyl-1H- pyrazol-4- ylazo)benzenesulfor		5R743 N; R51-53	Xi; N R: 43-51/53 S: (2-)22-24	-37-61	
611-106-00ekasodium 4,4'- dihydroxy-3,3'- bis[2- sulfonato-4- (4- sulfonatophenylazo phenylenebis[imino chloro-1,3,5- triazine-4,2- diyl)imino]]dinapht sulfonate	(6-	Xi; R41	Xi R: 41 S: (2-)26-39		
611-107-0 assium sodium 4-(4-chloro-6-(3,6-disulfonato-naphthalen-2-ylazo)-8-hydroxy-naphthalen-1-ylamino)-1,3,5-triazin-2-ylamino)-5-hydroxy-6-(4-(2-	412-490-7	R 43	Xi R: 43 S: (2-)22-24	-37	

sulfatoethanesulfon phenylazo)- naphthalene-1,7- disulfonate 611-108-0@8odium 5-((4- ((4- chloro-3- sulfonatophenyl)azo naphthyl)azo)-8-	413-600-66527-62-4	4R 52-53	R: 52/53 S: 61		
(phenylamino)-1-naphthalenesulfona	te 407-710-3	N;	N		
products of: copper(II) sulfate and tetrasodium 2,4- bis[6-(2- methoxy-5- sulfonatophenylazo hydroxy-7- sulfonato-2- naphthylaminol-6- (2- hydroxyethylamino triazine(2:1))-5-	R51-53	R: 51/53 S: 61		
611-110-0 @ ra- sodium/	408-210-8124605-8	2R943 N;	Xi; N R:		
lithium 4,4'- bis-(8- amino-3,6- disulfonato-1- naphthol-2- ylazo)-3- methylazobenzene		R51-53	43-51/53 S: (2-)24-28-	37-61	
611-111-0@isodium 2-[[4-(2-chloroethylsulfonyl [(2-hydrox y-5-sulfo-3- [3-[2-(2-(sulfooxy)ethylsulfo sulfobenzoato(3-)cu	nyl)ethylazo]-4-	R 43	Xi R: 43 S: (2-)22-24-	.37	

611-112-00etrasodium X 4- hydroxy-5- [4-[3-(2- sulfatoethane morpholin-4 yl-1,3,5- triazin-2- ylamino]-3- (1- sulfonatonar ylazo)naphth disulfonate	ohthalen-2-		R 43	Xi R: 43 S: (2-)22-24	-37	
611-113-00ithium sodium (2-(((5-((2,5-dichloropher hydroxypher (2-((4,5-dihydro-3-methyl-5-oxo-1-phenyl-1H-pyrazol-4-yl)azo)-5-sulfobenzoat chromate(2-)	nyl)azo)-2- nyl)methylene)ami to(3-))		R51-53	N R: 51/53 S: 24/25-61		
611-114-00i0nium sodium (4-((5-chloro-2-hydroxypher dihydro-5-methyl-3H-pyrazol-3-onato(2-)) (3-((4,5-dihydro-3-methyl-1-(4-methylpheny oxo-1H-pyrazol-4-yl)azo)-4-hydroxy-5-nitrobenzene-)) chromate(2-jul)	nyl)azo)-2,4- yl)-5- esulfonato(3		5X9n; R22 Xi; R41 R 52-53	Xn R: 22-41-52/ S: (2-)22-26		
611-115-0 0-i ithium bis(4-	414-290-3	5149564-65	5X8n; R22 R 52-53	Xn		

hydroxy-1	henyl)azo)	mate(3-)		R: 22-52/53 S: (2-)22-61		
triazin-2- ylamino)- hydroxy-3 (1- sulfonatpri ylazo)- naphthale disulfonat trisodium 5-{4- chloro-6- L2-(2,6- dichloro-3- cyanopyrr ylamino)- methyl- ethylamin triazin-2- ylamino]- hydroxy-3 (1- sulfonator ylazo)- naphthale disulfonat trisodium 5-{4- chloro-6- [2-(4,6- dichloro-3- cyanopyrr ylamino)-	5- imidin-4- ino]-1,3,5- 4- 3- naphthalend ne-2,7- te 5- imidin-4- 1- 1- 1- 1- 1- 1- 1- 1- 1- 1- 1- 1- 1-	3	Xi; R41 R 43	Xi R: 41-43 S: (2-)22-24	-26-37/39	
triazin-2- ylamino}	inoj-1,3,5- -4-					

ylazo)- naphthaler 2,7- disulfonate trisodium 5-{4- chloro-6- [2-(4,6- dichloro-5 cyanopyri ylamino)- methyl- ethylamin triazin-2- ylamino)-4 hydroxy-3 (1-	naphthalen-2- ne e i- midin-2- 1- o]-1,3,5- 4- i- naphthalen-2- ne-2,7-				
611-117-00-3-	415_100	-3149850-29	R343 Xi		
bis {6- fluoro-4- [1,5- disulfo-4- (3- aminocarb ethyl-6- hydroxy-4 methyl- pyrid-2- on-5- ylazo)- phenyl-2- ylamino]- triazin-2- ylamino) propane lithium-, sodium salt	oonyl-1- -	- Ψ 1 1 7 0 J U - Z 7	R: 43 S: (2-)22-	24-37	
611-118-00edium 1,2- bis[4- [4-(4-(4- sulfophen sulfophen ureido-	ylazo)-2-	1-8149850-3	Xi R: 43 S: (2-)22-	24-37	

phenyl- amino]-6- fluoro-1,3,5- triazin-2- ylamino]- propane, sodium salt						
611-119-0 @ asodium 4-[4- chloro-6- (4- methyl-2- sulfophenylamino)- triazin-2- ylamino]-6- (4,5- dimethyl-2- sulfophenylazo)-5- hydroxynaphthalend disulfonate	1,3,5-	4148878-2	2XI; R41 R 43	Xi R: 41-43 S: (2-)22-24	-26-37/39	
611-120-06-34-[5- amino-2- [4-(2- sulfoxyethylsulfony sulfo- phenylamino]-6- chloro-1,3,5- triazin-2- ylamino}-4- hydroxy-3- (1- sulfonaphthalen-2- ylazo)- naphthalene-2,7- disulfonicacid sodium salt		7157707-9 :o]-4-	4X3; R41 R 52-53	Xi R: 41-52/53 S: (2-)22-26	-39-61	
611-121-0049in component 6 (isomer): asym. 1:2 Cr(III)- complex of: A: 3- hydroxy-4- (2- hydroxy- naphthalene-1-	417-280-	930785-74	-Ki; R41 N; R50-53	Xi; N R: 41-50/53 S: (2-)26-39	-60-61	

	ylazo)naphthalene-	1 _					1
	sulfonic	1 -					
	acid,						
	Na-salt						
	and B:						
	1-[2-						
	hydroxy-5						
	-(4-						
	methoxy-	1					
	phenylazo)phenylaz naphthalene-2-	20]					
	ol						
	Main						
	component 8						
	(isomer):						
	asym. 1:2 Cr-						
	complex						
	of: A: 3-						
	hydrox						
	y-4-(2-						
	hydroxy-						
	naphthalene-1-						
	ylazo)-						
	naphthalene-1-						
	sulfonic						
	acid,						
	Na-salt						
	and B:						
	1-[2-						
	hydroxy-5-						
	(4-						
	methoxy-						
	phenylazo)-						
	phenylazo]-						
	naphthalene-2-						
	ol						
		445.050	-1-1-12-6-0	OT 17 D 44	***		
611-122-0	De kasodium	417-250-	\$151436-9		Xi		
	(di[N-			R 43	R: 41-43		
	(3-(4-				S:	26 27/20	
	[5-(5-				(2-)22-24	-26-37/39	
	amino-3-						
	methyl-1						
	nhanylnyrazal 4						
	phenylpyrazol-4-						
	yl- azo)-2,4-						
	disulfo-						
	anilino]-6-						
	chloro-1,3,5-						
	triazin-2-						
		1	I	l	l	l	

611-123-0 X	ylamino)phenyl)- sulfamoyl] (di- sulfo)- phthalocyaninato)ni)6-(2,4- bis(4- ((5-(4,6- bis(2- aminopropylamino) triazin-2- ylamino)-4- hydroxy-2,7- disulfonaphthalen-3 yl)azo)phenylamino triazin-6- ylamino)propyldieth lactate	424-310-4 -1,3,5-)-1,34-	4178452-6 um	6 X 9; R41	Xi R: 41 S: (2-)26-39		
611-124-0	mixture	424-320-9	180778-2	N;	Xi; N R:		
	of: pentasodium			R51-53	41-51/53 S:	C1	
	5- amino-3- (5- (4- chloro-6- [4-(2- sulfoxyethoxysulfor triazin-2-	ato)pheny	lamino]-1,	3,5-	(2-)26-39	-61	
	ylamino) -2-						
	sulfonatophenylazo) [5-(2,3- dibromopropionylar sulfonatophenylazo] hydrox ynaphthalene-2,7- disulfonate pentasodium	nino)-2-					
	5- amino-6- [5-(2- bromoacryloylaminosulfonatophenylazo] (5-{4- chloro-6- [4-(2- sulfoxyethoxysulfor	-3-	Jaminol 1	3.5.			
	triazin-2- ylamino}-2- sulfonatophenylazo		iaiiiiiO]-1,	٠,٠-			

triazin-2- ylamino}-2- sulfonatophe [5-(2,3- dibromoprop sulfonatophe hydroxynaph disulfonate	yl)phenylamino]-1 enylazo]-6- pionylamino)-2- enylazo]-4-	,3,5-			
yl- azo]-3- [2- oxido-4- (ethensulfon; methoxypher oxidonaphth; sulfonate copper (II) complex Disodium 6-[3- carboxy-4,5- dihydro-5- oxo-4-	enyl)pyrazolin-4- yl)-5- nylazo]-4- alene-2- lsulfonyl)-5- nylazo]-4-	Xi; R41 N; R51-53	Xi; N R: 41-51/53 S: (2-)26-39	-61	

	copper (II) complex;						
611-126-	00,6-bis- (2-(4-(4- amino- phenylamino)- phenylazo)-1,3- dimethyl-3H- imidazolium)-4- dimethylamino-1,3,3 triazine, dichloride		1174514-0	6X3; R41 N; R50-53	Xi; N R: 41-50/53 S: (2-)26-39	-60-61	
611-127-	pehtasodium 4- amino-6- (5-(4- (2-ethyl- phenylamino)-6- (2- sulfatoethanesulfony triazin-2- ylamino)-2- sulfonatophenylazo) hydroxy-3- (4-(2- sulfatoethanesulfony naphthalene-2,7- disulfonate	-5-		R5 Xi; R41 R 43 R 52-53	Xi R: 5-41-43-5 S: (2-)22-26	2/53 -36/37/39-	41-61
611-128-0	bis {6- chloro-4- {6-(4- vinylsulfonylphenyl disulfonicacid-5- hydroxynapht-4- ylamino]-1,3,5- triazin-2- yl)- N-(2- hydroxyethyl)ethane diamine, sodium salt	azo)-2,7-	9171599-8	R 43	Xi R: 41-43 S: (2-)22-24	-26-37/39	
611-129-	0A2 mixture of: 5- [(4-[(7- amino-1-	418-230-9	9163879-6	9E4 R2 Repr.Cat R62 Xn; R48/22	R:	2-62-51/53	3

acid 5-[(4- [(7- amino-1- hydroxy-3- sulfo-2- naphthyl)azo]- diethoxypheny [(3-	rl)azo]-2- nyl)azo]benzoic 2,5-	R 43 N; R51-53	S: (2-)26-35	-36/37-61	
611-130-0 testa- ammonium 2-[6- [7-(2- carboxylato- phenylazo)-8- hydroxy-3,6- disulfonato-1- naphthylamino hydroxy-1,3,5- triazin-2- ylamino]benzo	-	0-96XI; R36 N; R50-53	Xi; N R: 36-50/53 S: (2-)26-39	-60-61	
611-131-0 2 -\$2- hydroxy-3- (2- chlorophenyl) naphthylazo]- [2- hydroxy-3- (3- methylphenyl) naphthylazo]fl	carbamoyl-1-	Repr.Cat R61 R 53	2Ţ R: 61-53 S: 53-45-61		
611-132-0 entasodium bis {7- [4-(1- butyl-5- cyano-1,2- dihydro-2- hydroxy-4- methyl-6- oxo-3- pyridylazo)pho nitro-3,3'-	419-210-2 enylsulfonylamino]-5'-	Xi; R41 R 52-53	Xi R: 41-52/53 S: (2-)26-39	-61	

	disulfonat azobenzer diolato] chromate (III)	onaphthalo	ene-2-					
611-133-0	` '	enzene-4-e enzene-4- de ntly	419-260-3	5	Xi; R41 N; R51-53	Xi; N R: 41-51/53 S: (2-)26-39	-61	
	and 4'- amino-4-							

dip sul acid me wit ferr chl sod salt	ric oride, lium						
hyd [4- chl [4- dib sul tria yla sul ber sul- cop	odium α[2- droxy-3- oro-6- (2,3- romopropionylar fonatophenylami zin-2- mino]-5- fonatophenylazo nzylidenehydrazi fonatobenzoate, oper mplex	no]-1,3,5-]-	3	Xi; R41 N; R51-53	Xi; N R: 41-51/53 S: (2-)22-26	-39-61	
of: [[4] am ure [(2) (su aci wit 2,4 trif and par hyd to t cor vin der pot sod salt	duct 2- ino-2- idophenylazol-5- lfooxy)ethyl)sulf d h ,6- luoropyrimidine l tial drolysis he responding ylsulfonyl ivative,mixed assium/ lium	fonyl]]benz	enesulfoni		Xi R: 41-52/53 S: (2-)26-39	-61	
[4-	4-(2- moniopropylami droxy-3-	424-260-3 no)-6-	3	Repr.Cat.: R62 Xi; R41	3Xn; N R: 41-62-51/	53	

	methyl-2- methoxy-4- sulfamoylpl sulfonatona ylamino]-1, triazin-2- ylamino}-2 aminopropy formate	henylazo phth-7- ,3,5-)-2-		N; R51-53	S: (2-)22-26	-36/37/39-	51
611-137-(butyl-7- chloro-3- tridecyl-7,7 dihydro-1H pyrazolo[5, c]-1,2,4- triazole	[-	419-870-1	159038-1	6R153	R: 53 S: 61		
611-138-(02-(4- aminopheny tert- butyl-1H- pyrazolo[1, b] [1,2,4]triazo	5-	415-910-7	7152828-2	5R 6 43 N; R51-53	Xi; N R: 43-51/53 S: (2-)22-24	-37-61	
611-140-0	Q e 2 afenidin			68049-83	-T; R48/22 Repr. Cat. 2; R61 Repr. Cat. 3; R62 N; R50-53	T; N R: 61-48/22- S: 53-45-60-	R50/53	
612-184-0	(dibutylami methyl-2'- (phenylami n-1 (3H),9- (9H)- xanthenl-3- one	no)spiro[589331-94 ura	- R 52-53	R:52/53 S: 61		
612-185-0	0 0-0 -[4- ((heptadeca	fluorono		359493-72	- % i; R41	Xi; N		

	N,N,N-	o]propyl]- ummonium	ı		N; R50-53	R: 41-50/53 S: (2-)26-39	-60-61	
612-186-0	(7- hydroxy-8 methyl-5- phenylphe			8149057-6	4Xh; R48/22 Xi; R41 R 43 N; R50-53	Xn; N R: 41-43-48/ S: (2-)22-26	22-50/53 -36/37/39-0	60-61
612-187-0	0 2,3 ,4- trifluoroa	niline	407-170-9	93862-73-5	Xn; R21/22-4; Xi; R38-41 N; R51-53	21/22-38- S:	41-48/22-5 -36/37/39-0	
612-188-0	04,4'- (9H- fluoren-9- ylidene)bi chloroani	is(2-	407-560-9	9107934-6	8N9, R51-53	N R: 51/53 S: 61		
612-189-0	amino-2-	ethyl)pheno lloride		1135043-6	4Xth; R22 R 43 N; R50-53	Xn; N R: 22-43-50/ S: (2-)22-24	-37-60-61	
612-190-0	nethylene isopropylmethylani	-6-	415-150-6	616298-38	-Xn; R48/22 N; R51-53	Xn; N R: 48/22-51/ S: (2-)36-61	53	
612-191-0	OP Aymer of allylaming hydrochlo		415-050-2	271550-12	-Xn; R22 R 43	Xn R: 22-43 S: (2-)36/37		
612-192-0	isopropyl- (N-	-4- ninomethy		5154212-6	0X9n; R21/22 Xi; R38-41 N; R51-53	Xn; N R: 21/22-38- S: (2-)26-36		
612-193-0		inomethyl		718759-96 ne	-Kn; R21/22 C; R34 R 43 N; R50-53	C; N R: 21/22-34- S: (1/2-)26-3	43-50/53 36/37/39-45	5-60-61

	hydroxy-3 [(2- hydroxyer [2-(1-	thyl)- ecyl)amino)141890-3 ino]-	0X4n; R22 Xi; R41 N; R50-53	Xn; N R: 22-41-50/ S: (2-)26-39		
	4-		415-210-1		Xn; R20/22	Xn; N R:		
	1,5-	nzyl)amm nedisulfon	_		Xi; R41 N; R50-53	20/22-41- S: (2-)26-36	50/53 /39-60-61	
	6-0 chloro-o- toluidine [1] 4- chloro-o- toluidine hydrochlo	E	202-441-6 [1] 221-627-8 [2]	[1]	Carc.Cat R45 3Muta.Cat. R68 T; R23/24/2: N; R50-53	R: 3\\$5-23/24/ S: 53-45-60-	25-68-50/5 -61	3
612-197-0	0-4 5-	Е	205-282-0)137-17-7	Carc Cat	2T· N		
	trimethyla [1] 2,4,5- trimethyla hydrochlo [2]	aniline aniline	[1] -[2]	[1] 21436-97 [2]	R45	R: 45-23/24/	25-51/53	
	0,4'- thiodianil and its salts	E ine	205-370-9	9139-65-1	Carc.Cat R45 Xn; R22 N; R51-53	2T; N R: 45-22-51/ S: 53-45-61	53	
	04,4'- oxydianili and its salts p- aminophe ether		202-977-(0101-80-4	R45	R: 245-46-23/ S: 353-45-61	24/25-62-5	51/53
612-200-0	0 ,0 - diaminoai	nisole	210-406-1 [1] 254-323-9	[1] [1]	Carc.Cat.2 R45	2Ţ; N R: 45-22-68-	51/53	

(12.201.4	methoxy- phenylene [1] 2,4- diaminoa sulphate [2]	ediamine nisole	[2]	[2]	-Muta.Cat. R68 Xn; R22 N; R51-53	53-45-61		
612-201-(がが、N',N' tetrameth methylen	yl-4,4'-	202-959-2	2101-61-1	Carc.Cat.2 R45 N; R50-53	21; N R: 45-50/53 S: 53-45-60-	-61	
612-202-(0 , 4- dichloroa	niline	202-448-4	195-76-1	T; R23/24/2: Xi; R41 R43 N; R50-53	23/24/25- S:	41-43-50/5 36/37/39-45	
612-204-0	Basic Violet 3 4-[4,4'- bis(dimet benzhydr dien-1-	hylamino) ylidene]cyd imethylam	clohexa-2,	5548-62-9	R40	R: 22-40-41- S:	-50/53 /37/39-46-0	50-61
612-205-0	Basic Violet 3 with ≥ 0.1% of Michler's ketone (EC no. 202-027-3	E 5)	208-953-6	5548-62-9	R45	2T; N R: 45-22-41- S: 53-45-60-		
612-206-0	Mathoxado 3- anilino-5- methyl-5- (4- phenoxyp oxazolidin dione	henyl)-1,3	-	131807-5	7Xh; R48/22 N; R50-53	Xn; N R: 48/22-50/ S: (2-)46-60		
612-209-0 X	% - methoxy- toluidine	E m-	204-419-1	120-71-8	Carc.Cat. R45 Xn; R22	R: 45-22		

	<i>p</i> -							
	cresidine							
612-210-0	nitro-o- toluidine [1] 5- nitro-o- toluidine hydrochlo [2]	oride	202-765-8 [1] 256-960-8 [2]	899-55-8 [1] 851085-52 [2]	Carc.Cat R40 - 0 ; R23/24/2; R52-53	R: 23/24/25-		
612-211-0	[(benzotri yl)methyl	azole-1-	416-470-9)	Xi; R36 N; R51-53	Xi; N R: 36-51/53, S: (2-)26-61		
612-212-0	dichloro-4	1- nethylanilir)24279-39	- % n; R20/22 Xi; R38 R43 N; R50-53	Xn; N R: 20/22-38- S: (2-)24-37		
612-213-0	yl)-1,1-		-3-	2148348-1	3€4 R34 R52-53	C R: 34-52/53 S: (1/2-)23-2	26-36/37/3	9-45-61
612-214-0	diphenyle N,N-di-	thenyl)- nzenamine	421-390-2	289114-90	- P 53	R: 53 S: 61		
612-215-0	chloro-2-	lthio)anilin		5179104-3	2X6; R38 N; R51-53	Xi; N R: 38-51/53 S: (2-)37-61		
612-217-0	00-3 methoxy- propylam		422-550-4	137143-54	- F ; R11 C; R34 Xn; R22 R52-53	F; C R: 11-22-34- S: (1/2-)9-26	52/53 5-36/37/39-	-45-61
613-181-0	06,5- dimethyl- perhydro- pyrimidin one a-(4-		405-090-9	967485-29	R48/25 Xn; R22 Xi; R36 N; R50-53	T; N R: 22-36-48/ S: (1/2-)22-2	25-50/53 26-36/37-4	5-60-61

	α-(4-	nethylstyry nethyl)cinn		hydrazone				
613-182-0		nethyl)quii		765322-65	R40	R: 322-38-40- S:	-36/37/39-	
613-183-0	mixture of: 5-(N- methylper octadecyl oxazolidir one 5-(N-	n-2- rfluorohept -1,3-		ido)methyl		Xn; N R: 48/22-50/ S: (2-)36-60		
613-184-0	00:8rilotrie ol 2- ethylhexa	thyleneami noate	14d1Bicq570 _F 1	ane-2-	Xi; R36 R 43	Xi R: 36-43 S: (2-)24-26	-37	
613-185-0	02,3,5,6- tetrahydro methyl-21 cyclopent thiazol-3- one	I- a[d]-1,2-	407-630-9	82633-79	-T; R25 Xi; R41 R 43 N; R50-53	T; N R: 25-41-43- S: (1/2-)22-2		9-45-60-61
613-186-0	((R)-1- (tert- butyldime oxoazetid yl acetate	thylsiloxy		976855-69	-Xi; R36 R 43 N; R51-53	Xi; N R: 36-43-51/ S: (2-)24-26		
613-188-0 X				7116256-1	1½n; R22 Xi; R41 R 43 N; R51-53	Xn; N R: 22-41-43- S: (2-)22-24	-51/53	61
613-189-0	tetrakis(p	fonyl)-1,4,7	7,10-)52667-88	- R 43 N; R50-53	Xi; N R: 43-50/53		

						S: (2-)24-37	-60-61	
613-190-0	0 di® odium		414-040-4	149530-9	3128n · R22	Xn		
013 170 (1- amino-4- (2-(5- chloro-6- fluoro- pyrimidin ylamino- methyl)-4	_		7119330)	R 43	R: 22-43 S: (2-)22-24	-37	
	methyl-6- sulfo phenylam dioxo-9,1 dihydro- anthracen sulfonate	ino)-9,10- 0-						
613-191-0			421-150-7	7143860-0				
	ethyl-2- methyl-2-				R60 C: R34	R: 60-34-50/	53	
	(3-				N;	S:		
	methylbu oxazolidii				R50-53	53-45-60-	61	
613-193-0	[(6- hydroxy-4 tetramethy diazoniaii	ammonio) 1,4,8,8- yl-4,8- ndecane-1 amoyl)di[r	,11-	amoyl]-	N; R51-53	N R: 51/53 S: 61		
613-194-0	1		418-000-8	3163062-2	8 Xi ; R41	Xi		
	dichloro-3	3,10-				R: 41		
	bis {2- [4- fluoro-6- (2-					S: (2-)22-26	-39	
	triazin-2- ylaminojp [1,4]oxaz b.]phenox disulphon acid,	azine-4,11	o}benzo[5	,6]				
	lithium-, sodium salt.							
613-195-0		e)bis((4H-3		18600-59	- I R 43 R 53	Xi R: 43-53		

benzoxaz one)	ine-4-				S: (2-)24-37	-61	
613-196-06-\$[4-chloro-6-[2-[[4-fluoro-6-[[5-hydroxy-[(4-methoxy-sulfopher sulfo-2-naphthale triazin-2-yl]amino	2- yl)azo]-7- nyl]amino] -1- yl]amino]-]-1,3,5-	5168113-7	8X8; R41	(2-)24-37 Xi R: 41 S: (2-)26-39	-61	
(ethenyls hydroxy- naphtaler disulfonic acid, sodium salt				6D42	V;· NI		
triazine 2,4,6- tri(methy triazine [(2- butyl-4,6 dimethyl) triazine [(2,4- dibutyl-6	tricarbamo	1,3,5-)-1,3,5- oyl]-1,3,5-	187547-4	N; R51-53	Xi; N R: 43-51/53 S: (2-)24-37	-61	
triazine 613-199-00A X mixture of: 1,3,5- tris(3-	thylphenyl H)- ,4,6-	421-550-1		Carc.Cat.: R45 Repr.Cat.: R61 R 43 R 52-53	2Ț R: 245-61-43- S: 53-45-61	52/53	

			, .		ı		ı	ı
	trione a							
	mixture of							
	oligomers							
	of 3,5-]						
	bis(3-							
		hylphenyl)-1-					
	poly[3,5-	y - F - y -,	, -					
	bis(3-							
	aminome	hylphenyl)-2,4,6-					
	trioxo-1,3	,5-						
	(1H,3H,5)	H)-						
	triazin-1-							
	yl]-1,3,5-							
	(1H,3H,5							
	triazine-2	,4,6-						
	trione							
613-200-0			420-980-7	7	Xi; R41	Xi		
	product					R: 41		
	of:					S:	20	
	copper,					(2-)22-26	-39	
	(29H,31H	aninato(2-	`					
		N31,N32)						
	chlorosul		,					
	acid and	lario						
	3-(2-							
		thylsulfon	y					
	1)aniline,							
	sodium							
	salts							
613-201-0)()R9-5-		422-390-5	5143322-5	7R@pr.Cat.:	3Ţ; N		
	bromo-3-				R62	R:		
	(1-				T;		41-43-48/2	25-62-50/53
	methyl-2-				R39-48/2			
	pyrrolidir	iyl			Xn;	(1/2-)53-4	15-60-61	
	methyl)-1	H-			R20/22			
	indole				Xi; R41 R 43			
					N;			
					R50-53			
(12.202.0	0 4 4			122212.0		37		
613-202-0	py4 metroz	ine		123312-8	9@arc.Cat3			
	(ISO) (E)-4,5-				R40 R52-53	R: 40-52/53		
	(<i>E</i>)-4,3- dihydro-6				K32-33	40-52/55 S:		
	methyl-4-					(2-)36/37	-61	
	(3-					(2)30/37		
		thyleneam	ino)-1,2,4	_				
	triazin-3		, , ,					
	one							
-	ļ	ļ			ļ		L	

613-203-0 X	Opyraflufenethyl [1] pyraflufen [2]		-[1] -[2]	129630-1 [1] 129630-1 [2]	R50-53	N R: 50/53 S: 60-61		
613-204-0	Oxadiargyl (ISO) 3-[2,4-dichloro-5-(2-propynyloxy) (1,1-dimethylethy oxadiazol-2(3 one 5-tert-butyl-3-[2,4-dichloro-5-(prop-2-ynyloxy)pher oxadiazol-2(3 one)phenyl] -l)-1,3,4- 3 <i>H</i>)-	l-5- -	539807-15	-Repr.Cat 3; R63 Xn; R48/22 N; R50-53	Xn; N R: 48/22-63- S: (2-)36/37-		
613-205-0	00+0piconazol (+)-1- [2-(2,4- dichlorophen propyl-1,3- dioxolan-2- ylmethyl]-1 <i>H</i> triazole	yl)-4-	262-104-4	160207-90	-Kn; R22 R43 N; R50-53	Xn; N R: 22-43-50/ S: (2-)36/37-		
613-206-0	Committee (ISO) (I	-3,5-		161326-3	₽\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	N R: 50/53 S: 60-61		
613-207-0	Omlazalil sulphate, aqueous solution 1-[2- (allyloxy)eth (2,4- dichlorophen imidazolium	yl-2-	281-291-3	58594-72- 883918-57-		C; N R: 22-34-43- S: (2-)26-36	R22-34-4 / 30 / 99 -45- < C ≤ 50 %: Xn, N;	

S (; (; (; d iii h	nydrogen sulphate ±)-1- 2- allyloxy) 2,4- dichlorop midazolid nydrogen sulphate	henyl)]-1 <i>H</i>	<u>/-</u>				25 %
613-208-00	lm/azamox	ζ		114311-32	2N 9 , R50-53	N R: 50/53 S: 60-61	
d	9i2-1-(3- chloropro limethyl- piperidin nydrochlo		417-430-3	863645-17	- O ; R25 Xn; R48/22 R43 N; R51-53	T; N R: 25-43-48/ S: (1/2-)22-3	22-51/53 36/37-45-61
tı		pyl)-2,5,5- 1,3-	417-650-1	188128-57	- % n; R48/22 R52-53	Xn R: 48/22-52/ S: (2-)23-25	
f	nethyl-4- p-	ryl)pyridin fate		374401-04	- R 43 R52-53	Xi R: 43-52/53 S: (2-)22-24	-37-61
(tl		loxy)phen		3133467-4	1Xln; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)22-60	-61

613-213-00is-1- benzoyl-4- [(4- methylsulfo L- proline		16-040-0	120807-0	2R552-53	R: 52/53 S: 61		
613-214-004,N- X di-n- butyl-2- (1,2- dihydro-3- hydroxy-6- isopropyl-2 quinolylide dioxoindan carboxamic	2- ene)-1,3- -5-	16-260-7	147613-9	5R453	R: 53 S: 61		
613-215-00-5 chlorometh dimethoxyp chloride	yl-3,4-	16-440-5		Xn; R21/22-48 Xi; R38-41 R43 N; R51-53		41-43-48/2 /37/39-61	22-51/53
613-216-0 6-t ert- butyl-7- (6- diethylamin methyl-3- pyridylimin (3- methylpher c] [1,2,4]triaz	no-2- no)-3- nyl)pyrazolo	16-490-8 o[3,2-		N; R50-53	N R: 50/53 S: 60-61		
613-217-04-63- (3,5- di-tert- butyl-4- hydroxypho [2-[3- (3,5- di-tert- butyl-4- hydropheny -2,2,6,6- tetramethyl	enyl)propio yl)propiony	nyloxy]-		·R 53	R: 53 S: 61		
613-218-0 6- 1 hydroxyind		17-020-4	2380-86-1	Xn; R22 Xi; R41 R43 N; R51-53	Xn; N R: 22-41-43- S: (2-)24-26		

613-219-0	ethyl-3,5- bis(1- methylethyl)-2,3 tetrahydrooxazol c]-2,3,4,5- tetrahydrooxazol	,4,5- o[3,4-	779185-77	-&i R38 N; R51-53	Xi; N R: 38-51/53 S: (2-)37-61		
613-220-0	00-20ns- (4S,6S)-5,6- dihydro-6- methyl-4H- thieno[2,3- b]thiopyran-4- ol, 7,7- dioxide	417-290-	3147086-8	1X5n; R22	Xn R: 22 S: (2-)36		
613-221-0	0 2- 8 chloro-5- methyl- pyridine	418-050-	018368-64	- X n; R21/22 Xi; R38 R52-53	Xn R: 21/22-38- S: (2-)23-25	52/53	
613-222-0	04-81- oxo-2- propenyl)- morpholine	418-140-	15117-12-4	Xn; R22-48/22 Xi; R41 R43	22-41-43- S:	-48/22 -36/37/39	
613-223-0	isopropyl-3- (4- fluorophenyl)-1F indole		493957-49	-₽ 53	R: 53 S: 61		
613-224-0	02-4- dimercaptomethy dithiane		8136122-1	5Xln; R22 C; R34 R43 N; R50-53	C; N R: 22-34-43- S: (1/2-)26-3	-50/53 36/37/39-4	5-60-61
613-225-0 X	mixture of:[2- (anthraquinon-1- ylamino)-6- [(5- benzoylamino)- anthraquinone-1- ylamino]-4- phenyl]-1,3,5- triazine 2,6- bis-[(5- benzoylamino)-			Xn; R48/22 R53	Xn R: 48/22-53 S: (2-)22-36	-61	

	anthraquinglamino]-phenyl-1, triazine.	4-						
613-226-0	(ethyl(4- (4-(4-(4- (ethyl(2- pyridinoe methylpho phenylazo	enyl)amino n)-2- nzoylamin	3163831-6 o)-	7X1; R41 N; R50-53	Xi; N R: 41-50/53 S: (2-)26-39	-60-61	
613-227-(- 2H-1-	419-600-2]-6-	2	R 43 N; R51-53	Xi; N R: 43-51/53 S: (2-)24-28	-36/37-61	
613-228-0	0()+6-)- (R*,S*)-6 fluoro-3,4 dihydro-2 oxiranyl-2 benzopyra	 - 2H-1-	419-630-6	6	N; R51-53	N R: 51/53 S: 24-61		
613-230-0	Off-orasular (ISO) 2',6',8- trifluoro-s methoxy- triazolo[1 c] pyrimidin sulfonanil	5- 5- ,5- e-2-		145701-2	3NI; R50-53	N R: 50/53 S: 60-61		
613-233-0) (1,3 '-		423-230-7	756552-15	- % i; R41	Xi		
	(oxy- (bismethy bis-1,3- dioxolane				,	R: 41 S: (2-)26-39		
614-028-0	mixture of: 2- ethylhexy mono- D- glucopyra		414-420-0)	Xi; R41	Xi R: 41 S: (2-)26-39		

	12		i i				I I	
	2- ethylhexy	1						
	di-D-	1						
	glucopyra	noside						
(14.020.0			410 (40 (0.0704 14	₩ D22	V		
614-029-0	© -Ønstituti	onai	419-640-0)68784-14	- x n; R22	Xn R: 22		
	isomers of penta-					S: (2-)		
	O-allyl-					3. (2-)		
	β-D-							
	fructofura	nosvl-						
	α-D-							
	glucopyra	noside						
	Constituti							
	isomers							
	of hexa-							
	O-allyl-							
	β-D-							
	fructofura	nosyl-						
	α-D-							
	glucopyra Constituti	noside						
	isomers	Oliai						
	of hepta-							
	O-allyl-							
	β-D-							
	fructofura	nsovl-						
	α-D-	,						
	glucopyra	noside						
615-030-0)61 % ali	A			Xn;	Xn		
013 030 (salts,	11			R20/21/22			
	alkali				R32	20/21/22-	32-52/53	
	earth				R52-53	S:		
	salts and					(2-)13-61		
	other					,		
	salts of							
	thiocyanic							
	acid not							
	mentioned							
	elsewhere	;						
	in this Annex							
615-031-0		A	222-571-7	73535-84-(Xn; N		
	salt of				R20/21/22		22 51/52	
	thiocyanic				R32	20/21/22-	32-51/53	
	acid				N; R51-53	S:		
					K31-33	(2-)13-61		
615-032-0		A			Xn;	Xn; N		
	salts of				R20/21/22		22 50/52	
	thiocyanic				R32	20/21/22-	32-50/53	
	acid not	1			N;	S:	61	
	mentione	1			R50-53	(2-)13-60	TO 1	

	elsewhere							
	in this							
	Annex							
616-092-0	Pol ymeric	1	404-035-6	5	R 43	Xi		
010 072 (reaction	,	101 055 (,	R 53	R: 43-53		
	product				100	S:		
	of					(2-)24-37	-61	
	bicyclo[2]	2.11hepta-	2.5-			(=)= : = ;		
	diene,]v _k	_,-					
	ethene,							
	1,4-							
	hexadiene	,						
	1-							
	propene							
	with							
	N,N-							
	di-2-							
	propenylf	ormamide						
616-093-0	R -daction		406-620-1	129217-9	0R943	Xi; N		
	products				N;	R:		
	of:				R51-53	43-51/53		
	aniline-					S:		
	terephthal	aldehyde-				(2-)24-37	-61	
	0-							
	toluidine							
	condensat	e						
	with							
	maleic							
	anhydride							
616-094-0			406-370-3	358890-25		Xi		
	dicyclohe				R 53	R: 43-53		
	methylene					S:		
	phenylene	e)diurea				(2-)24-37	-61	
616-095-0	00-3'-		406-690-3	343136-14	- R 53	R: 53		
	dioctadec	yl-1,1'-				S: 61		
	methylene							
	phenylene	e)diurea						
(1,6,00,6,4	ANT CV A		400 110	1110402 0	7D2 5.2	D 52		
616-096-0		2	408-110-4	1110483-0	/K353	R: 53		
	hexadecyl					S: 61		
	hydroxypi	ор-1-						
	yl)- N-(2-							
	hydroxyet	hyl)palmit	amide					
		<i>J</i> -/Pullill		.00055 5=	D 50	D 72		
616-097-0) 0 4,3N'-1,4-	1 . (2	411-840-6	83372-55	- K 53	R: 53		
	phenylene	ebis(2-				S: 61		
	((2- methoxy-	4						
	memoxy-	+-						

	nitrophenyl)azo)-3- oxobutanamide						
616-098-0	chloro-3- ((2,2,3,3,3- pentafluoropropoxy phenyl-1H-1,2,4- triazole-3- carboxamide		7119126-1 enyl]-5-	5N, R51-53	N R: 51/53 S: 61		
616-099-	00-[4-[(4- hydroxyphenyl)sulfodimethyl- N-[5- [(methylsulfonyl)an [4- (1,1,3,3- tetramethylbutyl)ph oxopentanamide	onyl]pheno nino]-2-		OR153	R: 53 S: 61		
616-100-6	00-38- dimethyl-1,3- bis(trimethylsilyl)ur		710218-17	- X n; R22 Xi; R38	Xn R: 22-38 S: (2-)36/37		
616-101-0	tert- butyl-1,2,3,4- tetrahydro-3- isoquinolinecarboxa		149182-7	2X9n; R22 R 52-53	Xn R: 22-52/53 S:(2-)61		
616-102-0	mixture of: α- [3-(3- mercaptopropanoxy ω-[3-(3- mercaptopropanoxy poly- (oxyethylene- co- oxypropylene) 1,2-(or 1,3-)bis[α- (3- mercaptopropanoxy ω-oxy- poly(oxyethylene- co- oxypropylene)]-3- (or 2-)propanol 1,2,3- tris[α-	carbonylar	nino)methy	ylphenylan	ninocarbor	yloxy]-	

	amino)me ω-oxy- poly- (oxyethyle co-		laminocart	oonyl)-				
616-103-0	trans-4- (acetylam dihydro-6 methyl-7, dioxo-4H thieno[2,3 b]thiopyra sulfonami	- 7- - k- an-2-	415-030-3	3120298-3	8R643 N; R50-53	Xi; N R: 43-50/53 S: (2-)24-37	-60-61	
616-104-0 X	0e nalaxyl methyl N-(2,6-dimethylp (phenylac DL-alaninate	henyl)- <i>N</i> -	275-728-7	771626-11	-¥\; R50-53	N R: 50/53 S: 60-61		
616-105-0	00Morotolu 3-(3- chloro- <i>p</i> - tolyl)-1,1- dimethylu		239-592-2	215545-48	• Parc. Cat. 3; R40 Repr. Cat. 3; R63 N; R50-53	R: 40-63-50/ S:	/53 -26-46-60-	61
616-106-0	Menmedi methyl 3-(3- methylcar (ISO)	_	237-199-(xy)carbanil)13684-63 ate	- X I; R50-53	N R: 50/53 S: 60-61		
616-108-0	000dosulfu methyl- sodium	ron-		144550-3	6N7, R50-53	N R: 50/53 S: 60-61		
616-109-0	yl)-3-(2-	ypyrimidin nylimidazo 3-		141776-3	2NI; R50-53	N R: 50/53 S: 60-61		
616-110-0	03 2clanilid	e	419-150-7	7113136-7	7X9n; R22 N; R51-53	Xn; N R: 22-51/53		

dichloroanilinocarbonyl)	cyclopropanecarboxylic	S: (2-)61
616-111-06e8hexamid N-(2,3- dichlor-4- hydroxyphenyl)-1- methylcyclohexancarbox	-530-5126833-17A8 R51-53	N R: 51/53 S: 61
616-112-00-Assulfuron oxetan-3- yl 2- [(4,6- dimethylpyrimidin-2- yl)- carbamoylsulfamoyl]ben	144651-06X9h; R48/22 N; R50-53	Xn; N R: 48/22-50/53 S: (2-)46-60-61
616-113-0 des medipham ethyl 3-phenylcar bamoyloxyphe mate	-198-513684-56-N; R50-53 nylcarba	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
616-114-0 de tlecanamide, N,N'- (9,9', 10,10'- tetrahydro-9,9',10,10'- tetraoxo(1,1'- bianthracene)-4,4'- diyl)bis-	-010-2136897-58R 0 53	R: 53 S: 22-61
616-115-0N-(3- X acetyl-2- hydroxyphenyl)-4- (4- phenylbutoxy)benzamide	-150-9136450-06R153	R: 53 S: 61
616-116-0N5(4- dimethylaminopyridiniumethoxy-4- (1- methyl-5- nitroindol-3- ylmethyl)- N-(0- tolylsulfonyl)benzamidat		R: 53 S: 61

616-117-00NQ2-(3- acetyl-5- nitrothiophen-2- ylazo)-5- diethylaminophe		N; S:	N 2-50/53 22-36/37-60-61
616-118-0006(2',6'-dimethylphenyl) piperidinecarbox hydrochloride		R52-53 R: 22-5 S:	2/53
616-119-0 Q-(1- butyl-3,5- dioxo-2- phenyl- (1,2,4)- triazolidin-4- yl)-4,4- dimethyl-3- oxo- N-(2- methoxy-5- (2- (dodecyl-1- sulfonyl))propio phenyl)- pentanamide	418-060-5118020 nylamino)-	-93R253 R: 5 S: 6	
mixture of: N-(3- dimethylamino- methyl- phenyl)- benzamide	2-	R51-53 S:	N 2-51/53 36/37-61
616-121-0 Q-2 - dihydroxy- N-(2- methoxyphenyl)	419-090-1129205 benzamide	N; R51-53 R: 43-5 S:	
616-123-0 0 \-33- [[4-	414-740-096141-8	86-N; N R50-53 R: 5	0/53

	(diethylamino)-2- methylphenyl]imino oxo-1,4- cyclohexadienyljace	tamide		S: 60-61		
616-124-0)ในขาบท bis(trifluoromethyls	415-300-090 ulfonyl)imide	/	T R: 24/25-34- S: (1/2-)22-2	52/53 26-36/37/3	9-45-61
616-125-0	0- 4 yano- N-(1,1- dimethylethyl)andro diene-17- β- carboxamide	415-730-915 sta-3,5-	51338-1 IAS; R50-53	N R: 50/53 S: 60-61		
616-127-0	mixture of: N,N'- Ethane-1,2- diylbis(decanamide) 12- Hydroxy- N-[2-[1- oxydecyl)amino]eth N,N'- Ethane-1,2- diylbis(12- hydroxyoctadecanar	yl]octadecana	R43 N; R51-53	Xi; N R: 43-51/53 S: (2-)24-37	-61	
616-128-0	allyl-4,5- dicyanoimidazol-2- ylazo)-5- (dipropylamino)phe acetamide		23590-00Rb3	R: 53 S: 61		
616-129-0	bis(2,2,6,6- tetramethyl-4- piperidyl)isophthala		2774-15- X n; R22 Xi; R36	Xn R: 22-36 S: (2-)22-25	-26	
616-130-0	00H(3- (2-(4,4- dimethyl-2,5- dioxo- imidazolin-1- yl)-4,4- dimethyl-3- oxo- pentanoylamino)-4- methoxy-	421-780-215	50919-56 R5 3	R: 53 S: 61		

	phenyl)- octadecan	amide						
616-132-0	cyano-2- furfurylid dihydro-5 oxo-3-	- nyl]butane-		5130016-9	8N7, R50-53	N R: 50/53 S: 60-61		
616-133-0	cyclohexy S,S-	zo[b]tiophe		149118-6	6Xln; R22 Xi; R41 N; R50-53	Xn; N R: 22-41-50/ S: (2-)22-26	53 -39-60-61	
616-134-0	bis(diocty N,N'-	loxyphosp	•	thio)-	R52-53	R: 52/53 S: 61		
616-135-0	0(3 S ,4aS,8 [(2R,3S)- amino-2- hydroxy-4 phenylbut N-tert- butyldeca carboxam	3- 4- yl]- hydroisoqu)136522-1	7Xh; R22 R52-53	Xn R: 22-52/53 S: (2-)22-61		
616-142-0		sulfonylace		393629-90 opane	-Muta.Cat. R68 Xi; R41 R 43 R 52-53	R: 41-43-68- S:	-52/53 -36/37/39-	61
616-143-0	dihexaded N,N'- bis(2-	yl- thyl)propar		0149591-3	8X%n; Repr. Cat. 3; R62 Xi; R36 R53	Xn R: 62-36-53 S: (2-)26-36	/37-61	
617-018-0	mixture of: 1- methyl-1- (3-(1- methyleth methyl-1-	yl)phenyl) ylperoxide	ethyl-1-	371566-50	-D; R7 N; R51-53	O; N R: 7-51/53 S: (2-)3/7-14	1-36/37/39	-61

617-019-0	(4-(1-methylethyl)phenyl)methyl-1-phenylethylperoxide 31 % by weight 06-0 (phthalimido)peroxyacid	410-850-8128275-3	100 R7 Xi; R41 N; R50	O; Xi; N R: 7-41-50 S:
				(2-)3/7-14-26-36/37/39-61
617-020-0	di(prop-2,2- diyl)benzene bis(neodecanoylpero	420-060-5117663-1 exide)	IR310 O; R7 N; R51-53	O; N R: 7-10-51/53 S: (2-)7-14-36/37/39-47-61
650-042-0	product of: polyethylene- polyamine- (C 16- C18)- alkylamides with monothio- (C2)- alkyl phosphonates	417-450-2	Xi; R36/38 R43 R52-53	Xi R: 36/38-43-52/53 S: (2-)24-26-37-61
650-043-0 X	Reaction product of: 3,5-bis-tert-butylsalicylicacid and aluminiumsulfate	420-310-3	Xn; R22 N; R50-53	Xn; N R: 22-50/53 S: (2-)22-56-60-61
650-044-0	Inear and branched C14-15 alcohols ethoxylated, reaction product with epichlorohydrin	420-480-9158570-9	9Xii; R38 R43 N; R50-53	Xi; N R: 38-43-50/53 S: (2-)24-37-60-61
650-045-0	Reaction product of:	417-110-3	F; R11 Xi; R38-41	F; Xi; N R: 11-38-41-51/53

1,2,3- propanetr acid, 2- hydroxy, diethyl ester, 1- propanol and zirconium tetra-n- propanola				N; R51-53	S: (2-)9-16-2	26-37/39-6	1
disulfona	I- anin- N31,N32)dis			Xn; R22-48/22 N; R51-53	Xn; N 2R: 22-48/22- S: (2-)22-36		
650-047-0dibenzylphexafluor	henylsulfor 4 i oantimonate	นที•760-8	3134164-2	412 R48/25 Xn; R22 Xi; R41 R43 N; R51-53	S:	48/25-51/: 26-36/37/3	
650-048-0 Reaction product of: borax, hydrogen peroxide, acetic acid anhydride and acetic acid		20-070-1			O; C; N R: 27-20/21/2 S: (1/2-)3/7-		37/39-45-61
alkoyloxy hydrogen maleate, where alkoyl represents (by weight) 70 to 85% unsaturate octadecoy 0.5 to	ethyl	17-960-5		Xi; R38-41 R43 N; R50-53	Xi; N R: 38-41-43- S: (2-)24-26	-50/53 -37/39-60-	61

	saturated octadecoy and 2 to 18% saturated hexadecoy						
650-050-0	mixture of: 1- methyl-3- hydroxyp 3,5-[1,1- dimethyle hydroxyd cinnamate and/ or 3- hydroxyb [1,1- dimethyle hydroxyd 1,3- butanedio bis[3- (3'-(1,1- dimethyle hydroxy- phenyl)pr isomers 1,3- butanedio bis[3(3',5') (1,1- dimethyle	ropyl thyl]-4- ihydro- utyl3,5- thyl]-4- ihydrocinn l thyl)4'- opionate]			N; R51-53	N R: 51/53 S: 61	
650-055-0	sodium zirconium	n phosphate	422-570-3	3	N; R50-53	N R: 50/53 S: 60-61	
048-002-0	00a01mium (non- pyrophori [1] cadmium oxide (non- pyrophori [2]	c)	[1]	87440-43-9 [1] 21306-19-0 [2]	PCarc. Cat 2; R45)Muta. Cat. 3; R68 Repr. Cat. 3; R62-63	R:	53-68-50/53

		T; R48/23/2 T+; R26 N; R50-53	
048-011-00admium E X (pyrophoric)	231-152-87440-43-9	9Carc. Cat 2; R45 Muta. Cat. 3; R68 Repr. Cat. 3; R62-63 T; R48/23/2 T+; R26 F; R17 N; R50-53	R: 45-17-26-48/23/25-62-63-68-50/53 S: 53-45-7/8-43-60-61
609-006-00-3 C nitrotoluene	202-808-099-99-0	T; R23/24/2 R33 N; R51/53	T; N 5R: 23/24/25-33-51/53 S: (1/2-)28-37-45-61
609-065-0 Q- 5 E nitrotoluene	201-853-388-72-2	Carc. Cat 2; R45 Muta. Cat. 2; R46 Repr. Cat. 3; R62 Xn; R22 N; R51-53	T; N R: 45-46-22-62-51/53 S: 53-45-61
612-039-00-6 C ethoxyaniline o- phenetidine	202-356-494-70-2	T; R23/24/2 R33	T 5R: 23/24/25-33 S: (1/2-)28-36/37-45
612-207-0 6- 9 ethoxyaniline <i>p</i> - phenetidine	205-855-5156-43-4	Cat. 3; R68 Xn;	Xn R: 20/21/22-36-43-68 S: 2(2-)36/37-46

ANNEX2A U.K.

A.21.OXIDIZING PROPERTIES (LIQUIDS)

1. **METHOD** U.K.

1.1 INTRODUCTION U.K.

This test method is designed to measure the potential for a liquid substance to increase the burning rate or burning intensity of a combustible substance, or to form a mixture with a combustible substance which spontaneously ignites, when the two are thoroughly mixed. It is based on the UN test for oxidizing liquids (1) and is equivalent to it. However, as this method A.21 is primarily designed to satisfy the requirements of Dir 67/548, comparison with only one reference substance is required. Testing and comparison to additional reference substances may be necessary when the results of the test are expected to be used for other purposes.⁽³⁾

This test need not be performed when examination of the structural formula establishes beyond reasonable doubt that the substance is incapable of reacting exothermically with a combustible material.

It is useful to have preliminary information on any potential explosive properties of the substance before performing this test.

This test is not applicable to solids, gases, explosive or highly flammable substances, or organic peroxides.

This test may not need be performed when results for the test substance in the UN test for oxidizing liquids (1) are already available.

1.2 DEFINITIONS AND UNITS U.K.

Mean pressure rise time is the mean of the measured times for a mixture under test to produce a pressure rise from 690 kPa to 2070 kPa above atmospheric.

1.3 REFERENCE SUBSTANCE U.K.

65% (w/w) aqueous nitric acid (analytical grade) is required as a reference substance. (4)

Optionally, if the experimenter foresees that the results of this test may eventually be used for other purposes, testing of additional reference substances may also be appropriate. (5)

1.4 PRINCIPLE OF THE TEST METHOD U.K.

The liquid to be tested is mixed in a 1 to 1 ratio, by mass, with fibrous cellulose and introduced into a pressure vessel. If during mixing or filling spontaneous ignition occurs, no further testing is necessary.

If spontaneous ignition does not occur the full test is carried out. The mixture is heated in a pressure vessel and the mean time taken for the pressure to rise from 690 kPa to 2070 kPa above atmospheric is determined. This is compared with the mean pressure rise time for the 1:1 mixture of the reference substance(s) and cellulose.

1.5 QUALITY CRITERIA U.K.

In a series of five trials on a single substance no results should differ by more than 30 % from the arithmetic mean. Results that differ by more than 30 % from the mean should be discarded, the mixing and filling procedure improved and the testing repeated,

1.6 DESCRIPTION OF THE METHOD U.K.

1.6.1 **Preparation U.K.**

1.6.1.1 *Combustible substance* U.K.

Dried, fibrous cellulose with a fibre length between 50 and 250 μ m and a mean diameter of 25 μ m, ⁽⁶⁾ is used as the combustible material. It is dried to constant weight in a layer not more than 25 mm thick at 105 °C for 4 hours and kept in a desiccator, with desiccant, until cool and required for use. The water content of the dried cellulose should be less than 0.5% by dry mass⁽⁷⁾. If necessary, the drying time should be prolonged to achieve this. ⁽⁸⁾ The same batch of cellulose is to be used throughout the test.

1.6.1.2 Apparatus U.K.

1.6.1.2.1 Pressure vessel

A pressure vessel is required. The vessel consists of a cylindrical steel pressure vessel 89 mm in length and 60 mm in external diameter (see figure 1). Two flats are machined on opposite sides (reducing the cross-section of the vessel to 50 mm) to facilitate holding whilst fitting up the firing plug and vent plug. The vessel, which has a bore of 20 mm diameter is internally rebated at either end to a depth of 19 mm and threaded to accept 1" British Standard Pipe (BSP) or metric equivalent. A pressure take-off, in the form of a side arm, is screwed into the curved face of the pressure vessel 35 mm from one end and at 90° to the machined flats. The socket for this is bored to a depth of 12 mm and threaded to accept the 1/2" BSP (or metric equivalent) thread on the end of the side-arm. If necessary, an inert seal is fitted to ensure a gas-tight seal. The side-arm extends 55 mm beyond the pressure vessel body and has a bore of 6 mm. The end of the side-arm is rebated and threaded to accept a diaphragm type pressure transducer. Any pressure-measuring device may be used provided that it is not affected by the hot gases or the decomposition products and is capable of responding to rates of pressure rise of 690-2070 kPa in not more than 5 ms.

The end of the pressure vessel farthest from the side-arm is closed with a firing plug which is fitted with two electrodes, one insulated from, and the other earthed to, the plug body. The other end of the pressure vessel is closed by a bursting disk (bursting pressure approximately 2200 kPa) held in place with a retaining plug which has a 20 mm bore. If necessary, an inert seal is used with the firing plug to ensure a gas-tight fit. A support stand (figure 2) holds the assembly in the correct attitude during use. This usually comprises a mild steel base plate measuring 235 mm x 184 mm x 6 mm and a 185 mm length of square hollow section (S.H.S.) 70 mm x 70 mm x 4 mm.

A section is cut from each of two opposite sides at one end of the length of S.H.S. so that a structure having two flat sided legs surmounted by 86 mm length of intact box section results. The ends of these flat sides are cut to an angle of 60° to the horizontal and welded to the base plate. A slot measuring 22 mm wide x 46 mm deep is machined in one side of the upper end of the base section such that when the pressure vessel assembly is lowered, firing plug end first, into the box section support, the side-arm is accommodated in the slot. A piece of steel 30 mm wide and 6 mm thick is welded to the lower internal face of the box section to act as a spacer. Two 7 mm thumb screws, tapped into the opposite face, serve to hold the pressure vessel firmly in place. Two 12 mm wide strips of 6 mm thick steel, welded to the side pieces abutting the base of the box section, support the pressure vessel from beneath.

1.6.1.2.2 Ignition System

The ignition system consists of a 25 cm long Ni/Cr wire with a diameter 0.6 mm and a resistance of 3.85 ohm/m. The wire is wound, using a 5 mm diameter rod, in the shape of a coil and is attached to the firing plug electrodes. The coil should have one of the configurations shown in figure 3. The distance between the bottom of the vessel and the underside of the ignition coil should be 20 mm. If the electrodes are not adjustable, the ends of the ignition wire between the coil and the bottom of the vessel should be insulated by a ceramic sheath. The wire is heated by a constant current power supply able to deliver at least 10 A.

1.6.2 **Performance of the test**⁽⁹⁾ U.K.

The apparatus, assembled complete with pressure transducer and heating system but without the bursting disk in position, is supported firing plug end down. 2.5 g of the liquid to be tested is mixed with 2.5 g of dried cellulose in a glass beaker using a glass stirring rod⁽¹⁰⁾. For safety, the mixing should be performed with a safety shield between the operator and mixture. If the mixture ignites during mixing or filling, no further testing is necessary. The mixture is added, in small portions with tapping, to the pressure vessel making sure that the mixture is packe4 around the ignition coil and is in good contact with it. It is important that the coil is not distorted during the packing process as this may lead to erroneous results⁽¹¹⁾. The bursting disk is placed in position and the retaining plug is screwed in tightly. The charged vessel is transferred to the firing support stand, bursting disk uppermost, which should be located in a suitable, armoured fume cupboard or firing cell. The power supply is connected to the external terminals of the firing plug and 10 A applied. The time between the start of mixing and switching on the power should not exceed 10 minutes.

The signal produced by the pressure transducer is recorded on a suitable system which allows both evaluation and the generation of a permanent record of the time pressure profile obtained (e.g. a transient recorder coupled to a chart recorder). The mixture is heated until the bursting disk ruptures or until at least 60 s have elapsed. If the bursting disk does not rupture, the mixture should be allowed to cool before carefully dismantling the apparatus, taking precautions to allow for any pressurization which may occur. Five trials are performed with the test substance and the reference substance(s). The time taken for the pressure to rise from 690 kPa to 2070 kPa above atmospheric is noted. The mean pressure rise time is calculated.

In some cases, substances may generate a pressure rise (too high or too low), caused by chemical reactions not characterizing the oxidizing properties of the substance. In these cases, it may be necessary to repeat the test with an inert substance, e.g. diatomite (kieselguhr), in place of the cellulose in order to clarify the nature of the reaction.

2 DATA U.K.

Pressure rise times for both the test substance and the reference substance(s). Pressure rise times for the tests with an inert substance, if performed.

2.1 TREATMENT OF RESULTS U.K.

The mean pressure rise times for both the test substance and the reference substances(s) are calculated.

The mean pressure rise time for the tests with an inert substance (if performed) is calculated.

Some examples of results are shown in Table 1

TABLE 1

Examples of results⁰

Substance ^c	Mean pressure rise time for a 1:1 mixture with celulose (ms)			
Ammonium dichromate, saturated aqueous solution	20800			
Calcium nitrate, saturated aqueous solution	6700			
Ferric nitrate, saturated aqueous solution	4133			
Lithium perchlorate, saturated aqueous solution	1686			
Magnesium perchlorate, saturated aqueous solution	777			
Nickel nitrate, saturated aqueous solution	6250			
Nitric acid, 65 %	4767ª			
Perchloric acid, 50 %	121ª			
Perchloric acid, 55 %	59			
Potassium nitrate, 30 % aqueous solution	26690			
Silver nitrate, saturated aqueous solution	b			
Sodium chlorate, 40 % aqueous solution	2555ª			
Sodium nitrate, 45 % aqueous solution	4133			
Inert Substance				
Water:cellulose	b			
a Mean value from interlaboratory comparative trials				
b Maximum pressure of 2070 kPa not reached				
c Saturated solutions should be prepared at 20 °C				
d See reference (1) for classification under the UN transpo	rt scheme,			

3 REPORT U.K.

3.1 TEST REPORT U.K.

The test report should include the following information:

- the identity, composition, purity, etc of the substance tested;
- the concentration of the test substance;
- the drying procedure of the cellulose used
- the water content of the cellulose used
- the results of the measurements;
- the results from tests with an inert substance, if any;
- the calculated mean pressure rise times;
- any deviations from this method and the reasons for them;

- all additional information or remarks relevant to the interpretation of the results;
- 3.2 INTERPRETATION OF THE RESULTS⁽¹²⁾ U.K.

The test results are assessed on the basis of:

- a) whether the mixture of test substance and cellulose spontaneously ignites; and
- b) the comparison of the mean time taken for the pressure to rise from 690 kPa to 2070 kPa with that of the reference substance(s).

A liquid substance is to be considered as an oxidizer when:

- a) a 1:1 mixture, by mass, of the substance and cellulose spontaneously ignites; or
- b) a 1:1 mixture, by mass, of the substance and cellulose exhibits a mean pressure rise time less than or equal to the mean pressure rise time of a 1:1 mixture, by mass, of 65% (w/w) aqueous nitric acid and cellulose.

In order to avoid a false positive result, if necessary, the results obtained when testing the substance with an inert material should also be considered when interpreting the results.

4 **REFERENCES** U.K.

(1) Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria. 3rd revised edition. UN Publication No: ST/SG/AC.10/11/Rev. 3, 1999, page 342. Test 0.2: Test for oxidizing liquids.

Figure

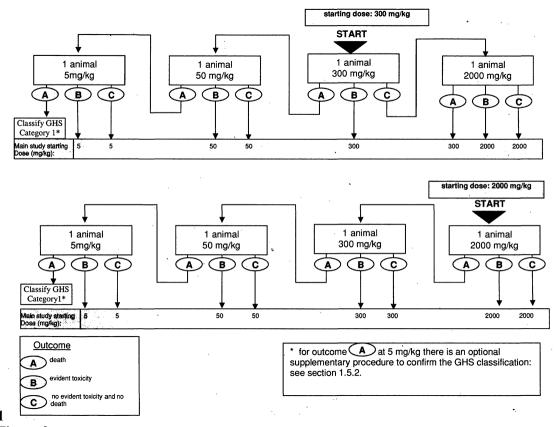


Figure 2

Support stand

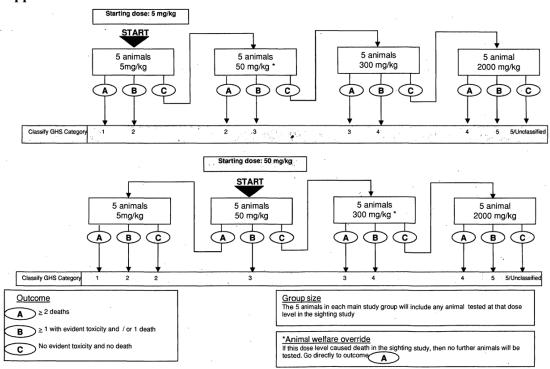
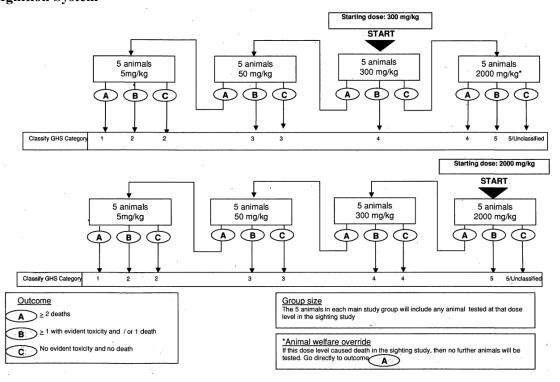


Figure 3

Ignition System



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ANNEX 2B U.K.

B.1 bis.ACUTE ORAL TOXICITY - FIXED DOSE PROCEDURE

1. **METHOD** U.K.

This test method is equivalent to OECD TG 420 (2001)

1.1 INTRODUCTION U.K.

Traditional methods for assessing acute toxicity use death of animals as an endpoint. In 1984, a new approach to acute toxicity testing was suggested by the British Toxicology Society based on the administration at a series of fixed dose levels (1). The approach avoided using death of animals as an endpoint, and relied instead on the observation of clear signs of toxicity at one of a series of fixed dose levels. Following UK (2) and international (3) *in vivo* validation studies the procedure was adopted as a testing method in 1992. Subsequently, the statistical properties of the Fixed Dose Procedure have been evaluated using mathematical models in a series of studies (4)(5)(6). Together, the *in vivo* and modelling studies have demonstrated that the procedure is reproducible, uses fewer animals and causes less suffering than the traditional methods and is able to rank substances in a similar manner to the other acute toxicity testing methods.

Guidance on the selection of the most appropriate test method for a given purpose can be found in the Guidance Document on Acute Oral Toxicity Testing (7). This Guidance Document also contains additional information on the conduct and interpretation of Testing Method B.1 bis.

It is a principle of the method that in the main study only moderately toxic doses are used, and that administration of doses that are expected to be lethal should be avoided. Also, doses that are known to cause marked pain and distress, due to corrosive or severely irritant actions, need not be administered. Moribund animals, or animals obviously in pain or showing signs of severe and enduring distress shall be humanely killed, and are considered in the interpretation of the test results in the same way as animals that died on test. Criteria for making the decision to kill moribund or severely suffering animals, and guidance on the recognition of predictable or impending death, are the subject of a separate Guidance Document (8).

The method provides information on the hazardous properties and allows the substance to be ranked and classified according to the Globally Harmonised System (GHS) for the classification of chemicals which cause acute toxicity (9).

The testing laboratory should consider all available information on the test substance prior to conducting the study. Such information will include the identity and chemical structure of the substance; its physico-chemical properties; the results of any other *in vitro* or *in vivo* toxicity tests on the substance; toxicological data on structurally related substances; and the anticipated use(s) of the substance. This information is necessary to satisfy all concerned that the test is relevant for the protection of human health, and will help in the selection of an appropriate starting dose.

1.2 DEFINITIONS U.K.

Acute oral toxicity: refers to those adverse effects occurring following oral administration of a single dose of a substance or multiple doses given within 24 hours.

Delayed death: means that an animal does not die or appear moribund within 48 hours but dies later during the 14-day observation period.

Dose: is the amount of test substance administered. Dose is expressed as weight of test substance per unit weight of test animal (e.g. mg/kg).

Evident toxicity: is a general term describing clear signs of toxicity following the administration of test substance (see (3) for examples) such that at the next highest fixed dose either severe pain and enduring signs of severe distress, moribund status (criteria are presented in the Humane Endpoints Guidance Document (8)), or probable mortality in most animals can be expected.

GHS: Globally Harmonised Classification System for Chemical Substances and Mixtures. A joint activity of OECD (human health and the environment), UN Committee of Experts on Transport of Dangerous Goods (physical-chemical properties) and ILO (hazard communication) and co-ordinated by the Interorganisation Programme for the Sound Management of Chemicals (IOMC).

Impending death: when moribund state or death is expected prior to the next planned time of observation. Signs indicative of this state in rodents could include convulsions, lateral position, recumbence, and tremor. (See the Humane Endpoint Guidance Document (8) for more details).

 LD_{50} (median lethal dose): is a statistically derived single dose of a substance that can be expected to cause death in 50 per cent of animals when administered by the oral route. The LD_{50} value is expressed in terms of weight of test substance per unit weight of test animal (mg/kg).

Limit dose: refers to a dose at an upper limitation on testing (2000 or 5000 mg/kg).

Moribund status: being in a state of dying or inability to survive, even if treated. (See the Humane Endpoint Guidance Document (8) for more details).

Predictable death: presence of clinical signs indicative of death at a known time in the future before the planned end of the experiment, for example: inability to reach water or food. (See the Humane Endpoint Guidance Document (8) for more details).

1.3 PRINCIPLE OF THE TEST METHOD U.K.

Groups of animals of a single sex are dosed in a stepwise procedure using the fixed doses of 5, 50, 300 and 2000 mg/kg (exceptionally an additional fixed dose of 5000 mg/kg may be considered, see section 1.6.2). The initial dose level is selected on the basis of a sighting study as the dose expected to produce some signs of toxicity without causing severe toxic effects or mortality. Clinical signs and conditions associated with pain, suffering, and impending death, are described in detail in a separate OECD Guidance Document (8). Further groups of animals may be dosed at higher or lower fixed doses, depending on the presence or absence of signs of toxicity or mortality. This procedure continues until the dose causing evident toxicity or no more than one death is identified, or when no effects are seen at the highest dose or when deaths occur at the lowest dose.

1.4 DESCRIPTION OF THE TEST METHOD U.K.

1.4.1 Selection of animal species U.K.

The preferred rodent species is the rat, although other rodent species may be used. Normally females are used (7). This is because literature surveys of conventional LD50 tests show that usually there is little difference in sensitivity between the sexes, but in those cases where differences are observed, females are generally slightly more sensitive (10). However, if knowledge of the toxicological or toxicokinetic properties of structurally related chemicals indicates that males are likely to be more sensitive then this sex should be used. When the test is conducted in males, adequate justification should be provided.

Healthy young adult animals of commonly used laboratory strains should be employed. Females should be nulliparous and non-pregnant. Each animal, at the commencement of its dosing,

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should be between 8 and 12 weeks old and its weight should fall in an interval within \pm 20% of the mean weight of any previously dosed animals.

1.4.2 Housing and feeding conditions U.K.

The temperature of the experimental animal room should be $22^{\circ}C$ ($\pm 3^{\circ}C$). Although the relative humidity should be at least 30% and preferably not exceed 70% other than during room cleaning the aim should be 50-60%. Lighting should be artificial, the sequence being 12 hours light, 12 hours dark. For feeding, conventional laboratory diets may be used with an unlimited supply of drinking water. Animals may be group-caged by dose, but the number of animals per cage must not interfere with clear observations of each animal.

1.4.3 **Preparation of animals U.K.**

The animals are randomly selected, marked to permit individual identification, and kept in their cages for at least 5 days prior to the start of dosing to allow for acclimatisation to the laboratory conditions.

1.4.4 **Preparation of doses U.K.**

In general test substances should be administered in a constant volume over the range of doses to be tested by varying the concentration of the dosing preparation. Where a liquid end product or mixture is to be tested however, the use of the undiluted test substance, i.e. at a constant concentration, may be more relevant to the subsequent risk assessment of that substance, and is a requirement of some regulatory authorities. In either case, the maximum dose volume for administration must not be exceeded. The maximum volume of liquid that can be administered at one time depends on the size of the test animal. In rodents, the volume should not normally exceed 1ml/100g of body weight: however in the case of aqueous solutions 2 ml/100g body weight can be considered. With respect to the formulation of the dosing preparation, the use of an aqueous solution/suspension/emulsion is recommended wherever possible, followed in order of preference by a solution/suspension/emulsion in oil (e.g. com oil) and then possibly solution in other vehicles. For vehicles other than water the toxicological characteristics of the vehicle should be known. Doses must be prepared shortly prior to administration unless the stability of the preparation over the period during which it will be used is known and shown to be acceptable.

1.5 PROCEDURE U.K.

1.5.1 Administration of doses U.K.

The test substance is administered in a single dose by gavage using a stomach tube or a suitable intubation canula. In the unusual circumstance that a single dose is not possible, the dose may be given in smaller fractions over a period not exceeding 24 hours.

Animals should be fasted prior to dosing (e.g. with the rat, food but not water should be withheld over-night; with the mouse, food but not water should be withheld for 3-4 hours). Following the period of fasting, the animals should be weighed and the test substance administered. After the substance has been administered, food may be withheld for a further 3-4 hours in rats or 1-2 hours in mice. Where a dose is administered in fractions over a period of time, it may be necessary to provide the animals with food and water depending on the length of the period.

1.5.2 **Sighting study** U.K.

The purpose of the sighting study is to allow selection of the appropriate starting dose for the main study. The test substance is administered to single animals in a sequential manner following the flow charts in Annex 1. The sighting study is completed when a decision on the starting dose for the main study can be made (or if a death is seen at the lowest fixed dose).

The starting dose for the sighting study is selected from the fixed dose levels of 5, 50, 300 and 2000 mg/kg as a dose expected to produce evident toxicity based, when possible, on evidence from *in vivo* and *in vitro* data from the same chemical and from structurally related chemicals. In the absence of such information, the starting dose will be 300 mg/kg.

A period of at least 24 hours will be allowed between the dosing of each animal. All animals should be observed for at least 14 days.

Exceptionally, and only when justified by specific regulatory needs, the use of an additional upper fixed dose level of 5000 mg/kg may be considered (see Annex 3). For reasons of animal welfare concern, testing of animals in GHS Category 5 ranges (2000-5000 mg/kg is discouraged and should only be considered when there is a strong likelihood that the results of such a test have a direct relevance for protecting human or animal health or the environment.

In cases where an animal tested at the lowest fixed dose level (5mg/kg) in the sighting study dies, the normal procedure is to terminate the study and assign the substance to GHS Category 1 (as shown in Annex 1). However, if further confirmation of the classification is required, an optional supplementary procedure may be conducted, as follows. A second animal is dosed at 5mg/kg. If this second animal dies, then GHS Category 1 will be confirmed and the study will be immediately terminated. If the second animal survives, then a maximum of three additional animals will be dosed at 5mg/kg. Because there will be a high risk of mortality, these animals should be dosed in a sequential manner to protect animal welfare. The time interval between dosing each animal should be sufficient to establish that the previous animal is likely to survive. If a second death occurs, the dosing sequence will be immediately terminated and no further animals will be dosed. Because the occurrence of a second death (irrespective of the number of animals tested at the time of termination) falls into outcome A (2 or more deaths), the classification rule of Annex 2 at the 5mg/kg fixed dose is followed (Category 1 if there are 2 or more deaths or Category 2 if there is no more than 1 death). In addition, Annex 4 gives guidance on the classification in the EU system until the new GHS is implemented.

1.5.3 **Main study** U.K.

1.5.3.1 Numbers of animals and dose levels

The action to be taken following testing at the starting dose level is indicated by the flow charts in Annex 2. One of three actions will be required; either stop testing and assign the appropriate hazard classification class, test at a higher fixed dose or test at a lower fixed dose. However, to protect animals, a dose level that caused death in the sighting study will not be revisited in the main study (see Annex 2). Experience has shown that the most likely outcome at the starting dose level will be that the substance can be classified and no further testing will be necessary.

A total of five animals of one sex will normally be used for each dose level investigated. The five animals will be made up of one animal from the sighting study dosed at the selected dose level together with an additional four animals (except, unusually, if a dose level used on the main study was not included in the sighting study).

The time interval between dosing at each level is determined by the onset, duration, and severity of toxic signs. Treatment of animals at the next dose should be delayed until one is confident of survival of the previously dosed animals. A period of 3 or 4 days between dosing at each dose level is recommended, if needed, to allow for the observation of delayed toxicity. The time interval may be adjusted as appropriate, e.g., in case of inconclusive response.

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When the use of an upper fixed dose of 5000 mg/kg is considered, the procedure outlined in Annex 3 should be followed (see also section 1.6.2).

1.5.3.2 Limit test

The limit test is primarily used in situations where the experimenter has information indicating that the test material is likely to be nontoxic, i.e., having toxicity only above regulatory limit doses. Information about the toxicity of the test material can be gained from knowledge about similar tested compounds or similar tested mixtures or products, taking into consideration the identity and percentage of components known to be of toxicological significance. In those situations where there is little or no information about its toxicity, or in which the test material is expected to be toxic, the main test should be performed.

Using the normal procedure, a sighting study starting dose of 2000 mg/kg (or exceptionally 5000 mg/kg) followed by dosing of a further four animals at this level serves as a limit test for this guideline.

1.6 OBSERVATIONS U.K.

Animals are observed individually after dosing at least once during the first 30 minutes, periodically during the first 24 hours, with special attention given during the first 4 hours, and daily thereafter, for a total of 14 days, except where they need to be removed from the study and humanely killed for animal welfare reasons or are found dead. However, the duration of observation should not be fixed rigidly. It should be determined by the toxic reactions, time of onset and length of recovery period, and may thus be extended when considered necessary. The times at which signs of toxicity appear and disappear are important, especially if there is a tendency for toxic signs to be delayed (11). All observations are systematically recorded, with individual records being maintained for each animal.

Additional observations will be necessary if the animals continue to display signs of toxicity. Observations should include changes in skin and fur, eyes and mucous membranes, and also respiratory, circulatory, autonomic and central nervous systems, and somatomotor activity and behaviour pattern. Attention should be directed to observations of tremors, convulsions, salivation, diarrhoea, lethargy, sleep and coma. The principles and criteria summarised in the Humane Endpoints Guidance Document should be taken into consideration (8). Animals found in a moribund condition and animals showing severe pain or enduring signs of severe distress should be humanely killed. When animals are killed for humane reasons or found dead, the time of death should be recorded as precisely as possible.

1.6.1 **Body weight** U.K.

Individual weights of animals should be determined shortly before the test substance is administered and at least weekly thereafter. Weight changes should be calculated and recorded. At the end of the test surviving animals are weighed and then humanely killed.

1.6.2 **Pathology** U.K.

All test animals (including those that die during the test or are removed from the study for animal welfare reasons) should be subjected to gross necropsy. All gross pathological changes should be recorded for each animal. Microscopic examination of organs showing evidence of gross pathology in animals surviving 24 or more hours after the initial dosing may also be considered because it may yield useful information.

2 DATA U.K.

Individual animal data should be provided. Additionally, all data should be summarised in tabular form, showing for each test group the number of animals used, the number of animals displaying signs of toxicity, the number of animals found dead during the test or killed for humane reasons, time of death of individual animals, a description and the time course of toxic effects and reversibility, and necropsy findings.

3 **REPORTING** U.K.

3.1 **Test report** U.K.

The test report must include the following information, as appropriate:

Test substance:

- physical nature, purity, and, where relevant, physico-chemical properties (including isomerisation);
- identification data, including CAS number.

Vehicle (if appropriate):

— justification for choice of vehicle, if other than water.

Test animals:

- species/strain used;
- microbiological status of the animals, when known;
- number, age and sex of animals (including, where appropriate, a rationale for use of males instead of females);
- source, housing conditions, diet etc.;

Test conditions:

- details of test substance formulation, including details of the physical form of the material administered;
- details of the administration of the test substance including dosing volumes and time of dosing;
- details of food and water quality (including diet type/source, water source);
- the rationale for the selection of the starting dose.

Results:

- tabulation of response data and dose level for each animal (i.e. animals showing signs of toxicity including mortality, nature, severity and duration of effects);
- tabulation of body weight and body weight changes;
- individual weights of animals at the day of dosing, in weekly intervals thereafter, and at time of death or sacrifice;
- date and time of death if prior to scheduled sacrifice.
- time course of onset of signs of toxicity and whether these were reversible for each animal;
- necropsy findings and histopathological findings for each animal; if available.

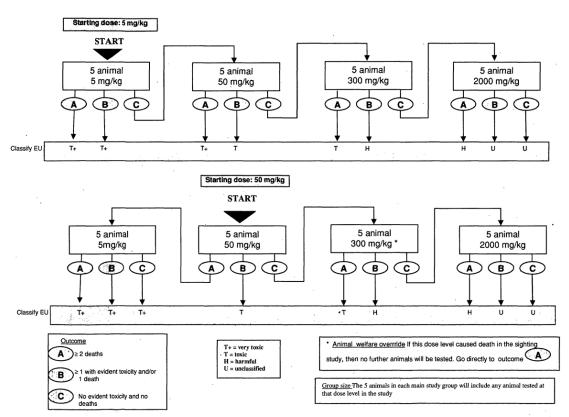
Discussion and interpretation of results.

Conclusions.

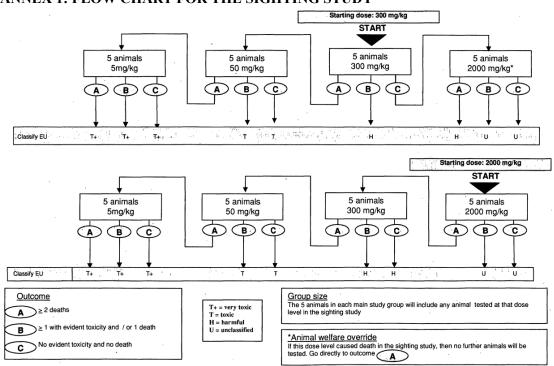
4 **REFERENCES** U.K.

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ANNEX 1: FLOW CHART FOR THE SIGHTING STUDY



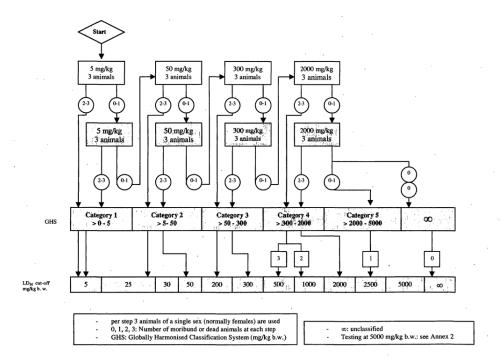
ANNEX 1: FLOW CHART FOR THE SIGHTING STUDY



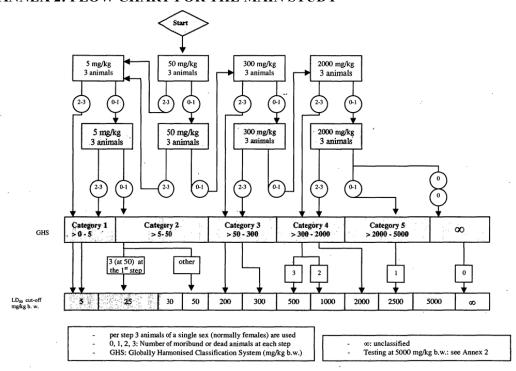
ANNEX 2: FLOW CHART FOR THE MAIN STUDY

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ANNEX 2: FLOW CHART FOR THE MAIN STUDY



ANNEX 3 CRITERIA FOR CLASSIFICATION OF TEST SUBSTANCES WITH EXPECTED LD₅₀ VALUES EXCEEDING 2000 MG/KG WITHOUT THE NEED FOR TESTING.

Criteria for hazard Category 5 are intended to enable the identification of test substances which are of relatively low acute toxicity hazard but which, under certain circumstances may present a danger to vulnerable populations. These substances are anticipated to have an oral or dermal

 LD_{50} in the range of 2000-5000 mg/kg or equivalent doses for other routes. Test substances could be classified in the hazard category defined by: 2000mg/kg <LD₅₀ < 5000mg/kg (Category 5 in the GHS) in the following cases:

- a) if directed to this category by any of the testing schemes of Annex 2, based on mortality incidences
- b) if reliable evidence is already available that indicates the LD⁵⁰ to be in the range of Category 5 values; or other animal studies or toxic effects in humans indicate a concern for human health of an acute nature.
- c) through extrapolation, estimation or measurement of data if assignment to a more hazardous class is not warranted, and
 - reliable information is available indicating significant toxic effects in humans, or
 - any mortality is observed when tested up to Category 4 values by the oral route, or
 - where expert judgement confirms significant clinical signs of toxicity, when tested up to Category 4 values, except for diarrhoea, piloerection or an ungroomed appearance, or
 - where expert judgement confirms reliable information indicating the potential for significant acute effects from the other animal studies.

TESTING AT DOSES ABOVE 2000 MG/KG

Exceptionally, and only when justified by specific regulatory needs, the use of an additional upper fixed dose level of 5000 mg/kg may be considered. Recognising the need to protect animal welfare, testing at 5000 mg/kg is discouraged and should only be considered when there is a strong likelihood that the results of such a test would have a direct relevance for protecting animal or human health (9).

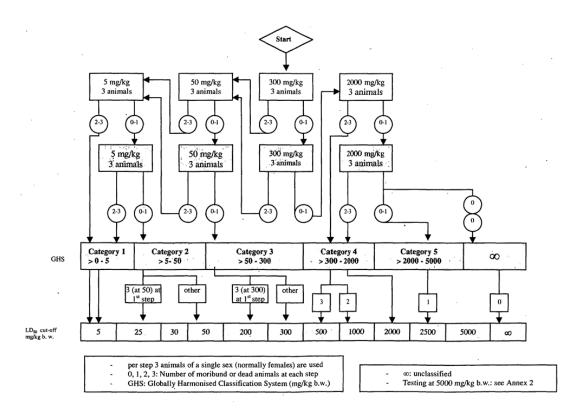
Sighting Study

The decision rules governing the sequential procedure presented in Annex 1 are extended to include a 5000 mg/kg dose level. Thus, when a sighting study starting dose of 5000 mg/kg is used outcome A (death) will require a second animal to be tested at 2000 mg/kg; outcomes B and C (evident toxicity or no toxicity) will allow the selection of 5000 mg/kg as the main study starting dose. Similarly, if a starting dose other than 5000 mg/kg is used then testing will progress to 5000 mg/kg in the event of outcomes B or C at 2000 mg/kg; a subsequent 5000 mg/kg outcome A will dictate a main study starting dose of 2000 mg/kg and outcomes B and C will dictate a main study starting dose of 5000 mg/kg.

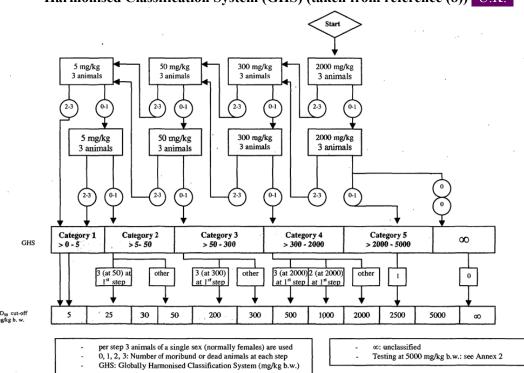
Main Study

The decision rules governing the sequential procedure presented in Annex 2 are extended to include a 5000 mg/kg dose level. Thus, when a main study starting dose of 5000 mg/kg is used, outcome A (≥2 deaths) will require the testing of a second group at 2000 mg/kg; outcome B (evident toxicity and/or ≤1 death) or C (no toxicity) will result in the substance being unclassified according to GHS. Similarly, if a starting dose other than 5000 mg/kg is used then testing will progress to 5000 mg/kg in the event of outcome C at 2000 mg/kg; a subsequent 5000 mg/kg outcome A will result in the substance being assigned to GHS Category 5 and outcomes B or C will lead to the substance being unclassified.

ANNEX TEST METHOD B.1 bis - Guidance on classification according to the EU scheme to cover the transition period until full implementation of the Globally Harmonised Classification System (GHS) (taken from reference (8)) U.K.



ANNEX TEST METHOD B. 1 bis - Guidance on classification according to the EU scheme to cover the transition period until full implementation of the Globally Harmonised Classification System (GHS) (taken from reference (8)) U.K.



ANNEX2C U.K.

B.1 tris.ACUTE ORAL TOXICITY - ACUTE TOXIC CLASS METHOD

1. **METHOD** U.K.

This test method is equivalent to OECD TG 423 (2001)

1.1 INTRODUCTION U.K.

The acute toxic class method (1) set out in this test is a stepwise procedure with the use of 3 animals of a single sex per step. Depending on the mortality and/or the moribund status of the animals, on average 2-4 steps may be necessary to allow judgement on the acute toxicity of the test substance. This procedure is reproducible, uses very few animals and is able to rank substances in a similar manner to the other acute toxicity testing methods. The acute toxic class method is based on biometric evaluations (2)(3)(4)(5) with fixed doses, adequately separated to enable a substance to be ranked for classification purposes and hazard assessment. The method as adopted in 1996 was extensively validated *in vivo* against LD₅₀ data obtained from the literature, both nationally (6) and internationally (7).

Guidance on the selection of the most appropriate test method for a given purpose can be found in the Guidance Document on Acute Oral Toxicity Testing (8). This Guidance Document also contains additional information on the conduct and interpretation of testing method B.1tris.

Test substances, at doses that are known to cause marked pain and distress due to corrosive or severely irritant actions, need not be administered. Moribund animals, or animals obviously in pain or showing signs of severe and enduring distress shall be humanely killed, and are considered in the interpretation of the test results in the same way as animals that died on test. Criteria for making the decision to kill moribund or severely suffering animals, and guidance on the recognition of predictable or impending death, are the subject of a separate Guidance Document (9).

The method uses pre-defined doses and the results allow a substance to be ranked and classified according to the Globally Harmonised System for the classification of chemicals which cause acute toxicity (10).

In principle, the method is not intended to allow the calculation of a precise LD50, but does allow for the determination of defined exposure ranges where lethality is expected since death of a proportion of the animals is still the major endpoint of this test. The method allows for the determination of an LD50 value only when at least two doses result in mortality higher than 0% and lower than 100%. The use of a selection of pre-defined doses, regardless of test substance, with classification explicitly tied to number of animals observed in different states improves the opportunity for laboratory to laboratory reporting consistency and repeatability.

The testing laboratory should consider all available information on the test substance prior to conducting the study. Such information will include the identity and chemical structure of the substance; its physico-chemical properties; the result of any other *in vivo* or *in vitro* toxicity tests on the substance; toxicological data on the structurally related substances; and the anticipated use(s) of the substance. This information is necessary to satisfy all concerned that the test is relevant for the protection of human health and will help in the selection of the most appropriate starting dose.

1.2 DEFINITIONS U.K.

Acute oral toxicity: refers to those adverse effects occurring following oral administration of a single dose of a substance or multiple doses given within 24 hours.

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Delayed death: means that an animal does not die or appear moribund within 48 hours but dies later during the 14-day observation period.

Dose: is the amount of test substance administered. Dose is expressed as weight of test substance per unit weight of test animal (e.g. mg/kg).

GHS: Globally Harmonised Classification System for Chemical Substances and Mixtures. A joint activity of OECD (human health and the environment), UN Committee of Experts on Transport of Dangerous Goods (physical-chemical properties) and ILO (hazard communication) and co-ordinated by the Interorganisation Programme for the Sound Management of Chemicals (IOMC).

Impending death: when moribund state or death is expected prior to the next planned time of observation. Signs indicative of this state in rodents could include convulsions, lateral position, recumbence, and tremor (See the Humane Endpoint Guidance Document (9) for more details).

LD₅₀ (**median lethal oral dose**): is a statistically derived single dose of a substance that can be expected to cause death in 50 per cent of animals when administered by the oral route. The LD50 value is expressed in terms of weight of test substance per unit weight of test animal (mg/kg).

Limit dose: refers to a dose at an upper limitation on testing (2000 or 5000 mg/kg).

Moribund status: being in a state of dying or inability to survive, even if treated (See the Humane Endpoint Guidance Document (9) for more details).

Predictable death: presence of clinical signs indicative of death at a known time in the future before the planned end of the experiment; for example: inability to reach water or food. (See the Humane Endpoint Guidance Document (9) for more details).

1.3 PRINCIPLE OF THE TEST U.K.

It is the principle of the test that, based on a stepwise procedure with the use of a minimum number of animals per step, sufficient information is obtained on the acute toxicity of the test substance to enable its classification. The substance is administered orally to a group of experimental animals at one of the defined doses. The substance is tested using a stepwise procedure, each step using three animals of a single sex (normally females). Absence or presence of compound-related mortality of the animals dosed at one step will determine the next step, i.e.;

- no further testing is needed,
- dosing of three additional animals, with the same dose
- dosing of three additional animals at the next higher or the next lower dose level.

Details of the test procedure are described in Annex 1. The method will enable a judgement with respect to classifying the test substance to one of a series of toxicity classes defined by fixed LD_{50} cut-off values.

1.4 DESCRIPTION OF THE METHOD U.K.

1.4.1 Selection of animal species U.K.

The preferred rodent species is the rat, although other rodent species may be used. Normally females are used (9). This is because literature surveys of conventional LD_{50} tests show that, although there is little difference in sensitivity between the sexes, in those cases where differences are observed females are generally slightly more - sensitive (11). However if knowledge of the toxicological or toxicokinetic properties of structurally related chemicals indicates that males are likely to be more sensitive, then this sex should be used. When the test is conducted in males adequate justification should be provided.

Healthy young adult animals of commonly used laboratory strains should be employed. Females should be nulliparous and non-pregnant. Each animal, at the commencement of its dosing, should be between 8 and 12 weeks old and its weight should fall in an interval within \pm 20% of the mean weight of any previously dosed animals.

1.4.2 Housing and feeding conditions U.K.

The temperature in the experimental animal room should be 22° (\pm 3°C). Although the relative humidity should be at least 30% and preferably not exceed 70% other than during room cleaning the aim should be 50-60%. Lighting should be artificial, the sequence being 12 hours light, 12 hours dark. For feeding, conventional laboratory diets may be used with an unlimited supply of drinking water. Animals may be group-caged by dose, but the number of animals per cage must not interfere with clear observations of each animal.

1.4.3 **Preparation of animals U.K.**

The animals are randomly selected, marked to permit individual identification, and kept in their cages for at least 5 days prior to dosing to allow for acclimatisation to the laboratory conditions.

1.4.4 **Preparation of doses U.K.**

In general test substances should be administered in a constant volume over the range of doses to be tested by varying the concentration of the dosing preparation. Where a liquid end product or mixture is to be tested however, the use of the undiluted test substance, i.e. at a constant concentration, may be more relevant to the subsequent risk assessment of that substance, and is a requirement of some regulatory authorities. In either case, the maximum dose volume for administration must not be exceeded. The maximum volume of liquid that can be administered at one time depends on the size of the test animal. In rodents, the volume should not normally exceed 1ml/100g of body weight: however in the case of aqueous solutions 2 ml/100g body weight can be considered. With respect to the, formulation of the dosing preparation, the use of an aqueous solution/suspension/emulsion is recommended wherever possible, followed in order of preference by a solution/suspension/emulsion in oil (e.g. corn oil) and then possibly solution in other vehicles. For vehicles other than water the toxicological characteristics of the vehicle should be known. Doses must be prepared shortly prior to administration unless the stability of the preparation over the period during which it will be used is known and shown to be acceptable.

1.5 PROCEDURE U.K.

1.5.1 Administration of doses U.K.

The test substance is administered in a single dose by gavage using a stomach tube or a suitable intubation canula. In the unusual circumstance that a single dose is not possible, the dose may be given in smaller fractions over a period not exceeding 24 hours.

Animals should be fasted prior to dosing (e.g. with the rat, food but not water should be withheld over-night, with the mouse, food but not water should be withheld for 3-4 hours). Following the period of fasting, the animals should be weighed and the test substance administered. After the substance has been administered, food may be withheld for a further 3-4 hours in rats or 1-2 hours in mice. Where a dose is administered in fractions over a period it may be necessary to provide the animals with food and water depending on the length of the period.

1.5.2 Number of animals and dose levels U.K.

Three animals are used for each step. The dose level to be used as the starting dose is selected from one of four fixed levels, 5, 50, 300 and 2000 mg/kg body weight. The starting dose level should be that which is most likely to produce mortality in some of the dosed animals. The flow

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charts of Annex 1 describe the procedure that should be followed for each of the starting doses. In addition, Annex 4 gives guidance on the classification in the EU system until the new GHS is implemented.

When available information suggests that mortality is unlikely at the highest starting dose level (2000 mg/kg body weight), then a limit test should be conducted. When there is no information on a substance to be tested, for animal welfare reasons it is recommended to use the starting dose of 300 mg/kg body weight.

The time interval between treatment groups is determined by the onset, duration, and severity of toxic signs. Treatment of animals at the next dose, should be delayed until one is confident of survival of the previously dosed animals.

Exceptionally, and only when justified by specific regulatory needs, the use of additional upper dose level of 5000 mg/kg body weight may be considered (see Annex 2). For reasons of animal welfare concern, testing of animals in GHS Category 5 ranges (2000-5000 mg/kg) is discouraged and should only be considered when there is a strong likelihood that the results of such a test would have a direct relevance for protecting human or animal health or the environment.

1.5.3 Limit test U.K.

The limit test is primarily used in situations where the experimenter has information indicating that the test material is likely to be nontoxic, i.e., having toxicity only above regulatory limit doses. Information about the toxicity of the test material can be gained from knowledge about similar tested compounds or similar tested mixtures or products, taking into consideration the identity and percentage of components known to be of toxicological significance. In those situations where there is little or no information about its toxicity, or in which the test material is expected to be toxic, the main test should be performed.

A limit test at one dose level of 2000 mg/kg body weight may be carried out with six animals (three animals per step). Exceptionally a limit test at one dose level of 5000 mg/kg may be carried out with three animals (see Annex 2). If test substance-related mortality is produced, further testing at the next lower level may need to be carried out.

1.6 OBSERVATIONS U.K.

Animals are observed individually after dosing at least once during the first 30 minutes, periodically during the first 24 hours, with special attention given during the first 4 hours, and daily thereafter, for a total of 14 days, except where they need to be removed from the study and humanely killed for animal welfare reasons or are found dead. However, the duration of observation should not be fixed rigidly. It should be determined by the toxic reactions, time of onset and length of recovery period, and may thus be extended when considered necessary. The times at which signs of toxicity appear and disappear are important, especially if there is a tendency for toxic signs to be delayed (12). All observations are systematically recorded with individual records being maintained for each animal.

Additional observations will be necessary if the animals continue to display signs of toxicity. Observations should include changes in skin and fur, eyes and mucous membranes, and also respiratory, circulatory, autonomic and central nervous systems, and somatomotor activity and behaviour pattern. Attention should be directed to observations of tremors, convulsions, salivation, diarrhoea, lethargy, sleep and coma. The principles and criteria summarised in the Humane Endpoints Guidance Document (9) should be taken into consideration. Animals found in a moribund condition and animals showing severe pain or enduring signs of severe distress should be humanely killed. When animals are killed for humane reasons or found dead, the time of death should be recorded as precisely as possible.

1.6.1 **Body weight** U.K.

Individual weights of animals should be determined shortly before the test substance is administered, and at least weekly thereafter. Weight changes should be calculated and recorded. At the end of the test surviving animals are weighed and humanely killed.

1.6.2 **Pathology U.K.**

All test animals (including those that die during the test or are removed from the study for animal welfare reasons) should be subjected to gross necropsy. All gross pathological changes should be recorded for each animal. Microscopic examination of organs showing evidence of gross pathology in animals surviving 24 or more hours may also be considered because it may yield useful information.

2. DATA U.K.

Individual animal data should be provided. Additionally, all data should be summarised in tabular form, showing for each test group the number of animals used, the number of animals displaying signs of toxicity, the number of animals found dead during the test or killed for humane reasons, time of death of individual animals, a description and the time course of toxic effects and reversibility, and necropsy findings.

3. **REPORTING** U.K.

3.1 **Test report** U.K.

The test report must include the following information, as appropriate:

Test substance

- physical nature, purity, and, where relevant, physico-chemical properties (including isomerisation);
- identification data, including CAS number.

Vehicle (if appropriate):

— justification for choice of vehicle, if other than water.

Test animals:

- species/strain used;
- microbiological status of the animals, when known;
- number, age, and sex of animals (including, where appropriate, a rationale for the use of males instead of females);
- source, housing conditions, diet etc.;

Test conditions:

- details of test substance formulation including details of the physical form of the material administered;
- details of the administration of the test substance including dosing volumes and time of dosing;
- details of food and water quality (including diet type/source, water source);
- the rationale for the selection of the starting dose.

Results:

— tabulation of response data and dose level for each animal (i.e. animals showing signs of toxicity including mortality; nature, severity, and duration of effects);

- tabulation of body weight and body weight changes;
 - individual weights of animals at the day of dosing, in weekly intervals thereafter, and at the time of death or sacrifice
- date and time of death if prior to scheduled sacrifice
- time course of onset of signs of toxicity, and whether these were reversible for each animal;
- necropsy findings and histopathological findings for each animal, if available.

Discussion and interpretation of results.

Conclusions.

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ANNEX PROCEDURE TO BE FOLLOWED FOR EACH OF THE STARTING 1 DOSES U.K.

GENERAL REMARKS

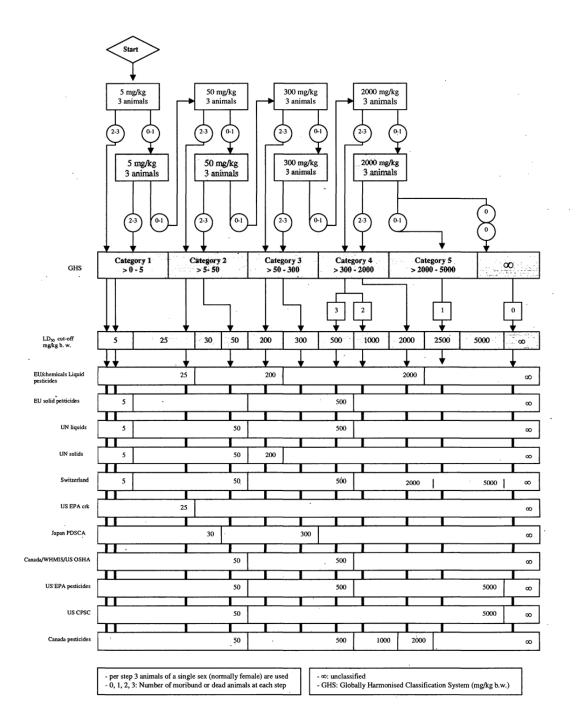
For each starting dose, the respective testing schemes as included in this Annex outline the procedure to be followed.

- Annex 1 a: Starting dose is 5 mg/kg bw
- Annex 1 b: Starting dose is 50 mg/kg bw
- Annex 1 c: Starting dose is: 300 mg/kg bw
- Annex 1 d: Starting dose is: 2000 mg/kg bw

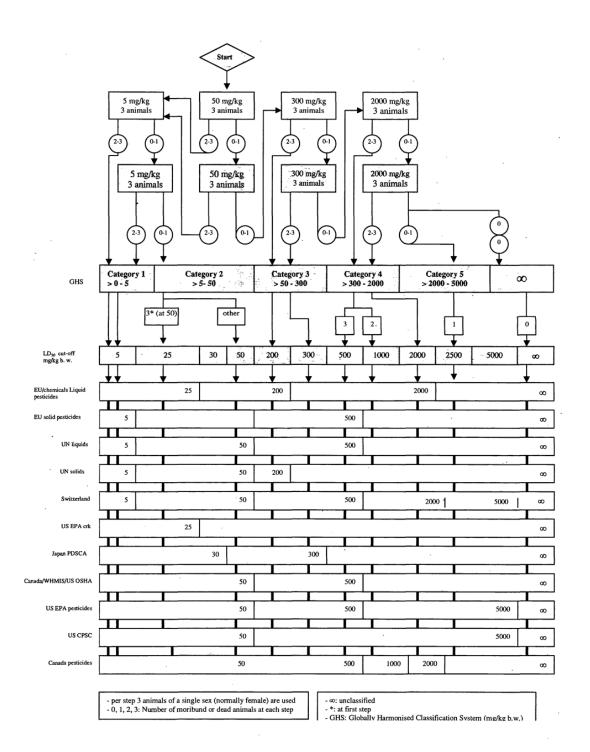
Depending on the number of humanely killed or dead animals, the test procedure follows the indicated arrows.

ANNEX TEST PROCEDURE WITH A STARTING DOSE OF 5 MG/KG BODY 1 A WEIGHT U.K.

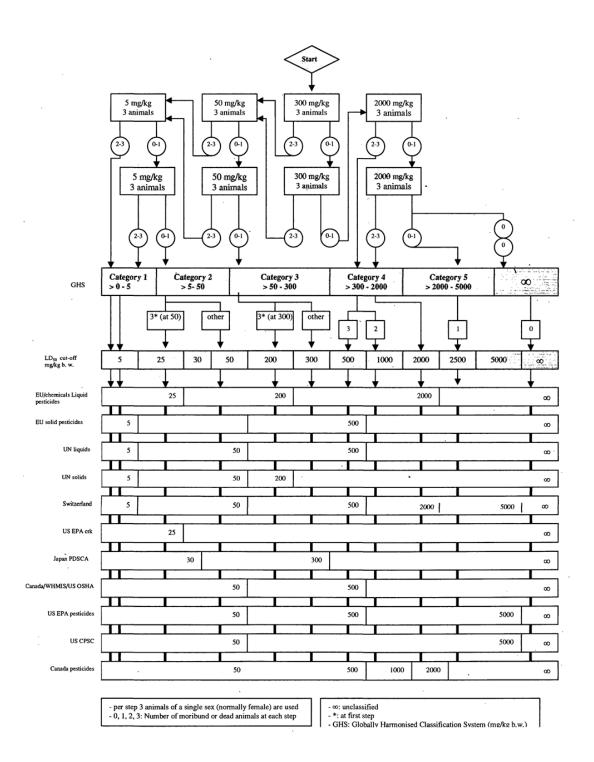
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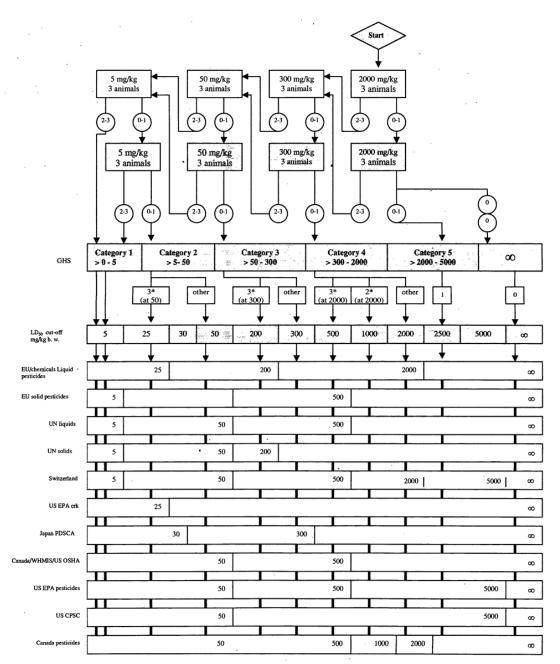
ANNEX TEST PROCEDURE WITH A STARTING DOSE OF 50 MG/KG BODY WEIGHT U.K. 1 B



ANNEX TEST PROCEDURE WITH A STARTING DOSE OF 300 MG/KG BODY 1 C WEIGHT U.K.



ANNEX TEST PROCEDURE WITH A STARTING DOSE OF 2000 MG/KG BODY 1 D WEIGHT U.K.



ANNEX 2 CRITERIA FOR CLASSIFICATION OF TEST SUBSTANCES WITH EXPECTED LD50 VALUES EXCEEDING 2000 MG/KG WITHOUT THE NEED FOR TESTING

Criteria for hazard Category 5 are intended to enable the identification of test substances which are of relatively low acute toxicity hazard but which, under certain circumstances may present a danger to vulnerable populations. These substances are anticipated to have an oral or dermal LD50 in the range of 2000-5000 mg/kg or equivalent doses for other routes. The test substance should be classified in the hazard category defined by: $2000 \text{mg/kg} < \text{LD}_{50} < 5000 \text{mg/kg}$ kg (Category 5 in the GHS) in the following cases:

a) If directed to this category by any of the testing schemes of Annex 1a-1d, based on mortality incidences;

- b) if reliable evidence is already available that indicates the LD₅₀ to be in the range of Category 5 values; or other animal studies or toxic effects in humans indicate a concern for human health of an acute nature.
- c) through extrapolation, estimation or measurement of data if assignment to a more hazardous class is not warranted, and
 - reliable information is available indicating significant toxic effects in humans, or
 - any mortality is observed when tested up to Category 4 values by the oral route, or
 - where expert judgement confirms significant clinical signs of toxicity, when tested up to Category 4 values, except for diarrhoea, piloerection or an ungroomed appearance, or
 - where expert judgement confirms reliable information indicating the potential for significant acute effects from the other animal studies.

TESTING AT DOSES ABOVE 2000 MG/KG

Recognising the need to protect animal welfare, testing of animals in Category 5 (5000 mg/kg) ranges is discouraged and should only be considered when there is a strong likelihood that results of such a test have a direct relevance for protecting human or animal health (10). No further testing should be conducted at higher dose levels.

When testing is required a dose of 5000mg/kg, only one step (i.e. three animals) is required. If the first animal dosed dies, then dosing proceeds at 2000mg/kg in accordance with the flow charts in Annex 1. If the first animal survives, two further animals are dosed. If only one of the three animals dies, the LD50 value is expected to exceed 5000mg/kg. If both animals die, then dosing proceeds at 2000mg/kg.

ANNEX TEST METHOD B.1 this: Guidance on classification according to EU scheme to cover the transition period until full implementation of the Globally Harmonised Classification System (GHS) (taken from reference (8)) U.K.

Erythema and Eschar Formation

N	No erythema	0
1	Very slight erythema (barely perceptible)	1
V	Vell defined erythema	2
N	Moderate to severe erythema	3
S	severe erythema (beef redness) to eschar formation preventing grading of erythema	4

Maximum possible: 4

Oedema Formation

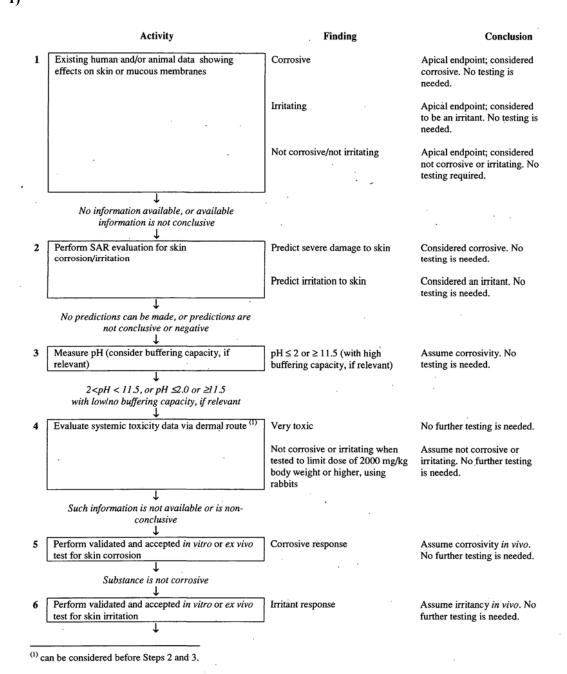
No oedema	
Very slight oederna (barely perceptible)	1
Slight oedema (edges of area well defined by definite raising)	
Moderate oedema (raised approximately 1 mm)	
Severe oedema (raised more than 1 mm and extending beyond area of exposure)	

Maximum possible: 4

Histopathological examination may be carried out to clarify equivocal responses.

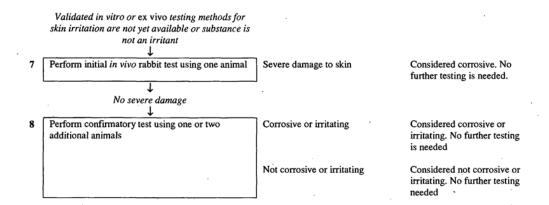
ANNEX TEST METHOD B.1 tris: Guidance on classification according to EU scheme to cover the transition period until full implementation of the Globally Harmonised Classification System (GHS) (taken from reference (8)) U.K.

(CONTINUED 1)



ANNEX TEST METHOD B.1 tris: Guidance on classification according to EU scheme to cover the transition period until full implementation of the Globally Harmonised (CONTINUES Diffication System (GHS) (taken from reference (8)) U.K.

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ANNEX TEST METHOD B.1 tris: Guidance on classififcation according to EU scheme to cover the transition period until full implementation of the Globally Harmonised (CONTINUES) fication System (GHS) (taken from reference (8)) U.K.

Cornea			
Opacity: degree of density (readings should be taken from most dense area)*			
No ulceration or opacity			
Maximum possible: 4			
* The area of corneal opacity should be noted			
Iris			
Normal Markedly deepened rugae, congestion, swelling, moderate circumcorneal hyperaemia; or injection; iris reactive to light (a sluggish reaction is considered to be an effect) Hemorrhage, gross destruction, or no reaction to light			
Maximum possible: 2			
Conjunctivae			
Redness (refers to palpebral and bulbar conjunctivae; excluding cornea and iris) Normal. Some blood vessels hyperaemic (injected) Diffuse, crimson colour; individual vessels not easily discernible Diffuse beefy red	1 2		
Maximum possible: 3			
Chemosis			
Swelling (refers to lids and/or nictating membranes)			
Normal Some swelling above normal Obvious swelling, with partial eversion of lids Swelling, with lids about half closed Swelling, with lids more than half closed	2		

ANNEX2D U.K.

Maximum possible: 4

B. 4.ACUTE TOXICITY: DERMAL IRRITATION/CORROSION

1. **METHOD** U.K.

This method is equivalent to the OECD TG 404 (2002).

1.1 INTRODUCTION U.K.

In the preparation of this updated method special attention was given to possible improvements in relation to animal welfare concerns and to the evaluation of all existing information on the test substance in order to avoid unnecessary testing in laboratory animals. This method includes the recommendation that prior to undertaking the described *in vivo* test for corrosion/irritation of the substance, a weight-of-the-evidence analysis be performed on the existing relevant data. Where insufficient data are available, they can be developed through application of sequential

testing (1). The testing strategy includes the performance of validated and accepted *in vitro* tests and is provided as an Annex to this method. In addition, where appropriate, the successive, instead of simultaneous, application of the three test patches to the animal in the initial *in vivo* test is recommended.

In the interest of both sound science and animal welfare, *in vivo* testing should not be undertaken until all available data relevant to the potential dermal corrosivity/irritation of the substance have been evaluated in a weight-of-the-evidence analysis. Such data will include evidence from existing studies in humans and/or laboratory animals, evidence of corrosivity/irritation of one or more structurally related substances or mixtures of such substances, data demonstrating strong acidity or alkalinity of the substance (2)(3), and results from validated and accepted *in vitro* or *ex vivo* tests (4)(5)(5a). This analysis should decrease the need for *in vivo* testing for dermal corrosivity/irritation of substances for which sufficient evidence already exists from other studies as to those two endpoints.

A preferred sequential testing strategy, which includes the performance of validated and accepted *in vitro* or *ex vivo* tests for corrosion/irritation, is included as an Annex to this Method. The strategy was developed at, and unanimously recommended by the participants of, an OECD workshop (6), and has been adopted as the recommended testing strategy in the Globally Harmonised System for the Classification of Chemical Substances (GHS) (7). It is recommended that this testing strategy be followed prior to undertaking *in vivo* testing. For new substances it is the recommended stepwise testing approach for developing scientifically sound data on the corrosivity/irritation of the substance. For existing substances with insufficient data on dermal corrosion/irritation, the strategy should be used to fill missing data gaps. The use of a different testing strategy or procedure, or a decision not to use a stepwise testing approach, should be justified.

If a determination of corrosivity or irritation cannot be made using a weight-of-the-evidence analysis, consistent with the sequential testing strategy, an *in vivo* test should be considered (see Annex).

1.2 DEFINITIONS U.K.

Dermal irritation: is the production of reversible damage of the skin following the application of a test substance for up to 4 hours.

Dermal corrosion: is the production of irreversible damage of the skin; namely, visible necrosis through the epidermis and into the dermis, following the application of a test substance for up to four hours. Corrosive reactions are typified by ulcers, bleeding, bloody scabs, and, by the end of observation at 14 days, by discoloration due to blanching of the skin, complete areas of alopecia, and scars. Histopathology should be considered to evaluate questionable lesions.

1.3 PRINCIPLE OF THE TEST METHOD U.K.

The substance to be tested is applied in a single dose to the skin of an experimental animal; untreated skin areas of the test animal serve as the control. The degree of irritation/corrosion is read and scored at specified intervals and is further described in order to provide a complete evaluation of the effects. The duration of the study should be sufficient to evaluate the reversibility or irreversibility of the effects observed.

Animals showing continuing signs of severe distress and/or pain at any stage of the test should be humanely killed, and the substance assessed accordingly. Criteria for making the decision to humanely kill moribund and severely suffering animals can be found in reference (8).

1.4 DESCRIPTION OF THE TEST METHOD U.K.

1.4.1 **Preparation for the** *in vivo* **test** U.K.

1.4.1.1 Selection of animal species U.K.

The albino rabbit is the preferable laboratory animal, and healthy young adult rabbits are used. A rationale for using other species should be provided.

1.4.1.2 Preparation of the animals U.K.

Approximately 24 hours before the test, fur should be removed by closely clipping the dorsal area of the trunk of the animals. Care should be taken to avoid abrading the skin, and only animals with healthy, intact skin should be used.

Some strains of rabbit have dense patches of hair that are more prominent at certain times of the year. Such areas of dense hair growth should not be used as test sites.

1.4.1.3 Housing and feeding conditions U.K.

Animals should be individually housed. The temperature of the experimental animal room should be 20°C (± 3°C) for rabbits. Although the relative humidity should be at least 30% and preferably not exceed 70%, other than during room cleaning, the aim should be 50-60%. Lighting should be artificial, the sequence being 12 hours light, 12 hours dark. For feeding, conventional laboratory diets may be used with an unrestricted supply of drinking water.

1.4.2 **Test procedure** U.K.

1.4.2.1 Application of the test substance U.K.

The test substance should be applied to a small area (approximately 6 cm²) of skin and covered with a gauze patch, which is held in place with non-irritating tape. In cases in which direct application is not possible (e.g., liquids or some pastes), the test substance should first be applied to the gauze patch, which is then applied to the skin. The patch should be loosely held in contact with the skin by means of a suitable semi-occlusive dressing for the duration of the exposure period. If the test substance is applied to the patch, it should be attached to the skin in such a manner that there is good contact and uniform distribution of the substance on the skin. Access by the animal to the patch and ingestion or inhalation of the test substance should be prevented.

Liquid test substances are generally used undiluted. When testing solids (which may be pulverised, if considered necessary), the test substance should be moistened with the smallest amount of water (or, where necessary, of another suitable vehicle) sufficient to ensure good skin contact. When vehicles other than water are used, the potential influence of the vehicle on irritation of the skin by the test substance should be minimal, if any.

At the end of the exposure period, which is normally 4 hours, residual test substance should be removed, where practicable, using water or an appropriate solvent without altering the existing response or the integrity of the epidermis.

1.4.2.2 Dose level U.K.

A dose of 0.5 ml. of liquid or 0.5 g of solid or paste is applied to the test site.

1.4.2.3 Initial test (In vivo dermal irritation/corrosion test using one animal) U.K.

It is strongly recommended that the *in vivo* test be performed initially using one animal, especially when the substance is suspected to have corrosion potential. This is in accordance with the sequential testing strategy (see Annex 1).

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When a substance has been judged to be corrosive on the basis of a weight-of-the-evidence analysis, no further animal testing is needed. For most substances suspected of being corrosive, further *in vivo* testing is normally not necessary. However, in those cases where additional data are felt warranted because of insufficient evidence, limited animal testing may be carried out using the following approach: Up to three tests patches are applied sequentially to the animal. The first patch is removed after three minutes. If no serious skin reaction is observed, a second patch is applied and removed after one hour. If the observations at this stage indicate that exposure can humanely be allowed to extend to four hours, a third patch is applied and removed after four hours, and the response is graded.

If a corrosive effect is observed after any of the three sequential exposures, the test is immediately terminated. If a corrosive effect is not observed after the last patch is removed, the animal is observed for 14 days, unless corrosion develops at an earlier time point.

In those cases in which the test substance is not expected to produce corrosion but may be irritating, a single patch should be applied to one animal for four hours.

1.4.2.4 Confirmatory test (In vivo dermal irritation test with additional animals) U.K.

If a corrosive effect is not observed in the initial test, the irritant or negative response should be confirmed using up to two additional animals, each with one patch, for an exposure period of four hours. If an irritant effect is observed in the initial test, the confirmatory test may be conducted in a sequential manner, or by exposing two additional animals simultaneously. In the exceptional case, in which the initial test is not conducted, two or three animals may be treated with a single patch, which is removed after four hours. When two animals are used, if both exhibit the same response, no further testing is needed. Otherwise, the third animal is also tested. Equivocal responses may need to be evaluated using additional animals.

1.4.2.5 *Observation period* U.K.

The duration of the observation period should be sufficient to evaluate fully the reversibility of the effects observed. However, the experiment should be terminated at any time that the animal shows continuing signs of severe pain or distress. To determine the reversibility of effects, the animals should be observed up to 14 days after removal of the patches. If reversibility is seen before 14 days, the experiment should be terminated at that time.

1.4.2.6 Clinical observations and grading of skin reactions U.K.

All animals should be examined for signs of erythema and oedema, and the responses scored at 60 minutes, and then at 24, 48 and 72 hours after patch removal. For the initial test in one animal, the test site is also examined immediately after the patch has been removed. Dermal reactions are graded and recorded according to the grades in the Table below. If there is damage to skin which cannot be identified as irritation or corrosion at 72 hours, observations may be needed until day 14 to determine the reversibility of the effects. In addition to the observation of irritation, all local toxic effects, such as defatting of the skin, and any systemic adverse effects (e.g., effects on clinical signs of toxicity and body weight), should be fully described and recorded. Histopathological examination should be considered to clarify equivocal responses.

The grading of skin responses is necessarily subjective. To promote harmonisation in grading of skin response and to assist testing laboratories and those involved in making and interpreting the observations, the personnel performing the observations need to be adequately trained in the scoring system used (see Table below). An illustrated guide for grading skin irritation and other lesions could be helpful (9).

2. DATA U.K.

2.1 PRESENTATION OF RESULTS U.K.

Study results should be summarised in tabular form in the final test report and should cover all items listed in section 3.1.

2.2 EVALUATION OF RESULTS U.K.

The dermal irritation scores should be evaluated in conjunction with the nature and severity of lesions, and their reversibility or lack of reversibility. The individual scores do not represent an absolute standard for the irritant properties of a material, as other effects of the test material are also evaluated. Instead, individual scores should be viewed as reference values, which need to be evaluated in combination with all other observations from the study.

Reversibility of dermal lesions should be considered in evaluating irritant responses. When responses such as alopecia (limited area), hyperkeratosis, hyperplasia and scaling, persist to the end of the 14-day observation period, the test substance should be considered an irritant.

3. **REPORTING** U.K.

3.1 TEST REPORT U.K.

The test report must include the following information:

Rationale for *in vivo* testing: weight-of-evidence analysis of pre-existing test data, including results from sequential testing strategy:

- description of relevant data available from prior testing;
 data derived at each stage of testing strategy;
 description of *in vitro* tests performed, including details of procedures, results obtained with test/reference substances;
- weight-of-the-evidence analysis for performing *in vivo* study.

Test substance:

- identification data (e.g., CAS number; source; purity; known impurities; lot number);
 physical nature and physicochemical properties (e.g. pH, volatility, stability);
- if mixture, composition and relative percentages of components.

Vehicle:

- identification, concentration (where appropriate), volume used;
- justification for choice of vehicle.

Test animals:

- species/strain used, rationale for using animals other than albino rabbit;
- number of animals of each sex;
- individual animal weights at start and conclusion of test;
- age at start of study;
- source of animals, housing conditions, diet, etc.

Test conditions:

- technique of patch site preparation;
- details of patch materials used and patching technique;
- details of test substance preparation, application, and removal.

Results:

- tabulation of irritation/corrosion response scores for each animal at all time points measured;
- descriptions of all lesions observed;
- narrative description of nature and degree of irritation or corrosion observed, and any histopathological findings;
- description of other adverse local (e.g., defatting of skin) and systemic effects in addition to dermal irritation or corrosion.

Discussion of results

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TABLE I:GRADING OF SKIN REACTIONS

	Activity	Finding	Conclusion
1	Existing human and/or animal data showing effects on eyes	Severe damage to eyes	Apical endpoint; consider corrosive to eyes. No testing is needed.
		Eye irritant	Apical endpoint; consider irritating to eyes. No testing is needed.
		Not corrosive/not irritating to eyes	Apical endpoint; considered non- corrosive and non-irritating to eyes. No testing required.
	Existing human and/or animal data showing corrosive effects on skin	Skin corrosive	Assume corrosivity to eyes. No testing is needed.
	Existing human and/or animal data showing severe irritant effects on skin	Severe skin irritant	Assume irritating to eyes. No testing is needed
	no information available, or available information is not conclusive		
2	Perform SAR for eye corrosion/irritation	Predict severe damage to eyes	Assume corrosivity to eyes. No testing is needed.
		Predict irritation to eyes	Assume irritating to eyes. No testing is needed.
	Perform SAR for skin corrosion	Predict skin corrosivity	Assume corrosivity to eyes. No testing is needed.
	No predictions can be made, or predictions are not conclusive or negative		
3	Measure pH (buffering capacity, if relevant)	pH \leq 2 or \geq 11.5 (with high buffering capacity, if relevant)	Assume corrosivity to eyes. No testing is needed.
	↓ 2 <ph 11.5,="" <="" buffering="" capacity,="" if="" low="" no="" or="" ph≤2.0="" relevant="" td="" with="" ↓'<="" ≥11.5=""><td></td><td></td></ph>		
4	Evaluate systemic toxicity via the dermal route	Very toxic at concentrations that would be tested in the eye.	Substance would be too toxic for testing. No testing is needed.
	Such information is not available, or substance is not very toxic		
5	Perform validated and accepted in vitro or ex vivo test for eye corrosion	Corrosive response	Assume corrosivity to eyes. No further testing is needed.
	1	•	

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GENERAL CONSIDERATIONS

In the interest of sound science and animal welfare, it is important to avoid the unnecessary use of animals and to minimise any testing that is likely to produce severe responses in animals. All information on a substance relevant to its potential skin corrosivity/irritancy should be evaluated prior to considering *in vivo* testing. Sufficient evidence may already exist to classify a test substance as to its dermal corrosion or irritation potential without the need to conduct testing in laboratory animals. Therefore, utilising a weight-of-the-evidence analysis and a sequential testing strategy, will minimise the need for *in vivo* testing, especially if the substance is likely to produce severe reactions.

It is recommended that a weight-of-the-evidence analysis be used to evaluate existing information regarding the skin irritation and corrosion of substances to determine whether additional studies, other than *in vivo* dermal studies, should be performed to help characterise such potential. Where further studies are needed, it is recommended that the sequential testing strategy be utilised to develop the relevant experimental data. For substances which have no testing history, the sequential testing strategy should be utilised to develop the data set needed to evaluate its dermal corrosion/irritation potential. The testing strategy described in this Annex was developed at an OECD workshop (1) and was later affirmed and expanded in the Harmonised Integrated Hazard Classification System for Human Health and Environmental Effects of Chemical Substances, as endorsed by the 28th Joint Meeting of the Chemicals Committee and the Working Party on Chemicals, in November 1998 (2).

Although this sequential testing strategy is not an integral part of testing method B.4, it expresses the recommended approach for the determination of skin irritation/corrosion characteristics. This approach represents both best practice and an ethical benchmark for *in vivo* testing for skin irritation/corrosion. The testing method provides guidance for the conduct of the *in vivo* test and summarises the factors that should be addressed before initiating such a test. The strategy provides an approach for the evaluation of existing data on the skin irritation/corrosion properties of test substances and a tiered approach for the generation of relevant data on substances for which additional studies are needed, or for which no studies have been performed. It also recommends the performance of validated and accepted *in vitro* or *ex vivo* tests for skin corrosion/irritation under specific circumstances.

DESCRIPTION OF THE EVALUATION AND TESTING STRATEGY

Prior to undertaking tests as part of the sequential testing strategy (Figure), all available information should be evaluated to determine the need for *in vivo* skin testing. Although significant information might be gained from the evaluation of single parameters (e.g., extreme pH), the totality of existing information should be considered. All relevant data on the effects of the substance in question, or its analogues, should be evaluated in making a weight-of-the-evidence decision, and a rationale for the decision should be presented. Primary emphasis should be placed upon existing human and animal data on the substance, followed by the outcome of *in vitro* or *ex vivo* testing. *In vivo* studies of corrosive substances should be avoided whenever possible. The factors considered in the testing strategy include:

Evaluation of existing human and animal data (Step 1). Existing human data, e.g. clinical or occupational studies and case reports, and/or animal test data, e.g. from single or repeated dermal exposure toxicity studies, should be considered first, because they provide information directly related to effects on the skin. Substances with known irritancy or corrosivity, and those with clear evidence of non-corrosivity or non-irritancy, need not be tested in *in vivo* studies.

Analysis of structure activity relationships (SAR) (Step 2). The results of testing of structurally related substances should be considered, if available. When sufficient human and/or animal data are available on structurally related substances or mixtures of such substances to indicate their skin corrosion/irritancy potential, it can be presumed that the test substance being evaluated

will produce the same responses. In those cases, the test substance may not need to be tested. Negative data from studies of structurally related substances or mixtures of such substances do not constitute sufficient evidence of non-corrosivity/non-irritancy of a substance under the sequential testing strategy. Validated and accepted SAR approaches should be used to identify both dermal corrosion and irritation potential.

Physicochemical properties and chemical reactivity (Step 3). Substances exhibiting pH extremes such as ≤ 2.0 and ≥ 11.5 may have strong local effects. If extreme pH is the basis for identifying a substance as corrosive to skin, then its acid/alkali reserve (or buffering capacity) may also be taken into consideration (3)(4). If the buffering capacity suggests that a substance may not be corrosive to the skin, then further testing should be undertaken to confirm this, preferably by the use of a validated and accepted *in vitro* or *ex vivo* test (see steps 5 and 6).

Dermal toxicity (Step 4). If a chemical has proven to be very toxic by the dermal route, an *in vivo* dermal irritation/corrosion study may not be practicable because the amount of test substance normally applied could exceed the very toxic dose and, consequently result in the death or severe suffering of the animals. In addition, when dermal toxicity studies utilising albino rabbits have already been performed up to the limit dose level of 2000 mg/kg body weight or higher, and no dermal irritation or corrosion has been seen, additional testing for skin irritation/corrosion may not be needed. A number of considerations should be borne in mind when evaluating acute dermal toxicity in previously performed studies. For example, reported information on dermal lesions may be incomplete. Testing and observations may have been made on a species other than the rabbit, and species may differ widely in sensitivity of their responses. Also the form of test substance applied to animals may not have been suitable for assessment of skin irritation/corrosion (e.g., dilution of substances for testing dermal toxicity (5). However, in those cases in which well-designed and conducted dermal toxicity studies have been performed in rabbits, negative findings may be considered sufficient evidence that the substance is not corrosive or irritating.

Results from in vitro or ex vivo tests (Steps 5 and 6). Substances that have demonstrated corrosive or severe irritant properties in a validated and accepted *in vitro* or *ex vivo* test (6)(7) designed for the assessment of these specific effects, need not be tested in animals. It can be presumed that such substances will produce similar severe effects *in vivo*.

In vivo test in rabbits (Steps 7 and 8). Should a weight-of the-evidence decision be made to conduct *in vivo* testing, it should begin with an initial test using one animal. If the results of this test indicate the substance to be corrosive to the skin, further testing should not be performed. If a corrosive effect is not observed in the initial test, the irritant or negative response should be confirmed using up to two additional animals for an exposure period of four hours. If an irritant effect is observed in the initial test, the confirmatory test may be conducted in a sequential manner, or by exposing the two additional animals simultaneously.

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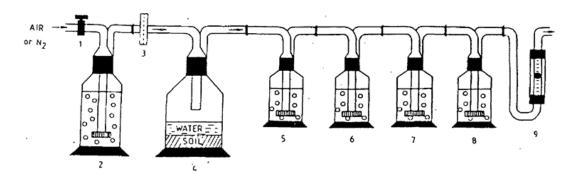
FIGURETESTING AND EVALUATION STRATEGY FOR DERMAL IRRITATION / CORROSION U.K.

Substance is not corrosive, or validated in vitro or ex vivo testing methods for eye corrosion are not yet available Perform validated and accepted in Assume irritancy to eyes. No further Irritant response vitro or ex vivo test for eye testing is needed. irritation Substance is not an irritant, or validated in vitro or ex vivo testing methods for eye irritation are not yet available Experimentally assess in vivo skin Assume corrosivity to eyes. No Corrosive or severe irritant irritation/corrosion potential (see response further testing is needed. Testing method B.4 including its Annex) Substance is not corrosive or severely irritating to skin 1 Perform initial in vivo rabbit eye Severe damage to eyes Consider corrosive to eyes. No test using one animal further testing is needed. No severe damage, or no response Perform confirmatory test using Corrosive or irritating Consider corrosive or irritating to one or two additional animals eyes. No further testing is needed Consider non-irritating and non-Not corrosive or irritating corrosive to eyes. No further testing is needed.

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- needle valve
- 2: gas washing bottle containing water
- 3: ultramembrane (sterile conditions only), pore size 0.2 µm
- soil metabolism flask (waterlogged only for anaerobic and paddy conditions;)
- ethylene glycol trap for organic volatile compounds
- sulphuric acid trap for alkaline volatile compounds
- 7, 8: sodium hydroxide trap for CO2 & other acidic volatiles
 - flow meter.



ANNEX 2E U.K.

B. 5.ACUTE TOXICITY: EYE IRRITATION/CORROSION

METHOD U.K. 1.

This method is equivalent to the OECD TG 405 (2002)

INTRODUCTION U.K. 1.1

In the preparation of this updated method special attention was given to possible improvements through the evaluation of all existing information on the test substance in order to avoid unnecessary testing in laboratory animals and thereby address animal welfare concerns. This method includes the recommendation that prior to undertaking the described in vivo test for acute eye irritation/corrosion, a weight-of-the-evidence analysis be performed (1) on the existing relevant data. Where insufficient data are available, it is recommended that they be developed through application of sequential testing (2)(3). The testing strategy includes the performance of validated and accepted in vitro tests and is provided as an Annex to the testing method. In addition, the use of an *in vivo* dermal irritation/corrosion test to predict eye corrosion prior to consideration of an *in vivo* eye test is recommended.

In the interest of both sound science and animal welfare, in vivo testing should not be considered until all available data relevant to the potential eye corrosivity/irritation of the substance has been evaluated in a weight-of-the-evidence analysis. Such data will include evidence from existing studies in humans and/or laboratory animals, evidence of corrosivity/irritation of one or more structurally related substances or mixtures of such substances, data demonstrating high acidity or alkalinity of the substance (4)(5), and results from validated and accepted in vitro or ex vivo tests for skin corrosion and irritation (6)(6a). The studies may have been conducted prior to, or as a result of, a weight-of-the-evidence analysis.

For certain substances, such an analysis may indicate the need for in vivo studies of the ocular corrosion/irritation potential of the substance. In all such cases, before considering the use of the in vivo eye test, preferably a study of the in vivo dermal effects of the substance should be conducted first and evaluated in accordance with testing method B.4 (7). The application of a

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weight-of-the-evidence analysis and the sequential testing strategy should decrease the need for *in vivo* testing for eye corrosivity/irritation of substances for which sufficient evidence already exists from other studies. If a determination of eye corrosion or irritation potential cannot be made using the sequential testing strategy, even after the performance of an *in vivo* study of dermal corrosion and irritation, an *in vivo* eye corrosion/irritation test may be performed.

A preferred sequential testing strategy, which includes the performance of validated *in vitro* or *ex vivo* tests for corrosion/irritation, is included in the Annex to this testing method. The strategy was developed at, and unanimously recommended by the participants of, an OECD workshop (8), and has been adopted as the recommended testing strategy in the Globally Harmonised System for the Classification of Chemical Substances (GHS) (9). It is recommended that this testing strategy be followed prior to undertaking *in vivo* testing. For new substances it is the recommended stepwise testing approach for developing scientifically sound data on the corrosivity/irritation of the substance. For existing substances with insufficient data on skin and eye corrosion/irritation, the strategy should be used to fill missing data gaps. The use of a different testing strategy or procedure, or the decision not to use a stepwise testing approach, should be justified.

1.2 DEFINITIONS U.K.

Eye irritation: is the production of changes in the eye following the application of a test substance to the anterior surface of the eye, which are fully reversible within 21 days of application.

Eye corrosion: is the production of tissue damage in the eye, or serious physical decay of vision, following application of a test substance to the anterior surface of the eye, which is not fully reversible within 21 days of application

1.3 PRINCIPLE OF THE TEST METHOD U.K.

The substance to be tested is applied in a single dose to one of the eyes of the experimental animal; the untreated eye serves as the control. The degree of eye irritation/corrosion is evaluated by scoring lesions of conjunctiva, cornea, and iris, at specific intervals. Other effects in the eye and adverse systemic effects are also described to provide a complete evaluation of the effects. The duration of the study should be sufficient to evaluate the reversibility or irreversibility of the effects.

Animals showing continuing signs of severe distress and/or pain at any stage of the test should be humanely killed, and the substance assessed accordingly. Criteria for making the decision to humanely kill moribund and severely suffering animals can be found in reference (10).

1.4 DESCRIPTION OF THE TEST METHOD U.K.

1.4.1 **Preparation for the** *in vivo* **test** U.K.

1.4.1.1 Selection of species U.K.

The albino rabbit is the preferable laboratory animal, and healthy young adult animals are used. A rationale for using other strains or species should be provided.

1.4.1.2 Preparation of animals U.K.

Both eyes of each experimental animal provisionally selected for testing should be examined within 24 hours before testing starts. Animals showing eye irritation, ocular defects, or pre-existing corneal injury should not be used.

1.4.1.3 *Housing and feeding conditions* U.K.

Animals should be individually housed. The temperature of the experimental animal room should be 20°C (± 3°C) for rabbits. Although the relative humidity should be at least 30% and preferably not exceed 70%, other than during room cleaning, the aim should be 50-60%. Lighting should be artificial, the sequence being 12 hours light, 12 hours dark. For feeding, conventional laboratory diets may be used with an unrestricted supply of drinking water.

1.4.2 **Test procedure U.K.**

1.4.2.1 Application of the test substance U.K.

The test substance should be placed in the conjunctival sac of one eye of each animal after gently pulling the lower lid away from the eyeball. The lids are then gently held together for about one second in order to prevent loss of the material. The other eye, which remains untreated, serves as a control.

1.4.2.2 Irrigation U.K.

The eyes of the test animals should not be washed for at least 24 hours following instillation of the test substance, except for solids (see section 1.4.2.3.2), and in case of immediate corrosive or irritating effects. At 24 hours a washout may be used if considered appropriate.

Use of a satellite group of animals to investigate the influence of washing is not recommended unless it is scientifically justified. If a satellite group is needed, two rabbits should be used. Conditions of washing should be carefully documented, e.g., time of washing; composition and temperature of wash solution; duration, volume, and velocity of application.

1.4.2.3 Dose level U.K.

1.4.2.3.1 Testing of liquids

For testing liquids, a dose of 0.1 ml is used. Pump sprays should not be used for instilling the substance directly into the eye. The liquid spray should be expelled and collected in a container prior to instilling 0.1 ml into the eye.

1.4.2.3.2 *Testing of solids*

When testing solids, pastes, and particulate substances, the amount used should have a volume of 0.1 ml or a weight of not more than 100 mg. The test material should be ground to a fine dust. The volume of solid material should be measured after gently compacting it, e.g., by tapping the measuring container. If the solid test substance has not been removed from the eye of the test animal by physiological mechanisms at the first observation time point of 1 hour after treatment, the eye may be rinsed with saline or distilled water.

1.4.2.3.3 Testing of aerosols

It is recommended that all pump sprays and aerosols be collected prior to instillation into the eye. The one exception is for substances in pressurised aerosol containers, which cannot be collected due to vaporisation. In such cases, the eye should be held open, and the test substance administered to the eye in a simple burst of about one second, from a distance of 10 cm directly in front of the eye. This distance may vary depending on the pressure of the spray and its contents. Care should be taken not to damage the eye from the pressure of the spray. In appropriate cases, there may be a need to evaluate the potential for "mechanical" damage to the eye from the force of the spray.

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An estimate of the dose from an aerosol can be made by simulating the test as follows: the substance is sprayed on to weighing paper through an opening the size of a rabbit eye placed directly before the paper. The weight increase of the paper is used to approximate the amount sprayed into the eye. For volatile substances, the dose may be estimated by weighing a receiving container before and after removal of the test material.

1.4.2.4 Initial test (In vivo eye irritation/corrosion test using one animal) U.K.

As articulated in the sequential testing strategy (see Annex 1), it is strongly recommended that the *in vivo* test be performed initially using one animal.

If the results of this test indicate the substance to be corrosive or a severe irritant to the eye using the procedure described, further testing for ocular irritancy should not be performed.

1.4.2.5 Local anaesthetics U.K.

Local anaesthetics may be used on a case-by-case basis. If the weight-of-the-evidence analysis indicates that the substance has the potential to cause pain, or initial testing shows that a painful reaction will occur, a local anaesthetic may be used prior to instillation of the test substance. The type, concentration, and dose of the local anaesthetic should be carefully selected to ensure that differences in reaction to the test substance will not result from its use. The control eye should be similarly anaesthetised.

1.4.2.6 Confirmatory test (In vivo eye irritation test with additional animals) U.K.

If a corrosive effect is not observed in the initial test, the irritant or negative response should be confirmed using up to two additional animals. If a severe irritant effect is observed in the initial test indicating a possible strong (irreversible) effect in the confirmatory testing, it is recommended that the confirmatory test be conducted in a sequential manner in one animal at a time, rather than exposing the two additional animals simultaneously. If the second animal reveals corrosive or severe irritant effects, the test is not continued. Additional animals may be needed to confirm weak or moderate irritant responses.

1.4.2.7 *Observation period* U.K.

The duration of the observation period should be sufficient to evaluate fully the magnitude and reversibility of the effects observed. However, the experiment should be terminated at any time that the animal shows continuing signs of severe pain or distress (9). To determine reversibility of effects, the animals should be observed normally for 21 days post administration of the test substance. If reversibility is seen before 21 days, the experiment should be terminated at that time.

1.4.2.7.1 Clinical observations and grading of eye reactions

The eyes should be examined at 1, 24, 48, and 72 hours after test substance application. Animals should be kept on test no longer than necessary once definitive information has been obtained. Animals showing continuing severe pain or distress should be humanely killed without delay, and the substance assessed accordingly. Animals with the following eye lesions post-instillation should be humanely killed: corneal perforation or significant corneal ulceration including staphyloma; blood in the anterior chamber of the eye; grade 4 corneal opacity which persists for 48 hours; absence of a light reflex (iridial response grade 2) which persists for 72 hours; ulceration of the conjunctival membrane; necrosis of the conjuctivae or nictitating membrane; or sloughing. This is because such lesions generally are not reversible

Animals that do not develop ocular lesions may be terminated not earlier than 3 days post instillation. Animals with mild to moderate lesions should be observed until the lesions clear, or for 21 days, at which time the study is terminated. Observations should be performed at 7, 14, and 21 days in order to determine the status of the lesions, and their reversibility or irreversibility.

The grades of ocular reaction (conjunctivae, cornea and iris) should be recorded at each examination (Table I). Any other lesions in the eye (e.g. pannus, staining) or adverse systemic effects should also be reported.

Examination of reactions can be facilitated by use of a binocular loupe, hand slit-lamp, biomicroscope, or other suitable device. After recording the observations at 24 hours, the eyes may be further examined with the aid of fluorescein.

The grading of ocular responses is necessarily subjective. To promote harmonisation of grading of ocular response and to assist testing laboratories and those involved in making and interpreting the observations, the personnel performing the observations need to be adequately trained in the scoring system used.

2. DATA U.K.

2.2 EVALUATION OF RESULTS U.K.

The ocular irritation scores should be evaluated in conjunction with the nature and severity of lesions, and their reversibility or lack of reversibility. The individual scores do not represent an absolute standard for the irritant properties of a material, as other effects of the test material are also evaluated. Instead, individual scores should be viewed as reference values and are only meaningful when supported by a full description and evaluation of all observations.

3. **REPORTING** U.K.

3.1 TEST REPORT U.K.

The test report must include the following information:

Rationale for *in vivo* testing: weight-of-the-evidence analysis of pre-existing test data, including results from sequential testing strategy

- description of relevant data available from prior testing;
- data derived in each step of testing strategy;
- description of *in vitro* tests performed, including details of procedures, results obtained with test/reference substances;
- description of *in vivo* dermal irritation/corrosion study performed, including results obtained;
- weight-of-the-evidence analysis for performing *in vivo* study

Test substance:

- identification data (e.g., CAS number, source, purity, known impurities, lot number);
- physical nature and physicochemical properties (e.g. pH, volatility, solubility, stability, reactivity with water);
- in case of a mixture, composition and relative percentages of components;
- if local anaesthetic is used, identification, purity, type, dose, and potential interaction with test substance.

Vehicle:

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identification, concentration (where appropriate), volume used;
justification for choice of vehicle.

Test animals:

- species/strain used, rationale for using animals other than albino rabbit;
- age of each animal at start of study;
- number of animals of each sex in test and control groups (if required);
- individual animal weights at start and conclusion of test;
- source, housing conditions, diet, etc.

Results:

- description of method used to score irritation at each observation time (e.g., hand slitlamp, biomicroscope, fluorescein);
- tabulation of irritant/corrosive response data for each animal at each observation time up to removal of each animal from the test;
- narrative description of the degree and nature of irritation or corrosion observed;
- description of any other lesions observed in the eye (e.g., vascularization, pannus formation, adhesions, staining);
- description of non-ocular local and systemic adverse effects, and histopathological findings, if any.

Discussion of results.

3.2 INTERPRETATION OF THE RESULTS U.K.

Extrapolation of the results of eye irritation studies in laboratory animals to humans is valid only to a limited degree. In many cases the albino rabbit is more sensitive than humans to ocular irritants or corrosives.

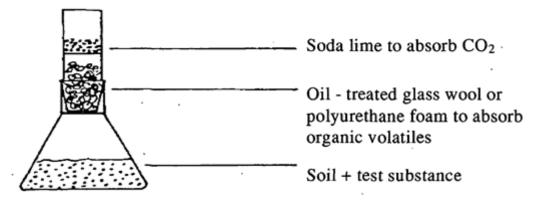
Care should be taken in the interpretation of data to exclude irritation resulting from secondary infection.

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TABLE I: GRADING OF OCULAR LESIONS



ANNEX A Sequential Testing Strategy for Eye Irritation and Corrosion U.K. GENERAL CONSIDERATIONS

In the interests of sound science and animal welfare, it is important to avoid the unnecessary use of animals, and to minimise testing that is likely to produce severe responses in animals. All information on a substance relevant to its potential ocular irritation/corrosivity should be evaluated prior to considering *in vivo* testing. Sufficient evidence may already exist to classify a test substance as to its eye irritation or corrosion potential without the need to conduct testing in laboratory animals. Therefore, utilizing a weight-of-the-evidence analysis and sequential testing strategy will minimise the need for *in vivo* testing, especially if the substance is likely to produce severe reactions.

It is recommended that a weight-of-the-evidence analysis be used to evaluate existing information pertaining to eye irritation and corrosion of substances and to determine whether additional studies, other than *in vivo* eye studies, should be performed to help characterise such potential. Where further studies are needed, it is recommended that the sequential testing strategy be utilised to develop the relevant experimental data. For substances which have no

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testing history, the sequential testing strategy should be utilised to develop the data needed to evaluate its eye corrosion/irritation. The testing strategy described in this Annex was developed at an OECD workshop (1). It was subsequently affirmed and expanded in the Harmonised Integrated Hazard Classification System for Human Health and Environmental Effects of Chemical Substances, as endorsed by the 28th Joint Meeting of the Chemicals Committee and the Working Party on Chemicals, in November 1998 (2).

Although this testing strategy is not an integrated part of testing method B.5, it expresses the recommended approach for the determination of eye irritation/corrosion properties. This approach represents both best practice and an ethical benchmark for *in vivo* testing for eye irritation/corrosion. The Testing method provides guidance for the conduct of the *in vivo* test and summarises the factors that should be addressed before considering such a test. The sequential testing strategy provides a weight-of-the-evidence approach for the evaluation of existing data on the eye irritation/corrosion properties of substances and a tiered approach for the generation of relevant data on substances for which additional studies are needed or for which no studies have been performed. The strategy includes the performance first of validated and accepted *in vitro* or *ex vivo* tests and then of testing method B.4 skin irritation/corrosion studies under specific circumstances (3)(4).

DESCRIPTION OF THE STEPWISE TESTING STRATEGY

Prior to undertaking tests as part of the sequential testing strategy (Figure), all available information should be evaluated to determine the need for *in vivo* eye testing. Although significant information might be gained from the evaluation of single parameters (e.g., extreme pH), the totality of existing information should be assessed. All relevant data on the effects of the substance in question, and its structural analogues, should be evaluated in making a weight-of-the-evidence decision, and a rationale for the decision should be presented. Primary emphasis should be placed upon existing human and animal data on the substance, followed by the outcome of *in vitro* or *ex vivo* testing. *In vivo* studies of corrosive substances should be avoided whenever possible. The factors considered in the testing strategy include:

Evaluation of existing human and animal data (Step 1). Existing human data, e.g. clinical and occupational studies, and case reports, and/or animal test data from ocular studies should be considered first, because they provide information directly related to effects on the eyes. Thereafter, available data from human and/or animal studies investigating dermal corrosion/irritation should be evaluated. Substances with known corrosivity or severe irritancy to the eye should not be instilled into the eyes of animals, nor should substances showing corrosive or irritant effects to the skin; such substances should be considered to be corrosive and/or irritating to the eyes as well. Substances with sufficient evidence of non-corrosivity and non-irritancy from previously performed ocular studies should also not be tested in *in vivo* eye studies.

Analysis of structure activity relationships (SAR) (Step 2). The results of testing of structurally related chemicals should be considered, if available. When sufficient human and/or animal data are available on structurally related substances or mixtures of such substances to indicate their eye corrrosion/irritancy potential, it can be presumed that the test substance will produce the same responses. In those cases, the substance may not need to be tested. Negative data from studies of structurally related substances or mixtures of such substances do not constitute sufficient evidence of non-corrosivity/non-irritancy of a substance under the sequential testing strategy. Validated and accepted SAR approaches should be used to identify the corrosion and irritation potential for both dermal and ocular effects.

Physicochemical properties and chemical reactivity (Step 3). Substances exhibiting pH extremes such as ≤ 2.0 or ≥ 11.5 may have strong local effects. If extreme pH is the basis for identifying a substance as corrosive or irritant to the eye, then its acid/alkaline reserve (buffering capacity) may also be taken into consideration (5)(6). If the buffering capacity suggests that a

substance may not be corrosive to the eye, then further testing should be undertaken to confirm this, preferably by the use of a validated and accepted *in vitro* or *ex vivo* test (see section step 5 and 6).

Consideration of other existing information (Step 4). All available information on systemic toxicity via the dermal route should be evaluated at this stage. The acute dermal toxicity of the test substance should also be considered. If the test substance has been shown to be very toxic by the dermal route, it may not need to be tested in the eye. Although there is not necessarily a relationship between acute dermal toxicity and eye irritation/corrosion, it can be assumed that if an agent is very toxic via the dermal route, it will also exhibit high toxicity when instilled into the eye. Such data may also be considered between Steps 2 and 3.

Results from in vitro or ex vivo tests (Steps 5 and 6). Substances that have demonstrated corrosive or severe irritant properties in an *in vitro* or *ex vivo* test (7)(8) that has been validated and accepted for the assessment specifically of eye or skin corrosivity/irritation, need not be tested in animals. It can be presumed that such substances will produce similar severe effects *in vivo*. If validated and accepted *in vitro/ex vivo* tests are not available, one should bypass Steps 5 and 6 and proceed directly to Step 7.

Assessment of in vivo dermal irritancy or corrosivity of the substance (Step 7). When insufficient evidence exists with which to perform a conclusive weight-of-the-evidence analysis of the potential eye irritation/corrosivity of a substance based upon data from the studies listed above, the *in vivo* skin irritation/corrosion potential should be evaluated first, using testing method B.4 (4) and its accompanying Annex (9). If the substance is shown to produce corrosion or severe skin irritation, it should be considered to be a corrosive eye irritant unless other information supports an alternative conclusion. Thus, an *in vivo* eye test would not need to be performed. If the substance is not corrosive or severely irritating to the skin, an *in vivo* eye test should be performed.

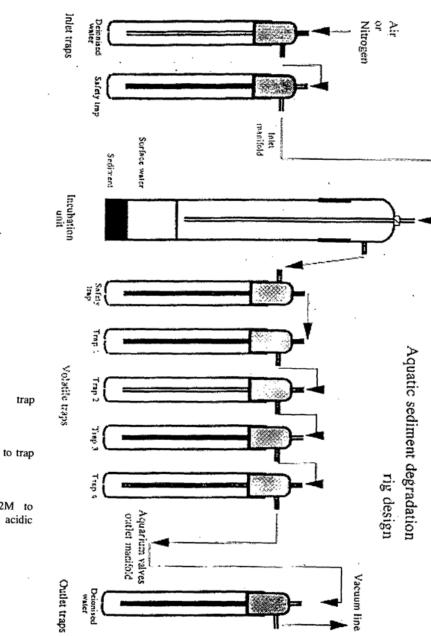
In vivo test in rabbits (Steps 8 and 9): *In vivo* ocular testing should begin with an initial test using one animal. If the results of this test indicate the substance to be a severe irritant or corrosive to the eyes, further testing should not be performed. If that test does not reveal any corrosive or severe irritant effects, a confirmatory test is conducted with two additional animals.

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FIGURETESTING AND EVALUATION STRATEGY FOR EYE IRRITATION/CORROSION U.K.

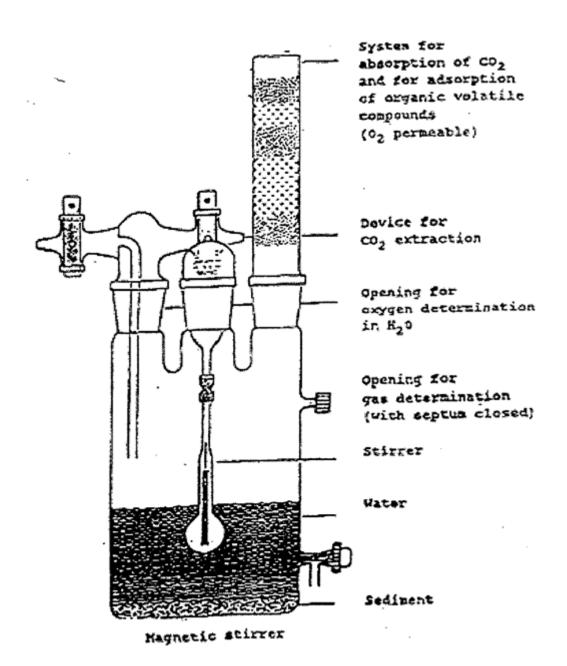


Safety trap, empty

Trap 1: ethyleneglycol to trap organic volatiles

Trap 2: sulphuric acid 0.1 M to trap alkaline volatiles

Traps 3 & 4: sodium hydroxide 2M to trap CO₂ and other acidic volatiles



ANNEX2F U.K.

B.31.PRENATAL DEVELOPMENTAL TOXICITY STUDY

1. **METHOD** U.K.

This method is a replicate of OECD TG 414 (2001).

1.1 INTRODUCTION U.K.

This method for developmental toxicity testing is designed to provide general information concerning the effects of prenatal exposure on the pregnant test animal and on the developing

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organism in utero; this may include assessment of maternal effects as well as death, structural abnormalities, or altered growth in the foetus. Functional deficits, although an important part of development, are not an integral part of this test method. They may be tested for in a separate study or as an adjunct to this study using the test method for developmental neurotoxicity. For information on testing for functional deficiencies and other postnatal effects the Test Method for the two-generation reproductive toxicity study and the developmental neurotoxicity study should be consulted as appropriate.

This test method may require specific adaptation in individual cases on the basis of specific knowledge on e.g. physicochemical or toxicological properties of the test substance. Such adaptation is acceptable, when convincing scientific evidence suggests that the adaptation will lead to a more informative test. In such a case, this scientific evidence should be carefully documented in the study report.

1.2 DEFINITIONS U.K.

Developmental toxicology: the study of adverse effects on the developing organism that may result from exposure prior to conception, during prenatal development, or postnatally to the time of sexual maturation. The major manifestations of developmental toxicity include 1) death of the organism, 2) structural abnormality, 3) altered growth, and 4) functional deficiency. Developmental toxicology was formerly often referred to as teratology.

Adverse effect: any treatment-related alteration from baseline that diminishes an organism's ability to survive, reproduce or adapt to the environment. Concerning developmental toxicology, taken in its widest sense it includes any effect which interferes with normal development of the conceptus, both before and after birth.

Altered growth: an alteration in offspring organ or body weight or size.

Alterations (anomalies): structural alterations in development that include both malformations and variations (28).

Malformation/Major Abnormality: Structural change considered detrimental to the animal (may also be lethal) and is usually rare.

Variation/Minor Abnormality: Structural change considered to have little or no detrimental effect on the animal; may be transient and may occur relatively frequently in the control population.

Conceptus: the sum of derivatives of a fertilised ovum at any stage of development from fertilisation until birth including the extra-embryonic membranes as well as the embryo or foetus.

Implantation (nidation): attachment of the blastocyst to the epithelial lining of the uterus, including its penetration through the uterine epithelium, and its embedding in the endometrium.

Embryo: the early or developing stage of any organism, especially the developing product of fertilisation of an egg after the long axis appears and until all major structures are present.

Embryotoxicity: detrimental to the normal structure, development, growth, and/or viability of an embryo.

Foetus: the unborn offspring in the post-embryonic period.

Foetotoxicity: detrimental to the normal structure, development, growth, and/or viability of a foetus.

Abortion: the premature expulsion from the uterus of the products of conception: of the embryo or of a nonviable foetus.

Resorption: a conceptus which, having implanted in the uterus, subsequently died and is being, or has been resorbed.

Early resorption: evidence of implantation without recognisable embryo/foetus

Late resorption: dead embryo or foetus with external degenerative changes

NOAEL: abbreviation for no-observed-adverse-effect level and is the highest dose or exposure level where no adverse treatment-related findings are observed.

1.3 REFERENCE SUBSTANCE U.K.

None.

1.4 PRINCIPLE OF THE TEST METHOD U.K.

Normally, the test substance is administered to pregnant animals at least from implantation to one day prior to the day of scheduled kill, which should be as close as possible to the normal day of delivery without risking loss of data resulting from early delivery. The test method is not intended to examine solely the period of organogenesis, (e.g. days 5-15 in the rodent, and days 6-18 in the rabbit) but also effects from preimplantation, when appropriate, through the entire period of gestation to the day before caesarean section. Shortly before caesarean section, the females are killed, the uterine contents are examined, and the foetuses are evaluated for externally visible anomalies and for soft tissue and skeletal changes.

1.5 DESCRIPTION OF THE TEST METHOD U.K.

1.5.1 Selection of animal species U.K.

It is recommended that testing be performed in the most relevant species, and that laboratory species and strains which are commonly used in prenatal developmental toxicity testing be employed. The preferred rodent species is the rat and the preferred non-rodent species is the rabbit. Justification should be provided if another species is used.

1.5.2 Housing and feeding conditions U.K.

The temperature in the experimental animal room should be 22 °C (\pm 3°) for rodents and 18 °C (\pm 3°) for rabbits. Although the relative humidity should be at least 30 % and preferably not exceed 70 % other than during room cleaning, the aim should be 50-60 %. Lighting should be artificial, the sequence being 12 hours light, 12 hours dark. For feeding, conventional laboratory diets may be used with an unlimited supply of drinking water. Mating procedures should be carried out in cages suitable for the purpose. While individual housing of mated animals is preferred, group housing in small numbers is also acceptable.

1.5.3 **Preparation of the animals U.K.**

Healthy animals, which have been acclimated to laboratory conditions for at least 5 days and have not been subjected to previous experimental procedures, should be used. The test animals should be characterised as to species, strain, source, sex, weight and/or age. The animals of all test groups should, as nearly as practicable, be of uniform weight and age. Young adult nulliparous female animals should be used at each dose level. The females should be mated with males of the same species and strain, and the mating of siblings should be avoided. For rodents day 0 of gestation is the day on which a vaginal plug and/or sperm are observed; for rabbits day 0 is usually the day of coitus or of artificial insemination, if this technique is used. Mated females

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should be assigned in an unbiased manner to the control and treatment groups. Cages should be arranged in such a way that possible effects due to cage placement are minimised. Each animal should be assigned a unique identification number. Mated females should be assigned in an unbiased manner to the control and treatment groups, and if the females are mated in batches, the animals in each batch should be evenly distributed across the groups. Similarly, females inseminated by the same male should be evenly distributed across the groups.

1.6 PROCEDURE U.K.

1.6.1 Number and sex of animals U.K.

Each test and control group should contain a sufficient number of females to result in approximately 20 female animals with implantation sites at necropsy. Groups with fewer than 16 animals with implantation sites may be inappropriate. Maternal mortality does not necessarily invalidate the study providing it does not exceed approximately 10 %.

1.6.2 **Preparation of doses U.K.**

If a vehicle or other additive is used to facilitate dosing, consideration should be given to the following characteristics: effects on the absorption, distribution, metabolism, and retention or excretion of the test substance; effects on the chemical properties of the test substance which may alter its toxic characteristics; and effects on the food or water consumption or the nutritional status of the animals. The vehicle should neither be developmentally toxic nor have effects on reproduction.

1.6.3 **Dosage** U.K.

Normally, the test substance should be administered daily from implantation (e.g., day 5 post mating) to the day prior to scheduled caesarean section. If preliminary studies, when available, do not indicate a high potential for preimplantation loss, treatment may be extended to include the entire period of gestation, from mating to the day prior to scheduled kill. It is well known that inappropriate handling or stress during pregnancy can result in prenatal loss. To guard against prenatal loss from factors which are not treatment-related, unnecessary handling of pregnant animals as well as stress from outside factors such as noise should be avoided.

At least three dose levels and a concurrent control should be used. Healthy animals should be assigned in an unbiased manner to the control and treatment groups. The dose levels should be spaced to produce a gradation of toxic effects. Unless limited by the physical/chemical nature or biological properties of the test substance, the highest dose should be chosen with the aim to induce some developmental and/or maternal toxicity (clinical signs or a decrease in body weight) but not death or severe suffering. At least one intermediate dose level should produce minimal observable toxic effects. The lowest dose level should not produce any evidence of either maternal or developmental toxicity. A descending sequence of dose levels should be selected with a view to demonstrating any dosage-related response and no-observed-adverse-effect level (NOAEL). Two- to fourfold intervals are frequently optimal for setting the descending dose levels, and the addition of a fourth test group is often preferable to using very large intervals (e.g. more than a factor of 10) between dosages. Although establishment of a maternal NOAEL is the goal, studies which do not establish such a level may also be acceptable (1).

Dose levels should be selected taking into account any existing toxicity data as well as additional information on metabolism and toxicokinetics of the test substance or related materials. This information will also assist in demonstrating the adequacy of the dosing regimen.

A concurrent control group should be used. This group should be a sham-treated control group or a vehicle-control group if a vehicle is used in administering the test substance. All groups

should be administered the same volume of either test substance or vehicle. Animals in the control group(s) should be handled in an identical manner to test group animals. Vehicle control groups should receive the vehicle in the highest amount used (as in the lowest treatment group).

1.6.4 Limit test U.K.

If a test at one dose level of at least 1000 mg/kg body weight/day by oral administration, using the procedures described for this study, produces no observable toxicity in either pregnant animals or their progeny and if an effect would not be expected based upon existing data (e.g., from structurally and/or metabolically related compounds), then a full study using three dose levels may not be considered necessary. Expected human exposure may indicate the need for a higher oral dose level to be used in the limit test. For other types of administration, such as inhalation or dermal application, the physico-chemical properties of the test substance often may indicate and limit the maximum attainable level of exposure (for example, dermal application should not cause severe local toxicity).

1.6.5 Administration of doses U.K.

The test substance or vehicle is usually administered orally by intubation. If another route of administration is used, the tester should provide justification and reasoning for its selection, and appropriate modifications may be necessary (2)(3)(4). The test substance should be administered at approximately the same time each day.

The dose to individual animals should normally be based on the most recent individual body weight determination. However, caution should be exercised when adjusting the dose during the last trimester of pregnancy. Existing data should be used for dose selection to prevent excess maternal toxicity. However, if excess toxicity is noted in the treated dams, those animals should be humanely killed. If several pregnant animals show signs of excess toxicity, consideration should be given to terminating that dose group. When the substance is administered by gavage, this should preferably be given as a single dose to the animals using a stomach tube or a suitable intubation canula. The maximum volume of liquid that can be administered at one time depends on the size of the test animal. The volume should not exceed 1 ml/100 g body weight, except in the case of aqueous solutions where 2 ml/100 g body weight may be used. When corn oil is used as a vehicle, the volume should not exceed 0.4 ml/100 g body weight. Variability in test volume should be minimised by adjusting the concentrations to ensure a constant volume across all dose levels.

1.6.6 **Observations of the dams U.K.**

Clinical observations should be made and recorded at least once a day, preferably at the same time(s) each day taking into consideration the peak period of anticipated effects after dosing. The condition of the animals should be recorded including mortality, moribundity, pertinent behavioural changes, and all signs of overt toxicity.

1.6.7 **Body weight and food consumption** U.K.

Animals should be weighed on day 0 of gestation or no later than day 3 of gestation if time-mated animals are supplied by an outside breeder, on the first day of dosing, at least every 3 days during the dosing period and on the day of scheduled kill.

Food consumption should be recorded at three-day intervals and should coincide with days of body weight determination.

1.6.8 **Post-mortem examination** U.K.

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Females should be killed one day prior to the expected day of delivery. Females showing signs of abortion or premature delivery prior to scheduled kill should be killed and subjected to a thorough macroscopic examination.

At the time of termination or death during the study, the dam should be examined macroscopically for any structural abnormalities or pathological changes. Evaluation of the dams during caesarean section and subsequent foetal analyses should be conducted preferably without knowledge of treatment group in order to minimise bias.

1.6.9 **Examination of uterine contents** U.K.

Immediately after termination or as soon as possible after death, the uteri should be removed and the pregnancy status of the animals ascertained. Uteri that appear non gravid should be further examined (e.g. by ammonium sulphide staining for rodents and Salewski staining or a suitable alternative method for rabbits) to confirm the non-pregnant status (5).

Gravid uteri including the cervix should be weighed. Gravid uterine weights should not be obtained from animals found dead during the study.

The number of corpora lutea should be determined for pregnant animals.

The uterine contents should be examined for numbers of embryonic or foetal deaths and viable foetuses. The degree of resorption should be described in order to estimate the relative time of death of the conceptus (see section 1.2).

1.6.10 Examination of foetuses U.K.

The sex and body weight of each foetus should be determined.

Each foetus should be examined for external alterations (6).

Foetuses should be examined for skeletal and soft tissue alterations (e.g. variations and malformations or anomalies) (7)(8)(9)(10)(11)(12)(13)(14)(15)(16)(17)(18)(19)(20)(21)(22) (23)(24). Categorisation of foetal alterations is preferable but not required. When categorisation is done, the criteria for defining each category should be clearly stated. Particular attention should be paid to the reproductive tract which should be examined for signs of altered development.

For rodents, approximately one-half of each litter should be prepared and examined for skeletal alterations. The remainder should be prepared and examined for soft tissue alterations, using accepted or appropriate serial sectioning methods or careful gross dissection techniques.

For non-rodents, e.g. rabbits, all foetuses should be examined for both soft tissue and skeletal alterations. The bodies of these foetuses are evaluated by careful dissection for soft tissue alterations, which may include procedures to further evaluate internal cardiac structure (25). The heads of one-half of the foetuses examined in this manner should be removed and processed for evaluation of soft tissue alterations (including eyes, brain, nasal passages and tongue), using standard serial sectioning methods (26) or an equally sensitive method. The bodies of these foetuses and the remaining intact foetuses should be processed and examined for skeletal alterations, utilising the same methods as described for rodents.

2 DATA U.K.

2.1 TREATMENT OF RESULTS U.K.

Data shall be reported individually for the dams as well as for their offspring and summarised in tabular form, showing for each test group and each generation the number of animals at the start of the test, the number of animals found dead during the test or killed for humane reasons,

the time of any death or humane kill, the number of pregnant females, the number of animals showing signs of toxicity, a description of the signs of toxicity observed, including time of onset, duration, and severity of any toxic effects, the types of embryo/foetal observations, and all relevant litter data.

Numerical results should be evaluated by an appropriate statistical method using the litter as the unit for data analysis. A generally accepted statistical method should be used; the statistical methods should be selected as part of the design of the study and should be justified. Data from animals that do not survive to the scheduled kill should also be reported. These data may be included in group means where relevant. Relevance of the data obtained from such animals, and therefore inclusion or exclusion from any group mean(s), should be justified and judged on an individual basis.

2.2 EVALUATION OF RESULTS U.K.

The findings of the Prenatal Developmental Toxicity Study should be evaluated in terms of the observed effects. The evaluation will include the following information:

- maternal and embryo/foetal test results, including the evaluation of the relationship, or lack thereof, between the exposure of the animals to the test substance and the incidence and severity of all findings;
- criteria used for categorising foetal external, soft tissue, and skeletal alterations if categorisation has been done;
- when appropriate, historical control data to enhance interpretation of study results;
- the numbers used in calculating all percentages or indices;
- adequate statistical analysis of the study findings, when appropriate, which should include sufficient information on the method of analysis, so that an independent reviewer/statistician can re-evaluate and reconstruct the analysis;

In any study which demonstrates the absence of any toxic effects, further investigations to establish absorption and bioavailability of the test substance should be considered.

2.3 INTERPRETATION OF RESULTS U.K.

A prenatal developmental toxicity study will provide information on the effects of repeated exposure to a substance during pregnancy on the dams and on the intrauterine development of their progeny. The results of the study should be interpreted in conjunction with the findings from subchronic, reproduction, toxicokinetic and other studies. Since emphasis is placed on both general toxicity in terms of maternal toxicity and on developmental toxicity endpoints, the results of the study will allow to a certain extent for the discrimination between developmental effects occurring in the absence of general toxicity and those which are only induced at levels that are also toxic to the maternal animal (27).

3 **REPORTING** U.K. TEST REPORT

The test report must include the following specific information:

Test substance:

- physical nature and, where relevant, physiochemical properties;
 identification including CAS number if known/established;
- purity.

Vehicle (if appropriate):

— justification for choice of vehicle, if other than water.

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Test anin	nals:
_	species and strain used;
_	number and age of animals;
	source, housing conditions, diet, etc.;
_	individual weights of animals at the start of the test.
Test cond	ditions:
_	rationale for dose level selection;
	details of test substance formulation/diet preparation, achieved concentration, stability and homogeneity of the preparation;
_	details of the administration of the test substance;
	conversion from diet/drinking water test substance concentration (ppm) to the actual dose (mg/kg body weight/day), if applicable;
	environmental conditions;
_	details of food and water quality.
Results:	
Maternal	toxic response data by dose, including but not limited to:
	the number of animals at the start of the test, the number of animals surviving, the number pregnant, and the number aborting, number of animals delivering early;
	day of death during the study or whether animals survived to termination;
_	data from animals that do not survive to the scheduled kill should be reported but not included in the inter-group statistical comparisons;
_	day of observation of each abnormal clinical sign and its subsequent course;
_	body weight, body weight change and gravid uterine weight, including, optionally, body weight change corrected for gravid uterine weight;
_	food consumption and, if measured, water consumption;
_	necropsy findings, including uterine weight;
_	NOAEL values for maternal and developmental effects should be reported.
Develop	mental endpoints by dose for litters with implants, including:
_	number of corpora lutea;
	number of implantations, number and percent of live and dead foetuses and resorptions;
_	number and percent of pre- and post-implantation losses.
Develop	mental endpoints by dose for litters with live foetuses, including:
—	number and percent of live offspring;
	sex ratio;
	foetal body weight, preferably by sex and with sexes combined;
	external, soft tissue, and skeletal malformations and other relevant alterations;
_	criteria for categorisation if appropriate;
_	total number and percent of foetuses and litters with any external, soft tissue, or skeletal alteration, as well as the types and incidences of individual anomalies and other relevant alterations.

Discussion of results.

Conclusions.

4 **REFERENCES** U.K.

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ANNEX2G U.K.

B.35.TWO-GENERATION REPRODUCTION TOXICITY STUDY

1. **METHOD** U.K.

This method is a replicate of the OECD TG 416 (2001).

1.1 INTRODUCTION U.K.

This method for two-generation reproduction testing is designed to provide general information concerning the effects of a test substance on the integrity and performance of the male and female reproductive systems, including gonadal function, the oestrus cycle, mating behaviour, conception, gestation, parturition, lactation, and weaning, and the growth and development

of the offspring. The study may also provide information about the effects of the test substance on neonatal morbidity, mortality, and preliminary data on prenatal and postnatal developmental toxicity and serve as a guide for subsequent tests. In addition to studying growth and development of the F1 generation, this test method is also intended to assess the integrity and performance of the male and female reproductive systems as well as growth and development of the F2 generation. For further information on developmental toxicity and functional deficiencies, either additional study segments can be incorporated into this protocol, consulting the methods for developmental toxicity and/or developmental neurotoxicity as appropriate, or these endpoints could be studied in separate studies, using the appropriate test methods.

1.2 PRINCIPLE OF THE TEST METHOD U.K.

The test substance is administered in graduated doses to several groups of males and females. Males of the P generation should be dosed during growth and for at least one complete spermatogenetic cycle (approximately 56 days in the mouse and 70 days in the rat) in order to elicit any adverse effects on spermatogenesis. Effects on sperm are determined by a number of sperm parameters (e.g., sperm morphology and motility) and in tissue preparation and detailed histopathology. If data on spermatogenesis are available from a previous repeated dose study of sufficient duration, e.g. a 90-day study, males of the P generation need not be included in the evaluation. It is recommended, however, that samples or digital recordings of sperm of the P generation are saved, to enable later evaluation. Females of the P generation should be dosed during growth and for several complete oestrus cycles in order to detect any adverse effects on oestrus cycle normality by the test substance. The test substance is administered to parental (P) animals during their mating, during the resulting pregnancies, and through the weaning of their F1 offspring. At weaning the administration of the substance is continued to F1 offspring during their growth into adulthood, mating and production of an F2 generation, until the F2 generation is weaned.

Clinical observations and pathological examinations are performed on all animals for signs of toxicity with special emphasis on effects on the integrity and performance of the male and female reproductive systems and on the growth and development of the offspring.

1.3 DESCRIPTION OF THE TEST METHOD U.K.

1.3.1 Selection of animal species U.K.

The rat is the preferred species for testing. If other species are used, justification should be given and appropriate modifications will be necessary. Strains with low fecundity or well-known high incidence of developmental defects should not be used. At the commencement of the study, the weight variation of animals used should be minimal and not exceed 20 % of the mean weight of each sex.

1.3.2 Housing and feeding conditions U.K.

The temperature in the experimental animal room should be $22\,^{\circ}C$ (± 3°). Although the relative humidity should be at least 30 % and preferably not exceed 70 % other than during room cleaning, the aim should be 50-60 %. Lighting should be artificial, the sequence being 12 hours light, 12 hours dark. For feeding, conventional laboratory diets may be used with an unlimited supply of drinking water. The choice of diet may be influenced by the need to ensure a suitable admixture of a test substance when administered by this method.

Animals may be housed individually or be caged in small groups of the same sex. Mating procedures should be carried out in cages suitable for the purpose. After evidence of copulation, mated females shall be single-caged in delivery or maternity cages. Mated rats may also be

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kept in small groups and separated one or two days prior to parturition. Mated animals shall be provided with appropriate and defined nesting materials when parturition is near.

1.3.3 **Preparation of animals U.K.**

Healthy young animals, which have been acclimated to laboratory conditions for at least 5 days and have not been subjected to previous experimental procedures, should be used. The test animals should be characterised as to species, strain, source, sex, weight and/or age. Any sibling relationships among the animals should be known so that mating of siblings is avoided. The animals should be randomly assigned to the control and treated groups (stratification by body weight is recommended). Cages should be arranged in such a way that possible effects due to cage placement are minimised. Each animal should be assigned a unique identification number. For the P generation, this should be done before dosing starts. For the F1 generation, this should be done at weaning for animals selected for mating. Records indicating the litter of origin should be maintained for all selected F1 animals. In addition, individual identification of pups as soon after birth as possible is recommended when individual weighing of pups or any functional tests are considered.

Parental (P) animals shall be about 5 to 9 weeks old at the start of dosing. The animals of all test groups shall, as nearly as practicable, be of uniform weight and age.

1.4 PROCEDURE U.K.

1.4.1 Number and sex of animals U.K.

Each test and control group should contain a sufficient number of animals to yield preferably not less than 20 pregnant females at or near parturition. For substances that cause undesirable treatment related effects (e.g. sterility, excessive toxicity at the high dose), this may not be possible. The objective is to produce enough pregnancies to assure a meaningful evaluation of the potential of the substance to affect fertility, pregnancy and maternal behaviour and suckling, growth and development of the F1 offspring from conception to maturity, and the development of their offspring (F2) to weaning. Therefore, failure to achieve the desired number of pregnant animals (i.e. 20) does not necessarily invalidate the study and should be evaluated on a case-by-case basis.

1.4.2 **Preparation of Doses U.K.**

It is recommended that the test substance be administered orally (by diet, drinking water or gavage) unless another route of administration (e.g. dermal or inhalation) is considered more appropriate.

Where necessary, the test substance is dissolved or suspended in a suitable vehicle. It is recommended that, wherever possible, the use of an aqueous solution/suspension be considered first, followed by consideration of a solution/emulsion in oil (e.g. corn oil) and then by possible solution in other vehicles. For vehicles other than water, the toxic characteristics of the vehicle must be known. The stability of the test substance in the vehicle should be determined.

1.4.3 **Dosage** U.K.

At least three dose levels and a concurrent control shall be used. Unless limited by the physical-chemical nature or biological effects of the test substance, the highest dose level should be chosen with the aim to induce toxicity but not death or severe suffering. In case of unexpected mortality, studies with a mortality rate of less than approximately 10 percent in the parental (P) animals would normally still be acceptable. A descending sequence of dose levels should be selected with a view to demonstrating any dosage related effect and no-observed-adverse-effects levels (NOAEL). Two to four fold intervals are frequently optimal for setting the descending

dose levels and addition of a fourth test group is often preferable to using very large intervals (e.g. more than a factor of 10) between dosages. For the dietary studies the dose interval should be not more than 3 fold. Dose levels should be selected taking into account any existing toxicity data, especially results from repeated dose studies. Any available information on metabolism and kinetics of the test compound or related materials should also be considered. In addition, this information will also assist in demonstrating the adequacy of the dosing regimen.

The control group shall be an untreated group or a vehicle-control group if a vehicle is used in administering the test substance. Except for treatment with the test substance, animals in the control group should be handled in an identical manner to the test group subjects. If a vehicle is used, the control group shall receive the vehicle in the highest volume used. If a test substance is administered in the diet, and causes reduced dietary intake or utilisation, then the use of a pair-fed control group may be considered necessary. Alternatively data from controlled studies designed to evaluate the effects of decreased food consumption on reproductive parameters may be used in lieu of a concurrent pair-fed control group.

Consideration should be given to the. following characteristics of vehicle and other additives: effects on the absorption, distribution, metabolism, or retention of the test substance; effects on the chemical properties of the test substance which may alter its toxic characteristics; and effects on the food or water consumption or the nutritional status of the animals.

1.4.4 Limit test U.K.

If an oral study at one dose level of at least 1000 mg/kg body weight/day or, for dietary or drinking water administration, an equivalent percentage in the diet or drinking water using the procedures described for this study, produces no observable toxic effects in either parental animals or their offspring and if toxicity would not be expected based upon data from structurally and /or metabolically related compounds, then a full study using several dose levels may not be considered necessary. The limit test applies except when human exposure indicates the need for a higher oral dose level to be used. For other types of administration, such as inhalation or dermal application, the physical-chemical properties of the test substance, such as solubility, often may indicate and limit the maximum attainable level of exposure.

1.4.5 Administration of doses U.K.

The animals should be dosed with the test substance on a 7-days-a-week basis. The oral route of administration (diet, drinking water, or gavage) is preferred. If another route of administration is used, justification shall be provided, and appropriate modifications may be necessary. All animals shall be dosed by the same method during the appropriate experimental period. When the test substance is administered by gavage, this should be done using a stomach tube. The volume of liquid administered at one time should not exceed 1 ml/100 g body weight (0.4 ml/100 g body weight is the maximum for corn oil), except in the case of aqueous solutions where 2 ml/100 g body weight may be used. Except for irritant or corrosive substances, which will normally reveal exacerbated effects with higher concentrations, variability in test volume should be minimised by adjusting the concentration to ensure a constant volume at all dose levels. In gavage studies, the pups will normally only receive test substance indirectly through the milk, until direct dosing commences for them at weaning. In diet or drinking water studies, the pups will additionally receive test substance directly when they commence eating for themselves during the last week of the lactation period.

For substances administered via the diet or drinking water, it is important to ensure that the quantities of the test substance involved do not interfere with normal nutrition or water balance. When the test substance is administered in the diet either a constant dietary concentration (ppm) or a constant dose level in terms of the body weight of the animal may be used; the alternative used must be specified. For a substance administered by gavage, the dose should be given at

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similar times each day, and adjusted at least weekly to maintain a constant dose level in terms of animal body weight. Information regarding placental distribution should be considered when adjusting the gavage dose based on weight.

1.4.6 Experimental schedules U.K.

Daily dosing of the parental (P) males and females shall begin when they are 5 to 9 weeks old. Daily dosing of the F1 males and females shall begin at weaning; it should be kept in mind that in cases of test substance administration via diet or drinking water, direct exposure of the F1 pups to the test substance may already occur during the lactation period. For both sexes (P and F1), dosing shall be continued for at least 10 weeks before the mating period. Dosing is continued in both sexes during the 2 week mating period. Males should be humanely killed and examined when they are no longer needed for assessment of reproductive effects. For parental (P) females, dosing should continue throughout pregnancy and up to the weaning of the F1 offspring. Consideration should be given to modifications in the dosing schedule based on available information on the test substance, including existing toxicity data, induction of metabolism or bioaccumulation. The dose to each animal should normally be based on the most recent individual body weight determination. However, caution should be exercised when adjusting the dose during the last trimester of pregnancy.

Treatment of the P and F1 males and females shall continue until termination. All P and F1 adult males and females should be humanely killed when they are no longer needed for assessment of reproductive effects. F1 offspring not selected for mating and all F2 offspring should be humanely killed after weaning.

1.4.7 **Mating procedure** U.K.

1.4.7.1 Parental (P) mating U.K.

For each mating, each female shall be placed with a single male from the same dose level (1:1 mating) until copulation occurs or 2 weeks have elapsed. Each day, the females shall be examined for presence of sperm or vaginal plugs. Day 0 of pregnancy is defined as the day a vaginal plug or sperm are found. In case pairing is unsuccessful, re-mating of females with proven males of the same group could be considered. Mating pairs should be clearly identified in the data. Mating of siblings should be avoided.

1.4.7.2 *F1 mating* U.K.

For mating the F1 offspring, at least one male and one female should be selected at weaning from each litter for mating with other pups of the same dose level but different litter, to produce the F2 generation. Selection of pups from each litter should be random when no significant differences are observed in body weight or appearance between the litter mates. In case these differences are observed, the best representatives of each litter should be selected. Pragmatically, this is best done on a body weight basis but it may be more appropriate on the basis of appearance. The F1 offspring should not be mated until they have attained full sexual maturity.

Pairs without progeny should be evaluated to determine the apparent cause of the infertility. This may involve such procedures as additional opportunities to mate with other proven sires or dams, microscopic examination of the reproductive organs, and examination of the oestrous cycles or spermatogenesis.

1.4.7.3 Second mating U.K.

In certain instances, such as treatment-related alterations in litter size or the observation of an equivocal effect in the first mating, it is recommended that the P or F1 adults be remated to produce a second litter. It is recommended to remate females or males, which have not

produced a litter with proven breeders of the opposite sex. If production of a second litter is deemed necessary in either generation, animals should be remated approximately one week after weaning of the last litter.

1.4.7.4 Litter size U.K.

Animals shall be allowed to litter normally and rear their offspring to weaning. Standardisation of litter sizes is optional. When standardisation is done, the method used should be described in detail.

1.5 OBSERVATIONS U.K.

1.5.1 Clinical observations U.K.

A general clinical observation should be made each day and, and in the case of gavage dosing its timing should take into account the anticipated peak period of effects after dosing. Behavioural changes, signs of difficult or prolonged parturition and all signs of toxicity should be recorded. An additional, more detailed examination of each animal should be conducted on at least a weekly basis and could conveniently be performed on an occasion when the animal is weighed. Twice daily, during the weekend once daily when appropriate, all animals should be observed for morbidity and mortality.

1.5.2 Body weight and food/water consumption of parent animals U.K.

Parental animals (P and F1) shall be weighed on the first day of dosing and at least weekly thereafter. Parental females (P and F1) shall be weighed at a minimum on gestation days 0, 7, 14, and 20 or 21, and during lactation on the same days as the weighing of litters and on the day the animals are killed. These observations should be reported individually for each adult animal. During the premating and gestation periods food consumption shall be measured weekly at a minimum. Water consumption shall be measured weekly at a minimum if the test substance is administered in the water.

1.5.3 **Oestrus cycle** U.K.

Estrous cycle length and normality are evaluated in P and F1 females by vaginal smears prior to mating, and optionally during mating, until evidence of mating is found. When obtaining vaginal/cervical cells, care should be taken to avoid disturbance of mucosa and subsequently, the induction of pseudopregnancy (1).

1.5.4 **Sperm parameters U.K.**

For all P and F1 males at termination, testis and epididymis weight shall be recorded and one of each organ reserved for histopathological examination (see section 1.5.7, 1.5.8.1). Of a subset of at least ten males of each group of P and F1 males, the remaining testes and epididymides should be used for enumeration of homogenisation-resistant spermatids and cauda epididymal sperm reserves, respectively. For this same subset of males, sperm from the cauda epididymides or vas deferens should be collected for evaluation of sperm motility and sperm morphology. If treatment-related effects are observed or when there is evidence from other studies of possible effects on spermatogenesis, sperm evaluation should be conducted in all males in each dose group; otherwise enumeration may be restricted to control and high-dose P and F1 males.

The total number of homogenisation-resistant testicular spermatids and cauda epididymal sperm should be enumerated (2)(3). Cauda sperm reserves can be derived from the concentration and volume of sperm in the suspension used to complete the qualitative evaluations, and the number of sperm recovered by subsequent mincing and/or homogenising of the remaining cauda tissue. Enumeration should be performed on the selected subset of males of all dose groups immediately after killing the animals unless video or digital recordings are made, or unless the specimens are

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freezed and analysed later. In these instances, the controls and high dose group may be analysed first. If no treatment-related effects (e.g., effects on sperm count, motility, or morphology) are seen the other dose groups need not be analysed. When treatment-related effects are noted in the high-dose group, then the lower dose groups should also be evaluated.

Epididymal (or ductus deferens) sperm motility should be evaluated or video taped immediately after sacrifice. Sperm should be recovered while minimising damage, and diluted for motility analysis using acceptable methods (4). The percentage of progressively motile sperm should be determined either subjectively of objectively. When computer-assisted motion analysis is performed (5)(6)(7)(8)(9)(10) the derivation of progressive motility relies on user-defined thresholds for average path velocity and straightness or linear index. If samples are videotaped (11) or the images are otherwise recorded at the time of necropsy, subsequent analysis of only control and high-dose P and F1 males may be performed unless treatment-related effects are observed; in that case, the lower dose groups should also be evaluated. In the absence of a video or digital image, all samples in all treatment groups should be analysed at necropsy.

A morphological evaluation of an epididymal (or vas deferens) sperm sample should be performed. Sperm (at least 200 per sample) should be examined as fixed, wet preparations (12) and classified as either normal or abnormal. Examples of morphologic sperm abnormalities would include fusion, isolated heads, and misshapen heads and/or tails. Evaluation should be performed on the selected subset of males of all dose groups either immediately after killing the animals, or, based on the video or digital recordings, at a later time. Smears, once fixed, can also be read at a later time. In these instances, the controls and high dose group may be analysed first. If no treatment-related effects (e.g., effects on sperm morphology) are seen the other dose groups need not be analysed. When treatment-related effects are noted in the high-dose group, then the lower dose groups should also be evaluated.

If any of the above sperm evaluation parameters have already been examined as part of a systemic toxicity study of at least 90 days, they need not necessarily be repeated in the two-generation study. It is recommended, however, that samples or digital recordings of sperm of the P generation are saved, to enable later evaluation, if necessary.

1.5.5 **Offspring** U.K.

Each litter should be examined as soon as possible after delivery (lactation day 0) to establish the number and sex of pups, stillbirths, live births, and the presence of gross anomalies. Pups found dead on day 0, if not macerated, should preferably be examined for possible defects and cause of death and preserved. Live pups should be counted and weighed individually at birth (lactation day 0) or on day 1, and on regular weigh days thereafter, e.g., on days 4, 7, 14, and 21 of lactation. Physical or behavioural abnormalities observed in the dams or offspring should be recorded.

Physical development of the offspring should be recorded mainly by body weight gain. Other physical parameters (e.g. ear and eye opening, tooth eruption, hair growth) may give supplementary information, but these data should preferably be evaluated in the context of data on sexual maturation (e.g. age and body weight at vaginal opening or balano-preputial separation) (13). Functional investigations (e.g. motor activity, sensory function, reflex ontogeny) of the F1 offspring before and/or after weaning, particularly those related to sexual maturation, are recommended if such investigations are not included in separate studies. The age of vaginal opening and preputial separation should be determined for F1 weanlings selected for mating. Anogenital distance should be measured at postnatal day 0 in F2 pups if triggered by alterations in F1 sex ratio or timing of sexual maturation.

Functional observations may be omitted in groups that otherwise reveal clear signs of adverse effects (e.g., significant decrease in weight gain, etc.). If functional investigations are made, they should not be done on pups selected for mating.

1.5.6 Gross necropsy U.K.

At the time of termination or death during the study, all parental animals (P and F1), all pups with external abnormalities or clinical signs, as well as one randomly selected pup/sex/litter from both the F1 and F2 generation, shall be examined macroscopically for any structural abnormalities or pathological changes. Special attention should be paid to the organs of the reproductive system. Pups that are humanely killed in a moribund condition and dead pups, when not macerated, should be examined for possible defects and/or cause of death and preserved.

The uteri of all primiparous females should be examined, in a manner which does not compromise histopathological evaluation, for the presence and number of implantation sites.

1.5.7 **Organ weights U.K.**

At the time of termination, body weight and the weight of the following organs of all P and F1 parental animals shall be determined (paired organs should be weighed individually):

- Uterus, ovaries;
- Testes, epididymides (total and cauda);
- Prostate:
- Seminal vesicles with coagulating glands and their fluids and prostate (as one unit);
- Brain, liver, kidneys, spleen, pituitary, thyroid and adrenal glands and known target organs.

Terminal body weights should be determined for F1 and F2 pups that are selected for necropsy. The following organs from the one randomly selected pup/sex/litter (see section 1.5.6) shall be weighed: Brain, spleen and thymus.

Gross necropsy and organ weight results should be assessed in context with observations made in other repeated dose studies, when feasible.

1.5.8 **Histopathology** U.K.

1.5.8.1 Parental Animals U.K.

The following organs and tissues of parental (P and F1) animals, or representative samples thereof, shall be fixed and stored in a suitable medium for histopathological examination.

- Vagina, uterus with cervix, and ovaries (preserved in appropriate fixative);
- One testis (preserved in Bouin's or comparable fixative), one epididymis, seminal vesicles, prostate, and coagulating gland;
- Previously identified target organ(s) from all P and F1 animals selected for mating.

Full histopathology of the preserved organs and tissues listed above should be performed for all high dose and control P and F1 animals selected for mating. Examination of the ovaries of the P animals is optional. Organs demonstrating treatment-related changes should also be examined in the low- and mid-dose groups to aid in the elucidation of the NOAEL. Additionally, reproductive organs of the low- and mid-dose animals suspected of reduced fertility, e.g., those that failed to mate, conceive, sire, or deliver healthy offspring, or for which oestrus cyclicity or sperm number, motility, or morphology were affected, should be subjected to histopathological evaluation. All gross lesions such as atrophy or tumours shall be examined.

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Detailed testicular histopathological examination (e.g. using Bouin's fixative, paraffin embedding and transverse sections of 4-5 µm thickness) should be conducted in order to identify treatment-related effects such as retained spermatids, missing germ cell layers or types, multinucleated giant cells or sloughing of spermatogenic cells into the lumen (14). Examination of the intact epididymis should include the caput, corpus, and cauda, which can be accomplished by evaluation of a longitudinal section. The epididymis should be evaluated for leukocyte infiltration, change in prevalence of cell types, aberrant cell types, and phagocytosis of sperm. PAS and haematoxylin staining may be used for examination of the male reproductive organs.

The postlactational ovary should contain primordial and growing follicles as well as the large corpora lutea of lactation. Histopathological examination should detect qualitative depletion of the primordial follicle population. A quantitative evaluation of primordial follicles should be conducted for F1 females; the number of animals, ovarian section selection, and section sample size should be statistically appropriate for the evaluation procedure used. Examination should include enumeration of the number of primordial follicles, which can be combined with small growing follicles, for comparison of treated and control ovaries (15)(16)(17)(18)(19).

1.5.8.2 Weanlings U.K.

Grossly abnormal tissue and target organs from all pups with external abnormalities or clinical signs, as well as from the one randomly selected pup/sex/litter from both the F1 and F2 generation which have not been selected for mating, shall be fixed and stored in a. suitable medium for histopathological examination. Full histopathological characterisation of preserved tissue should be performed with special emphasis on the organs of the reproductive system.

2 DATA U.K.

2.1 TREATMENT OF RESULTS U.K.

Data shall be reported individually and summarised in tabular form, showing for each test group and each generation the number of animals at the start of the test, the number of animals found dead during the test or killed for humane reasons, the time of any death or humane kill, the number of fertile animals, the number of pregnant females, the number of animals showing signs of toxicity, a description of the signs of toxicity observed, including time of onset, duration, and severity of any toxic effects, the types of parental and offspring observations, the types of histopathological changes, and all relevant litter data.

Numerical results should be evaluated by an appropriate, generally accepted statistical method; the. statistical methods should be selected as part of the design of the study and should be justified. Dose-response statistical models may be useful for analysing data. The report should include sufficient information on the method of analysis and the computer program employed, so that an independent reviewer/statistician can re-evaluate and reconstruct the analysis.

2.2 EVALUATION OF RESULTS U.K.

The findings of this two-generation reproduction toxicity study should be evaluated in terms of the observed effects including necropsy and microscopic findings. The evaluation will include the relationship, or lack thereof, between the dose of the test substance and the presence or absence, incidence and severity of abnormalities, including gross lesions, identified target organs, affected fertility, clinical abnormalities, affected reproductive and litter performance, body weight changes, effects on mortality and any other toxic effects. The physico-chemical properties of the test substance, and when available, toxicokinetics data should be taken into consideration when evaluating test results.

A properly conducted reproduction toxicity test should provide a satisfactory estimation of a no-effect level and an understanding of adverse effects on reproduction, parturition, lactation, postnatal development including growth and sexual development.

2.3 INTERPRETATION OF RESULTS U.K.

A two-generation reproduction toxicity study will provide information on the effects of repeated exposure to a substance during all phases of the reproductive cycle. In particular, the study provides information on the reproductive parameters, and on development, growth, maturation and survival of offspring. The results of the study should be interpreted in conjunction with the findings from subchronic, prenatal developmental and toxicokinetic and other available studies. The results of this study can be used in assessing the need for further testing of a chemical. Extrapolation of the results of the study to man is valid to a limited degree. They are best used to provide information on no-effect-levels and permissible human exposure (20)(21)(22)(23).

3 **REPORTING U.K.** TEST REPORT

The test report must include the following information:

absorption data (if available);

litter and pup weight data;

animals;

Test subs	stance:
_	physical nature and, where relevant, physicochemical properties;
	identification data;
	purity.
Vehicle ((if appropriate):
	justification for choice of vehicle if other than water.
Test anir	mals:
_	species/strain used;
_	number, age and sex of animals;
_	source, housing conditions, diet, nesting materials, etc.;
	individual weights of animals at the start of the test.
Test con	ditions:
_	rationale for dose level selection;
_	details of test substance formulation/diet preparation, achieved concentrations;
_	stability and homogeneity of the preparation;
_	details of the administration of the test substance;
	conversion from diet/drinking water test substance concentration (ppm) to the achieved dose (mg/kg body weight/day), if applicable;
	details of food and water quality.
Results:	
_	food consumption, and water consumption if available, food efficiency (body weight gain per gram of food consumed), and test material consumption for P and F1 animals,

except for the period of cohabitation and for at least the last third of lactation;

body weight at sacrifice and absolute and relative organ weight data for the parental

body weight data for P and F1 animals selected for mating;

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- nature, severity and duration of clinical observations (whether reversible or not);
- time of death during the study or whether animals survived to termination;
- toxic response data by sex and dose, including indices of mating, fertility, gestation, birth, viability, and lactation; the report should indicate the numbers used in calculating these indices;
- toxic or other effects on reproduction, offspring, post-natal growth, etc.;
- necropsy findings;
- detailed description of all histopathological findings;
- number of P and F1 females cycling normally and cycle length;
- total cauda epididymal sperm number, percent progressively motile sperm, percent morphologically normal sperm, and percent of sperm with each identified abnormality;
- time-to-mating, including the number of days until mating;
- gestation length;
- number of implantations, corpora lutea, litter size;
- number of live births and post-implantation loss;
- number of pups with grossly visible abnormalities, if determined the number of runts should be reported;
- data on physical landmarks in pups and other post natal developmental data; physical landmarks evaluated should be justified;
- data on functional observations in pups and adults, as applicable;
- tatistical treatment of results, where appropriate.

Discussion of results.

Conclusions, including NOAEL values for maternal and offspring effects.

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ANNEX 2H U.K.

B.42.SKIN SENSITISATION: LOCAL LYMPH NODE ASSAY

1. **METHOD** U.K.

This test method is equivalent to the OECD TG 429 (2002)

1.1 INTRODUCTION U.K.

The Local Lymph Node Assay (LLNA) has been sufficiently validated and accepted to justify its adoption as a new Method (1)(2)(3). This is the second method for assessing skin sensitisation potential of chemicals in animals. The other method (B.6) utilises guinea pig tests, notably the guinea pig maximisation test and the Buehler test (4).

The LLNA provides an alternative method for identifying skin sensitising chemicals and for confirming that chemicals lack a significant potential to cause skin sensitisation. This does not necessarily imply that in all instances the LLNA should be used in place of guinea pig test, but rather that the assay is of equal merit and may be employed as an alternative in which positive and negative results generally no longer require further confirmation.

The LLNA provides certain advantages with regard to both scientific progress and animal welfare. It studies the induction phase of skin sensitisation and provides quantitative data suitable for dose response assessment. The details of the validation of the LLNA and a review of the associated work have been published (5)(6)(7)(8). In addition, it should be noted that the mild/moderate sensitisers, which are recommended as suitable positive control substances for guinea pig test methods, are also appropriate for use with the LLNA (6)(8)(9).

The LLNA is an *in vivo* method and, as a consequence, will not eliminate the use of animals in the assessment of contact sensitising activity. It has, however, the potential to reduce the number of animals required for this purpose. Moreover, the LLNA offers a substantial refinement of the way in which animals are used for contact sensitisation testing. The LLNA is based upon consideration of immunological events stimulated by chemicals during the induction phase of sensitisation. Unlike guinea pig tests the LLNA does not require that challenged-induced dermal hypersensitivity reactions be elicited. Furthermore, the LLNA does not require the use of an adjuvant, as is the case for the guinea pig maximisation test. Thus, the LLNA reduces animal distress. Despite the advantages of the LLNA over traditional guinea pig tests, it should be recognised that there are certain limitations that may necessitate the use of traditional guinea pigs tests (e.g., false negative findings in the LLNA with certain metals, false positive findings with certain skin irritants)(10).

See also Introduction part B.

1.2 PRINCIPLE OF THE TEST METHOD U.K.

The basic principle underlying the LLNA is that sensitisers induce a primary proliferation of lymphocytes in the lymph node draining the site of chemical application. This proliferation is proportional to the dose applied (and to the potency of the allergen) and provides a simple means of obtaining an objective, quantitative measurement of sensitisation. The LLNA assesses this proliferation as a dose-response relationship in which the proliferation in test groups is compared to that in vehicle treated controls. The ratio of the proliferation in treated groups to that in vehicular controls, termed the Stimulation Index, is determined, and must be at least three before a test substance can be further evaluated as a potential skin sensitiser. The methods described here are based on the use of radioactive labelling to measure cell proliferation. However, other endpoints for assessment of proliferation may be employed provided there is justification and appropriate scientific support, including full citations and description of the methodology.

1.3 DESCRIPTION OF THE TEST METHOD U.K.

1.3.1 **Preparations U.K.**

1.3.1.1 *Housing and feeding conditions* U.K.

Animals should be individually housed. The temperature of the experimental animals room should be 22°C (±3°C). Although the relative humidity should be at least 30% and preferably not exceed 70% other than during room cleaning, the aim should be 50-60%. Lighting should be artificial, the sequence being 12 hours light, 12 hours dark. For feeding, conventional laboratory diets may be used with an unlimited supply of drinking water.

1.3.1.2 Preparation of animals U.K.

The animals are randomly selected, marked to permit individual identification (but not by any form of ear marking), and kept in their cages for at least 5 days prior to the start of dosing to allow for acclimatisation to the laboratory conditions. Prior to the start of treatment all animals are examined to ensure that they have no observable skin lesions.

1.3.2 **Test Conditions** U.K.

1.3.2.1 Experimental animals U.K.

The mouse is the species of choice for this test. Young adult female mice of CBA/Ca or CBA/J strain, which are nulliparous and non-pregnant are used. At the start of the study, animals should be between 8-12 weeks old, and the weight variation of the animals should be minimal and not exceed 20% of the mean weight. Other strains and males may be used when sufficient data are generated to demonstrate that significant strain and/or gender-specific differences in the LLNA response do not exist.

1.3.2.2 Reliability check U.K.

Positive controls are used to demonstrate appropriate performance of the assay and competency of the laboratory to successfully conduct the assay. The positive control should produce a positive LLNA response at an exposure level expected to give an increase in the stimulation index (SI) >3 over the negative control group. The positive control dose should be chosen such that the induction is clear but not excessive. Preferred substances are hexyl cinnamic aldehyde (CAS No 101-86-0, EINECS No 202-983-3) and mercaptobenzothiazole (CAS No 149-30-4, EINECS No 205-736-8). There may be circumstances in which, given adequate justification, other control substances, meeting the above criteria, may be used. While ordinarily a positive control group may be required in each assay, there may be situations in which test laboratories will have available historic positive control data to show consistency of a satisfactory response over a six-month or more extended period. In those situations, less frequent testing with positive controls may be appropriate at intervals no greater than 6 months. Although the positive control substance should be tested in the vehicle that is known to elicit a consistent response (e.g., acetone: olive oil), there may be certain regulatory situations in which testing in non-standard vehicle (clinically/chemically relevant formulation) will also be necessary. In such situation the possible interaction of a positive control with this unconventional vehicle should be tested.

1.3.2.3 *Number of animals, dose levels and vehicle selection.* U.K.

A minimum of four animals is used per dose group, with a minimum of three concentrations of the test substance, plus a negative control group treated only with the vehicle for the test substance, and, as appropriate, a positive control. In those cases in which individual animal data are to be collected, a minimum of five animals per dose group are used. Except for absence of

treatment with the test substance, animals in the control groups should be handled and treated in a manner identical to that of animals in the treatment groups.

Dose and vehicle selection should be based on the recommendations given in reference (1). Doses are selected from the concentration series 100%, 50%, 25%, 10%, 5%, 2.5%, 1%, 0.5% etc. Existing acute toxicity and dermal irritation data should be considered, where available, in selecting the three consecutive concentrations so that the highest concentration maximises exposure whilst avoiding systemic toxicity and excessive local skin irritation (2)(11).

The vehicle should be selected on the basis of maximising the test concentrations and solubility whilst producing a solution/suspension suitable for application of the test substance. In order of preference, recommended vehicles are acetone/olive oil (4:1 v/v), dimethylformamide, methyl ethyl ketone, propylene glycol and dimethyl sulphoxide (2)(10), but others may be used if sufficient scientific rationale is provided. In certain situations it may be necessary to use a clinically relevant solvent or the commercial formulation in which the test substance is marketed as an additional control. Particular care should be taken to ensure that hydrophilic materials are incorporated into a vehicle system, which wets the skin and does not immediately run off. Thus, wholly aqueous vehicles are to be avoided.

1.3.3 **Test procedure U.K.**

1.3.3.1 Experimental schedule U.K.

The experimental schedule of the assay is as follows:

• Day 1:

Individually identify and record the weight of each animal. Open application of $25\mu l$ of the appropriate dilution of the test substance, the vehicle alone, or the positive control (as appropriate), to the dorsum of each ear.

• Days 2 and 3:

Repeat the application procedure carried out on day 1.

• Days 4 and 5:

No treatment.

• *Day 6:*

Record the weight of each animal. Inject $250\mu l$ of phosphate-buffered saline (PBS) containing $20 \mu Ci$ (7.4e + 8 Bq) of 3H -methyl thymidine into all test and control mice via the tail vein. Alternatively inject $250 \mu L$ PBS containing $2 \mu Ci$ (7.4e + 7 Bq) of 125 I-iododeoxyuridine and 10^{-5} M fluorodeoxyuridine into all mice via the tail vein.

Five hours later, the animals are killed. The draining auricular lymph nodes from each ear are excised and pooled in PBS for each experimental group (pooled treatment group approach); alternatively pairs of lymph nodes from individual animals may be excised and pooled in PBS for each animal (individual animal approach). Details and diagrams of the node identification and dissection can be found in Annex I of reference 10.

1.3.3.2 Preparation of cell suspensions U.K.

A single cell suspension of lymph node cells (LNC) either from pooled treatment groups or bilaterally from individual animals is prepared by gentle mechanical disaggregation through 200 μ m-mesh stainless steel gauze. Lymph node cells are washed twice with an excess of PBS and precipitated with 5% trichloroacetic acid (TCA) at 4 °C for 18h (2). Pellets are either resuspended in 1 ml TCA and transferred to scintillation vials containing 10 ml of scintillation fluid for ³H-counting, or transferred directly to gamma counting tubes for ¹²⁵I-counting.

1.3.3.3 Determination of cell proliferation (incorporated radioactivity) U.K.

Incorporation of ³H-methyl thymidine is measured by β-scintillation counting as disintegrations per minute (DPM). Incorporation of ¹²⁵I-iododeoxyuridine is measured by ¹²⁵I-counting and also is expressed as DPM. Depending on the approach used, the incorporation will be expressed as DPM/treatment group (pooled approach) or the DPM/animal (individual approach).

1.3.3.4 *Observations* U.K.

1.3.3.4.1 Clinical observations

Animals should be carefully observed once daily for any clinical signs, either of local irritation at the application site or of systemic toxicity. All observations are systematically recorded with individual records being maintained for each animal.

1.3.3.4.2 Body Weights

As stated in section 1.3.3.1, individual animal body weights should be measured at the start of the test and at the scheduled kill of the animals.

1.3.4 Calculation of results U.K.

Results are expressed as the Stimulation Index (SI). When using the pooled approach, the SI is obtained by dividing the pooled radioactive incorporation for each treatment group by the incorporation of the pooled vehicle control group; this yields a mean SI. When using the individual approach, the SI is derived by dividing the mean DPM/animal within each test substance group and the positive control group by the mean DPM/animal for the solvent/vehicle control group. The average SI for vehicle treated controls is then 1.

Use of the individual approach to calculate the SI will enable the performance of a statistical analysis of the data. In choosing an appropriate method of statistical analysis the investigator should maintain an awareness of possible inequalities of variances and other related problems that may necessitate a data transformation or a non-parametric statistical analysis. An adequate approach for interpreting the data is to evaluate all individual data of treated and vehicle controls, and derive from these the best fitting dose response curve, taking confidence limits into account (8)(12)(13). However, the investigator should be alert to possible "outlier" responses for individual animals within a group that may necessitate the use of an alternative measure of response (e.g., median rather than mean) or elimination of the outlier.

The decision process with regard to a positive response includes a stimulation index ≥ 3 together with consideration of dose-response and, where appropriate, statistical significance (3)(6)(8) (12)(14).

If it is necessary to clarify the results obtained, consideration should be given to various properties of the test substance, including whether it has a structural relationship to known skin sensitisers, whether it causes excessive skin irritation and the nature of the dose response seen. These and other considerations are discussed in detail elsewhere (7).

2 DATA U.K.

Data should be summarised in tabular form showing the mean and individual DPM values and stimulation indexes for each dose (including vehicle control) group.

3 **REPORTING** U.K. TEST REPORT

The test report should contain the following information:

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Test	SII	กราล	nce:

- identification data (e.g., CAS number, if available; source; purity; known impurities; lotnumber);
- physical nature and physicochemical properties (e.g., volatility, stability, solubility);
- if mixture, composition and relative percentages of components.

Vehicle:

- identification data [purity; concentration (where appropriate); volume used]
- iustification for choice of vehicle.

Test animals:

- strain of mice used;
- microbiological status of the animals, when known;
- number, age and sex of animals;
- source of animals, housing conditions, diet, etc.

Test conditions:

- details of test substance preparation and application;
- justification for dose selection, including results from range finding study, if conducted; vehicle and test substance concentrations used and the total amount of substance applied
- details of food and water quality (including diet type/source, water source).

Reliability check:

- a summary of the results of the latest reliability check including information on substance, concentration and vehicle used.
- concurrent and/or historical positive and negative control data for testing laboratory

Results:

- individual weights of animals at the start of dosing and at scheduled kill.
- a table of mean (pooled approach) and individual (individual approach) DPM values as well as the range of values for both approaches and the stimulation indices for each dose (including vehicle control) group.
- statistical analysis where appropriate
- time course of onset and signs of toxicity, including dermal irritation at site of administration, if any, for each animal.

Discussion of results:

 A brief commentary on the results, the dose-response analysis, and statistical analyses, where appropriate, with a conclusion as to whether the test substance should be considered a skin sensitiser.

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B.43. NEUROTOXICITY STUDY IN RODENTS U.K.

1. **METHOD** U.K.

This method is equivalent of OECD TG 424 (1997).

This Test Method has been designed to obtain the information necessary to confirm or to further characterise the potential neurotoxicity of chemicals in adult animals. It can either be combined with existing Test Methods for repeated dose toxicity studies or to be carried out as a separate study. It is recommended that the OECD Guidance Document on Neurotoxicity

Testing Strategies and Methods (1) be consulted to assist in the design of studies based on this Test Method. This is particularly important when modifications of the observations and test procedures as recommended for routine use of this Method are considered. The Guidance Document has been prepared to facilitate the selection of other test procedures for use in specific circumstances. The assessment of developmental neurotoxicity is not the subject of this Method.

1.1 INTRODUCTION U.K.

In the assessment and evaluation of the toxic characteristics of chemicals, it is important to consider the potential for neurotoxic effects. Already the Test Method for repeated dose systemic toxicity includes observations that screen for potential neurotoxicity. This Test Method can be used to design a study to obtain further information on, or to confirm, the neurotoxic effects observed in the repeated dose systemic toxicity studies. However, consideration of the potential neurotoxicity of certain classes of chemicals may suggest that they may be more appropriately evaluated using this Method without prior indications of the potential neurotoxicity from repeated dose systemic toxicity studies. Such considerations include, for example:

- observation of neurological signs or neuropathological lesions in toxicity studies other than repeated dose systemic toxicity studies, or
- structural relationship or other information linking them to known neurotoxicants.

In addition there may be other instances when use of this Test Method is appropriate; for further details see (1).

This Method has been developed so that it can be tailored to meet particular needs to confirm the specific histopathological and behavioural neurotoxicity of a chemical as well as provide a characterization and quantification of the neurotoxic responses.

In the past, neurotoxicity was equated with neuropathy involving neuropathological lesions or neurological dysfunctions, such as seizure, paralysis or tremor. Although neuropathy is an important manifestation of neurotoxicity, it is now clear that there are many other signs of nervous system toxicity (e.g. loss of motor coordination, sensory deficits, learning and memory dysfunctions) that may not be reflected in neuropathy or other types of studies.

This neurotoxicity Test Method is designed to detect major neurobehavioural and neuropathological effects in adult rodents. While behavioural effects, even in the absence of morphological changes, can reflect an adverse impact on the organism, not all behavioural changes are specific to the nervous system. Therefore, any changes observed should be evaluated in conjunction with correlative histopathological, haematological or biochemical data as well as data on other types of systemic toxicity. The testing called for in this Method to provide a characterization and quantification of the neurotoxic responses includes specific histopathological and behavioural procedures that may be further supported by electrophysiological and/or biochemical investigations (1)(2)(3)(4).

Neurotoxicants may act on a number of targets within the nervous system and by a variety of mechanisms. Since no single array of tests is capable of thoroughly assessing the neurotoxic potential of all substances, it may be necessary to utilize other *in vivo* or *in vitro* tests specific to the type of neurotoxicity observed or anticipated.

This Test Method can also be used, in conjunction with the guidance set out in the OECD Guidance Document on Neurotoxicity Testing Strategies and Methods (1) to design studies intended to further characterize or increase the sensitivity of the dose-response quantification in order or better estimate a no-observed-adverse effect level or to substantiate known or suspected hazards of the chemical. For example, studies may be designed to identify and evaluate the neurotoxic mechanism(s) or supplement the data already available from the use of basic neurobehavioural and neuropathological observation procedures. Such studies need not

replicate data that would be generated from the use of the standard procedures recommended in this Method, if such data are already available and are not considered necessary for the interpretation of the results of the study.

This neurotoxicity study, when used alone or in combination, provides information that can:

- identify whether the nervous system is permanently or reversibly affected by the chemical tested;
- contribute to the characterization of the nervous system alterations associated with exposure to the chemical, and to understanding the underlying mechanism.
- determine dose-and time-response relationships in order to estimate a no-observedadverse-effect level (which can be used to establish safety criteria for the chemical).

This Test Method uses oral administration of the test substance. Other routes of administration (e.g. dermal or inhalation) may be more appropriate, and may require modification of the procedures recommended. Considerations of the choice of the route of administration depend on the human exposure profile and available toxicological or kinetic information.

1.2 DEFINITIONS U.K.

Adverse effect: is any treatment-related alteration from baseline that diminishes an organism's ability to survive, reproduce or adapt to the environment.

Dose: is the amount of test substance administered. Dose is expressed as weight (g, mg) or as weight of test substance per unit weight of the test animal (e.g. mg/Kg), or as constant dietary concentrations (ppm).

Dosage: is a general term comprising of dose, its frequency and the duration of dosing.

Neurotoxicity: is an adverse change in the structure or function of the nervous system that results from exposure to a chemical, biological or physical agent.

Neurotoxicant: is any chemical, biological or physical agent having the potential to cause neurotoxicity.

NOAEL: is the abbreviation for no-observed-adverse effect level and is the highest dose level where no adverse treatment-related findings are observed.

1.3 PRINCIPLE OF THE TEST METHOD U.K.

The test chemical is administered by the oral route across a range of doses to several groups of laboratory rodents. Repeated doses are normally required, and the dosing regimen may be 28 days, subchronic (90 days) or chronic (1 year or longer). The procedures set out in this Test Method may also be used for an acute neurotoxicity study. The animals are tested to allow the detection or the characterization of behavioural and/or neurological abnormalities. A range of behaviours that could be affected by neurotoxicants is assessed during each observation period. At the end of the test, a subset of animals of each sex from each group are perfused *in situ* and sections of the brain, spinal cord, and peripheral nerves are prepared and examined.

When the study is conducted as a stand-alone study to screen for neurotoxicity or to characterize neurotoxic effects, the animals in each group not used for perfusion and subsequent histopathology (see Table 1) can be used for specific neurobehavioural, neuropathological, neurochemical or electrophysiological procedures that may supplement the data obtained from the standard examinations required by this Method (1). These supplemental procedures can be particularly useful when empirical observations or anticipated effects indicate a specific type or target of a chemical's neurotoxicity. Alternatively, the remaining animals can be used for evaluations such as those called for in Test Methods for repeated dose toxicity studies in rodents.

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When the procedures of this Test Method are combined with those of other Test Methods, a sufficient number of animals is needed to satisfy the requirements for the observations of both studies.

1.4 DESCRIPTION OF THE TEST METHOD U.K.

1.4.1 Selection of animal species U.K.

The preferred rodent species is the rat, although other rodent species, with justification, may be used. Commonly used laboratory strains of young adult healthy animals should be employed. The females should be nulliparous and non-pregnant. Dosing should normally begin as soon as possible after weaning, preferably not later than when animals are six weeks, and, in any case, before the animals are nine weeks age. However, when this study is combined with other studies this age requirement may need adjustment. At the commencement of the study the weight variation of animals used should not exceed \pm 20 % of the mean weight of each sex. Where a repeated dose study of short duration is conducted as a preliminary to a long term study, animals from the same strain and source should be used in both studies.

1.4.2 Housing and feeding conditions U.K.

The temperature in the experimental animal room should be 22 $^{\circ}$ C (\pm 3 $^{\circ}$ C). Although the relative humidity should be at least 30 % and preferably not exceed 70 % other than during room cleaning, the aim should be 50-60 %. Lighting should be artificial, the sequence being 12 hours light, 12 hours dark. Loud intermittent noise should be kept to a minimum. For feeding, conventional laboratory diets may be used with an unlimited supply of drinking water. The choice of diet may be influenced by the need to ensure a suitable admixture of a test substance when administered by this method. Animals may be housed individually, or be caged in small groups of the same sex.

1.4.3 **Preparation of animals U.K.**

Healthy young animals are randomly assigned to the treatment and control groups. Cages should be arranged in such a way that possible effects due to cage placement are minimized. The animals are identified uniquely and kept in their cages for at least (5) five days prior the start of the study to allow for acclimatization to the laboratory conditions.

1.4.4 Route of administration and preparation of doses U.K.

This Test Method specifically addresses the oral administration of the test substance. Oral administration may be by gavage, in the diet, in drinking water or by capsules. Other routes of administration (e.g. dermal or inhalation) can be used but may require modification of the procedures recommended. Considerations of the choice of the route of administration depend on the human exposure profile and available toxicological or kinetic information. The rationale for choosing the route of administration as well as resulting modifications to the procedures of this Test Method should be indicated.

Where necessary, the test substance may be dissolved or suspended in a suitable vehicle. It is recommended that the use of an aqueous solution/suspension be considered first, followed by consideration of a solution/suspension in oil (e.g., corn oil) and then by possible solution/suspension in other vehicle. The toxic characteristics of the vehicle must be known. In addition, consideration should be given to the following characteristics of the vehicle: effects of the vehicle on absorption, distribution, metabolism, or retention of the test substance which may alter its toxic characteristics; and effects on the food or water consumption or the nutritional status of the animals.

1.5 PROCEDURES U.K.

1.5.1 Number and sex animals U.K.

When the study is conducted as a separate study, at least 20 animals (10 females and 10 males) should be used in each dose and control group for the evaluation of detailed clinical and functional observations. At least five males and five females, selected from these 10 males and 10 females, should be perfused *in situ* and used for detailed neurohistopathology at the end of the study. In cases where only a limited number of animals in a given dose group are observed for signs of neurotoxic effects, consideration should be given to the inclusion of these animals in those selected for perfusion. When the study is conducted in combination with a repeated dose toxicity study, adequate numbers of animals should be used to meet the objectives of both studies. The minimum numbers of animals per group for various combinations of studies are given in Table 1. If interim kills or recovery groups for observation of reversibility, persistence or delayed occurrence of toxic effects post treatment are planned or when supplemental observations are considered, then the number of animals should be increased to ensure that the number of animals required for observation and histopathology are available.

1.5.2 Treatment and control group U.K.

At least three dose groups and a control group should generally be used, but if from the assessment of other data, no effects would be expected at a repeated dose of 1000 mg/kg body weight/day, a limit test may be performed. If there are no suitable data available, a range finding study may be performed to aid in the determination of the doses to be used. Except for treatment with the test substance, animals in the control group should be handled in an identical manner to the test group subjects. If a vehicle is used in administering the test substance, the control group should receive the vehicle at the highest volume used.

1.5.3 **Reliability check** U.K.

The laboratory performing the study should present data demonstrating its capability to carry out the study and the sensitivity of the procedures used. Such data should provide evidence of the ability to detect and quantify, as appropriate, changes in the different end points recommended for observation, such as autonomic signs, sensory reactivity, limb grip strength and motor activity. Information on chemicals that cause different types of neurotoxic responses and could be used as positive control substances can be found in references 2 to 9. Historical data may be used if the essential aspects of the experimental procedures remain the same. Periodic updating of historical data is recommended. New data that demonstrate the continuing sensitivity of the procedures should be developed when some essential element of the conduct of the test or procedures has been changed by the performing laboratory.

1.5.4 **Dose selection** U.K.

Dose levels should be selected by taking into account any previously observed toxicity and kinetic data available for the test compound or related materials. The highest dose level should be chosen with the aim of inducing neurotoxic effects or clear systemic toxic effects. Thereafter, a descending sequence of dose levels should be selected with a view to demonstrating any dose-related response and no-observed-adverse effect (NOAEL) at the lowest dose level. In principle, dose levels should be set so that primary toxic effects on the nervous system can be distinguished from effects related to systemic toxicity. Two to three intervals are frequently optimum and addition of a fourth test group is often preferable to using very large intervals (e.g., more than a factor of 10) between dosages. Where there is a reasonable estimation of human exposure this should also be taken into account.

1.5.5 Limit test U.K.

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If a study at one dose level of at least 1000 mg/kg body weight/day, using the procedures described, produces no observable neurotoxic effects and if toxicity would not be expected based upon data from structurally related compounds, then a full study using three dose levels may not be considered necessary. Expected human exposure may indicate the need for a higher oral dose level to be used in the limit test. For other types of administration, such as inhalation or dermal application, the physical chemical properties of the test substance often may dictate the maximum attainable level of exposure. For the conduct of an oral acute study, the dose for a limit test should be at least 2000 mg/kg.

1.5.6 Administration of doses U.K.

The animals are dosed with the test substance daily, seven days each week, for a period at least 28 days; use of a five-day dosing regime or a shorter exposure period needs to be justified. When the test substance is administered by gavage, this should be done in a single dose using a stomach tube or a suitable intubation cannula. The maximum volume of a liquid that can be administered at one time depends on the size of the test animals. The volume should not exceed 1 ml/100 g body weight. However in the case of aqueous solutions, the use of up to 2 ml/100 g body weight can be considered. Except for irritating or corrosive substances, which will normally reveal exacerbated effects with higher concentrations, variability in test volume should be minimized by adjusting the concentration to ensure a constant volume at all dose levels.

For substances administered via the diet or drinking water, it is important to ensure that the quantities of the test substance involved do not interfere with normal nutrition or water balance. When the test substance is administered in the diet either a constant dietary concentration (ppm) or a constant dose level in terms of the animals' body weight may be used; the alternative used must be specified. For a substance administered by gavage, the dose should be given at similar times each day, and adjusted as necessary to maintain a constant dose level in terms of animal body weight. Where a repeat dose study is used as a preliminary to a long term study, a similar diet should be used in both studies. For acute studies, if a single dose is not possible, the dose may be given in smaller fractions over a period not exceeding 24 hours.

1.6 OBSERVATION U.K.

1.6.1 Frequency of observations and tests U.K.

In repeated dose studies, the observation period should cover the dosage period. In acute studies, 14-day post-treatment period should be observed. For animals in satellite groups which are kept without exposure during a post-treatment period, observations should cover this period as well.

Observations should be made with sufficient frequency to maximize the probability of detection of any behavioural and/or neurological abnormalities. Observations should be made preferably at the same times each day with consideration given to the peak period of anticipated effects after dosing. The frequency of clinical observations and functional tests is summarized in Table 2. If kinetic or other data generated from previous studies indicates the need to use different time points for observations, tests or post-observation periods, an alternative schedule should be adopted in order to achieve maximum information. The rationale for changes to the schedule should be provided.

1.6.1.1 Observations of general health condition and mortality/morbidity U.K.

All animals should be carefully observed at least once daily with respect to their health condition as well as at least twice daily for morbidity and mortality.

1.6.1.2 Detailed clinical observations U.K.

Detailed clinical observations should be made on all animals selected for this purpose (see Table 1) once before the first exposure (to allow for within-subject comparisons) and at different intervals thereafter, dependant on the duration of the study (see Table 2). Detailed clinical observations on satellite recovery groups should be made at the end of the recovery period. Detailed clinical observations should be made outside the home cage in a standard arena. They should be carefully recorded using scoring systems that include criteria or scoring scales for each measurement in the observations. The criteria or scales used should be explicitly defined by the testing laboratory. Effort should be made to ensure that variations in the test conditions are minimal (not systematically related to treatment) and that observations are conducted by trained observers unaware of the actual treatment.

it is recommended that the observations be carried out in a structured fashion in which well-defined criteria (including the definition of the normal "range") are systematically applied to each animal at each observation time. The "normal range" should be adequately documented. All observed signs should be recorded. Whenever feasible, the magnitude of the observed signs should also be recorded. Clinical observations should include, but not be limited to, changes in skin, fur, eyes, mucous membranes, occurrence of secretions and excretions and autonomic activity (e.g., lacrimation, piloerection, pupil size, unusual respiratory pattern and/or mouth breathing, any unusual signs of urination or defecation, and discoloured urine).

Any unusual responses with respect to body position, activity level (e.g., decreased or increased exploration of the standard arena) and co-ordination of movement should also be noted. Changes in gait (e.g., waddling, ataxia), posture (e.g., hunched-back) and reactivity to handling, placing or other environmental stimuli, as well as the presence of clonic or tonic movements, convulsions or tremors, stereotypes (e.g., excessive grooming, unusual head movements, repetitive circling) or bizarre behaviour (e.g., biting or excessive licking, self mutilation, walking backwards, vocalization) or aggression should be recorded.

1.6.1.3 Functional tests U.K.

Similar to the detailed clinical observations, functional tests should also be conducted once prior to exposure and frequently thereafter in all animals selected for this purpose (see Table 1). The frequency of functional testing is also dependent on the study duration (see Table 2). In addition to the observation periods as set out in Table 2, functional observations on satellite recovery groups should also be made as close as possible to the terminal kill. Functional tests should include sensory reactivity to stimuli of different modalities [e.g., auditory, visual and proprioceptive stimuli (5)(6)(7)], assessment of limb grip strength (8) and assessment of motor activity (9). Motor activity should be measured with an automated device capable of detecting both decreases and increases in activity. If another defined system is used it should be quantitative and its sensitivity and reliability should be demonstrated. Each device should be tested to ensure reliability across time and consistency between devices. Further details of the procedures that can be followed are given in the respective references. If there are no data (e.g. structure-activity, epidemiological data, other toxicology studies) to indicate the potential neurotoxic effects, the inclusion of more specialized tests of sensory and motor function or learning and memory to examine these possible effects in greater details should be considered. More information on more specialized tests and their use is provided in (1).

Exceptionally, animals that reveal signs of toxicity to an extent that would significantly interfere with the functional test may be omitted from that test. Justification for the elimination of animals from a functional test should be provided.

1.6.2 **Body weight and food/water consumption** U.K.

For studies up to 90 days duration, all animals should be weighed at least once a week and measurements should be made of food consumption (water consumption, when the test

substance is administered by that medium) at least weekly. For long term studies, all animals should be weighed at least once at week for the first 13 weeks and at least once every 4 weeks thereafter. Measurements should be made of food consumption (water consumption, when the test substance is administered by that medium) at least weekly for the first 13 weeks and then at approximately three-month intervals unless the health status or body weight changes dictate otherwise.

1.6.3 **Ophthalmology** U.K.

For studies longer than 28 days duration, ophthalmologic examination, using an ophthalmoscope or an equivalent suitable instrument, should be made prior to the administration of the test substance and at the termination of the study, preferably on all animals, but at least on animals in the high dose and control groups. If changes in the eyes are detected or, if clinical signs indicate the need, all animals should be examined. For long term studies, an ophthalmologic examination should also be carried out at 13 weeks. Ophthalmologic examinations need not to be conducted if this data is already available from others studies of similar duration and at similar dose levels.

1.6.4 Haematology and clinical biochemistry U.K.

When the neurotoxicity study is carried out in combination with a repeated dose systemic toxicity study, haematological examinations and clinical biochemistry determinations should be carried out as set out in the respective Method of the systemic toxicity study. Collection of samples should be carried out in such a way that any potential effects on neurobehaviour are minimized.

1.6.5 **Histopathology** U.K.

The neuropathological examination should be designed to complement and extend the observations made during the *in vivo* phase of the study. Tissues from at least 5 animals/sex/group (see Table 1 and next paragraph) should be fixed *in situ*, using generally recognized perfusion and fixation techniques (see reference 3, chapter 5 and reference 4, chapter 50). Any observable gross changes should be recorded. When the study is conducted as a standalone study screen for neurotoxicity or to characterize neurotoxic effects, the remainder of the animals may be used either for specific neurobehavioural (10)(11), neuropathological (10)(11) (12)(13), neurochemical (10)(11)(14)(15) or electrophysiological (10)(11)(16)(17) procedures that may supplement the procedures and examinations described here, or to increase the number of subjects examined for histophatology. These supplementary procedures are of particular use when empirical observations or anticipated effects indicate a specific type or target of neurotoxicity (2)(3). Alternatively, the remainder of the animals can also be used for routine pathological evaluations as described in Method for repeated dose studies.

A general staining procedure, such as haematoxylin and eosin (H&E), should be performed on all tissue specimens embedded in paraffin and microscopic examination should be carried out. If signs of peripheral neuropathy are observed or suspected, plastic-embedded samples of peripheral nerve tissue should be examined. Clinical signs may also suggest additional sites for examination or the use of special staining procedures. Guidance on additional sites to be examined can be found in (3)(4). Appropriate special stains to demonstrate specific types of pathological change may also be helpful (18).

Representative sections of the central and peripheral nervous system should be examined histologically (see reference 3, chapter 5 and reference 4, chapter 50). The areas examined should normally include: the forebrain, the centre of the cerebrum, including a section through the hippocampus, the midbrain, the cerebellum, the pons, the medulla oblongata, the eye with optic nerve and retina, the spinal cord at the cervical and lumbar swellings, the dorsal root

ganglia, the dorsal and ventral root fibres, the proximal sciatic nerve, the proximal tibial nerve (at the knee) and the tibial nerve calf muscle branches. The spinal cord and peripheral nerve sections should include both cross or transverse and longitudinal sections. Attention should be given to the vasculature of the nervous system. A sample of skeletal muscle, particularly calf muscle, should also be examined. Special attention should be paid to sites with cellular and fibre structure and pattern in the CNS and PNS known to be particularly affected by neurotoxicants.

Guidance on neurophatological alterations that typically result from toxicant exposure can be found in the references (3)(4). A stepwise examination of tissue samples is recommended in which sections from the high dose group are first compared with those of the control group. If no neurophatological alterations are observed in the samples from these groups, subsequent analysis is not required. If neuropathological alterations are observed in the high dose group, sample from each of the potentially affected tissues from the intermediate and low dose groups should then be coded and examined sequentially.

If any evidence of neuropathological alterations is found in the qualitative examination, then a second examination should be performed on all regions of the nervous system showing these alterations. Sections from all dose groups from each of the potentially affected regions should be coded and examined at random without knowledge of the code. The frequency and severity of each lesion should be recorded. After all regions from all dose groups have been rated, the code can be broken and statistical analysis performed to evaluate dose-response relationships. Examples of different degrees of severity of each lesion should be described.

The neuropathological findings should be evaluated in the context of behavioural observations and measurements, as well as other data from preceding and. concurrent systemic toxicity studies of the test substance.

2 DATA U.K.

2.1 TREATMENT OF RESULTS U.K.

Individual data should be provided. Additionally, all data should be summarized in tabular form showing for each test or control group the number of animals at the start of the test, the number of animals found dead during the test or killed for humane reasons and the time of any death or humane kill, the number showing signs of toxicity, a description of the signs of toxicity observed, including time of onset, duration, type and severity of any toxic effects, the number of animals showing lesions, including the type and severity of the lesion(s).

2.2 EVALUATION AND INTERPRETATION OF RESULTS U.K.

The findings of the study should be evaluated in terms of the incidence, severity and correlation of neurobehavioural and neuropathological effects (neurochemical or electrophysiological effects as well if supplementary examinations are included) and any other adverse effects observed. When possible, numerical results should be evaluated by an appropriate and generally acceptable statistical method. The statistical methods should be selected during the design of the study.

3 **REPORTING U.K.**

TEST REPORT

The test report must include the following information:

Test substance:

- physical nature (including isomerism, purity and physicochemical properties);
- identification data.

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Vehicle (if appropriate):
_ `	justification for choice of vehicle.
Test anin	nals:
	species/strain used;
_	number, age and sex of animals;
	source, housing conditions, acclimatization, diet, etc;
_	individual weights of animals at the start of the test.
Test cond	ditions:
	details of test substance formulation/diet preparation, achieved concentration, stability and homogeneity of the preparation:
_	specification of the doses administered, including details of the vehicle, volume and physical form of the material administered;
	details of the administration of the test substance;
_	rationale for dose levels selected;
_	rationale for the route and duration of the exposure;
	conversion from diet/drinking water test substance concentration (ppm) to the actual dose (mg/kg body weight/day), if applicable;
_	details of the food and water quality.
Observat	tion and Test Procedures:
	details of the assignment of animals in each group to the perfusion subgroups;
_	details of scoring systems, including criteria and scoring scales for each measurement in the detailed clinical observations;
_	details on the functional tests for sensory reactivity to stimuli of different modalities (e.g., auditory, visual and proprioceptive); for assessment of limb grip strength; for motor activity assessment (including
	details of automated devices for detecting activity); and other procedures used;
	details of ophthalmologic examinations and, if appropriate, haematological examinations and clinical biochemistry tests with relevant base-line values;
	details for specific neurobehavioural, neuropathological, neurochemical or electrophysiological procedures.
Results:	
_	body weight/body weight changes including body weight at kill;
	food consumption and water consumption, as appropriate;
	toxic response data by sex and dose level, including signs of toxicity or mortality;
_	nature, severity and duration (time of onset and subsequent course) of the detailed clinical observations (whether reversible or not);
	a detailed description of all functional test results;
_	necropsy findings;
_	a detailed description of all neurobehavioural, neuropathological, and neurochemical or electrophysiological findings, if available;
_	absorption and metabolism data, if available;
_	statistical treatment of results, where appropriate.
Discussion	on of results;

dose response information;

- relationship of any other toxic effects to a conclusion about the neurotoxic potential of the test chemical;
- no-observed-adverse effect level.

Conclusions:

a specific statement of the overall neurotoxicity of the test chemical is encouraged.

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TABLE 1:

Minimum numbers of animals needed per group when the neurotoxicity study is conducted separately or in combination whit studies

	NEUROTOXICITY STUDY CONDUCTED AS:					
	Separate study	Combined study with the 28-day study	Combined study with the 90-day study	Combined study with the chronic toxicity study		
Total number of animals per group	10males and 10 females	10males and 10 females	15males and 15 females	25males and 25 females		
Number of animals selected for functional testing including detailed clinical observations	10males and 10 females	10males and 10 females	10males and 10 females	10males and 10 females		
Number of animals selected per perfusion in situ and neurohistopatholo	5males and 5 females	5males and 5 females	5males and 5 females	5males and 5 females		
Number of animals selected for repeated dose/subchronic/ chronic toxicity observations, haematology, clinical biochemistry, histopathology, etc. as indicate		5males and 5 females	10males ^a and 10 females ^a	20males*and 20 females*		

a Includes five animals selected for functional testing and detailed clinical observations as part of the neurotoxicity study

in the respective Guidelines			
Supplemental observations, as appropriate	5males and 5 females		

a Includes five animals selected for functional testing and detailed clinical observations as part of the neurotoxicity study

TABLE 2:

Frequency of clinical observation and functional tests

Type of observations		Study duration							
		Acute		28-day		90-day		Chronic	
In all animals	General health condition	daily		daily		daily		daily	
	Mortality/ morbidity	Twice d	aily	Twice da	aily	Twice d	aily	Twice d	aily
In animals selected for functional observations	Detailed clinical observations		prior to first expos within 8 hours of dosin at estime of peak effect at day 7 and 14 after dosin	sure n— g ate	prior to first expos once week theres	sure — Iy	prior to first exposonce durin the first or secon week of exposonont there:	g nd sure hly	prior to first exposure once at the end of the first month of exposure every three months thereaft
	Functional tests	_	prior to first expos within 8 hours of dosin at estim	sure n—	prior to first expos durin the fourtl week of treatm as	sure g	prior to first expos once durin the first or secor week	sure — g	prior to first exposur once at the end of the first

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	_	time of peak effect at day 7 and 14 after dosing	close as possible to the end of the exposure period	of exposure monthly thereafter	month of exposure every three months thereafter
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ANNEX 2I U.K.

C.21.SOIL MICROORGANISMS: NITROGEN TRANSFORMATION TEST

1. **METHOD** U.K.

This test method is a replicate of OECD TG 216 (2000).

1.1 INTRODUCTION U.K.

This Testing method describes a laboratory method designed to investigate the long-term effects of chemicals, after a single exposure, on nitrogen transformation activity of soil microorganisms. The test is principally based on the recommendations of the European and Mediterranean Plant Protection Organization (1). However, other guideline, including those of the German Biologische Bundesanstalt (2), the US Environmental Protection Agency (3) SET AC (4) and the International Organization for Standardization (5), were also taken into account. An OECD Workshop on soil/sediment Selection held at Belgirate, Italy, in 1995 (6) agreed on the number and type of soils for use in this test. Recommendations for collection, handling and storage of soil sample are based on an ISO Guidance Document (7) and recommendations from the Belgirate Workshop. In the assessment and evaluation of toxic characteristics of test substances, determination of effects on soil microbial activity may be required, e.g. when data on the potential side effects of crop protection products on soil microflora are required or when exposure of soil microorganisms to chemicals other than crop protection products is expected. The nitrogen transformation test is carried out to determine the effects of such chemicals on soil microflora. If agrochemicals (e.g. crop protection products, fertilisers, forestry chemicals) are tested, both nitrogen transformation and carbon transformation tests are conducted. If non agrochemicals are tested, the nitrogen transformation test is sufficient. However, if EC₅₀ values of the nitrogen transformation test for such chemicals fall within the range found for commercially available nitrification inhibitors (e.g. nitrapyrin), a carbon transformation test can be conducted to gain further information.

Soils consist of living and non-living components which exist in complex and heterogeneous mixtures. Microorganisms play an important role in break-down and transformation of organic matter in fertile soils with many species contributing to different aspects of soil fertility. Any long-term interference with these biochemical processes could potentially interfere with nutrient cycling and this could alter soil fertility. Transformation of carbon and nitrogen occurs in all fertile soils. Although the microbial communities responsible for these processes differ from soil to soil, the pathways of transformation are essentially the same.

This Testing method described is designed to detect long-term adverse effects of a substance on the process of nitrogen transformation in aerobic surface soils. The test method also allows estimation of the effects of substances on carbon transformation by the soil microflora. Nitrate formation takes place subsequent to the degradation of carbon-nitrogen bonds. Therefore, if equal rates of nitrate production are found in treated and control soils, it is highly probable that the major carbon degradation pathways are intact and functional. The substrate chosen for the test (powdered lucerne meal) has a favourable carbon to nitrogen ratio (usually between 12/1 and 16/1). Because of this, carbon starvation is reduced during the test and if microbial communities are damaged by a chemical, they might recover within 100 days.

The tests from which this Testing Method was developed were primarily designed for substances for which the amount reaching the soil can be anticipated. This is the case, for example, for crop protection products for which the application rate in the field is known. For agrochemicals, testing of two doses relevant to the anticipated or predicted application rate is sufficient. Agrochemicals can be tested as active ingredients (a.i.) or as formulated products. However, the test is not limited to agrochemicals. By changing both the amounts of test substance applied to the soil, and the way in which the data are evaluated, the test can also be used for chemicals for which the amount expected to reach the soil is not known. Thus, with chemicals other than agrochemicals, the effects of a series of concentrations on nitrogen transformation are determined. The data from these tests are used to prepare a dose-response curve and calculate EC_x values, where x is defined % effect.

1.2 DEFINITIONS U.K.

Nitrogen transformation: is the ultimate degradation by microorganisms of nitrogencontaining organic matter, via the process of ammonification and nitrification, to the respective inorganic end-product nitrate.

 EC_x (Effective Concentration): is the concentration of the test substance in soil that results in a x percent inhibition of nitrogen transformation to nitrate.

 EC_{50} (Median Effective Concentration): is the concentration of the test substance in soil that results in a 50 percent (50%) inhibition of nitrogen transformation to nitrate.

1.3 REFERENCE SUBSTANCES U.K.

None.

1.4 PRINCIPLE OF THE TEST METHOD U.K.

Sieved soil is amended with powdered plant meal and either treated with the test substance or left untreated (control). If agrochemicals are tested, a minimum of two test concentrations are recommended and these should be chosen in relation to the highest concentration anticipated in the field. After 0, 7, 14 days and 28 days of incubation, samples of treated and control soils are extracted with an appropriate solvent, and the quantities of nitrate in the extracts are determined. The rate of nitrate formation in treated samples is compared with the rate in the controls, and the percent deviation of the treated from the control is calculated. All tests run for at least 28 days. If, on the 28th day, differences between treated and untreated soils are equal to or greater than 25%, measurements are continued to a maximum of 100 days. If non agrochemicals are tested, a series of concentrations of the test substance are added to samples of the soil, and the quantities of nitrate formed in treated and control samples are measured after 28 days of incubation. Results from tests with multiple concentrations are analysed using a regression model, and the EC_X values are calculated (i.e. EC_{50} , EC_{25} and/or EC_{10}). See definitions.

1.5 VALIDITY OF THE TEST U.K.

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Evaluations of test results with agrochemicals are based on relatively small differences (i.e. average value $\pm 25\%$) between nitrate concentrations in control and treated soil samples, so large variations in the controls can lead to false results. Therefore, the variation between replicate control samples should be less than $\pm 15\%$.

1.6 DESCRIPTION OF THE TEST METHOD U.K.

1.6.1 **Apparatus** U.K.

Test containers made of chemically inert material are used. They should be of a suitable capacity in compliance with the procedure used for incubation of soils, i.e. incubation in bulk or as a series of individual soil samples (see section 1.7.1.2). Care should be taken both to minimise water loss and to allow gas exchange during the test (e.g. the test containers may be covered with perforated polyethylene foil). When volatile substances are tested, sealable and gas-tight containers should be used. These should be of a size such that approximately one quarter of their volume is filled with the soil sample.

Standard laboratory equipment including the following is used:

- agitation device: mechanical shaker or equivalent equipment;
- centrifuge (3000 g) or filtration device (using nitrate-free filter paper);
- instrument of adequate sensitivity and reproducibility for nitrate analysis.

1.6.2 Selection and number of soils U.K.

One single soil is used. The recommended soil characteristics are as follows:

- sand content: not less than 50% and not greater than 75%;
- pH: 5.5 -7.5;
- organic carbon content: 0.5 1.5%;
- the microbial biomass should be measured (8)(9) and its carbon content should be at least 1 % of the total soil organic carbon.

In most cases, a soil with these characteristics represents a worst case situation, since adsorption of the test chemical is minimum and its availability to the microflora is maximum. Consequently, tests with other soils are generally unnecessary. However, in certain circumstances, e.g. where the anticipated major use of the test substance is in particular soils such as acidic forest soils, or for electrostatically charged chemicals, it may be necessary to use an additional soil.

1.6.3 Collection and storage of soil samples U.K.

1.6.3.1 Collection U.K.

Detailed information on the history of the field site from where the test soil is collected should be available. Details include exact location, vegetation cover, dates of treatments with crop protection products, treatments with organic and inorganic fertilisers, additions of biological materials or accidental contaminations. The site chosen for soil collection should be one which allows long-term use. Permanent pastures, fields with annual cereal crops (except maize) or densely sown green manures are suitable. The selected sampling site should not have been treated with crop protection products for a minimum of one year before sampling. Also, no organic fertiliser should have been applied for at least six months. The use of mineral fertiliser is only acceptable when in accordance with the requirements of the crop and soil samples should not be taken until at least three months after fertiliser application. The use of soil treated with fertilisers with known biocidal effects (e.g. calcium cyanamide) should be avoided.

Sampling should be avoided during or immediately following long periods (greater than 30 days) of drought or water logging. For ploughed soils, samples should be taken from a depth of

0 down to 20 cm. For grassland (pasture) or other soils where ploughing does not occur over longer periods (at least one growing season), the maximum depth of sampling may be slightly more than 20 cm (e.g. to 25 cm).

Soil samples should be transported using containers and under temperature conditions which guarantee that the initial soil properties are not significantly altered.

1.6.3.2 Storage U.K.

The use of soils freshly collected from the field is preferred. If storage in the laboratory cannot be avoided, soils may be stored in the dark at $4\pm2^{\circ}$ C for a maximum of three months. During the storage of soils, aerobic conditions must be ensured. If soils are collected from areas where they are frozen for at least three months per year, storage for six months at minus 18° C to minus 22° C can be considered. The microbial biomass of stored soils is measured prior to each experiment and the carbon in the biomass should be at least 1 % of the total soil organic carbon content (see section 1.6.2).

1.6.4 Handling and preparation of soil for the test U.K.

1.6.4.1 *Pre-incubation* U.K.

If the soil was stored (see section 1.6.3.2), pre-incubation is recommended for a period between 2 and 28 days. The temperature and moisture content of the soil during pre-incubation should be similar to that used in the test (see sections 1.6.4.2 and 1.7.1.3).

1.6.4.2 Physical-chemical characteristics U.K.

The soil is manually cleared of large objects (e.g. stones, parts of plants, etc.) and then moist sieved without excess drying to a particle size less than or equal to 2 mm. The moisture content of the soil sample should be adjusted with distilled or deionised water to a value between 40% and 60% of the maximum water holding capacity.

1.6.4.3 Amendment with organic substrate U.K.

The soil should be amended with a suitable organic substrate, e.g. powdered lucerne-grass-green meal (main component: *Medicago sativa*) with a C/N ratio between 12/1 and 16/1. The recommended lucerne-soil ratio is 5 g of lucerne per kilogram of soil (dry weight).

1.6.5 Preparation of the test substance for the application to soil U.K.

The test substance is normally applied using a carrier. The carrier can be water (for water soluble substances) or an inert solid such as fine quartz sand (particle size: 0.1 -0.5mm). Liquid carriers other than water (e.g. organic solvents such as acetone, chloroform) should be avoided since they can damage the microflora. If sand is used as a carrier, it can be coated with the test substance dissolved or suspended in an appropriate solvent. In such cases, the solvent should be removed by evaporation before mixing with the soil. For an optimum distribution of the test substance in soil, a ratio of 10 g of sand per kilogram of soil (dry weight) is recommended. Control samples are treated with an equivalent amount of water and/or quartz sand only.

When testing volatile chemicals, losses during treatment should be avoided as far as possible and an attempt should be made to ensure homogeneous distribution in the soil (e.g. the test substance should be injected into the soil at several places).

1.6.6 **Test concentrations** U.K.

If agrochemicals are tested, at least two concentrations should be used. The lower concentration should reflect at least the maximum amount expected to reach the soil under practical conditions

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whereas the higher concentration should be a multiple of the lower concentration. The concentrations of test substance added to soil are calculated assuming uniform incorporation to a depth of 5 cm and a soil bulk density of 1.5. For agrochemicals that are applied directly to soil, or for chemicals for which the quantity reaching the soil can be predicted, the test concentrations recommended are the maximum Predicted Environmental Concentration (PEC) and five times that concentration. Substances that are expected to be applied to soils several times in one season should be tested at concentrations derived from multiplying the PEC by the maximum anticipated number of applications. The upper concentration tested, however, should not exceed ten times the maximum single application rate. If non-agrochemicals are tested, a geometric series of at least five concentrations is used. The concentrations tested should cover the range needed to determine the EC_x values.

1.7 PERFORMANCE OF THE TEST U.K.

1.7.1 **Conditions of exposure** U.K.

1.7.1.1 *Treatment and control* U.K.

If agrochemicals are tested, the soil is divided into three portions of equal weight. Two portions are mixed with the carrier containing the product, and the other is mixed with the carrier without the product (control). A minimum of three replicates for both treated and untreated soils is recommended. If non-agrochemicals are tested, the soil is divided into six portions of equal weight. Five of the samples are mixed with the carrier containing the test substance, and the sixth sample is mixed with the carrier without the chemical. Three replicates for both treatments and control are recommended. Care should be taken to ensure homogeneous distribution of the test substance in the treated soil samples. During mixing, compacting or balling of the soil should be avoided.

1.7.1.2 *Incubation of soil samples* U.K.

Incubation of soil samples can be performed in two ways: as bulk samples of each treated and untreated soil or as a series of individual and equally sized subsamples of each treated and untreated soil. However, when volatile substances are tested, the test should only be performed with a series of individual subsamples. When soils are incubated in bulk, large quantities of each treated and untreated soils are prepared and subsamples to be analysed are taken as needed during the test. The amount initially prepared for each treatment and control depends on the size of the subsamples, the number of replicates used for analysis and the anticipated maximum number of sampling times. Soils incubated in bulk should be thoroughly mixed before subsampling. When soils are incubated as a series of individual soil samples, each treated and untreated bulk soil is divided into the required number of subsamples, and these are utilised as needed. In the experiments where more than two sampling times can be anticipated, enough subsamples should be prepared to account for all replicates and all sampling times. At least three replicate samples of the test soil should be incubated under aerobic conditions (see section 1.7.1.1). During all tests, appropriate containers with sufficient headspace should be used to avoid development of anaerobic conditions. When volatile substances are tested, the test should only be performed with a series of individual subsamples.

1.7.1.3 *Test conditions and duration* U.K.

The test is carried out in the dark at room temperature of $20\pm2^{\circ}$ C. The moisture content of soil samples should be maintained during the test between 40% and 60% of the maximum water holding capacity of the soil (see section 1.6.4.2) with a range of $\pm5\%$. Distilled, deionized water can be added as needed.

The minimum duration of tests is 28 days. If agrochemicals are tested, the rates of nitrate formation in treated and control samples are compared. If these differ by more than 25% on

day 28, the test is continued until a difference equal to or less than 25% is obtained, or for a maximum of 100 days, whichever is shorter. For non-agrochemicals, the test is terminated after 28 days. On day 28, the quantities of nitrate in treated and control soil samples are determined and the EC_x values are calculated.

1.7.2 Sampling and analysis of soils U.K.

1.7.2.1 *Soil sampling schedule* U.K.

If agrochemicals are tested, soil samples are analysed for nitrate on days 0, 7, 14 and 28. If a prolonged test is required, further measurements should be made at 14 days intervals after day 28.

If non-agrochemicals are tested, at least five test concentrations are used and soil samples are analysed for nitrate at the beginning (day 0) and at the end of the exposure period (28 days). An intermediate measurement, e.g. at day 7, may be added if deemed necessary. The data obtained on day 28 are used to determine EC_x value for the chemical. If desired, data from day 0 control samples can be used to report the initial quantity of nitrate in the soil.

1.7.2.2 Analysis of soil samples U.K.

The amount of nitrate formed in each treated and control replicate is determined at each sampling time. Nitrate is extracted from soil by shaking samples with a suitable extraction solvent, e.g. a 0.1 M potassium chloride solution. A ratio of 5 ml of KC1 solution per gram dry weight equivalent of soil is recommended. To optimise extraction, containers holding soil and extraction solution should not be more than half full. The mixtures are shaken at 150 rpm for 60 minutes. The mixtures are centrifuged or filtered and the liquid phases are analysed for nitrate. Particle-free liquid extracts can be stored prior to analysis at minus 20±5 °C for up to six months.

2 DATA U.K.

2.1 TREATMENT OF RESULTS U.K.

If tests are conducted with agrochemicals, the quantity of nitrate formed in each replicate soil sample should be recorded, and the mean values of all replicates should be provided in tabular form. Nitrogen transformation rates should be evaluated by appropriate and generally acceptable statistical methods (e.g. F-test, 5% significance level). The quantities of nitrate formed are expressed in mg nitrate/kg dry weight soil/day. The nitrate formation rate in each treatment is compared with that in the control, and the percent deviation from the control is calculated.

If tests are conducted with non-agrochemicals, the quantity of nitrate formed in each replicate is determined, and a dose-response curve is prepared for estimation of the EC_x values. The quantities of nitrate (i.e. mg nitrate/kg dry weight soil) found in the treated samples after 28 days are compared to that found in the control. From these data, the % inhibition values for each test concentration are calculated. These percentages are plotted against concentration, and statistical procedures are then used to calculate the EC_x values. Confidence limits (p = 0.95) for the calculated EC_x are also determined using standard procedures (10)(11)(12).

Test substances that contain high quantities of nitrogen may contribute to the quantities of nitrate formed during the test. If these substances are tested at a high concentration (e.g. chemicals which are expected to be used in repeated applications) appropriate controls must be included in the test (i.e. soil plus test substance but without plant meal). Data from these controls must be accounted for in the EC_x calculations.

2.2 INTERPRETATION OF RESULTS U.K.

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When results from tests with agrochemicals are evaluated, and the difference in the rates of nitrate formation between the lower treatment (i.e. the maximum predicted concentration) and control is equal to or less than 25% at any sampling time after day 28, the product can be evaluated as having no long-term influence on nitrogen transformation in soils. When results from tests with chemicals other than agrochemicals are evaluated, the EC_{50} , EC_{25} and/or EC_{10} values are used.

3 **REPORTING U.K.**

The test	report must include the following information:
Complet —	re identification of the soil used including: geographical reference of the site (latitude, longitude); information on the history of the site (i.e. vegetation cover, treatments with crop
	protection products, treatments with fertilisers, accidental contamination, etc.); use pattern (e.g. agricultural soil, forest, etc.);
	depth of sampling (cm);
	sand/silt/clay content (% dry weight);
	pH (in water);
	organic carbon content (% dry weight);
	nitrogen content (% dry weight);
	initial nitrate concentration (mg nitrate/kg dry weight);
_	cation exchange capacity (mmol/kg);
	micfobial biomass in terms of percentage of the total organic carbon;
	reference of the methods used for the determination of each parameter;
_	all information relating to the collection and storage of soil samples;
	details of pre-incubation of soil if any.
Test sub	stance:
<u> </u>	physical nature and, where relevant, physical-chemical properties; chemical identification data, where relevant, including structural formula, purity (i.e. for crop protection products the percentage of active ingredient), nitrogen content.
Substrate	a·
	source of substrate;
	composition (i.e. lucerne meal, lucerne-grass-green meal);
	carbon, nitrogen content (% dry weight);
	sieve size (mm).
Test con	ditions:
_	details of the amendment of soil with organic substrate;
_	number of concentrations of test chemical used and, where appropriate, justification of the selected concentrations;
	details of the application of test substance to soil;
	incubation temperature;
	soil moisture content at the beginning and during the test;
	method of soil incubation used (i.e. as bulk or as a series of individual subsamples);
	number of replicates;
_	sampling times;

method used for extraction of nitrate from soil;

Results:

- analytical procedure and equipment used to analyse nitrate;
- tabulated data including individual and mean values for nitrate measurements;
- variation between the replicates in treated and control samples;
- explanations of corrections made in the calculations, if relevant;
- the percent variation in nitrate formation rates at each sampling time or, if appropriate, the EC₅₀ value with 95 per cent confidence limit, other EC_x (i.e. EC₂₅ or EC₁₀) with confidence intervals, and a graph of the dose-response curve;
- statistical treatment of results:
- all information and observations helpful for the interpretation of the results.

4 **REFERENCES** U.K.

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C.22. SOIL MICROORGANISMS: CARBON TRANSFORMATION TEST U.K.

1. **METHOD** U.K.

This method is a replicate of OECD TG 217 (2000).

1.1 INTRODUCTION U.K.

This Testing method describes a laboratory method designed to investigate long term potential effects of a single exposure of crop protection products and possibly other chemicals on

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carbon transformation activity of soil microorganisms. The test is principally based on the recommendations of the European and Mediterranean Plant Protection Organization (1). However, other guideline, including those of the German Biologische Bundesanstalt (2), the US Environmental Protection Agency (3) and SET AC (4), were also taken into account. An OECD Workshop on Soil/Sediment Selection held at Belgirate, Italy, in 1995 (5) agreed on the number and type of soils for use in this test. Recommendations for collection, handling and storage of soil sample are based on an ISO Guidance Document (6) and recommendations from the Belgirate Workshop.

In the assessment and evaluation of toxic characteristics of test substances, determination of effects on soil microbial activity may be required, e.g. when data on the potential side effects of crop protection products on soil microflora are required or when exposure of soil microorganisms to chemicals other than crop protection products is expected. The carbon transformation test is carried out to determine the effects of such chemicals on soil microflora. If agrochemicals (e.g. crop protection products, fertilisers, forestry chemicals) are tested, both carbon transformation and nitrogen transformation tests are conducted. If non-agrochemicals are tested, the nitrogen transformation test is sufficient. However, if EC_{50} values of the nitrogen transformation test for such chemicals fall within the range found for commercially available nitrification inhibitors (e.g. nitrapyrin), a carbon transformation test can be conducted to gain further information.

Soils consist of living and non-living components which exist in complex and heterogeneous mixtures. Microorganisms play an important role in breakdown and transformation of organic matter in fertile soils with many species contributing to different aspects of soil fertility. Any long-term interference with these biochemical processes could potentially interfere with nutrient cycling and this could alter the soil fertility. Transformation of carbon and nitrogen occurs in all fertile soils. Although the microbial communities responsible for these processes differ from soil to soil, the pathways of transformation are essentially the same.

This Testing Method is designed to detect long-term adverse effects of a substance on the process of carbon transformation in aerobic surface soils. The test is sensitive to changes in size and activity of microbial communities responsible for carbon transformation since it subjects these communities to both chemical stress and carbon starvation. A sandy soil low in organic matter is used. This soil is treated with the test substance and incubated under conditions that allow rapid microbial metabolism. Under these conditions, sources of readily available carbon in the soil are rapidly depleted. This causes carbon starvation which both kills microbial cells and induces dormancy and/or sporulation. If the test runs for more than 28 days, the sum of these reactions can be measured in (untreated soil) controls as a progressive loss of metabolically active microbial biomass (7). If the biomass in carbon-stressed soil, under the conditions of the test, is affected by the presence of a chemical, it may not return to the same level as the control. Hence, disturbances caused by the test substance at any time during the test will often last until the end of the test.

The tests from which this Testing Method was developed were primarily designed for substances for which the amount reaching the soil can be anticipated. This is the case, for example, for crop protection products for which the application rate in the field is known. For agrochemicals, testing of two doses relevant to the anticipated or predicted application rate is sufficient. Agrochemicals can be tested as active ingredients (a.i.) or as formulated products. However, the test is not limited to chemicals with predictable environmental concentrations. By changing both the amounts of test substance applied to the soil, and the way in which the data are evaluated, the test can also be used for chemicals for which the amount expected to reach the soil is not known. Thus, with non-agrochemicals, the effects of a series of concentrations on carbon transformation are determined. The data from these tests are used to prepare a dose-response curve and calculate EC_x values, where x is defined % effect.

1.2 DEFINITIONS U.K.

Carbon transformation: is the degradation by microorganisms of organic matter to form inorganic end-product carbon dioxide.

 EC_x (Effective Concentration): is the concentration of the test substance in soil that results in a x percent inhibition of carbon transformation in carbon dioxide.

EC₅₀ (Median Effective Concentration): is the concentration of test substance in soil that results in a 50 per cent inhibition of carbon transformation in carbon dioxide.

1.3 REFERENCE SUBSTANCES U.K.

None.

1.4 PRINCIPLE OF THE TEST METHOD U.K.

Sieved soil is either treated with the test substance or left untreated (control). If agrochemicals are tested, a minimum of two test concentrations are recommended and these should be chosen in relation to the highest concentration anticipated in the field. After 0, 7, 14 and 28 days incubation, samples of treated and control soils are mixed with glucose, and glucose-induced respiration rates are measured for 12 consecutive hours. Respiration rates are expressed as carbon dioxide released (mg carbon dioxide/kg dry soil/h) or oxygen consumed (mg oxygen/kg soil/h). The mean respiration rate in the treated soil samples is compared with that in control and the percent deviation of the treated from the control is calculated. All tests run for at least 28 days. If, on the 28th day, differences between treated and untreated soils are equal to or greater than 25% measurements are continued in 14 day intervals for a maximum of 100 days. If chemicals other than agrochemicals are tested, a series of concentrations of the test substance are added to samples of the soil, and glucose induced respiration rates (i.e. the mean of the quantities of carbon dioxide formed or oxygen consumed) are measured after 28 days. Results from tests with a series of concentrations are analysed using a regression model, and the EC_x values are calculated (i.e. EC₅₀, EC₂₅ and/or EC₁₀). See definitions.

1.5 VALIDITY OF THE TEST U.K.

Evaluations of test results with agrochemicals are based on relatively small differences (i.e. average value $\pm 25\%$) between the carbon dioxide released or the oxygen consumed in (or by) control and treated soil samples, so large variations in the controls can lead to false results. Therefore, the variation between replicate control samples should be less than $\pm 15\%$.

1.6 DESCRIPTION OF THE TEST METHOD U.K.

1.6.1 **Apparatus** U.K.

Test containers made of chemically inert material are used. They should be of a suitable capacity in compliance with the procedure used for incubation of soils, i.e. incubation in bulk or as a series of individual soil samples (see section 1.7.1.2). Care should be taken both to minimise water loss and to allow gas exchange during the test (e.g. the test containers may be covered with perforated polyethylene foil). When volatile substances are tested, sealable and gas-tight containers should be used. These should be of a size such that approximately one quarter of their volume is filled with the soil sample.

For determination of glucose-induced respiration, incubation systems and instruments for measurement of carbon dioxide production or oxygen consumption are required. Examples of such systems and instruments are found in the literature (8) (9) (10) (11).

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1.6.2 Selection and number of soils U.K.

One single soil is used. The recommended soil characteristics are as follows:

- sand content: not less than 50% and not greater than 75%;
- pH: 5.5 7.5;
- organic carbon content: 0.5 -1.5%;
- the microbial biomass should be measured (12)(13) and its carbon content should be at least 1% of the total soil organic carbon.

In most cases, a soil with these characteristics represents a worst case situation, since adsorption of the test chemical is minimised and its availability to the microflora is maximum. Consequently, tests with other soils are generally unnecessary. However, in certain circumstances, e.g. where the anticipated major use of the test substance is in particular soils such as acidic forest soils, or for electrostatically charged chemicals, it may be necessary to substitute an additional soil.

1.6.3 Collection and storage of soil samples U.K.

1.6.3.1 Collection U.K.

Detailed information on the history of the field site from where the test soil is collected should be available. Details include exact location, vegetation cover, dates of treatments with crop protection products, treatments with organic and inorganic fertilisers, additions of biological materials or accidental contaminations. The site chosen for soil collection should be one which allows long-term use. Permanent pastures, fields with annual cereal crops (except maize) or densely sown green manures are suitable. The selected sampling site should not have been treated with crop protection products for a minimum of one year before sampling. Also, no organic fertiliser should have been applied for at least six months. The use of mineral fertiliser is only acceptable when in accordance with the requirements of the crop and soil samples should not be taken until at least three months after fertiliser application. The use of soil treated with fertilisers with known biocidal effects (e.g. calcium cyanamide) should be avoided.

Sampling should be avoided during or immediately following long periods (greater than 30 days) of drought or water logging. For ploughed soils, samples should be taken from a depth of 0 down to 20 cm. For grassland (pasture) or other soils where ploughing does not occur over longer periods (at least one growing season), the maximum depth of sampling may be slightly more than 20 cm (e.g. to 25 cm). Soil samples should be transported using containers and under temperature conditions which guarantee that the initial soil properties are not significantly altered.

1.6.3.2 Storage U.K.

The use of soils freshly collected from the field is preferred. If storage in the laboratory cannot be avoided, soils may be stored in the dark at 4 ± 2 °C for a maximum of three months. During the storage of soils, aerobic conditions must be ensured. If soils are collected from areas where they are frozen for at least three months per year, storage for six months at minus 18 °C can be considered. The microbial biomass of stored soils is measured prior to each experiment and the carbon in the biomass should be at least 1 % of the total soil organic carbon content (see section 1.6.2).

1.6.4 Handling and preparation of soil for the test U.K.

1.6.4.1 *Pre-incubation* U.K.

If the soil was stored (see sections 1.6.4.2 and 1.7.1.3), pre-incubation is recommended for a period between 2 and 28 days. The temperature and moisture content of the soil during pre-incubation should be similar to that used in the test (see sections 1.6.4.2 and 1.7.1.3).

1.6.4.2 Physical-chemical characteristics U.K.

The soil is manually cleared of large objects (e.g. stones, parts of plants, etc.) and then moist sieved without excess drying to a particle size less than or equal to 2 mm. The moisture content of the soil sample should be adjusted with distilled or deionised water to a value between 40% and 60% of the maximum water holding capacity.

1.6.5 Preparation of the test substance for the application to soil U.K.

The test substance is normally applied using a carrier. The carrier can be water (for water soluble substances) or an inert solid such as fine quartz sand (particle size: 0.1-0.5 mm). Liquid carriers other than water (e.g. organic solvents such as acetone, chloroform) should be avoided since they can damage the microflora. If sand is used as a carrier, it can be coated with the test substance dissolved or suspended in an appropriate solvent. In such cases, the solvent should be removed by evaporation before mixing with the soil. For an optimum distribution of the test substance in soil, a ratio of 10 g of sand per kilogram of soil (dry weight) is recommended. Control samples are treated with the equivalent amount of water and/or quartz sand only.

When testing volatile chemicals, losses during treatment should be avoided and an attempt should be made to ensure homogeneous distribution in the soil (e.g. the test substance should be injected into the soil at several places).

1.6.6 **Test concentrations** U.K.

If crop protection products or other chemicals with predictable environmental concentrations are tested, at least two concentrations should be used. The lower concentration should reflect at least the maximum amount expected to reach the soil under practical conditions whereas the higher concentration should be a multiple of the lower concentration. The concentrations of test substance added to soil are calculated assuming uniform incorporation to a depth of 5 cm and a soil bulk density of 1.5. For agrochemicals that are applied directly to soil, or for chemicals for which the quantity reaching the soil can be predicted, the test concentrations recommended are the Predictable Environmental Concentration (PEC) and five times that concentration. Substances that are expected to be applied to soils several times in one season should be tested at concentrations derived from multiplying the PEC by the maximum anticipated number of applications. The upper concentration tested, however, should not exceed ten times the maximum single application rate.

If non-agrochemicals are tested, a geometric series of at least five concentrations is used. The concentrations tested should cover the range needed to determine the EC_x values.

1.7 PERFORMANCE OF THE TEST U.K.

1.7.1 Conditions of exposure U.K.

1.7.1.1 *Treatment and control* U.K.

If agrochemicals are tested, the soil is divided into three portions of equal weight. Two portions are mixed with the carrier containing the product, and the other is mixed with the carrier without the product (control). A minimum of three replicates for both treated and untreated soils is recommended. If non-agrochemicals are tested, the soil is divided into six portions of equal weight. Five of the samples are mixed with the carrier containing the test substance, and the sixth sample is mixed with the carrier without the chemical. Three replicates for both treatments and control are recommended. Care should be taken to ensure homogeneous distribution of

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the test substance in the treated soil samples. During mixing, compacting or balling of the soil should be avoided.

1.7.1.2 Incubation of soil samples U.K.

Incubation of soil samples can be performed in two ways: as bulk samples of each treated and untreated soil or as a series of individual and equally sized subsamples of each treated and untreated soil. However, when volatile substances are tested, the test should only be performed with a series of individual subsamples. When soils are incubated in bulk, large quantities of each treated and untreated soils are prepared and subsamples to be analysed are taken as needed during the test. The amount initially prepared for each treatment and control depends on the size of the subsamples, the number of replicates used for analysis and the anticipated maximum number of sampling times. Soils incubated in bulk should be thoroughly mixed before subsampling. When soils are incubated as a series of individual soil samples, each treated and untreated bulk soil is divided into the required number of subsamples, and these are utilised as needed. In the experiments where more than two sampling times can be anticipated, enough subsamples should be prepared to account for all replicates and all sampling times. At least three replicate samples of the test soil should be incubated under aerobic conditions (see section 1.7.1.1). During all tests, appropriate containers with sufficient headspace should be used to avoid development of anaerobic conditions. When volatile substances are tested, the test should only be performed with a series of individual subsamples.

1.7.1.3 *Test conditions and duration* U.K.

The test is carried out in the dark at room temperature of $20\pm2^{\circ}$ C. The moisture content of soil samples should be maintained during the test between 40% and 60% of the maximum water holding capacity of the soil (see section 1.6.4.2) with a range of $\pm5\%$. Distilled, deionised water can be added as needed.

The minimum duration of tests is 28 days. If agrochemicals are tested, the quantities of carbon dioxide released or oxygen consumed in treated and control samples are compared. If these differ by more than 25% on day 28, the test is continued until a difference equal to or less than 25% is obtained, or for a maximum of 100 days, whichever is shorter. If non-agrochemicals are tested, the test is terminated after 28 days. On day 28, the quantities of carbon dioxide released or oxygen consumed in treated and control soil samples are determined and the EC_x values are calculated.

1.7.2 Sampling and analysis of soils U.K.

1.7.2.1 *Soil sampling schedule* U.K.

If agrochemicals are tested, soil samples are analysed for glucose-induced respiration rates on days 0, 7, 14 and 28. If a prolonged test is required, further measurements should be made at 14 days intervals after day 28.

If non-agrochemicals are tested, at least five test concentrations are used and soil samples are analysed for glucose-induced respiration at the beginning (day 0) and at the end of the exposure period (28 days). An intermediate measurement, e.g. at day 7, may be added if deemed necessary. The data obtained on day 28 are used to determine EC_x value for the chemical. If desired, data from day 0 control samples can be used to estimate the initial quantities of metabolically active microbial biomass in the soil (12).

1.7.2.2 Measurement of glucose-induced respiration rates U.K.

The glucose-induced respiration rate in each treated and control replicate is determined at each sampling time. The soil samples are mixed with a sufficient amount of glucose to elicit an

immediate maximum respiratory response. The amount of glucose needed to elicit a maximum respiratory response from a given soil can be determined in a preliminary test using a series of concentrations of glucose (14). However, for sandy soils with 0.5-1.5% organic carbon, 2000 mg to 4000 mg glucose per kg dry weight soil is usually sufficient. The glucose can be ground to a powder with clean quartz sand (10 g sand/kg dry weight soil) and homogeneously mixed with the soil.

The glucose amended soil samples are incubated in a suitable apparatus for measurement of respiration rates either continuously, every hour, or every two hours (see section 1.6.1) at 20 \pm 2 °C. The carbon dioxide released or the oxygen consumed is measured for 12 consecutive hours and measurements should start as soon as possible, i.e. within 1 to 2 hours after glucose supplement. The total quantities of carbon dioxide released or oxygen consumed during the 12 hours are measured and mean respiration rates are determined.

2 DATA U.K.

2.1 TREATMENT OF RESULTS U.K.

If agrochemicals are tested, the carbon dioxide released from, or oxygen consumed by each replicate soil sample should be recorded, and the mean values of all replicates should be provided in tabular form. Results should be evaluated by appropriate and generally acceptable statistical methods (e.g. F-test, 5% significance level). Glucose-induced respiration rates are expressed in mg carbon dioxide/kg dry weight soil/h or mg oxygen/dry weight soil/h. The mean carbon dioxide formation rate or mean oxygen consumption rate in each treatment is compared with that in control, and the percent deviation from the control is calculated.

If tests are conducted with non-agrochemicals, the quantities of carbon dioxide released or oxygen consumed by each replicate is determined, and a dose-response curve is prepared for estimation of the EC_x values. The glucose-induced respiration rates (i.e. mg carbon dioxide/kg dry weight soil/h or mg oxygen/dry weight soil/h) found in the treated samples after 28 days are compared to that found in control. From these data, the % inhibition values for each test concentration are calculated. These percentages are plotted against concentration, and statistical procedures are used to calculate the EC_x values. Confidence limits (p = 0.95) for the calculated EC_x are also determined using standard procedures (15)(16)(17).

2.2 INTERPRETATION OF RESULTS U.K.

When results from tests with agrochemicals are evaluated, and the difference in respiration rates between the lower treatment (i.e. the maximum predicted concentration) and control is equal to or less than 25% at any sampling time after day 28, the product can be evaluated as having no long-term influence on carbon transformation in soils. When results from tests with chemicals other than agrochemicals are evaluated, the EC_{50} , EC_{25} and/or EC_{10} values are used.

3 **REPORTING U.K.** TEST REPORT

The test report must include the following information:

Complete identification of the soil used including:

- geographical reference of the site (latitude, longitude);
- information on the history of the site (i.e. vegetation cover, treatments with crop protection products, treatments with fertilisers, accidental contamination, etc.)
- use pattern (e.g. agricultural soil, forest, etc.);
- depth of sampling (cm);
- sand/silt/clay content (% dry weight);

ANNEX 2I

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	pH (in water);
	organic carbon content (% dry weight);
	nitrogen content (% dry weight);
_	cation exchange capacity (mmol/kg);
_	initial microbial biomass in terms of percentage of the total organic carbon;
_	reference of the methods used for the determination of each parameter;
	all information relating to the collection and storage of soil samples;
	details of pre-incubation of soil if any.
Test subs	stance:
	physical nature and, where relevant, physical-chemical properties;
_	chemical identification data, where relevant, including structural formula, purity (i.e. for crop protection products the percentage of active ingredient), nitrogen content.
Test con	ditions:
	details of the amendment of soil with organic substrate;
_	number of concentrations of test chemical used and, where appropriate, justification of the selected concentrations;
	details of the application of test substance to soil;
_	incubation temperature;
_	soil moisture content at the beginning and during the test;
_	method of soil incubation used (i.e. as bulk or as a series of individual subsamples);
	number of replicates;
_	sampling times.
Results:	
	method and equipment used for measurement of respiration rates;
_	tabulated data including individual and mean values for quantities of carbon dioxide or oxygen;
	variation between the replicates in treated and control samples;
	explanations of corrections made in the calculations, if relevant;
_	the percent variation of glucose-induced respiration rates at each sampling time or, if appropriate, the EC_{50} with 95 per cent confidence limit, other EC_x (i.e. EC_{25} or EC_{10}) with confidence intervals, and a graph of the dose-response curve;
	statistical treatment of results, where appropriate;
	all information and observations helpful for the interpretation of the results.
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C.23. AEROBIC AND ANAEROBIC TRANSFORMATION IN SOIL U.K.

1. **METHOD** U.K.

This Test Method is a replicate of the OECD TG 307 (2002)

1.1 INTRODUCTION U.K.

This Test Method is based on existing guidelines (1)(2)(3)(4)(5)(6)(7)(8)(9). The method described in this Test Method is designed for evaluating aerobic and anaerobic transformation of chemicals in soil. The experiments are performed to determine (i) the rate of transformation of the test substance, and (ii) the nature and rates of formation and decline of transformation products to which plants and soil organisms may be exposed. Such studies are required for chemicals which are directly applied to soil or which are likely to reach the soil environment.

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The results of such laboratory studies can also be used to develop sampling and analysis protocols for related field studies.

Aerobic and anaerobic studies with one soil type are generally sufficient for the evaluation of transformation pathways (8)(10)(11). Rates of transformation should be determined in at least three additional soils (8)(10).

An OECD Workshop on soil and sediment selection, held at Belgirate, Italy in 1995 (10) agreed, in particular, on the number and types of soils for use in this test. The types of soils tested should be representative of the environmental conditions where use or release will occur. For example, chemicals that may be released in subtropical to tropical climates should be tested with Ferrasols or Nitosols (FAO system). The Workshop also made recommendations relating to collection, handling and storage of soil samples, based on the ISO Guidance (15). The use of paddy (rice) soils is also considered in this method.

1.2 DEFINITIONS U.K.

Test substance: any substance, whether the parent compound or relevant transformation products.

Transformation products: all substances resulting from biotic or abiotic transformation reactions of the test substance including CO₂ and products that are in bound residues.

Bound residues: "Bound residues" represent compounds in soil, plant or animal, which persist in the matrix in the form of the parent substance or its metabolite(s)/transformation products after extraction. The extraction method must not substantially change the compounds themselves or the structure of the matrix. The nature of the bond can be clarified in part by matrix-altering extraction methods and sophisticated analytical techniques. To date, for example, covalent ionic and sorptive bonds, as well as entrapments, have been identified in this way. In general, the formation of bound residues reduces the bioaccessibility and the bioavailability significantly (12) [modified from IUPAC 1984 (13)].

Aerobic transformation: reactions occurring in the presence of molecular oxygen (14).

Anaerobic transformation: reactions occurring under exclusion of molecular oxygen (14).

Soil: is a mixture of mineral and organic chemical constituents, the latter containing compounds of high carbon and nitrogen content and of high molecular weights, animated by small (mostly micro-) organisms. Soil may be handled in two states:

- (a) undisturbed, as it has developed with time, in characteristic layers of a variety of soil types;
- (b) disturbed, as it is usually found in arable fields or as occurs when samples are taken by digging and used in this test method (14).

Mineralisation: is the complete degradation of an organic compound to CO_2 and H_2O under aerobic conditions, and CH_4 , CO_2 and H_2O under anaerobic conditions. In the context of this test method, when ^{14}C -labelled compound is used, mineralisation means extensive degradation during which a labelled carbon atom is oxidised with release of the appropriate amount of $^{14}CO_2$ (14).

Half-life: t_{0.5}, is the time taken for 50% transformation of a test substance when the transformation can be described by first-order kinetics; it is independent of the concentration.

DT₅₀ (**Disappearance Time 50**): is the time within which the concentration of the test substance is reduced by 50%; it is different from the half-life to 5 when transformation does not follow first order kinetics.

DT₇₅ (**Disappearance Time 75**): is the time within which the concentration of the test substance is reduced by 75%.

 DT_{90} (Disappearance Time 90): is the time within which the concentration of the test substance is reduced by 90%.

1.3 REFERENCE SUBSTANCES U.K.

Reference substances should be used for the characterisation and/or identification of transformation products by spectroscopic and chromatographic methods.

1.4 APPLICABILITY OF THE TEST U.K.

The method is applicable to all chemical substances (non-labelled or radiolabelled) for which an analytical method with sufficient accuracy and sensitivity is available. It is applicable to slightly volatile, non-volatile, water-soluble or water-insoluble compounds. The test should not be applied to chemicals which are highly volatile from soil (e.g. fumigants, organic solvents) and thus cannot be kept in soil under the experimental conditions of this test.

1.5 INFORMATION ON THE TEST SUBSTANCE U.K.

Non-labelled or labelled test substance can be used to measure the rate of transformation. Labelled material is required for studying the pathway of transformation and for establishing a mass balance. ¹⁴C-labelling is recommended but the use of other isotopes, such as ¹³C, ¹⁵N, ³H, ³²P, may also be useful. As far as possible, the label should be positioned in the most stable part(s) of the molecule⁽¹³⁾. The purity of the test substance should be at least 95 %.

Before carrying out a test on aerobic and anaerobic transformation in soil, the following information on the test substance should be available:

- (a) solubility in water (Method A.6)
- (b) solubility in organic solvents;
- (c) vapour pressure (Method A.4) and Henry's law constant;
- (d) n-octanol/water partition coefficient (Method A.8);
- (e) chemical stability in dark (hydrolysis) (Method C.7);
- (f) pK_a if a molecule is liable to protonation or deprotonation [OECD Guideline 112] (16).

Other useful information may include data on toxicity of the test substance to soil microorganisms [Testing Methods C.21 and C.22] (16).

Analytical methods (including extraction and clean-up methods) for quantification and identification of the test substance and its transformation products should be available.

1.6 PRINCIPLE OF THE TEST METHOD U.K.

Soil samples are treated with the test substance and incubated in the dark in biometer-type flasks or in flow-through systems under controlled laboratory conditions (at constant temperature and soil moisture). After appropriate time intervals, soil samples are extracted and analysed

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for the parent substance and for transformation products. Volatile products are also collected for analysis using appropriate absorption devices. Using ¹⁴C-labelled material, the various mineralisation rates of the test substance can be measured by trapping evolved ¹⁴CO₂ and a mass balance, including the formation of soil bound residues, can be esstablished.

1.7 QUALITY CRITERIA U.K.

1.7.1 **Recovery U.K.**

Extraction and analysis of, at least, duplicate soil samples immediately after the addition of the test substance gives a first indication of the repeatability of the analytical method and of the uniformity of the application procedure for the test substance. Recoveries for later stages of the experiments are given by the respective mass balances. Recoveries should range from 90% to 110% for labelled chemicals (8) and from 70% to 110% for non-labelled chemicals (3).

1.7.2 Repeatability and sensitivity of analytical method U.K.

Repeatability of the analytical method (excluding the initial extraction efficiency) to quantify test substance and transformation products can be checked by duplicate analysis of the same extract of the soil, incubated long enough for formation of transformation products.

The limit of detection (LOD) of the analytical method for the test substance and for the transformation products should be at least 0.01 mg-kg⁻¹ soil (as test substance) or 1% of applied dose whichever is lower. The limit of quantification (LOQ) should also be specified.

1.7.3 Accuracy of transformation data U.K.

Regression analysis of the concentrations of the test substance as a function of time gives the appropriate information on the reliability of the transformation curve and allows the calculation of the confidence limits for half-lives (in the case of pseudo first order kinetics) or DT_{50} values and, if appropriate, DT_{75} and DT_{90} values.

1.8 DESCRIPTION OF THE TEST METHOD U.K.

1.8.1 Equipment and chemical reagents U.K.

Incubation systems consist of static closed systems or suitable flow-through systems (7)(17). Examples of suitable flow-through soil incubation apparatus and biometer-type flask are shown in Figures 1 and 2, respectively. Both types of incubation systems have advantages and limitations (7)(17).

Standard laboratory equipment is required and especially the following:

- Analytical instruments such as GLC, HPLC, TLC-equipment, including the appropriate detection systems for analysing radiolabelled or non-labelled substances or inverse isotopes dilution method;
- Instruments for identification purposes (e.g. MS, GC-MS, HPLC-MS, NMR, etc.);
- Liquid scintillation counter;
- Oxidiser for combustion of radioactive material;
- Centrifuge;
- Extraction apparatus (for example, centrifuge tubes for cold extraction and Soxhlet apparatus for continuous extraction under reflux);
- Instrumentation for concentrating solutions and extracts (e.g. rotating evaporator);
- Water bath:
- Mechanical mixing device (e.g. kneading machine, rotating mixer).

Chemical reagents used include, for example:

- NaOH, analytical grade, 2 mol dm-⁻³, or other appropriate base (e.g. KOH, ethanolamine);
- H₂SO₄, analytical grade, 0.05 mol dm-⁻³;
- Ethylene glycol, analytical grade;
- Solid absorption materials such as soda lime and polyurethane plugs;
- Organic solvents, analytical grade, such as acetone, methanol, etc.;
- Scintillation liquid.

1.8.2 **Test substance application U.K.**

For addition to and distribution in soil, the test substance can be dissolved in water (deionised or distilled) or, when necessary, in minimum amounts of acetone or other organic solvents (6) in which the test substance is sufficiently soluble and stable. However, the amount of solvent selected should not have a significant influence on soil microbial activity (see sections 1.5 and 1.9.2-1.9.3 The use of solvents which inhibit microbial activity, such as chloroform, dichloromethane and other halogenated solvents, should be avoided.

The test substance can also be added as a solid, e.g. mixed in quartz sand (6) or in a small subsample of the test soil which has been air-dried and sterilised. If the test substance is added using a solvent the solvent should be allowed to evaporate before the spiked sub-sample is added to the original non-sterile soil sample.

For general chemicals, whose major route of entry into soil is through sewage sludge/farming application, the test substance should be first added to sludge which is then introduced into the soil sample (see sections 1.9.2 and 1.9.3)

The use of formulated products is not routinely recommended. However, e.g. for poorly soluble test substances, the use of formulated material may be an appropriate alternative.

1.8.3 **Soils** U.K.

1.8.3.1 Soil selection U.K.

To determine the transformation pathway, a representative soil can be used; a sandy loam or silty loam or loam or loamy sand [according to FAO and USDA classification (18)] with a pH of 5.5-8.0, an organic carbon content of 0.5-2.5% and a microbial biomass of at least 1% of total organic carbon is recommended (10).

For transformation rate studies at least three additional soils should be used representing a range of relevant soils. The soils should vary in their organic carbon content, pH, clay content and microbial biomass (10).

All soils should be characterised, at least, for texture (% sand, % silt, % clay) [according to FAO and USDA classification (18)], pH, cation exchange capacity, organic carbon, bulk density, water retention characteristic⁽¹⁴⁾ and microbial biomass (for aerobic studies only). Additional information on soil properties may be useful in interpreting the results. For determination of the soil characteristics the methods recommended in references (19)(20)(21)(22)(23) can be used. Microbial biomass should be determined by using the substrate-induced respiration (SIR) method (25)(26) or alternative methods (20).

1.8.3.2 Collection, handling, and storage of soils U.K.

Detailed information on the history of the field site from where the test soil is collected should be available. Details include exact location, vegetation cover, treatments with chemicals,

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treatments with organic and inorganic fertilisers, additions of biological materials or other contamination. If soils have been treated with the test substance or its structural analogues within the previous four years, these should not be used for transformation studies (10)(15).

The soil should be freshly collected from the field (from the A horizon or top 20 cm layer) with a soil water content which facilitates sieving. For soils other than those from paddy fields, sampling should be avoided during or immediately following long periods (> 30 days) of drought, freezing or flooding (14). Samples should be transported in a manner which minimises changes in soil water content and should be kept in the dark with free access of air, as much as possible. A loosely-tied polyethylene bag is generally adequate for this purpose.

The soil should be processed as soon as possible after sampling. Vegetation, larger soil fauna and stones should be removed prior to passing the soil through a 2 mm sieve which removes small stones, fauna and plant debris. Extensive drying and crushing of the soil before sieving should be avoided (15).

When sampling in the field is difficult in winter (soil frozen or covered by layers of snow), it may be taken from a batch of soil stored in the greenhouse under plant cover (e.g. grass or grass-clover mixtures). Studies with soils freshly collected from the field are strongly preferred, but if the collected and processed soil has to be stored prior to the start of the study storage conditions must be adequate and for a limited time only $(4 \pm 2^{\circ}\text{C})$ for a maximum of three months) to maintain microbial activity⁽¹⁵⁾. Detailed instructions on collection, handling and storage of soils to be used for biotransformation experiments can be found in (8)(10)(15)(26)(27).

Before the processed soil is used for this test, it should be pre-incubated to allow germination and removal of seeds, and to re-establish equilibrium of microbial metabolism following the change from sampling or storage conditions to incubation conditions. A pre-incubation period between 2 and 28 days approximating the temperature and moisture conditions of the actual test is generally adequate (15). Storage and pre-incubation time together should not exceed three months.

- 1.9 PERFORMANCE OF THE TEST U.K.
- 1.9.1 **Test conditions U.K.**
- 1.9.1.1 *Test temperature* U.K.

During the whole test period, the soils should be incubated in the dark at a constant temperature representative of the climatic conditions where use or release will occur. A temperature of 20 \pm 2 °C is recommended for all test substances which may reach the soil in temperate climates. The temperature should be monitored.

For chemicals applied or released in colder climates (e.g. in northern countries, during autumn/winter periods), additional soil samples should be incubated but at a lower temperature (e.g. 10 ± 2 °C).

1.9.1.2 *Moisture content* U.K.

For transformation tests under aerobic conditions, the soil moisture content⁽¹⁶⁾ should be adjusted to and maintained at a pF between 2.0 and 2.5 (3). The soil moisture content is expressed as mass of water per mass of dry soil and should be regularly controlled (e.g. in 2 week intervals) by weighing of the incubation flasks and water losses compensated by adding water (preferably sterile-filtered tap water). Care should be given to prevent or minimise losses of test substance and/or transformation products by volatilisation and/or photodegradation (if any) during moisture addition.

For transformation tests under anaerobic and paddy conditions, the soil is water-saturated by flooding.

1.9.1.3 Aerobic incubation conditions U.K.

In the flow-through systems, aerobic conditions will be maintained by intermittent flushing or by continuously ventilating with humidified air. In the biometer flasks, exchange of air is maintained by diffusion.

1.9.1.4 Sterile aerobic conditions U.K.

To obtain information on the relevance of abiotic transformation of a test substance, soil samples may be sterilised (for sterilisation methods see references 16 and 29), treated with sterile test substance (e.g. addition of solution through a sterile filter) and aerated with humidified sterile air as described in section 1.9.1.3. For paddy soils, soil and water should be sterilised and the incubation should be carried out as described in section 1.9.1.6.

1.9.1.5 Anaerobic incubation conditions U.K.

To establish and maintain anaerobic conditions, the soil treated with the test substance and incubated under aerobic conditions for 30 days or one half-life or DT50 (whichever is shorter) is then water-logged (1-3 cm water layer) and the incubation system flushed with an inert gas (e.g. nitrogen or argon)⁽¹⁷⁾. The test system must allow for measurements such as pH, oxygen concentration and redox potential and include trapping devices for volatile products. The biometer-type system must be closed to avoid entrance of air by diffusion.

1.9.1.6 *Paddy incubation conditions* U.K.

To study transformation in paddy rice soils, the soil is flooded with a water layer of about 1-5 cm and the test substance applied to the water phase (9). A soil depth of at least 5 cm is recommended. The system is ventilated with air as under aerobic conditions. pH, oxygen concentration and redox potential of the aqueous layer should be monitored and reported. A preincubation period of at least two weeks is necessary before commencing transformation studies (see section 1.8.3.2).

1.9.1.7 *Test duration* U.K.

The rate and pathway studies should normally not exceed 120 days⁽¹⁸⁾ (3)(6)(8), because thereafter a decrease of the soil microbial activity with time would be expected in an artificial laboratory system isolated from natural replenishment. Where necessary to characterise the decline of the test substance and the formation and decline of major transformation products, studies can be continued for longer periods (e.g. 6 or 12 months) (8). Longer incubation periods should be justified in the test report and accompanied by biomass measurements during and at the end of these periods.

1.9.2 **Performance of the test U.K.**

About 50 to 200 g of soil (dry weight basis) are placed into each incubation flask (see Figures 1 and 2 in Annex 3) and the soil treated with the test substance by one of the methods described in section 1.8.2. When organic solvents are used for the application of the test substance, they should be removed from soil by evaporation. Then the soil is thoroughly mixed with a spatula and/or by shaking of the flask. If the study is conducted under paddy field conditions, soil and water should be thoroughly mixed after application of the test substance. Small aliquots (e.g. 1 g) of the treated soils should be analysed for the test substance to check for uniform distribution. For alternative method, see below.

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The treatment rate should correspond to the highest application rate of a crop protection product recommended in the use instructions and uniform incorporation to an appropriate depth in the field (e.g. top 10 cm layer⁽¹⁹⁾ of soil). For example, for chemicals foliarly or soil applied without incorporation, the appropriate depth for computing how much chemical should be added to each flask is 2.5 cm. For soil incorporated chemicals, the appropriate depth is the incorporation depth specified in the use instructions. For general chemicals, the application rate should be estimated based on the most relevant route of entry; for example, when the major route of entry in soil is through sewage sludge, the chemical should be dosed into the sludge at a concentration that reflects the expected sludge concentration and the amount of sludge added to the soil should reflect normal sludge loading to agricultural soils. If this concentration is not high enough to identify major transformation products, incubation of separate soil samples containing higher rates may be helpful, but excessive rates influencing soil microbial functions should be avoided (see sections 1.5 and 1.8.2).

Alternatively, a larger batch (i.e. 1 to 2 kg) of soil can be treated with the test substance, carefully mixed in an appropriate mixing machine and then transferred in small portions of 50 to 200 g into the incubation flasks (for example with the use of sample splitters). Small aliquots (e.g. 1 g) of the treated soil batch should be analysed for the test substance to check for uniform distribution. Such a procedure is preferred since it allows for more uniform distribution of the test substance into the soil.

Also untreated soil samples are incubated under the same conditions (aerobic) as the samples treated with the test substance. These samples are used for biomass measurements during and at the end of the studies.

When the test substance is applied to the soil dissolved in organic solvent(s), soil samples treated with the same amount of solvent(s) are incubated under the same conditions (aerobic) as the samples treated with the test substance. These samples are used for biomass measurements initially, during and at the end of the studies to check for effects of the solvent(s) on microbial biomass.

The flasks containing the treated soil are either attached to the flow-through system described in Figure 1 or closed with the absorption column shown in Figure 2 (see Annex 3).

1.9.3 Sampling and measurement U.K.

Duplicate incubation flasks are removed at appropriate time intervals and the soil samples extracted with appropriate solvents of different polarity and analysed for the test substance and/ or transformation products. A well-designed study includes sufficient flasks so that two flasks are sacrificed at each sampling event. Also, absorption solutions or solid absorption materials are removed at various time intervals (7-day intervals during the first month and after one month in 17 -day intervals) during and at the end of incubation of each soil sample and analysed for volatile products. Besides a soil sample taken directly after application (0-day sample) at least 5 additional sampling points should be included. Time intervals should be chosen in such a way that pattern of decline of the test substance and patterns of formation and decline of transformation products can be established (e.g. 0,1, 3,7 days; 2, 3 weeks; 1, 2, 3 months, etc.).

When using ¹⁴C-labelled test substance, non-extractable radioactivity will be quantified by combustion and a mass balance will be calculated for each sampling interval.

In the case of anaerobic and paddy incubation, the soil and water phases are analysed together for test substance and transformation products or separated by filtration or centrifugation before extraction and analysis.

1.9.4 **Optional tests** U.K.

Aerobic, non-sterile studies at additional temperatures and soil moistures may be useful for the estimation of the influence of temperature and soil moisture on the rates of transformation of a test substance and/or its transformation products in soil.

A further characterisation of non-extractable radioactivity can be attempted using, for example, supercritical fluid extraction.

2 DATA U.K.

2.1 TREATMENT OF RESULTS U.K.

The amounts of test substance, transformation products, volatile substances (in % only), and non-extractable should be given as % of applied initial concentration and, where appropriate, as mgkg- 1 soil (based on soil dry weight) for each sampling interval. A mass balance should be given in percentage of the applied initial concentration for each sampling interval. A graphical presentation of the test substance concentrations against time will allow an estimation of its transformation half-life or DT_{50} . Major transformation products should be identified and their concentrations should also be plotted against time to show their rates of formation and decline. A major transformation product is any product representing $\geq 10\%$ of applied dose at any time during the study.

The volatile products trapped give some indication of the volatility potential of a test substance and its transformation products from soil.

More accurate determinations of half-lives or DT_{50} values and, if appropriate, DT_{75} and DT_{90} values should be obtained by applying appropriate kinetic model calculations. The half-life and DT_{50} values should be reported together with the description of the model used, the order of kinetics and the determination coefficient (r^2). First order kinetics is favoured unless $r^2 < 0.7$. If appropriate, the calculations should also be applied to the major transformation products. Examples of appropriate models are described in references 31 to 35.

In the case of rate studies carried out at various temperatures, the transformation rates should be described as a function of temperature within the experimental temperature range using the Arrhenius relationship of the form:

$$k = A \cdot e^{-B/T}$$
 or $lnk=ln A - \frac{B}{T}$

where ln A and B are regression constants from the intercept and slope, respectively, of a best fit line generated from linearly regressing ln k against 1/T, k is the rate constant at temperature T and T is the temperature in Kelvin. Care should be given to the limited temperature range in which the Arrehenius relationship will be valid in case transformation is governed by microbial action.

2.2 EVALUATION AND INTERPRETATION OF RESULTS U.K.

Although the studies are carried out in an artificial laboratory system, the results will allow estimation of the rate of transformation of the test substance and also of rate of formation and decline of transformation products under field conditions (36)(37).

A study of the transformation pathway of a test substance provides information on the way in which the applied substance is structurally changed in the soil by chemical and microbial reactions.

3 **REPORTING** U.K.

TEST REPORT

The test report must include:

Test substance:

- common name, chemical name, CAS number, structural formula (indicating position of label(s) when radiolabelled material is used) and relevant physical-chemical properties (see section 1.5);
- purity (impurities) of test substance;
- radiochemical purity of labelled chemical and specific activity (where appropriate);

Reference substances:

 chemical name and structure of reference substances used for the characterisation and/ or identification of transformation product;

Test soils:

- details of collection site;
- date and procedure of soil sampling;
- properties of soils, such as pH, organic carbon content, texture (% sand, % silt, % clay), cation exchange capacity, bulk density, water retention characteristic, and microbial biomass:
- length of soil storage and storage conditions (if stored);

Test conditions:

- dates of the performance of the studies;
- amount of test substance applied;
- solvents used and method of application for the test substance;
- weight of soil treated initially and sampled at each interval for analysis;
- description of the incubation system used;
- air flow rates (for flow-through systems only);
- temperature of experimental set-up;
- soil moisture content during incubation;
- microbial biomass initially, during and at the end of the aerobic studies;
- pH, oxygen concentration and redox potential initially, during and at the end of the anaerobic and paddy studies;
- method(s) of extraction;
- methods for quantification and identification of the test substance and major transformation products in soil and absorption materials;
- number of replicates and number of controls.

Results:

- result of microbial activity determination;
- repeatability and sensitivity of the analytical methods used;
- rates of recovery (% values for a valid study are given in section 1.7.1);
- tables of results expressed as % of applied initial dose and, where appropriate, as mgkg-¹ soil (on a dry weight basis);
- mass balance during and at the end of the studies;
- characterisation of non-extractable (bound) radioactivity or residues in soil;
- quantification of released CO₂ and other volatile compounds;

- plots of soil concentrations versus time for the test substance and, where appropriate, for major transformation products;
- half-life or DT₅₀, DT₇₅ and DT₉₀ for the test substance and, where appropriate, for major transformation products including confidence limits;
- estimation of abiotic degradation rate under sterile conditions;
- an assessment of transformation kinetics for the test substance and, where appropriate, for major transformation products;
- proposed pathways of transformation, where appropriate;
- discussion and interpretation of results;
- raw data (i.e. sample chromatograms, sample calculations of transformation rates and means used to identify transformation products).

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ANNEX WATER TENSION, FIELD CAPACITY (FC) AND WATER HOLDING 1 CAPACITY (WHC)(1) U.K.

Height of Water Column [cm]	pF ^a	bar ^b	Remarks
10 ⁷	7	104	Dry Soil
1.6 · 10 ⁴	4.2	16	Wilting point
104	4	10	
$\frac{10^4}{10^3}$	3	1	
$6 \cdot 10^2$	2.8	0.6	
$3.3 \cdot 10^2$	2.5	0.33°	
10 ²	2	0.1	Range of Field
60	1.8	0.06	capacity ^d WHC
33	1.5	0.033	(approximation)
10	1	0.01	Water saturated soil
1	0	0.001	

 $[\]mathbf{a}$ pF = log of cm water column.

Water tension is measured in cm water column or in bar. Due to the large range of suction tension it is expressed simply as pF value which is equivalent to the logarithm of cm water column.

b $1 \text{ bar} = 10^5 \text{Pa}.$

c Corresponds to an approximate water content of 10% in sand, 35% in loam and 45% in clay.

d Field capacity is not constant but varies with soil type between pF 1.5 and 2.5.

Field capacity is defined as the amount of water which can be stored against gravity by a natural soil 2 days after a longer raining period or after sufficient irrigation. It is determined in undisturbed soil in situ in the field. The measurement is thus not applicable to disturbed laboratory soil samples. FC values determined in disturbed soils may show great systematic variances.

Water holding capacity (WHC) is determined in the laboratory with undisturbed and disturbed soil by saturating a soil column with water by capillary transport. It is particularly useful for disturbed soils and can be up to 30 % greater than field capacity (1). It is also experimentally easier to determine than reliable FC-values.

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ANNEX SOIL MOISTURE CONTENTS (g water per 100 g dry soil) OF VARIOUS SOIL

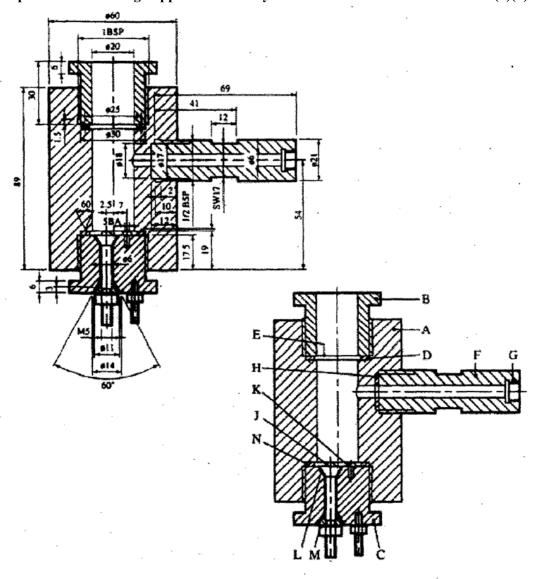
TYPES FROM VARIOUS COUNTRIES U.K.

Soil Moisture Content at			Soil Type	Country
		WHC ^a	pF = 1.8	pF = 2.5
Sand	Germany	28.7	8.8	3.9
Loamy sand	Germany	50.4	17.9	12.1
Loamy sand	Switzerland	44.0	35.3	9.2
Silt loam	Switzerland	72.8	56.6	28.4
Clay loam	Brazil	69.7	38.4	27.3
Clay loam	Japan	74.4	57.8	31.4
Sandy loam	Japan	82.4	59.2	36.0
Silt loam	USA	47.2	33.2	18.8
Sandy loam	USA	40.4	25.2	13.3

a Water Holding Capacity

ANNEX 3
Figure 1

Example of a flow-through apparatus to study transformation of chemicals in soil (1)(2)

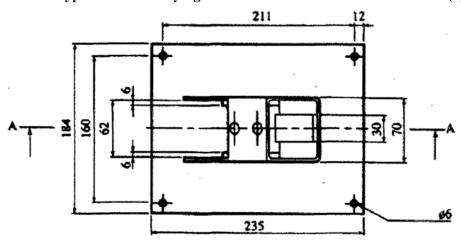


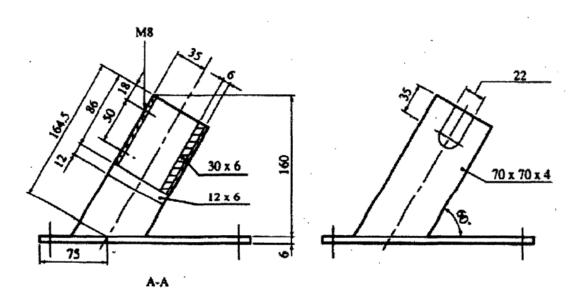
Pressure vessel

Figure 2

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Example of a biometer-type flask for studying the transformation of chemicals in soil (3)





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C.24. AEROBIC AND ANAEROBIC TRANSFORMATION IN AQUATIC SEDIMENT SYSTEMS U.K.

1. **METHOD** U.K.

This test method is a replicate of the OECD TG 308 (2002).

1.1 INTRODUCTION U.K.

Chemicals can enter shallow or deep surface waters by such routes as direct application, spray drift, run-off, drainage, waste disposal, industrial, domestic or agricultural effluent and atmospheric deposition. This Testing Method describes a laboratory method to assess aerobic and anaerobic transformation of organic chemicals in aquatic sediment systems. It is based on existing Guidelines (1)(2)(3)(4)(5)(6). An OECD Workshop on Soil/Sediment Selection, held in Belgirate, Italy in 1995 (7) agreed, in particular, on the number and type of sediments for use in this test. It also made recommendations relating to collection, handling and storage of sediment samples, based on the ISO Guidance (8). Such studies are required for chemicals which are directly applied to water or which are likely to reach the aqueous environment by the routes described above.

The conditions in natural aquatic sediment systems are often aerobic in the upper water phase. The surface layer of sediment can be either aerobic or anaerobic, whereas the deeper sediment is usually anaerobic. To encompass all of these possibilities both aerobic and anaerobic tests are described in this document. The aerobic test simulates an aerobic water column over an aerobic sediment layer that is underlain with an anaerobic gradient. The anaerobic test simulates a completely anaerobic water-sediment system. If circumstances indicate that it is necessary to deviate significantly from these recommendations, for example by using intact sediment cores or sediments that may have been exposed to the test substance, other methods are available for this purpose (9).

1.2 DEFINITIONS U.K.

Standard International (SI) units should be used in any case.

Test substance: any substance, whether the parent or relevant transformation products.

Transformation products: all substances resulting from biotic and abiotic transformation reactions of the test substance including CO_2 and bound residues.

Bound residues: "Bound residues" represent compounds in soil, plant or animal that persist in the matrix in the form of the parent substance or its metabolite(s) after extractions. The extraction method must not substantially change the compounds themselves or the structure of the matrix. The nature of the bond can be clarified in part by matrix-altering extraction methods and sophisticated analytical techniques. To date, for example, covalent ionic and sorptive bonds, as well as entrapments, have been identified in this way. In general, the formation of bound residues reduces the bioaccessibility and the bioavailability significantly (10) [modified from IUPAC 1984(11)].

Aerobic transformation: (oxidising): reactions occurring in the presence of molecular oxygen (12).

Anaerobic transformation: (reducing): reactions occurring under exclusion of molecular oxygen (12).

Natural waters: are surface waters obtained from ponds, rivers, streams, etc.

Sediment: is a mixture of mineral and organic chemical constituents, the latter containing compounds of high carbon and nitrogen content and of high molecular masses. It is deposited by natural water and forms an interface with that water.

Mineralisation: is the complete degradation of an organic compound to CO₂, H₂O under aerobic conditions, and CH₄, CO₂ and H₂O under anaerobic conditions. In the context of this test method, when radiolabelled compound is used, mineralisation means extensive degradation of a molecule during which a labelled carbon atom is oxidised or reduced quantitatively with release of the appropriate amount of ¹⁴CO₂ or ¹⁴CH₄, respectively.

Half-life, t_{0.5}, is the time taken for 50% transformation of a test substance when the transformation can be described by first-order kinetics; it is independent of the initial concentration.

DT₅₀ (Disappearance Time 50): is the time within which the initial concentration of the test substance is reduced by 50%.

DT₇₅ (Disappearance Time 75): is the time within which the initial concentration of the test substance is reduced by 75%.

DT₉₀ (**Disappearance Time 90**): is the time within which the initial concentration of the test substance is reduced by 90%.

1.3 REFERENCE SUBSTANCES U.K.

Reference substances should be used for the identification and quantification of transformation products by spectroscopic and chromatographic methods.

1.4 INFORMATION ON THE TEST SUBSTANCE U.K.

Non-labelled or isotope-labelled test substance can be used to measure the rate of transformation although labelled material is preferred. Labelled material is required for studying the pathway of transformation and for establishing a mass balance. ¹⁴C-labelling is recommended, but the use of other isotopes, such as ¹³C, ¹⁵N, ³H, ³²P, may also be useful. As far as possible, the label should be positioned in the most stable part(s) of the molecule⁽²⁰⁾. The chemical and/or radiochemical purity of the test substance should be at least 95 %.

Before carrying out a test, the following information about the test substance should be available:

- (a) solubility in water (Method A.6);
- (b) solubility in organic solvents;
- (c) vapour pressure (Method A.4) and Henry's Law constant;
- (d) n-octanol/water partition coefficient (Method A.8);
- (e) adsorption coefficient (K_d , K_f or K_{oc} , where appropriate) (Method C.18);
- (f) hydrolysis (Method C.7);
- (g) dissociation constant (pK_a) [OECD Guideline 112] (13);
- (h) chemical structure of the test substance and position of the isotope-label(s), if applicable.

Note: The temperature at which these measurements were made should be reported.

Other useful information may include data on toxicity of the test substance to microorganisms, data on ready and/or inherent biodegradability, and data on aerobic and anaerobic transformation in soil.

Analytical methods (including extraction and clean-up methods) for identification and quantification of the test substance and its transformation products in water and in sediment should be available (see section 1.7.2).

1.5 PRINCIPLE OF THE TEST METHOD U.K.

The method described in this test employs an aerobic and an anaerobic aquatic sediment (see Annex 1) system which allows:

- (i) the measurement of the transformation rate of the test substance in a water-sediment system,
- (ii) the measurement of the transformation rate of the test substance in the sediment,
- (iii) the measurement of the mineralisation rate of the test substance and /or its transformation products (when ¹⁴C-labelled test substance is used),
- (iv) the identification and quantification of transformation products in water and sediment phases including mass balance (when labelled test substance is used),
- (v) the measurement of the distribution of the test substance and its transformation products between the two phases during a period of incubation in the dark (to avoid, for example, algal blooms) at constant temperature. Half-lives, DT₅₀, DT₇₅ and DT₉₀ values are determined where the data warrant, but should not be extrapolated far past the experimental period (see section 1.2).

At least two sediments and their associated waters are required for both the aerobic and the anaerobic studies respectively (7). However, there may be cases where more than two aquatic sediments should be used, for example, for a chemical that may be present in freshwater and/or marine environments.

1.6 APPLICABILITY OF THE TEST U.K.

The method is generally applicable to chemical substances (unlabelled or labelled) for which an analytical method with sufficient accuracy and sensitivity is available. It is applicable to slightly volatile, non-volatile, water-soluble or poorly water-soluble compounds. The test should not be applied to chemicals which are highly volatile from water (e.g. fumigants, organic solvents) and thus cannot be kept in water and/or sediment under the experimental conditions of this test.

The method has been applied so far to study the transformation of chemicals in fresh waters and sediments, but in principle can also be applied to estuarine/marine systems. It is not suitable to simulate conditions in flowing water (e.g. rivers) or the open sea.

1.7 QUALITY CRITERIA U.K.

1.7.1 **Recovery U.K.**

Extraction and analysis of, at least, duplicate water and sediment samples immediately after the addition of the test substance gives a first indication of the repeatability of the analytical method and of the uniformity of the application procedure for the test substance. Recoveries for later stages of the experiments are given by the respective mass balances (when labelled material is used). Recoveries should range from 90% to 110% for labelled chemicals (6) and from 70% to 110% for non-labelled chemicals.

1.7.2 Repeatability and sensitivity of analytical method U.K.

Repeatability of the analytical method (excluding the initial extraction efficiency) to quantify test substance and transformation products can be checked by duplicate analysis of the same

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extract of the water or the sediment samples which were incubated sufficiently long enough for formation of transformation products.

The limit of detection (LOD) of the analytical method for the test substance and for the transformation products should be at least 0.01 mgkg⁻¹ in water or sediment (as test substance) or 1% of the initial amount applied to a test system whichever is lower. The limit of quantification (LOQ) should also be specified.

1.7.3 Accuracy of transformation data U.K.

Regression analysis of the concentrations of the test substance as a function of time gives the appropriate information on the accuracy of the transformation curve and allows the calculation of the confidence limits for half-lives (if pseudo first-order kinetics apply) or DT_{50} values and, if appropriate, DT_{75} and DT_{90} values.

1.8 DESCRIPTION OF THE METHOD U.K.

1.8.1 **Test system and apparatus U.K.**

The study should be performed in glass containers (e.g. bottles, centrifuge tubes), unless preliminary information (such as n-octanol-water partition coefficient, sorption data, etc.) indicates that the test substance may adhere to glass, in which case an alternative material (such as Teflon) may have to be considered. Where the test substance is known to adhere to glass, it may be possible to alleviate this problem using one or more of the following methods:

- determine the mass of test substance and transformation products sorbed to glass;
- ensure a solvent wash of all glassware at the end of the test;
- use of formulated products (see also section 1.9.2);
- use an increased amount of co-solvent for addition of test substance to the system; if a co-solvent is used it should be a co-solvent that does not solvolyse the test substance.

Examples of typical test apparatus, i.e. gas flow-through and biometer-type systems, are shown in Annexes 2 and 3, respectively (14). Other useful incubation systems are described in reference 15. The design of the experimental apparatus should permit the exchange of air or nitrogen and the trapping of volatile products. The dimensions of the apparatus must be such that the requirements of the test are complied with (see section 1.9.1). Ventilation may be provided by either gentle bubbling or by passing air or nitrogen over the water surface. In the latter case gentle stirring of the water from above may be advisable for better distribution of the oxygen or nitrogen in the water. CO₂-free air should not be used as this can result in increases in the pH of the water. In either case, disturbance of the sediment is undesirable and should be avoided as far as possible. Slightly volatile chemicals should be tested in a biometer-type system with gentle stirring of the water surface. Closed vessels with a headspace of either atmospheric air or nitrogen and internal vials for the trapping of volatile products can also be used (16). Regular exchange of the headspace gas is required in the aerobic test in order to compensate for the oxygen consumption by the biomass.

Suitable traps for collecting volatile transformation products include but are not restricted to 1 moldm^{-3} solutions of potassium hydroxide or sodium hydroxide for carbon dioxide⁽²¹⁾ and ethylene glycol, ethanolamine or 2% paraffin in xylene for organic compounds. Volatiles formed under anaerobic conditions, such as methane, can be collected, for example, by molecular sieves. Such volatiles can be combusted, for example, to CO_2 by passing the gas through a quartz tube filled with CuO at a temperature of 900 °C and trapping the CO_2 formed in an absorber with alkali (17).

Laboratory instrumentation for chemical analysis of test substance and transformation products is required (e.g. gas liquid chromatography (GLC), high performance liquid chromatography (HPLC), thin-layer chromatography (TLC), mass spectroscopy (MS), gas chromatographymass spectroscopy (GC-MS), liquid chromatography-mass spectrometry (LC-MS), nuclear magnetic resonance (NMR), etc.), including detection systems for radiolabelled or non-labelled chemicals as appropriate. When radiolabelled material is used a liquid scintillation counter and combustion oxidiser (for the combustion of sediment samples prior to analysis of radioactivity) will also be required.

Other standard laboratory equipment for physical-chemical and biological determinations (see section Table 1, section 1.8.2.2), glassware, chemicals and reagents are required as appropriate.

1.8.2 Selection and number of aquatic sediments U.K.

The sampling sites should be selected in accordance with the purpose of the test in any given situation. In selecting sampling sites, the history of possible agricultural, industrial or domestic inputs to the catchment and the waters upstream must be considered. Sediments should not be used if they have been contaminated with the test substance or its structural analogues within the previous 4 years.

1.8.2.1 Sediment selection U.K.

Two sediments are normally used for the aerobic studies (7). The two sediments selected should differ with respect to organic carbon content and texture. One sediment should have a high organic carbon content (2.5-7.5%) and a fine texture, the other sediment should have a low organic carbon content (0.5-2.5%) and a coarse texture. The difference between the organic carbon contents should normally be at least 2%. "Fine texture" is defined as a [clay + silt] content of <50%. The difference in [clay + silt] content for the two sediments should normally be at least 20%. In cases, where a chemical may also reach marine waters, at least one of the water-sediment systems should be of marine origin.

For the strictly anaerobic study, two sediments (including their associated waters) should be sampled from the anaerobic zones of surface water bodies (7). Both the sediment and the water phases should be handled and transported carefully under exclusion of oxygen.

Other parameters may be important in the selection of sediments and should be considered on a case-by-case basis. For example, the pH range of sediments would be important for testing chemicals for which transformation and/or sorption may be pH-dependent. pH-dependency of sorption might be reflected by the pK_a of the test substance.

1.8.2.2 Characterisation of water-sediment samples U.K.

Key parameters that must be measured and reported (with reference to the method used) for both water and sediment, and the stage of the test at which those parameters are to be determined are summarised in the Table hereafter. For information, methods for determination of these parameters are given in references (18)(19)(20)(21).

In addition, other parameters may need to be measured and reported on a case by case basis (e.g. for freshwater: particles, alkalinity, hardness, conductivity, NO₃/PO₄ (ratio and individual values); for sediments: cation exchange capacity, water holding capacity, carbonate, total nitrogen and phosphorus; and for marine systems: salinity). Analysis of sediments and water for nitrate, sulfate, bioavailable iron, and possibly other electron acceptors may be also useful in assessing redox conditions, especially in relation to anaerobic transformation.

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MEASUREMENT OF PARAMETERS FOR CHARACTERISATION OF WATER-SEDIMENT SAMPLES (7)(22)(23)

Parameter	Stage of test procedure					
	field sampling	post- handling	start of acclimation	start of test	during test	end of test
Water						
Origin/ source	X					
Temperature	X					
PH	X		X	X	X	X
TOC			X	X		X
O ₂ concentratio	X n ^a		X	X	X	X
Redox Potential ^a			X	X	X	X
Sediment	1	I		1	1	<u>'</u>
Origin/ source	X					
Depth of layer	X					
PH		X	X	X	X	X
Particle size distribution		X				
TOC		X	X	X		X
Microbial biomass ^b		X		X		X
Redox potential ^a	Observation (colour/smell)		X	X	X	X

a Recent research results have shown that measurements of water oxygen concentrations and of redox potentials have neither a mechanistic nor a predictive value as far as growth and development of microbial populations in surface waters are concerned (24)(25). Determination of the biochemical oxygen demand (BOD, at field sampling, start and end of test) and of concentrations of micro/macro nutrients Ca, Mg and Mn (at start and end of test) in water and the measurement of total N and total P in sediments (at field sampling and end of test) may be better tools to interpret and evaluate aerobic biotransformation rates and routes.

1.8.3 Collection, Handling and Storage U.K.

1.8.3.1 Collection U.K.

b Microbial respiration rate method (26), fumigation method (27) or plate count measurements (e.g. bacteria, actinomycetes, fungi and total colonies) for aerobic studies; methanogenesis rate for anaerobic studies.

The draft ISO guidance on sampling of bottom sediment (8) should be used for sampling of sediment. Sediment samples should be taken from the entire 5 to 10 cm upper layer of the sediment. Associated water should be collected from the same site or location and at the same time as the sediment. For the anaerobic study, sediment and associated water should be sampled and transported under exclusion of oxygen (28)(see section 1.8.2.1). Some sampling devices are described in the literature (8)(23).

1.8.3.2 *Handling* U.K.

The sediment is separated from the water by filtration and the sediment wet-sieved to a 2 mm-sieve using excess location water that is then discarded. Then known amounts of sediments and water are mixed at the desired ratio (see section 1.9.1) in incubation flasks and prepared for the acclimation period (see section 1.8.4). For the anaerobic study, all handling steps have to be done under exclusion of oxygen (29)(30)(31)(32)(33).

1.8.3.3 Storage U.K.

Use of freshly sampled sediment and water is strongly recommended, but if storage is necessary, sediment and water should be sieved as described above and stored together, water-logged (6-10 cm water layer), in the dark, at $4 \pm 2^{\circ}C^{(23)}$ for a maximum of 4 weeks (7)(8)(23). Samples to be used for aerobic studies should be stored with free access of air (e.g. in open containers), whereas those for anaerobic studies under exclusion of oxygen. Freezing of sediment and water and drying-out of the sediment must not occur during transportation and storage.

1.8.4 Preparation of the sediment/water samples for the test U.K.

A period of acclimation should take place prior to adding the test substance, with each sediment/ water sample being placed in the incubation vessel to be used in the main test, and the acclimation to be carried out under exactly the same conditions as the test incubation (see section 1.9.1). The acclimation period is the time needed to reach reasonable stability of the system, as reflected by pH, oxygen concentration in water, redox potential of the sediment and water, and macroscopic separation of phases. The period of acclimation should normally last between one week and two weeks and should not exceed four weeks. Results of determinations performed during this period should be reported.

1.9 PERFORMANCE OF THE TEST U.K.

1.9.1 **Test conditions** U.K.

The test-should be performed in the incubation apparatus (see section 1.8.1) with a water sediment volume ratio between 3:1 and 4:1, and a sediment layer of 2.5 cm (\pm 0.5 cm). A minimum amount of 50 g of sediment (dry weight basis) per incubation vessel is recommended.

The test should be performed in the dark at a constant temperature in the range of 10 to 30 $^{\circ}$ C. A temperature of $(20 \pm 2)^{\circ}$ C is appropriate. Where appropriate, an additional lower temperature (e.g. 10° C) may be considered on a case-by-case basis, depending on the information required from the test. Incubation temperature should be monitored and reported.

1.9.2 Treatment and application of test substance U.K.

One test concentration of chemical is used⁽²⁴⁾. For crop protection chemicals applied directly to water bodies, the maximum dosage on the label should be taken as, the maximum application rate calculated on the basis of the surface area of the water in the test vessel. In all other cases, the concentration to be used should be based on predictions from environmental emissions. Care must be taken to ensure that an adequate concentration of test substance is applied in order to characterise the route of transformation and the formation and decline of transformation

products. It may be necessary to apply higher doses (e.g. 10 times) in situations where test substance concentrations are close to limits of detection at the start of the study and/or where major transformation products could not readily be detected when present at 10% of the test substance application rate. However, if higher test concentrations are used they should not have a significant adverse effect on the microbial activity of the water-sediment system. In order to achieve a constant concentration of test substance in vessels of differing dimensions an adjustment to the quantity of the material applied may be considered appropriate, based on the depth of the water column in the vessel in relation to the depth of water in the field (which is assumed to be 100 cm, but other depths can be used). See Annex 4 for an example calculation.

Ideally the test substance should be applied as an aqueous solution into the water phase of the test system. If unavoidable, the use of low amounts of water miscible solvents (such as acetone, ethanol) is permitted for application and distribution of the test substance, but this should not exceed 1% v/v and should not have adverse effects on microbial activity of the test system. Care should be exercised in generating the aqueous solution of the test substance use of generator columns and pre-mixing may be appropriate to ensure complete homogeneity. Following addition of the aqueous solution to the test system, gentle mixing of the water phase is recommended, disturbing the sediment as little as possible.

The use of formulated products is not routinely recommended as the formulation ingredients may affect the distribution of the test substance and/or transformation products between water and sediment phases. However, for poorly water-soluble test substances, the use of formulated material may be an appropriate alternative.

The number of incubation vessels depends on the number of sampling times (see section 1.9.3). A sufficient number of test systems should be included so that two systems may be sacrificed at each sampling time. Where control units of each aquatic sediment system are employed, they should not be treated with the test substance. The control units can be used to determine the microbial biomass of the sediment and the total organic carbon of the water and sediment at the termination of the study. Two of the control units (i.e. one control unit of each aquatic sediment) can be used to monitor the required parameters in the sediment and water during the acclimation period (see Table in section 1.8.2.2). Two additional control units have to be included in case the test substance is applied by means of a solvent to measure adverse effects on the microbial activity of the test system.

1.9.3 Test duration and sampling U.K.

The duration of the experiment should normally not exceed 100 days (6), and should continue until the degradation pathway and water/sediment distribution pattern are established or when 90 % of the test substance has dissipated by transformation and/or volatilisation. The number of sampling times should be at least six (including zero time), with an optional preliminary study (see section 1.9.4) being used to establish an appropriate sampling regime and the duration of the test, unless sufficient data is available on the test substance from previous studies. For hydrophobic test substances, additional sampling points during the initial period of the study may be necessary in order to determine the rate of distribution between water and sediment phases.

At appropriate sampling times, whole incubation vessels (in replicate) are removed for analysis. Sediment and overlying water are analysed separately (25). The surface water should be carefully removed with minimum disturbance of the sediment. The extraction and characterisation of the test substance and transformation products should follow appropriate analytical procedures. Care should be taken to remove material that may have adsorbed to the incubation vessel or to interconnecting tubing used to trap volatiles.

1.9.4 Optional preliminary test U.K.

If duration and sampling regime cannot be estimated from other relevant studies on the test substance, an optional preliminary test may be considered appropriate, which should be performed using the same test conditions proposed for the definitive study. Relevant experimental conditions and results from the preliminary test, if performed, should be briefly reported.

1.9.5 Measurements and analysis U.K.

Concentration of the test substance and the transformation products at every sampling time in water and sediment should be measured and reported (as a concentration and as percentage of applied). In general, transformation products detected at >10% of the applied radioactivity in the total water-sediment system at any sampling time should be identified unless reasonably justified otherwise. Transformation products for which concentrations are continuously increasing during the study should also be considered for identification, even if their concentrations do not exceed the limits given above, as this may indicate persistence. The latter should be considered on a case by case basis, with justifications being provided in the report.

Results from gases/volatiles trapping systems (CO₂ and others, i.e. volatile organic compounds) should be reported at each sampling time. Mineralisation rates should be reported. Non-extractable (bound) residues in sediment are to be reported at each sampling point.

2 DATA U.K.

2.1 TREATMENT OF RESULTS U.K.

Total mass balance or recovery (see section 1.7.1) of added radioactivity is to be calculated at every sampling time. Results should be reported as a percentage of added radioactivity. Distribution of radioactivity between water and sediment should be reported as concentrations and percentages, at every sampling time.

Half-life, DT₅₀ and, if appropriate, DT₇₅ and DT₉₀ of the test substance should be calculated along with their confidence limits (see section 1.7.3). Information on the rate of dissipation of the test substance in the water and sediment can be obtained through the use of appropriate evaluation tools. These can range from application of pseudo-first order kinetics, empirical curve-fitting techniques which apply graphical or numerical solutions and more complex assessments Using, for example, single- or multi-compartment models. Further details can be obtained from the relevant published literature (35)(36)(37).

All approaches have their strengths and weaknesses and vary considerably in complexity. An assumption of first-order kinetics may be an oversimplification of the degradation and distribution processes, but when possible gives a term (the rate constant or half-life) which is easily understood and of value in simulation modelling and calculations of predicted environmental concentrations. Empirical approaches or linear transformations can result in better fits of curves to data and therefore allow better estimation of half-lives, DT_{50} and, if appropriate, DT_{75} and DT_{90} values., The use of the derived constants, however, is limited. Compartment models can generate a number of useful constants of value in risk assessment that describe the rate of degradation in different compartments and the distribution of the chemical. They should also be used for estimation of rate constants for the formation and degradation of major transformation products. In all cases, the method chosen must be justified and the experimenter should demonstrate graphically and/or statistically the goodness of fit.

3 **REPORTING** U.K.

3.1 TEST REPORT U.K.

The report must include the following information:

Test substance:

- common name, chemical name, CAS number, structural formula (indicating position of the label(s) when radiolabelled material is used) and relevant physical-chemical properties;
- purity (impurities) of test substance;
- radiochemical purity of labelled chemical and molar activity (where appropriate).

Reference substances

 chemical name and structure of reference substances used for the characterisation and/ or identification of transformation products

Test sediments and waters:

- location and description of aquatic sediment sampling site(s) including, if possible, contamination history;
- all information relating to the collection, storage (if any) and acclimation of watersediment systems;
- characteristics of the water-sediment samples as listed in Table in section 1.8.2.2.

Test conditions:

- test system used (e.g. flow-through, biometer, way of ventilation, method of stirring, water volume, mass of sediment, thickness of both water and sediment layer, dimension of test vessels, etc.)
- application of test substance to test system: test concentration used, number of replicates and controls mode of application of test substance (e.g. use of solvent if any), etc.
- incubation temperature;
- sampling times;
- extraction methods and efficiencies as well as analytical methods and detection limits;
- methods for characterisation/identification of transformation products;
- deviations from the test protocol or test conditions during the study.

Results:

- raw data figures of representative analyses (all raw data have to be stored in the GLP-archive);
- repeatability and sensitivity of the analytical methods used;
- rates of recovery (% values for a valid study are given in section 1.7.1);
- tables of results expressed as % of the applied dose and in mgkg⁻¹ in water, sediment and total system (% only) for the test substance and, if appropriate, for transformation products and non-extractable radioactivity;
- mass balance during and at the end of the studies;
- a graphical representation of the transformation in the water and sediment fractions and in total system (including mineralisation);
- mineralisation rates:
- half-life, DT₅₀ and, if appropriate, DT₇₅ and DT₉₀ values for the test substance and, where appropriate, for major transformation products including confidence limits in water, sediment and in total system;
- an assessment of the transformation kinetics of the test substance and, where appropriate, the major transformation products;

- a proposed pathway of transformation, where appropriate;
- discussion of results.

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ANNEX GUIDANCE ON THE AEROBIC AND THE ANAEROBIC TEST 1 SYSTEMS U.K. Aerobic test system

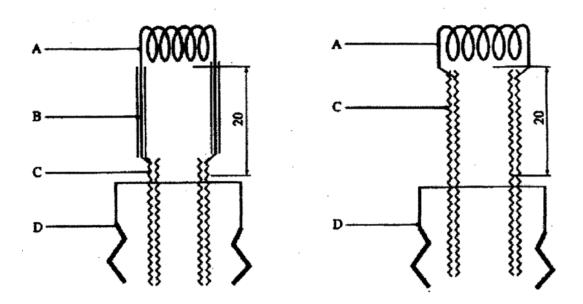
The aerobic test system described in this test method consists of an aerobic water layer (typical oxygen concentrations range from 7 to 10 mgl^{-1}) and a sediment layer, aerobic at the surface and anaerobic below the surface (typical average redox potentials (E_h) in the anaerobic zone of the sediment range from -80 to -190 mV). Moistened air is passed over the surface of the water in each incubation unit to maintain sufficient oxigen in the head space.

Anaerobic test system

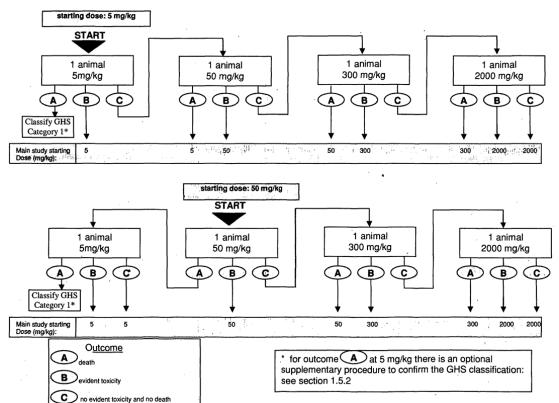
For the anaerobic test system, the test procedure is essentially the same as that outlined for the aerobic system with the exception that moistened nitrogen is passed above the surface of the water in each incubation unit to maintain a head space of nitrogen. The sediment and water are regarded as anaerobic once the redox potential (E_h) is lower than -100 mV.

In the anaerobic test, assessment of mineralisation includes measurement of evolved carbon dioxide and methane.

ANNEX EXAMPLE OF A GAS FLOW-THROUGH APPARATUS U.K.



ANNEX EXAMPLE OF A BIOMETER APPARATUS U.K.



ANNEX EXAMPLE CALCULATION FOR APPLICATION DOSE TO TEST 4 VESSELS U.K.

Cylinder internal diameter:	= 8 cm

Commission Directive 2004/73/EC of 29 April 2004 adapting to technical progress for the...

ANNEX 21

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Water column depth not including sediment:	= 12 cm
Surface area: 3.142 x 4 ²	$= 50.3 \text{ cm}^2$
Application rate: 500 g test substance/ha corresponds to 5 μg/cm ²	
Total μg: 5 x 50.3	= 251.5 μg
Adjust quantity in relation to a depth of 100 cm: 12 x 251.5 ÷ 100	= 30.18 μg
Volume of water column: 50.3 x 12	= 603 ml
Concentration in water: 30.18 ÷ 603	= 0.050 μg/ml or 50 μg/l

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- (1) OJ 196,16.8.1967, p. 1. Directive as last amended by Commission Directive 2001/59/EC (OJ L 225, 6.8.2001, p. 1)
- (2) OJ L 358, 18.12.1986, p. 1. Directive as last amended by Directive 2003/65/EC of the European Parliament and of the Council (OJ L 230,16.9.2003, p. 32).
- (3) As, for example, in the framework of UN transport regulations.
- (4) The acid should be titrated before testing to confirm its concentration.
- (5) E.g.: 50% (w/w) perchloric acid and 40% (w/w) sodium chlorate are used in reference 1.
- (6) e.g. Whatman Column Chromatographic Cellulose Powder CF 11, catalogue no 4021 050
- (7) Confirmed by (e.g.) Karl-Fisher titration
- (8) Alternatively, this water content can also be achieved by (e.g.) heating at 105 °C under vacuum for 24 h
- (9) Mixtures of oxidizers with cellulose must be treated as potentially explosive and handled with due care
- (10) In practice this can be achieved by preparing a 1:1 mixture of the liquid to be tested and cellulose in a greater amount than needed for the trial and transferring 5 ± 0.1 g to the pressure vessel. The mixture is to be freshly prepared for each trial.
- (11) In particular, contact between the adjacent turns of the coil must be avoided.
- (12) See reference 1 for interpretation of the results under the UN transport regulations using several reference substances.
- (13) For example, if the test substance contains one ring, labelling on this ring is required; if the test substance contains two or more rings, separate studies may be needed to evaluate the fate of each labelled ring and to obtain suitable information on formation of transformation products.
- (14) Water retention characteristic of a soil can be measured as field capacity, as water holding capacity or as water suction tension (pF). For explanations see Annex 1. It should be reported in the test report whether water retention characteristics and bulk density of soils were determined in undisturbed field samples or in disturbed (processed) samples.
- (15) Recent research results indicate that soils from temperate zones can also be stored at -20°C for more than three months (28)(29) without significant losses of microbial activity.
- (16) The soil should neither be too wet nor too dry to maintain adequate aeration and nutrition of soil microflora. Moisture contents recommended for optimal microbial growth range from 40-60% water holding capacity (WHC) and from 0.1-0.33 bar (6). The latter range is equivalent to a pFrange of 2.0 2.5. Typical moisture contents of various soil types are given in Annex 2.
- (17) Aerobic conditions are dominant in surface soils and even in sub-surface soils as shown in an EU sponsored research project [K. Takagi et al. (1992). Microbial diversity and activity in subsoils: Methods, field site, seasonal variation in subsoil temperatures and oxygen contents. Proc. Internat. Symp. Environm. Aspects Pesticides Microbiol., 270-277, 17-21 August 1992, Sigtuna, Sweden]. Anaerobic conditions may only occur occasionally during flooding of soils after heavy rainfalls or when paddy conditions are established in rice fields.
- (18) Aerobic studies might be terminated much before 120 days provided that ultimate transformation pathway and ultimate mineralisation are clearly reached at that time. Termination of the test is possible after 120 days, or when at least 90% of the test substance is transformed, but only if at least 5% CO₂ is formed.
- (19) Calculation of the initial concentration on an area basis using the following equation:

C_{soil}= Initial concentration in soil [mg·kg⁻¹]

A = Application rate $[kg \cdot ha^{-1}]$; l = thickness of field soil layer [m]; <math>d = dry bulk density of soil $[kg \cdot m^{-3}]$.

As a rule of thumb, an application rate of 1 kg·ha⁻¹ results in a soil concentration of approximately 1 mg·kg⁻¹ in a 10 cm layer (assuming a bulk density of 1 g·cm⁻³).

- (20) For example, if the substance contains one ring, labelling on this ring is required; if the test substance contains two or more rings, separate studies may be needed to evaluate the fate of each labelled ring and to obtain suitable information on formation of transformation products.
- (21) As these alkaline absorption solutions also absorb the carbon dioxide from the ventilation air and that formed by respiration in aerobic experiments, they have to be exchanged in regular intervals to avoid their saturation and thus loss of their absorption capacity.
- (22) [Clay + silt] is the mineral fraction of the sediment with particle size of $< 50 \mu m$
- (23) Recent studies have shown that storage at 4 °C can lead to a decrease of the organic carbon content of the sediment which may possibly result in a decrease of microbial activity (34).
- (24) Test with a second concentration can be useful for chemicals that reach surface waters by different entry routes resulting in significantly different concentrations, as long as the lower concentration can be analysed with sufficient accuracy.
- (25) In cases where rapid re-oxidation of anaerobic transformation products may readily occur, anaerobic conditions should be maintained during sampling and analyses.