Commission Directive 2006/129/EC of 8 December 2006 amending and correcting Directive 96/77/EC laying down specific purity criteria on food additives other than colours and sweeteners (Text with EEA relevance)

COMMISSION DIRECTIVE 2006/129/EC

of 8 December 2006

amending and correcting Directive 96/77/EC laying down specific purity criteria on food additives other than colours and sweeteners

(Text with EEA relevance)

THE COMMISSION OF THE EUROPEAN COMMUNITIES,

Having regard to the Treaty establishing the European Community,

Having regard to Council Directive 89/107/EEC of 21 December 1988 on the approximation of the laws of the Member States concerning food additives authorised for use in foodstuffs intended for human consumption⁽¹⁾, and in particular Article 3(3)(a) thereof,

After consulting the Scientific Committee on Food and the European Food Safety Authority,

Whereas:

- (1) Commission Directive 96/77/EC⁽²⁾ of 2 December 1996 laying down specific purity criteria on food additives other than colours and sweeteners sets out the purity criteria for the additives mentioned in Directive 95/2/EC of the European Parliament and of the Council of 20 February 1995 on food additives other than colours and sweeteners⁽³⁾.
- (2) It is appropriate to withdraw the purity criteria for E 216 propyl p-hydroxybenzoate and E 217 sodium propyl p-hydroxybenzoate which are no longer permitted for use as food additives.
- (3) A number of language versions of Directive 96/77/EC contain errors regarding the following substances: E 307 alpha-tocopherol, E 315 erythorbic acid, E 415 xanthan gum. Those errors need to be corrected. In addition it is necessary to take into account the specifications and analytical techniques for additives as set out in the Codex Alimentarius as drafted by the Joint FAO/WHO Expert Committee on Food Additives (JECFA). In particular where appropriate, the specific purity criteria have been adapted to reflect the limits for individual heavy metals of interest. For reasons of clarity the whole text concerning those substances should be replaced.
- (4) The level of sulphated ash in the purity criteria for E 472c citric acid esters of monoand diglycerides of fatty acids should be amended in order to cover partially or wholly neutralised products.
- (5) It is necessary to ensure that E 559 aluminium silicate is produced from raw kaolinitic clay which is free from unacceptable dioxin contamination. The presence of dioxin in the raw kaolinitic clay should therefore be restricted to the lowest possible level.

- (6) It is necessary to adopt specifications for the new food additives authorised through Directive 2006/52/EC of the European Parliament and of the Council of 5 July 2006 amending Directive 95/2/EC on food additives other than colours and sweeteners and Directive 94/35/EC on sweeteners for use in foodstuffs: E 319 tertiary-butylhydroquinone (TBHQ), E 426 soybean hemicellulose, E 462 ethyl cellulose, E 586 4-hexylresorcinol, E 1204 pullulan and E 1452 starch aluminium octenyl succinate.
- (7) Directive 96/77/EC should therefore be amended and corrected accordingly.
- (8) The measures provided for in this Directive are in accordance with the opinion of the Standing Committee on the Food Chain and Animal Health,

HAS ADOPTED THIS DIRECTIVE:

Article 1

The Annex to Directive 96/77/EC is amended and corrected in accordance with the Annex to this Directive.

Article 2

1 Member States shall bring into force the laws, regulations and administrative provisions necessary to comply with this Directive by 15 February 2008 at the latest. They shall forthwith communicate to the Commission the text of those provisions and a correlation table between those provisions and this Directive.

When Member States adopt those provisions, they shall contain a reference to this Directive or be accompanied by such a reference on the occasion of their official publication. Member States shall determine how such reference is to be made.

2 Member States shall communicate to the Commission the text of the main provisions of national law which they adopt in the field covered by this Directive.

Article 3

This Directive shall enter into force on the 20th day following that of its publication in the *Official Journal of the European Union*.

Article 4

This Directive is addressed to the Member States.

Done at Brussels, 8 December 2006.

For the Commission

Markos KYPRIANOU

Member of the Commission

Status: This is the original version (as it was originally adopted).

ANNEX

The Annex to Directive 96/77/EC is amended and corrected as follows:

- 1. The texts concerning E 216 propyl p-hydroxybenzoate and E 217 sodium propyl phydroxybenzoate are deleted.
- The text concerning E 307 alpha-tocopherol is replaced by the following: 2.

$ \begin{array}{c c} \textbf{Synonyms} & DL-\alpha\text{-Tocopherol} \\ \hline \textbf{Definition} \\ \hline \textbf{Chemical name} & DL-5,7,8\text{-Trimethyltocol} \\ DL-2,5,7,8\text{-Tetramethyl-2-}(4',8',12'\text{-trimethyltridecyl})\text{-}6\text{-chromanol} \\ \hline \textbf{Einecs} & 233\text{-}466\text{-}0 \\ \hline \textbf{Chemical formula} & C_{29}H_{50}O_{2} \\ \hline \textbf{Molecular weight} & 430,71 \\ \hline \textbf{Assay} & \textbf{Content not less than 96 \%} \\ \hline \textbf{Description} & Slightly yellow to amber, nearly odourless, clear, viscous oil which oxidizes and darkens on exposure to air or light \\ \hline \textbf{Identification} \\ \hline \textbf{A.} & Solubility tests & Insoluble in water, freely soluble in ethanol, miscible in ether \\ \hline \textbf{B.} & \textbf{Spectrophotometry} & In absolute ethanol the maximum absorption is about 292 nm \\ \hline \textbf{Purity} \\ \hline \textbf{Refractive index} & n_D^{20} 1,503 \text{ to } 1,507 \\ \hline \textbf{Specific absorption E}^{1\%_{1\text{cm}}} \text{ in ethanol} & E^{1\%_{1\text{cm}}} \text{ (292 nm) } 72 \text{ to } 76 \\ (0,01\ \text{g in } 200\ \text{ml of absolute ethanol}) \\ \hline \textbf{Sulphated ash} & \text{Not more than } 0,1\% \\ \hline \textbf{Specific rotation} & [\alpha]^{25}_{D}0^{\circ} \pm 0,05^{\circ} \text{ (1 in } 10 \text{ solution in } \text{chloroform)} \\ \hline \textbf{Lead} & \text{Not more than } 2\text{mg/kg} \\ \hline \end{array}$	E 307 ALPHA-TOCOPHEROL		
$ \begin{array}{c} \text{Chemical name} & \begin{array}{c} DL-5,7,8-\text{Trimethyltocol} \\ DL-2,5,7,8-\text{Tetramethyl-2-}(4',8',12'-\text{trimethyltridecyl})-6-\text{chromanol} \\ \end{array} \\ \text{Einecs} & 233-466-0 \\ \text{Chemical formula} & C_{29}H_{50}O_{2} \\ \text{Molecular weight} & 430,71 \\ \text{Assay} & \text{Content not less than 96 \%} \\ \text{\textbf{Description}} & \begin{array}{c} \text{Slightly yellow to amber, nearly odourless, clear, viscous oil which oxidizes and darkens on exposure to air or light} \\ \text{\textbf{Insoluble in water, freely soluble in ethanol, miscible in ether}} \\ \text{\textbf{B}.} & \text{Spectrophotometry} & \begin{array}{c} \text{In absolute ethanol the maximum absorption is about 292 nm} \\ \text{\textbf{Purity}} \\ \text{\textbf{Refractive index}} & \begin{array}{c} n_{D}^{20} 1,503 \text{ to } 1,507 \\ \text{\textbf{Specific absorption E}}^{1~\%}_{1~\text{cm}} \text{ in ethanol} \\ \text{\textbf{0},01~g in 200 ml of absolute ethanol} \\ \text{\textbf{Sulphated ash}} & \text{\textbf{Not more than 0,1 \%}} \\ \text{\textbf{Specific rotation}} & [\alpha]^{25}_{D}0^{\circ} \pm 0,05^{\circ} \text{ (1 in 10 solution in chloroform)} \\ \end{array}$	Synonyms	DL-α-Tocopherol	
$\begin{array}{c c} DL-2,5,7,8\text{-Tetramethyl-}2-(4',8',12'-\text{trimethyltridecyl})\text{-}6\text{-}\text{chromanol} \\ \hline Einecs & 233\text{-}466\text{-}0 \\ \hline Chemical formula & C_{29}H_{50}O_2 \\ \hline \text{Molecular weight} & 430,71 \\ \hline \text{Assay} & \text{Content not less than 96 \%} \\ \hline \textbf{Description} & \text{Slightly yellow to amber, nearly odourless, clear, viscous oil which oxidizes and darkens on exposure to air or light} \\ \hline \textbf{Identification} \\ \hline \textbf{A.} & \text{Solubility tests} & \text{Insoluble in water, freely soluble in ethanol, miscible in ether} \\ \hline \textbf{B.} & \text{Spectrophotometry} & \text{In absolute ethanol the maximum absorption is about 292 nm} \\ \hline \textbf{Purity} \\ \hline \textbf{Refractive index} & \text{n}_D^{20} \text{ 1,503 to 1,507} \\ \hline \textbf{Specific absorption E}^{1\%}_{1\text{cm}} \text{ in ethanol} & \text{E}^{1\%}_{1\text{cm}} \text{ (292 nm) 72 to 76} \\ \text{(0,01 g in 200 ml of absolute ethanol)} \\ \hline \textbf{Sulphated ash} & \text{Not more than 0,1 \%} \\ \hline \textbf{Specific rotation} & [\alpha]^{25}_{D}0^{\circ} \pm 0,05^{\circ} \text{ (1 in 10 solution in chloroform)} \\ \hline \end{array}$	Definition		
$ \begin{array}{c} C_{1} \\ C_{2} \\ Molecular \ weight \\ Assay \\ Content \ not \ less \ than 96 \% \\ \hline \textbf{Description} \\ Slightly \ yellow \ to \ amber, \ nearly \ odourless, \ clear, \ viscous \ oil \ which \ oxidizes \ and \ darkens \ on \ exposure \ to \ air \ or \ light \\ \hline \textbf{Identification} \\ A. Solubility \ tests \\ B. Spectrophotometry \\ \hline \textbf{In absolute ethanol the maximum} \ absorption \ is \ about \ 292 \ nm \\ \hline \textbf{Purity} \\ \hline \textbf{Refractive index} \\ \hline \textbf{Specific absorption } E^{1\%}_{1 \ cm} \ in \ ethanol \\ \hline \textbf{Specific absorption } E^{1\%}_{1 \ cm} \ in \ ethanol \\ \hline \textbf{Sulphated ash} \\ \hline \textbf{Not more than 0,1 \%} \\ \hline \textbf{Specific rotation} \\ \hline \ [\alpha]^{25}_{D}0^{\circ} \pm 0,05^{\circ} \ (1 \ in \ 10 \ solution \ in \ chloroform) \\ \hline \end{array}$	Chemical name	DL-2,5,7,8-Tetramethyl-2-(4',8',12'-	
Molecular weight Assay Content not less than 96 % Slightly yellow to amber, nearly odourless, clear, viscous oil which oxidizes and darkens on exposure to air or light Identification A. Solubility tests Insoluble in water, freely soluble in ethanol, miscible in ether B. Spectrophotometry In absolute ethanol the maximum absorption is about 292 nm Purity Refractive index $n_D^{20} 1,503 \text{ to } 1,507$ Specific absorption $E^{1 \%}_{1 \text{ cm}}$ in ethanol $E^{1 \%}_{1 \text{ cm}}$ (292 nm) 72 to 76 (0,01 g in 200 ml of absolute ethanol) Sulphated ash Not more than 0,1 % Specific rotation $[\alpha]^{25}_{D}0^{\circ} \pm 0,05^{\circ}$ (1 in 10 solution in chloroform)	Einecs	233-466-0	
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DescriptionSlightly yellow to amber, nearly odourless, clear, viscous oil which oxidizes and darkens on exposure to air or lightIdentificationInsoluble in water, freely soluble in ethanol, miscible in etherB. SpectrophotometryIn absolute ethanol the maximum absorption is about 292 nmPurityRefractive index n_D^{20} 1,503 to 1,507Specific absorption $E^{1\%}_{1cm}$ in ethanol $E^{1\%}_{1cm}$ (292 nm) 72 to 76 (0,01 g in 200 ml of absolute ethanol)Sulphated ashNot more than 0,1 %Specific rotation $[\alpha]^{25}_{D}0^{\circ} \pm 0,05^{\circ}$ (1 in 10 solution in chloroform)	Molecular weight	430,71	
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A. Solubility tests ethanol, miscible in ether B. Spectrophotometry In absolute ethanol the maximum absorption is about 292 nm Purity Refractive index $n_D^{20} 1,503 \text{ to } 1,507$ Specific absorption $E^{1\%}_{1\text{cm}}$ in ethanol $E^{1\%}_{1\text{cm}}$ (292 nm) 72 to 76 (0,01 g in 200 ml of absolute ethanol) Sulphated ash Not more than 0,1 % Specific rotation $[\alpha]^{25}_D0^{\circ} \pm 0,05^{\circ}$ (1 in 10 solution in chloroform)	Identification		
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Refractive index $ n_D^{20} \ 1,503 \ to \ 1,507 $ Specific absorption $E^{1\%}_{1cm}$ in ethanol $ E^{1\%}_{1cm} \ (292 \ nm) \ 72 \ to \ 76 $ $(0,01 \ g \ in \ 200 \ ml \ of \ absolute \ ethanol) $ Sulphated ash $ Not \ more \ than \ 0,1 \ \% $ Specific rotation $ [\alpha]^{25}_{D}0^{\circ} \pm 0,05^{\circ} \ (1 \ in \ 10 \ solution \ in \ chloroform) $	B. Spectrophotometry		
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$(0,01 \text{ g in } 200 \text{ ml of absolute ethanol})$ Sulphated ash Not more than 0,1 % $[\alpha]^{25}_{D}0^{\circ} \pm 0,05^{\circ} \text{ (1 in } 10 \text{ solution in chloroform)}$	Refractive index	n_D^{20} 1,503 to 1,507	
Specific rotation $ [\alpha]^{25}{}_{D}0^{\circ} \pm 0,05^{\circ} \text{ (1 in 10 solution in chloroform)} $	Specific absorption E ¹ % _{1 cm} in ethanol		
chloroform)	Sulphated ash	Not more than 0,1 %	
Lead Not more than 2 mg/kg	Specific rotation		
	Lead	Not more than 2 mg/kg	

3. The text concerning E 315 erythorbic acid is replaced by the following:

E 315 ERYTHORBIC ACID	
Synonyms	Isoascorbic acid D-Araboascorbic acid
Definition	
Chemical name	D-Erythro-hex-2-enoic acid γ-lactone

		Isoascorbic acid D-Isoascorbic acid
Einec	S	201-928-0
Chem	ical formula	C ₆ H ₈ O ₆
Molec	cular weight	176,13
Assay	,	Content not less than 98 % on the anhydrous basis
Descr	ription	White to slightly yellow crystalline solid which darkens gradually on exposure to light
Iden	tification	
A.	Melting range	About 164 °C to 172 °C with decomposition
В.	Positive test for ascorbic acid/colour reaction	
Puri	ty	
Loss	on drying	Not more than 0,4 % after drying under reduced pressure on silica gel for 3 hours
Sulph	ated ash	Not more than 0,3 %
Speci	fic rotation	$[\alpha]^{25}_{D}10\%$ (w/v) aqueous solution between – 16,5° to – 18,0°
Oxala	te	To a solution of 1 g in 10 ml of water add 2 drops of glacial acetic acid and 5 ml of 10 % calcium acetate solution. The solution should remain clear
Lead		Not more than 2 mg/kg

4. The following text concerning E 319 tertiary-butylhydroquinone (TBHQ) is inserted after E 316 sodium erythorbate:

E 319 TERTIARY-BUTYLHYDROQUINONE (TBHQ)		
Synonyms	ТВНО	
Definition		
Chemical names	Tert-butyl-1,4-benzenediol 2-(1,1-Dimethylethyl)-1,4-benzenediol	
Einecs	217-752-2	
Chemical formula	$C_{10}H_{14}O_2$	
Molecular weight	166,22	
Assay	Content not less than 99 % of C ₁₀ H ₁₄ O ₂	

Descrip	otion	White crystalline solid having a characteristic odour
Identi	fication	·
A.	Solubility	Practically insoluble in water; soluble in ethanol
В.	Melting point	Not less than 126,5 °C
C.	Phenolics	Dissolve about 5 mg of the sample in 10 ml of methanol and add 10,5 ml of dimethylamine solution (1 in 4). A red to pink colour is produced
Purity		
Tertiary	v-Butyl-p-benzoquinone	Not more than 0,2 %
2,5- <i>Di-tertiary</i> -butyl hydroquinone		Not more than 0,2 %
Hydroxyquinone		Not more than 0,1 %
Toluene	2	Not more than 25 mg/kg
Lead		Not more than 2 mg/kg

5. The text concerning E 415 xanthan gum is replaced by the following:

Xanthan gum is a high molecular weight polysaccharide gum produced by a pure-culture fermentation of a carbohydrate with natural strains of <i>Xanthomonas campestris</i> , purified by recovery with ethanol or propan-2-ol, dried and milled. It contains D-glucose and D-mannose as the dominant hexose units, along with D-glucuronic acid and pyruvic acid, and is prepared as the sodium, potassium or calcium salt. Its solutions are neutral			
Approximately 1 000 000			
234-394-2			
Yields, on dried basis, not less than 4,2 % and not more than 5 % of CO ₂ corresponding to between 91 % and 108 % of xanthan gum			
Cream-coloured powder			
Identification			
Soluble in water. Insoluble in ethanol			
Not more than 15 % (105 °C, 2^{1} / ₂ hours)			

Total ash	Not more than 16 % on the anhydrous basis determined at 650 °C after drying at 105 °C for four hours
Pyruvic acid	Not less than 1,5 %
Nitrogen	Not more than 1,5 %
Ethanol and propan-2-ol	Not more than 500 mg/kg singly or in combination
Lead	Not more than 2 mg/kg
Total plate count	Not more than 5 000 colonies per gram
Yeast and mould	Not more than 300 colonies per gram
E. coli	Absent in 5 g
Salmonella spp.	Absent in 10 g
Xanthomonas campestris	Viable cells absent in 1 g

6. The following text concerning E 426 soybean hemicellulose is inserted after E 425(ii) konjac glucomannan:

E 426 SOYBEAN HEMICELLULOSE		
Synonyms		
Definition	Soybean hemicellulose is a refined water-soluble polysaccharide obtained from natural strain soybean fibre by hot water extraction	
Chemical names	Water soluble soybean polysaccharides Water soluble soybean fibre	
Assay	Not less than 74 % carbohydrate	
Description	Free flowing spray-dried white powder	
Identification	1	
A. Solubility	Soluble in hot and cold water without gel formation	
pH of 1 % solution	$5,5 \pm 1,5$	
B. Viscosity of 10 % solution	Not more than 200 mPa.s	
Purity		
Loss on drying	Not more than 7 % (105 °C, 4 h)	
Protein	Not more than 14 %	
Total ash	Not more than 9,5 % (600 °C, 4 h)	
Arsenic	Not more than 2 mg/kg	
Lead	Not more than 5 mg/kg	
Mercury	Not more than 1 mg/kg	

Status: This is the original version (as it was originally adopted).

Cadmium	Not more than 1 mg/kg
Standard plate count	Not more than 3 000 colonies per gram
Yeast and mould	Not more than 100 colonies per gram
E. Coli	Negative in 10 g

The following text concerning E 462 ethyl cellulose is inserted after E 461 methyl 7. cellulose:

E 462 ETHYL CELLULOSE		
Synor	ıyms	Cellulose ethyl ether
Defin	ition	Ethyl cellulose is cellulose obtained directly from fibrous plant material and partially etherified with ethyl groups
Chem	ical names	Ethyl ether of cellulose
Chem	ical formula	The polymers contain substituted anhydroglucose units with the following general formula: C ₆ H ₇ O ₂ (OR ₁)(OR ₂) where R ₁ and R ₂ may be any of the following: — H — CH ₂ CH ₃
Assay		Content not less than 44 % and not more than 50 % of ethoxyl groups (-OC ₂ H ₅) on the dried basis (equivalent to not more than 2,6 ethoxyl groups per anhydroglucose unit)
Descr	iption	Slightly hygroscopic, white to off white, odourless and tasteless powder
Ident	tification	
A.	Solubility	Practically insoluble in water, in glycerol and in propane-1,2-diol but soluble in varying proportions in certain organic solvents depending upon the ethoxyl content. Ethyl cellulose containing less than 46 to 48 % of ethoxyl groups is freely soluble in tetrahydrofuran, in methyl acetate, in chloroform and in aromatic hydrocarbon ethanol mixtures. Ethyl cellulose containing 46 to 48 % or more of ethoxyl groups is freely soluble in ethanol, in methanol, in toluene, in chloroform and in ethyl acetate
В.	Film forming test	Dissolve 5 g of the sample in 95 g of an 80:20 (w/w) mixture of toluene ethanol. A clear, stable, slightly yellow

	solution is formed. Pour a few ml of the solution onto a glass plate and allow the solvent to evaporate. A thick, tough, continuous, clear film remains. The film is flammable
Purity	
Loss on drying	Not more than 3 % (105 °C, 2 h)
Sulphated ash	Not more than 0,4 %
pH of a 1 % colloidal solution	Neutral to litmus
Arsenic	Not more than 3 mg/kg
Lead	Not more than 2 mg/kg
Mercury	Not more than 1 mg/kg
Cadmium	Not more than 1 mg/kg

8. The text concerning E 472c citric acid esters of mono- and diglycerides of fatty acids is replaced by the following:

	2c CITRIC ACID ESTERS OF I	MONO- AND DIGLYCYERIDES OF
Syno	nyms	Citrem Citric acid esters of mono- and diglycerides Citroglycerides Mono- and diglycerides of fatty acids esterified with citric acid
Defin	aition	Esters of glycerol with citric acid and fatty acids occurring in food oils and fats. They may contain small amounts of free glycerol, free fatty acids, free citric acid and free glycerides. They may be partially or wholly neutralised with sodium hydroxide or with potassium hydroxide
Desci	ription	Yellowish or light brown liquids to waxy solids or semi-solids
Iden	tification	
A.	Positive test for glycerol, for fatty acids and for citric acid	
В.	Solubility	Insoluble in cold water Dispersible in hot water Soluble in oils and fats In soluble in cold ethanol
Puri	ty	
Acids	s other than citric and fatty acids	Not detectable

Status: This is the original version (as it was originally adopted).

Free glycerol	Not more than 2 %
Total glycerol	Not less than 8 % and not more than 33 %
Total citric acid	Not less than 13 % and not more than 50 %
Sulphated ash (determined at 800 ± 25 °C)	Non-neutralised products: not more than 0,5 % Partially or wholly neutralised products: not more than 10 %
Lead	Not more than 2 mg/kg
Free fatty acids	Not more than 3 % estimated as oleic acid

Purity criteria apply to the additive free of sodium, potassium and calcium salts of fatty acids, however, these substances may be present up to a maximum level of 6 % (expressed as sodium oleate).

9. The text concerning E 559 aluminium silicate (kaolin) is replaced by the following:

Synonyme	Kaolin, light or heavy
Synonyms	Kaomi, fight of fleavy
Definition	Aluminium silicate hydrous (kaolin) is a purified white plastic clay composed of kaolinite, potassium aluminium silicate, feldspar and quartz. Processing should not include calcination. The raw kaolinitic clay used in the production of aluminium silicate shall have a level of dioxin which does not make it injurious to health or unfit for human consumption
Einecs	215-286-4 (kaolinite)
Chemical formula	Al ₂ Si ₂ O ₅ (OH) ₄ (kaolinite)
Molecular weight	264
Assay	Content not less than 90 % (sum of silica and alumina, after ignition) Silica (SiO ₂) Between 45 % and 55 % Alumina (A1 ₂ O ₃) Between 30 % and 39 %
Description	Fine, white or greyish white, unctuous powder. Kaolin is made up of loose aggregations of randomly oriented stacks of kaolinite flakes or of individual hexagonal flakes.

A.	Positive test for alumina and for silicate	
B.	X-ray diffraction	Characteristic peaks at 7,18/3,58/2,38/1,78 Å
C.	IR absorption	Peaks at 3 700 and 3 620 cm ⁻¹
Purity	y	
Loss of	n ignition	Between 10 and 14 % (1 000 °C, constant weight)
Water	soluble matter	Not more than 0,3 %
Acid soluble matter		Not more than 2 %
Iron		Not more than 5 %
Potassium oxide (K ₂ O)		Not more than 5 %
Carbon		Not more than 0,5 %
Arsenic		Not more than 3 mg/kg
Lead		Not more than 5 mg/kg
Mercury		Not more than 1 mg/kg

10. The following text concerning E 586 4-hexylresorcinol is inserted after E 578 calcium gluconate:

E 586 4-HEXYLRESORCINOL			
Synor	nyms	4-Hexyl-1,3-benzenediol Hexylresorcinol	
Defin	Definition		
Chem	ical names	4-Hexylresorcinol	
Einecs		205-257-4	
Chemical formula		$C_{12}H_{18}O_2$	
Mole	cular weight	197,24	
Assay		Not less than 98,0 % on the dried basis	
Description		White powder	
Iden	Identification		
A.	Solubility	Freely soluble in ether and acetone; very slightly soluble in water	
В.	Nitric acid test	To 1 ml of a saturated solution of the sample, add 1 ml of nitric acid. A light red colour appears	
C.	Bromine test	To 1 ml of saturated solution of the sample, add 1 ml of bromine TS. A	

	yellow, flocculent precipitate dissolves producing a yellow solution
D. Melting range	62 to 67 °C
Purity	
Acidity	Not more than 0,05 %
Sulphated ash	Not more than 0,1 %
Resorcinol and other phenols	Shake about 1 g of the sample with 50 ml of water for a few minutes, filter, and to the filtrate add 3 drops of ferric chloride TS. No red or blue colour is produced
Nickel	Not more than 2 mg/kg
Lead	Not more than 2 mg/kg
Mercury	Not more than 3 mg/kg

11. The following text concerning E 1204 pullulan is inserted after E 1200 polydextrose:

E 1204 PULLULAN		
Definition	on	Linear, neutral glucan consisting mainly of maltotriose units connected by -1,6 glycosidic bonds. It is produced by fermentation from a food grade hydrolysed starch using a non-toxin producing strain of <i>Aureobasidium pullulans</i> . After completion of the fermentation, the fungal cells are removed by microfiltration, the filtrate is heat-sterilised and pigments and other impurities are removed by adsorption and ion exchange chromatography
Einecs		232-945-1
Chemical formula		$(C_6H_{10}O_5)_x$
Assay		Not less than 90 % of glucan on the dried basis
Description		White to off-white odourless powder
Identifi	cation	
A.	Solubility	Soluble in water, practically insoluble in ethanol.
В.	pH of 10 % solution	5,0 to 7,0
C.	Precipitation with polyethylene glycol 600	Add 2 ml of polyethylene glycol 600 to 10 ml of a 2 % aqueous solution of pullulan. A white precipitate is formed

D.	Depolymerisation with pullulanase	Prepare two test tubes each with 10 ml of a 10 % pullulan solution. Add 0,1 ml pullulanase solution having activity 10 units/g to one test tube, and 0,1 ml water to the other. After incubation at about 25 °C for 20 min, the viscosity of the pullulanase-treated solution is visibly lower than that of the untreated solution
Purity		
Loss on drying		Not more than 6 % (90 °C, pressure not more than 50 mm Hg, 6 h)
Mono-, di- and oligosaccharides		Not more than 10 % expressed as glucose
Viscosity		100 to 180 mm ² /s (10 % w/w aqueous solution at 30 °C)
Lead		Not more than 1 mg/kg
Yeast and moulds		Not more than 100 colonies per gram
Coliforn	ns	Absent in 25 g
Salmonella		Absent in 25 g

12. The following text concerning E 1452 starch aluminium octenyl succinate is inserted after E 1451 acetylated oxidised starch:

E 1452 STARCH ALUMINIUM OCTENYL SUCCINATE		
Synony	vms	SAOS
Definit	ion	Starch aluminium octenyl succinate is starch esterified with octenylsuccinic anhydride and treated with aluminium sulphate
Descrij	otion	White or nearly white powder or granules or (if pregelatinised) flakes, amorphous powder or coarse particles
Identi	fication	
A.	If not pregelatinised: by miscroscopic observation	
B.	Iodine staining positive (dark blue to light red colour)	
Purity		
	ues expressed on an anhydrous accept for loss on drying)	
Loss on drying		Not more than 21 %
Octenylsuccinyl groups		Not more than 3 %

Status: This is the original version (as it was originally adopted).

Octenylsuccinic acid residue	Not more than 0,3 %
Sulphur dioxide	Not more than 50 mg/kg for modified cereal starches Not more than 10 mg/kg for the other modified starches unless otherwise specified
Arsenic	Not more than 1 mg/kg
Lead	Not more than 2 mg/kg
Mercury	Not more than 0,1 mg/kg
Aluminium	Not more than 0,3 %

- (1) OJ L 40, 11.2.1989, p. 27. Directive as last amended by Regulation (EC) No 1882/2003 of the European Parliament and of the Council (OJ L 284, 31.10.2003, p. 1).
- (2) OJ L 339, 30.12.1996, p. 1. Directive as last amended by Directive 2004/45/EC (OJ L 113, 20.4.2004, p. 19).
- (3) OJ L 61, 18.3.1995, p. 1. Directive as last amended by Directive 2006/52/EC (OJ L 204, 26.7.2006, p. 10).